

# Denmark Group Meeting

## Palladium catalyzed Dearomatization Reaction & Electrophilic rearrangement of amides

**Bo Peng**  
**11<sup>th</sup> Feb. 2014**





# Palladium catalyzed Dearomatization Reaction

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## Electrophilic rearrangement of amides



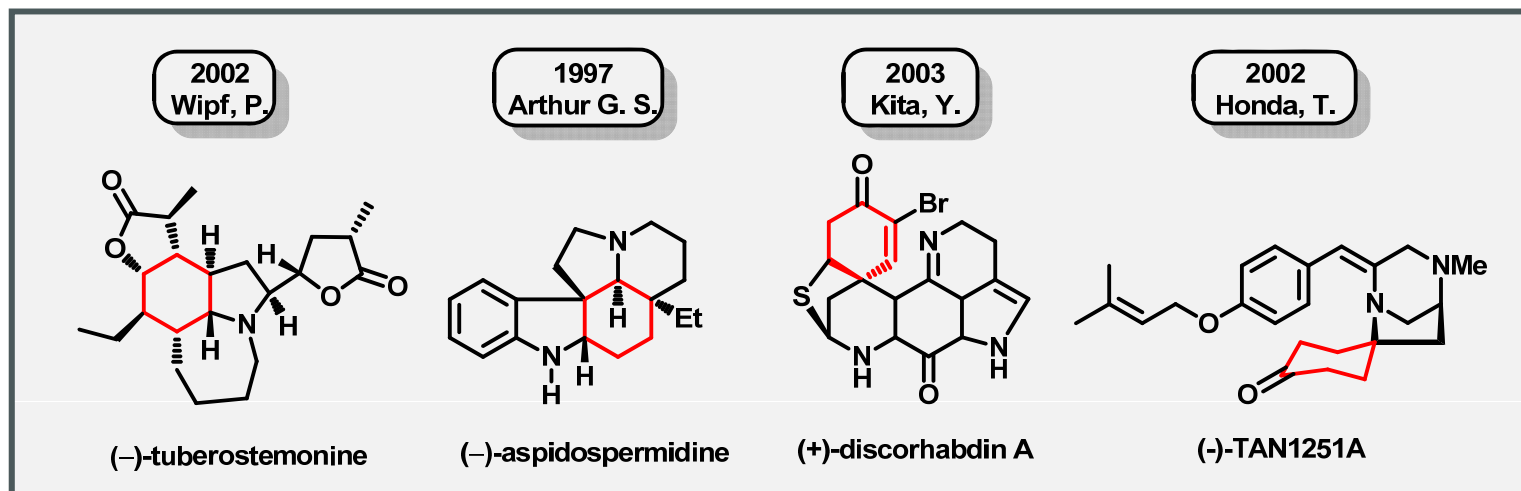


## Palladium catalyzed Dearomatization Reaction: background

### Dearomatization Reaction

Starting material  
Aromatic compounds

product  
Alicyclic compounds



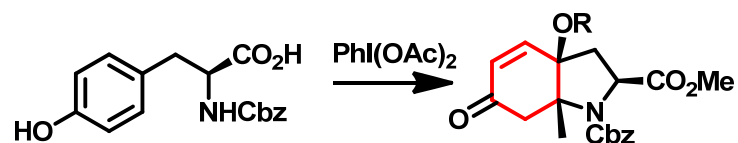
### Natural products

a) Roche, S. P.; Porco, J. A., Jr. *Angew. Chem., Int. Ed.* **2011**, *50*, 4068-4093; b). Pouysegou, L.; Deffieux, D.; Quideau, S. *Tetrahedron* **2010**, *66*, 2235–2261.

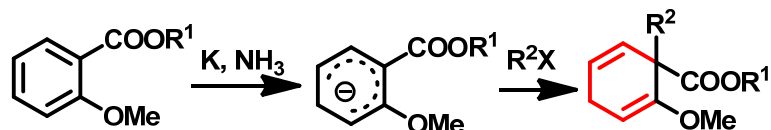


# Palladium catalyzed Dearomatization Reaction: background

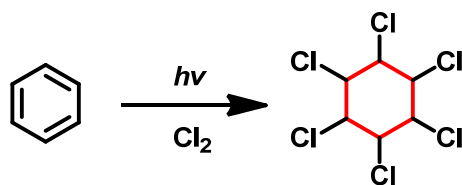
## 1. oxidative dearomatization



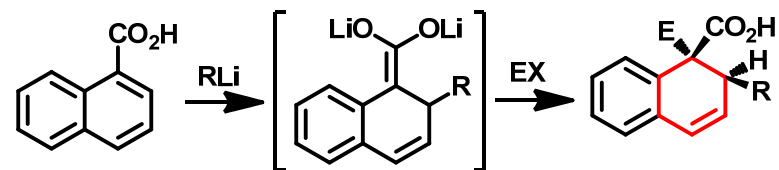
## 2. reductive dearomatization



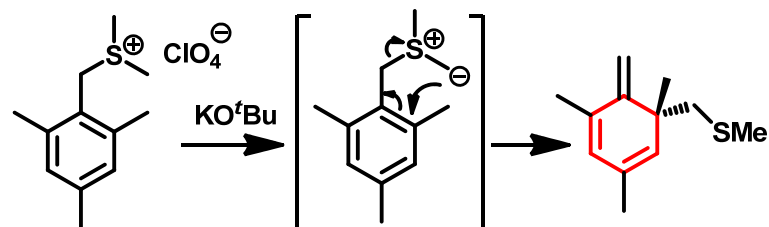
## 3. photocatalyzed dearomatization



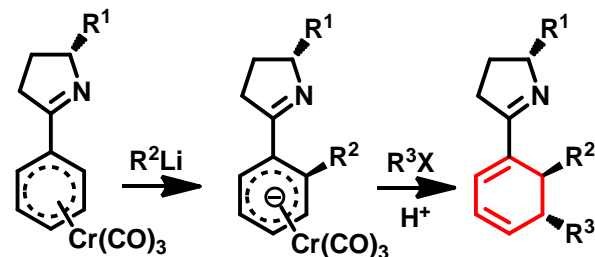
## 4. nucleophilic additive dearomatization



## 5. $\sigma$ -rearrangement dearomatization



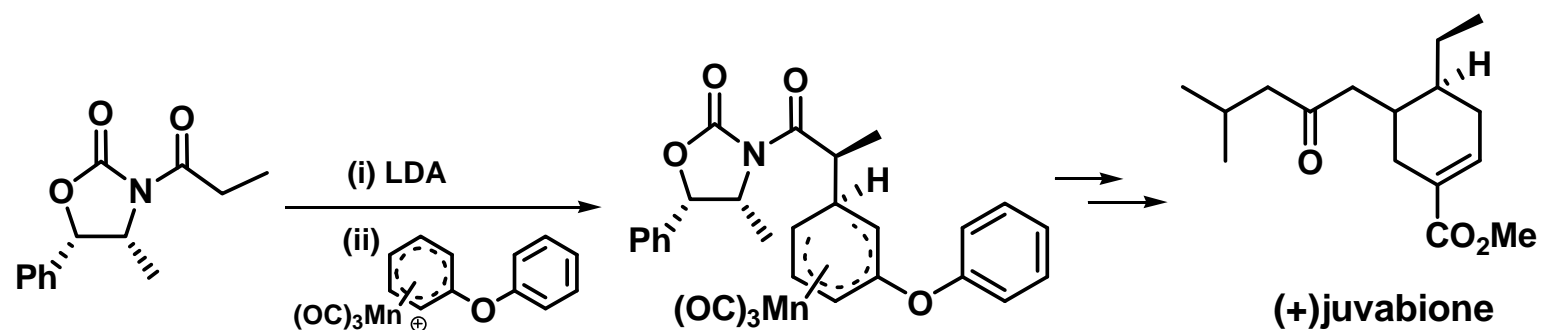
## 6. transition metals promoted dearomatization





## Palladium catalyzed Dearomatization Reaction: background

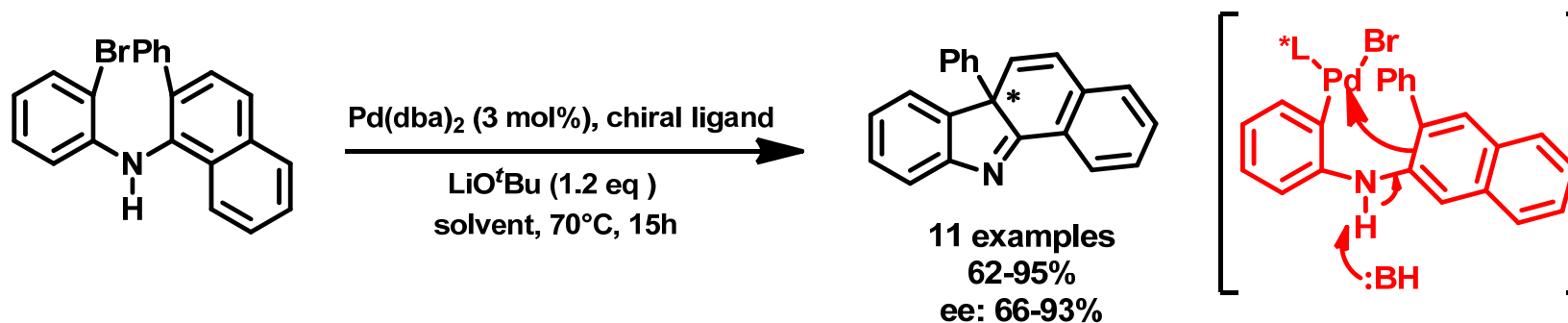
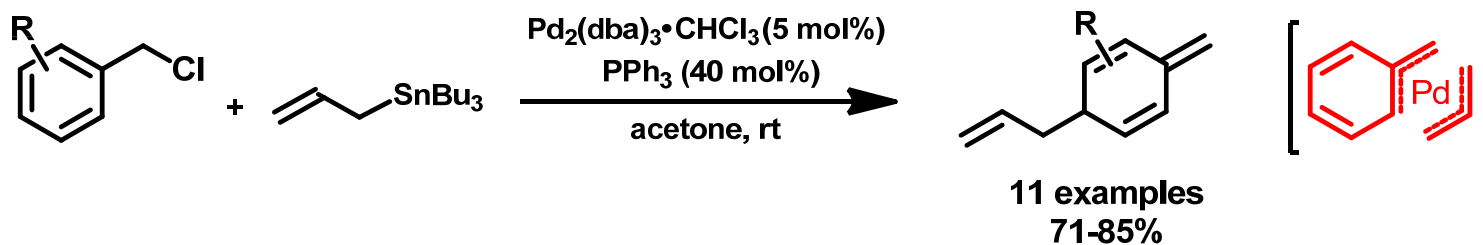
### Application of $[(\eta^6\text{-Arene}) \text{Mn}(\text{CO})_3]^+$





## Palladium catalyzed Dearomatization Reaction: background

### Palladium catalyzed dearomatization reactions

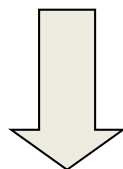
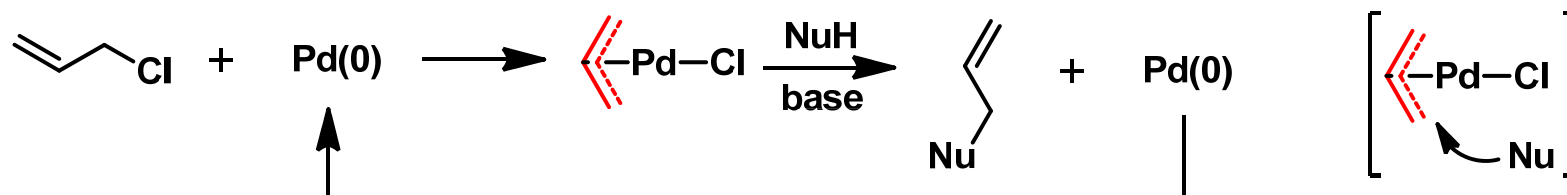


a). Bao, M.; Nakamura, H.; Yamamoto, Y. *J. Am. Chem. Soc.*, **2001**, *123*, 759; b). Fortanet, J. G.; Kessler, F.; Buchwald, S. L. *J. Am. Chem. Soc.*, **2009**, *131*, 6676

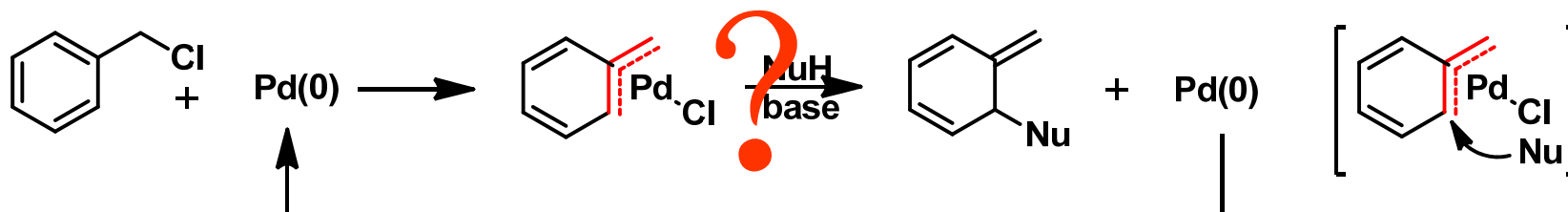


## Palladium catalyzed Dearomatization Reaction: background

### Tsuji-Trost reaction



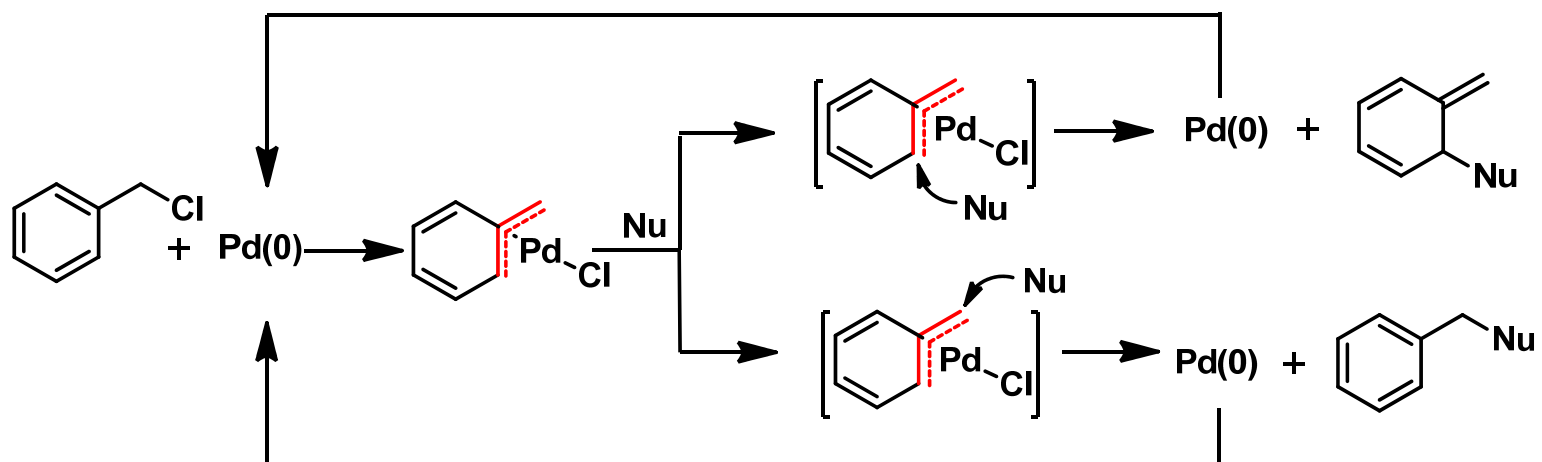
### Palladium catalyzed Nucleophilic Dearomatization ?





## Palladium catalyzed Dearomatization Reaction: background

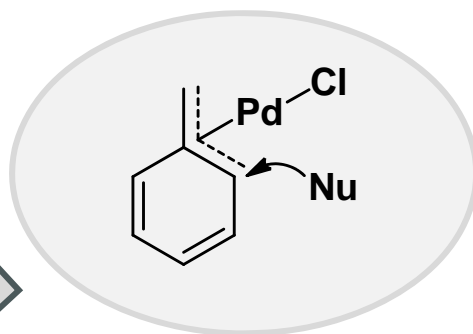
### Regional Selectivity ?



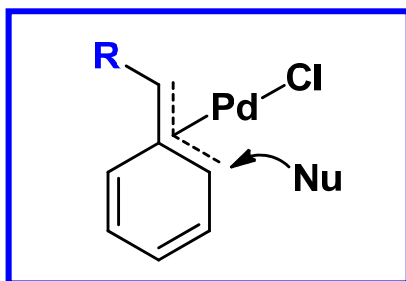


# Palladium catalyzed Dearomatization Reaction: background

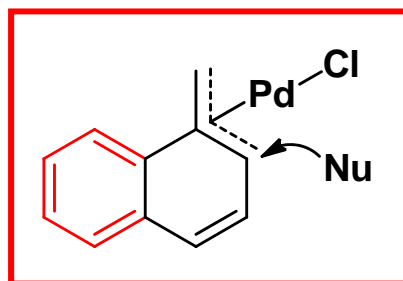
## Possible solutions



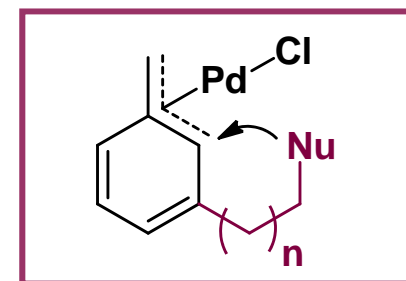
**A: introducing R group**



**B: reducing aromaticity**

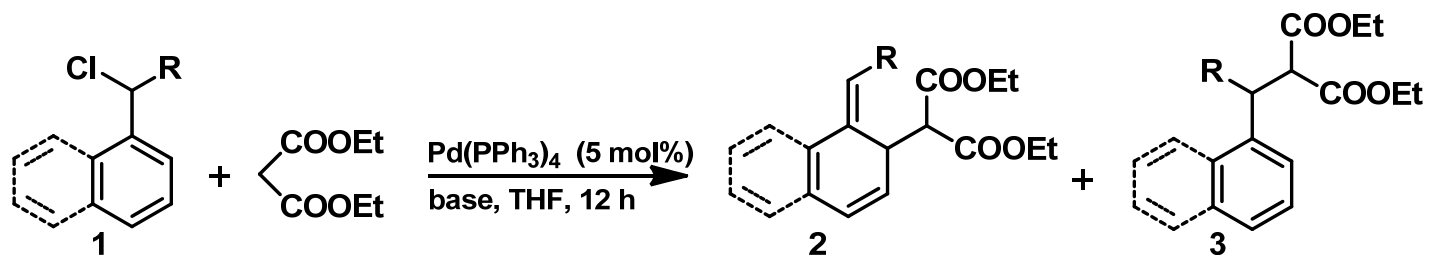


**C: intramolecular nucleophile**





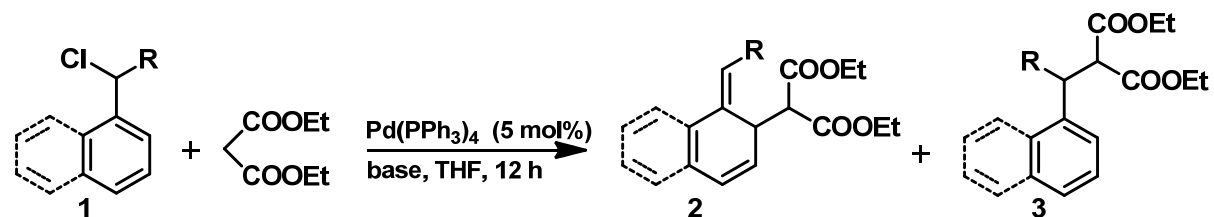
## Palladium catalyzed Dearomatization Reaction: **exploration**



entry	substrate	base	temp [°C]	product	yield [%]
1		NaH	rt	—	NR
2		NaH	50		> 99
3		NaH	rt		> 99
4		NaH	-20 - rt		> 99



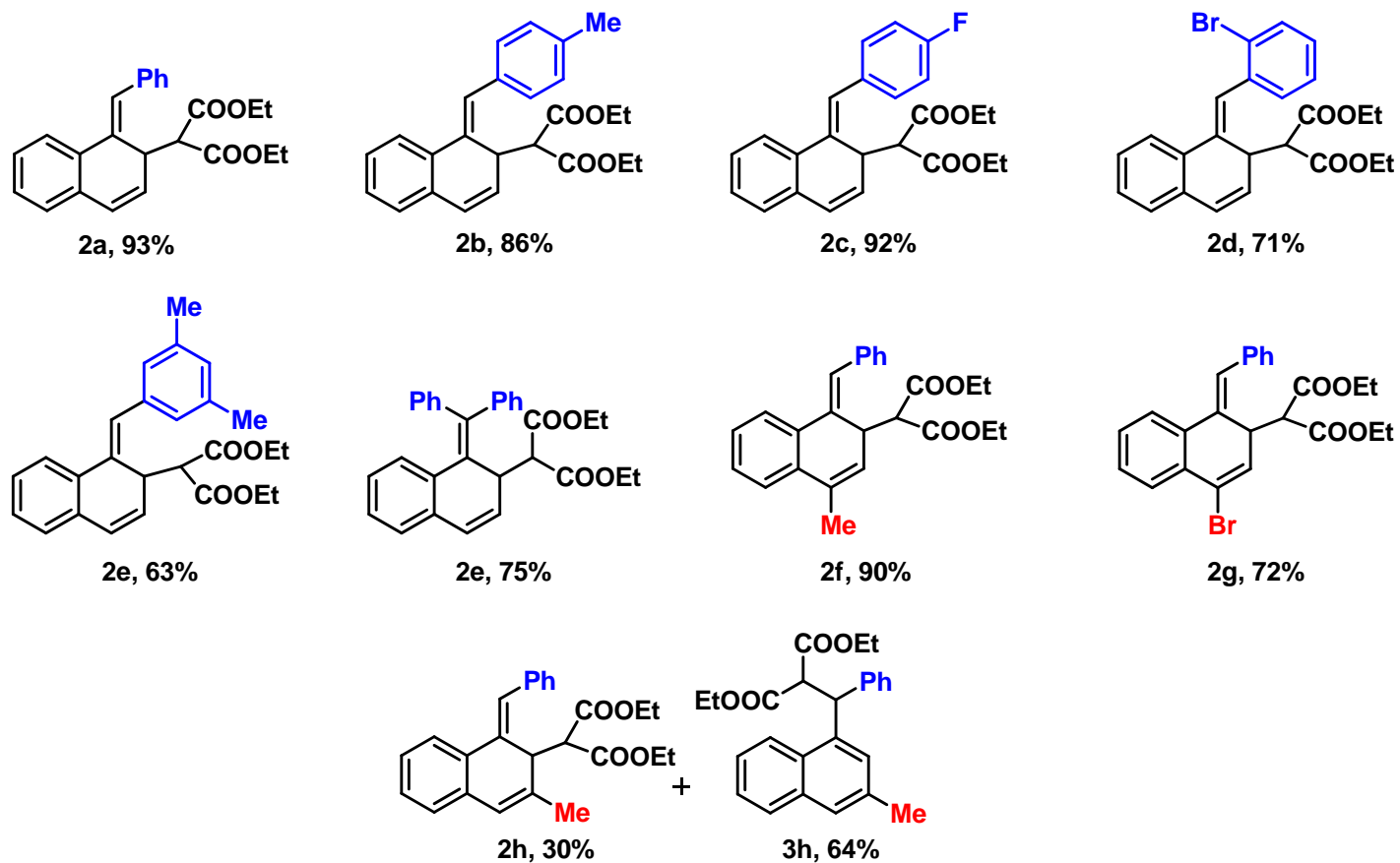
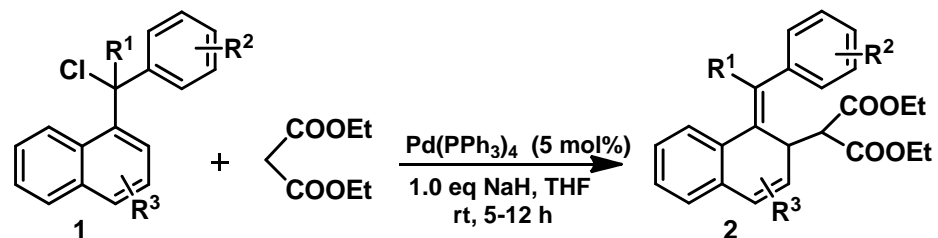
# Palladium catalyzed Dearomatization Reaction: exploration



entry	substrate	base	temp [°C]	product	yield [%]
5		K <sub>3</sub> PO <sub>4</sub>	-20 - rt		95 (5/95)
6		Cs <sub>2</sub> CO <sub>3</sub>	-20 - rt		87 (7/93)
7		NaH	rt		85 (37/16/47)
8		NaH	rt		90



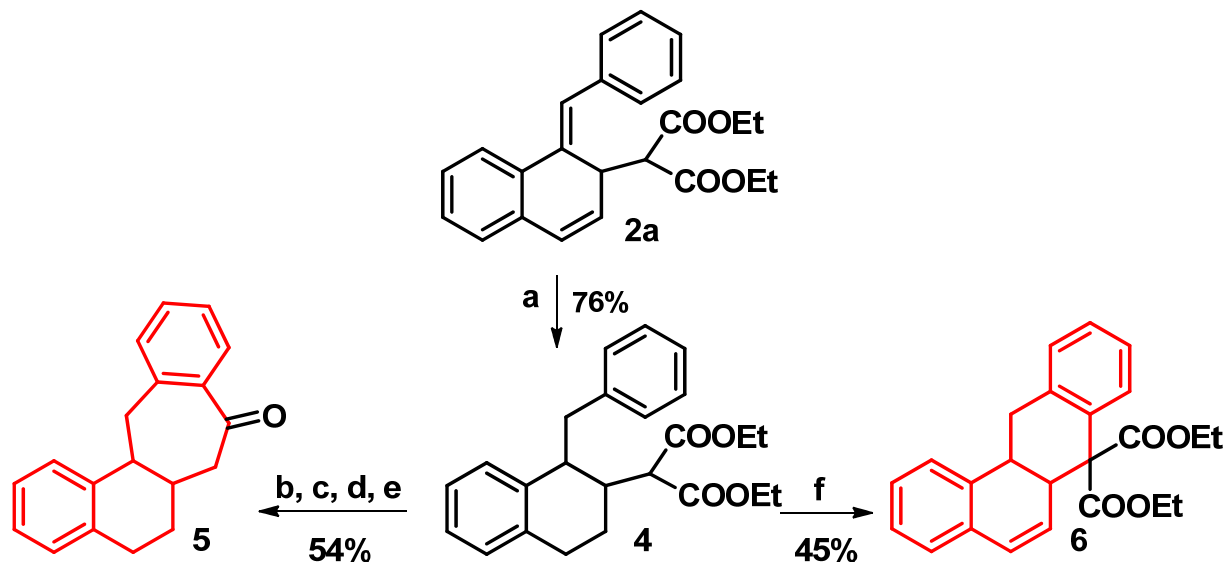
# Palladium catalyzed Dearomatization Reaction: scope study



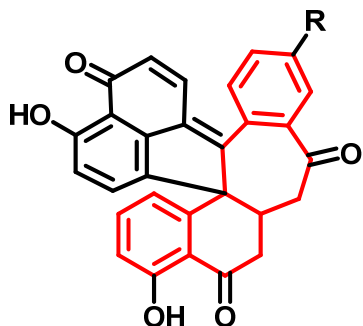


# Palladium catalyzed Dearomatization Reaction: application

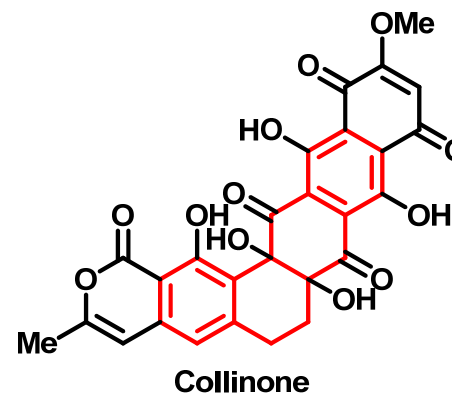
## Application to Tetracyclic Compounds



Reaction conditions: (a) Pd/C, H<sub>2</sub> (1 atm.), THF, rt; (b) KOH, EtOH, rt; (c) HCl, H<sub>2</sub>O, rt; (d) 170 °C; (e) PCl<sub>5</sub>, SnCl<sub>4</sub>, DCM, -20 °C; (f) Mn(OAc)<sub>3</sub>·H<sub>2</sub>O, HOAc, 140 °C

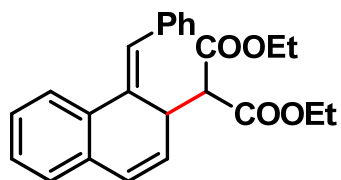
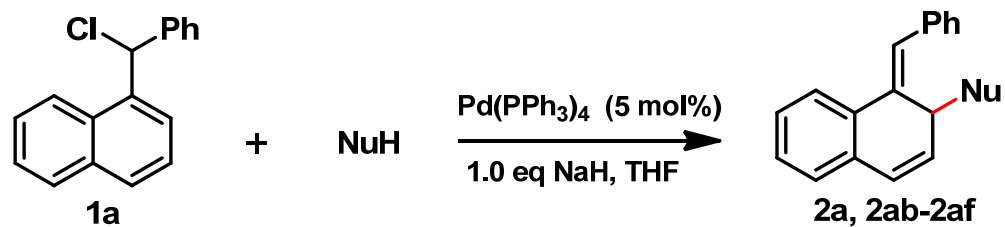


R = H, dalesconol A  
R = OH, dalesconol B

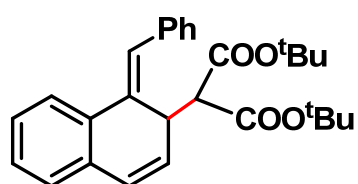




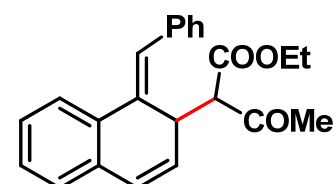
## Palladium catalyzed Dearomatization Reaction: Nu scope



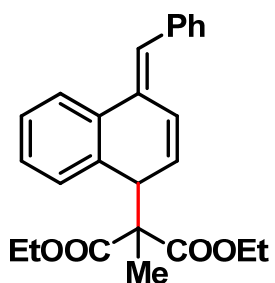
rt, 6 h  
2a, 93%



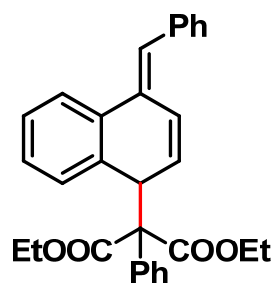
rt, 7 h  
2ab, 91%



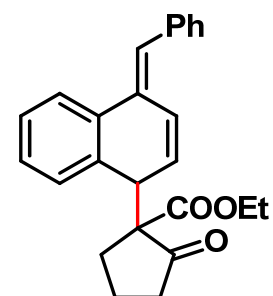
50 °C, 72 h  
2ac, 42%, 1/1 dr



rt, 6 h  
2ad, 92%



rt, 96 h  
2ae, 67%

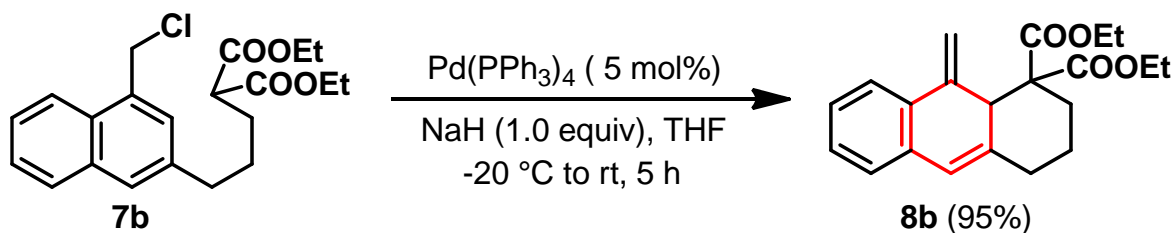
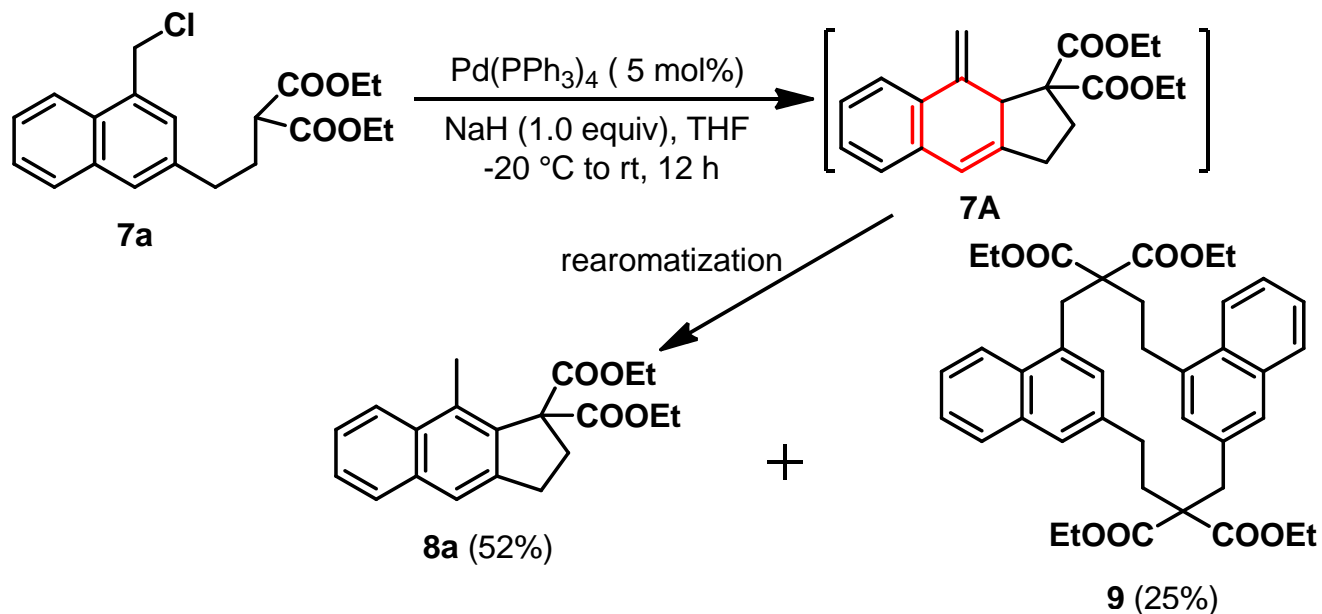


50 °C, 12 h  
2af, 71%, 3/2 dr



# Palladium catalyzed Dearomatization Reaction: intramolecular

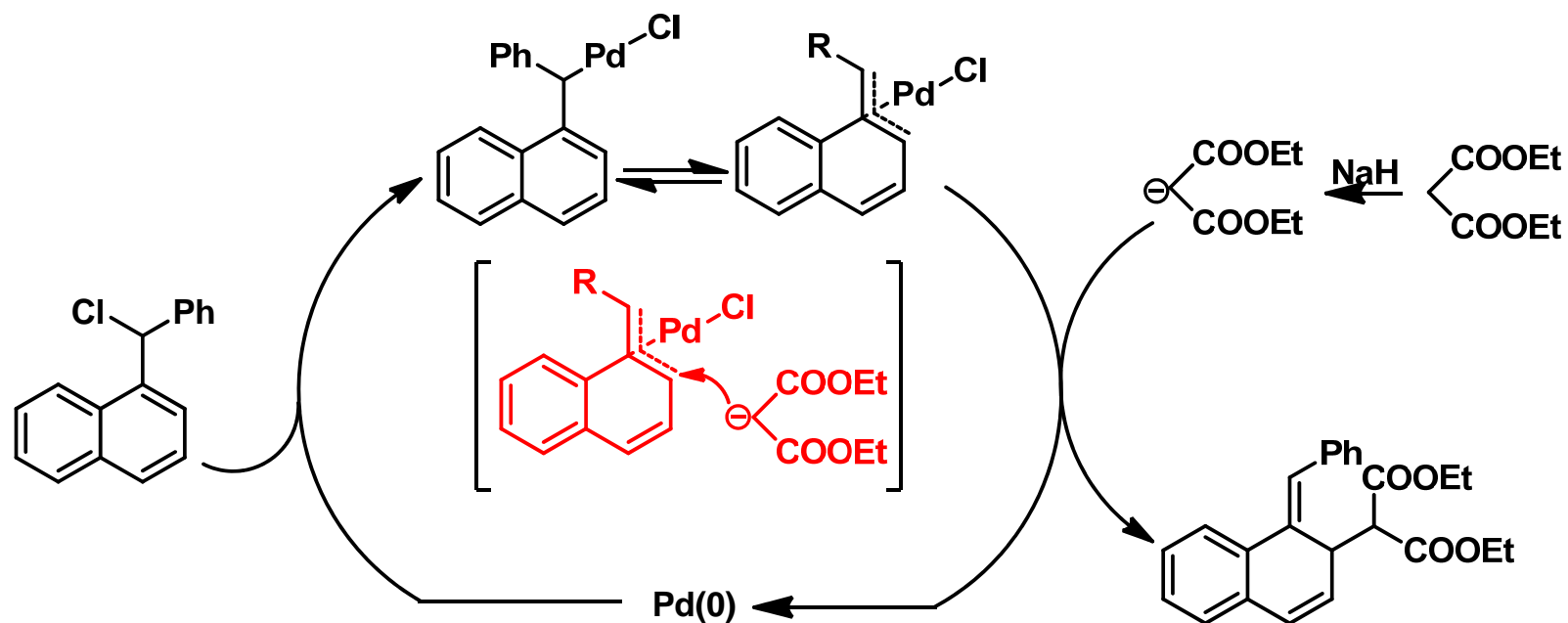
## Intramolecular palladium catalyzed dearomatization





# Palladium catalyzed Dearomatization Reaction: **mechanism**

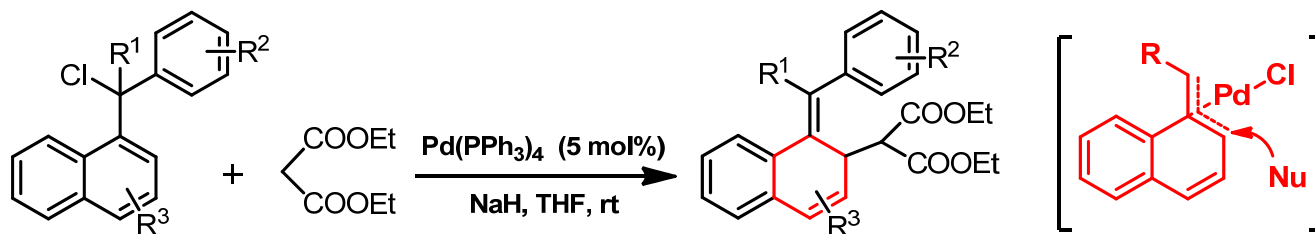
## Possible mechanism



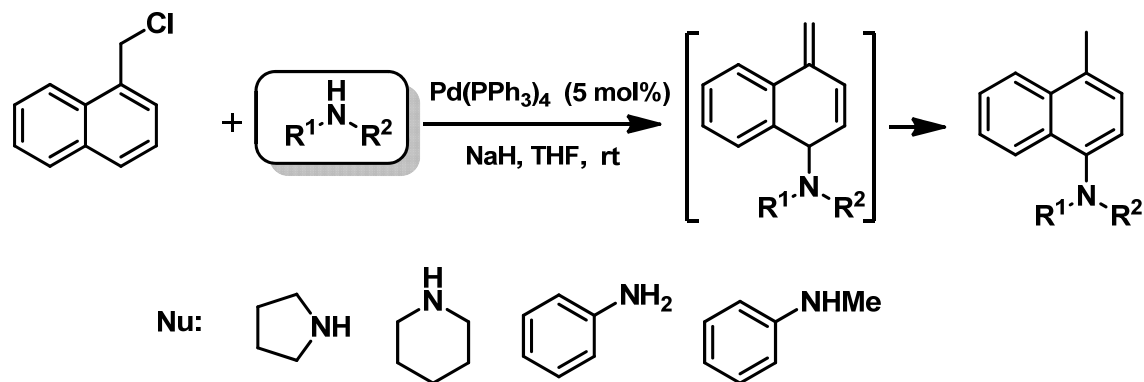


# Palladium catalyzed Dearomatization Reaction

## summary



## recent progress



a) Peng, B.; Zhang, S.; Yu, X.; Feng, X.; Bao, M. *Org. Lett.* **2011**, *13*, 5402-5405; b). Zhang, S.; Wang, Y.; Feng, X. *J. Am. Chem. Soc.* **2012**, *134*, 5492-5495.

# Palladium catalyzed Dearomatization Reaction

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**Electrophilic rearrangement of amides**

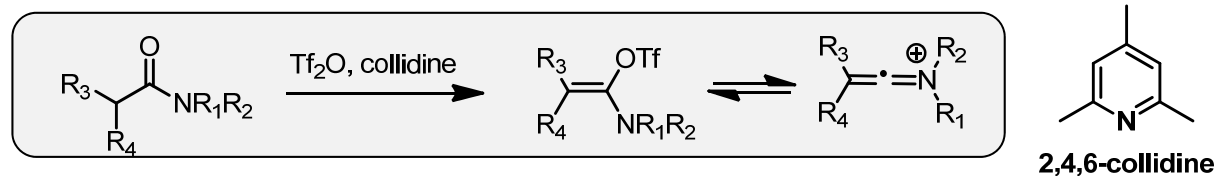




# Electrophilic rearrangement of amides: background

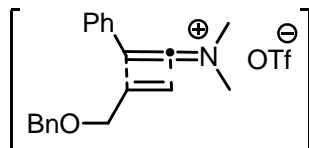
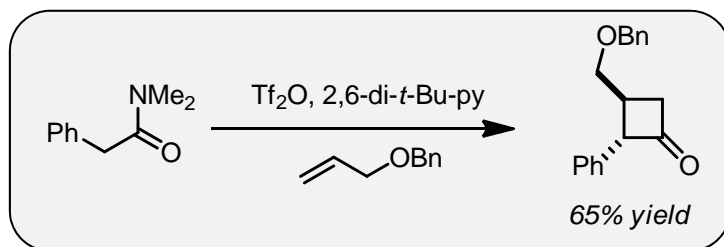
## Keteniminium Chemistry

### Generation of Keteniminium Salt

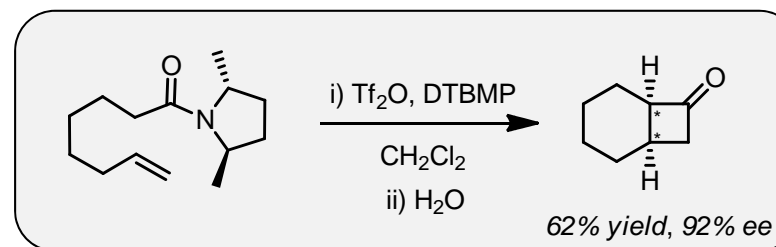


### [2+2] cycloaddition of Keteniminium Salt

#### Intermolecular



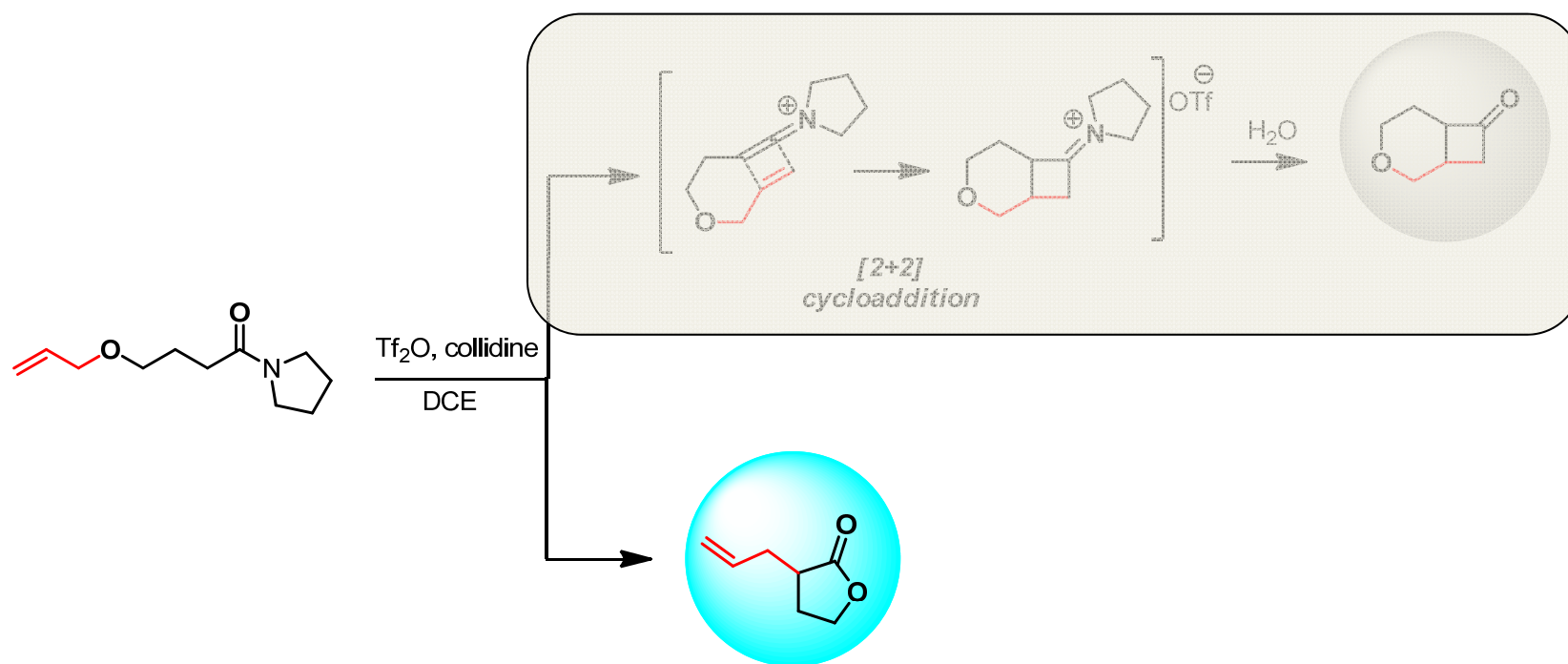
#### Intramolecular





# Electrophilic rearrangement of amides: background

## Rational design VS Serendipitous result

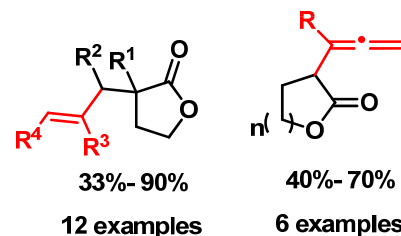
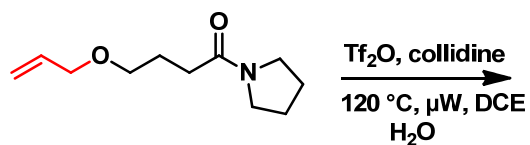




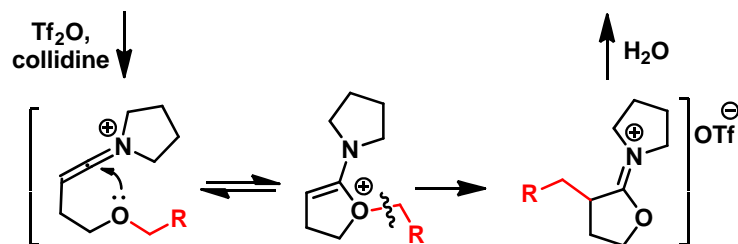
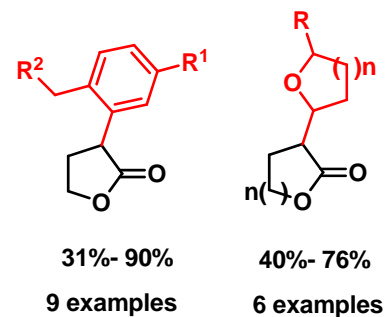
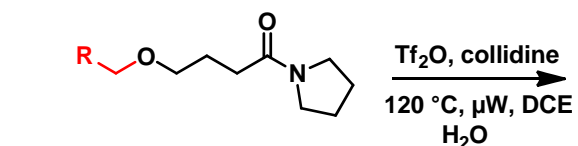
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## Electrophilic rearrangement of amides: background

2009 Claire, Vivi



2010 Vivi, Claire



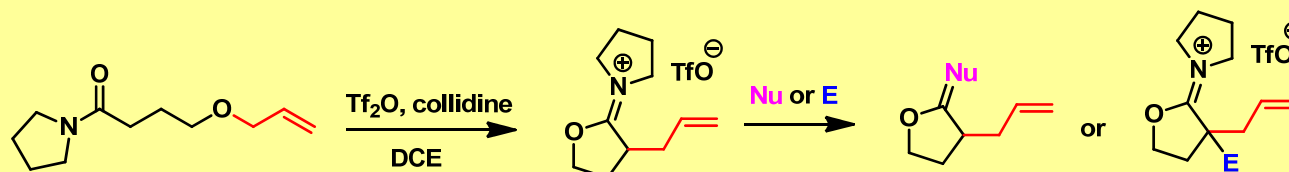
*Keteniminium induced Claisen Rearrangement*

b). Madelaine C., Valerio V., Maulide N., *Angew. Chem. Ed.* **2010**, 49, 1583-1586; c). Valerio V., Madelaine C., Maulide N., *Chem. Eur. J.* **2011**, 17, 4742-4745.

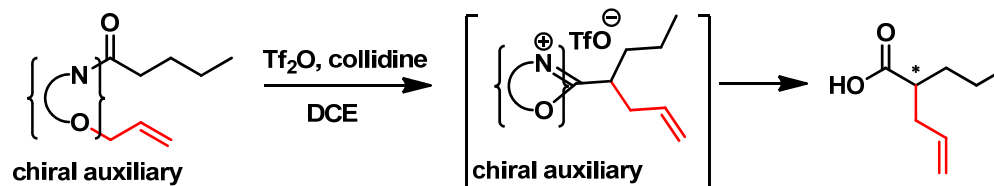


# Electrophilic rearrangement of amides: content

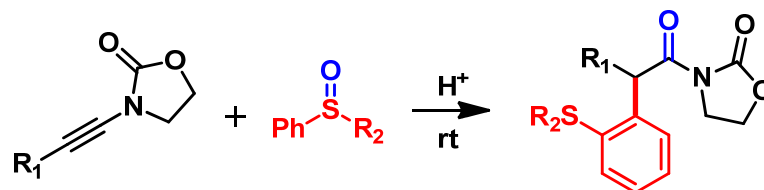
## Part 1. Nucleophilic/electrophilic capture of in-situ generated iminium ethers



## Part 2. Development of Asymmetric $\alpha$ -Allylation



## Part 3. Development of A Brønsted Acid-Catalyzed Redox Arylation

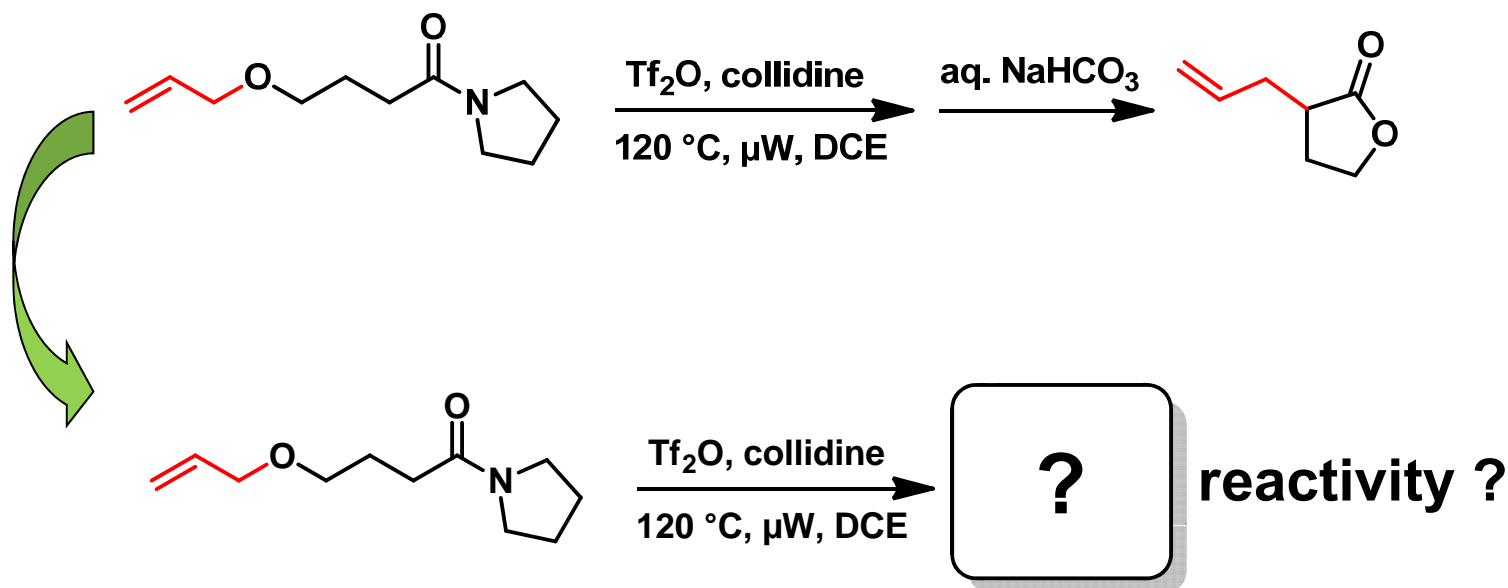




# Part 1. Nucleophilic/electrophilic capture of in-situ generated iminium ethers

With Daniel and Igor

*Previous discovery:*



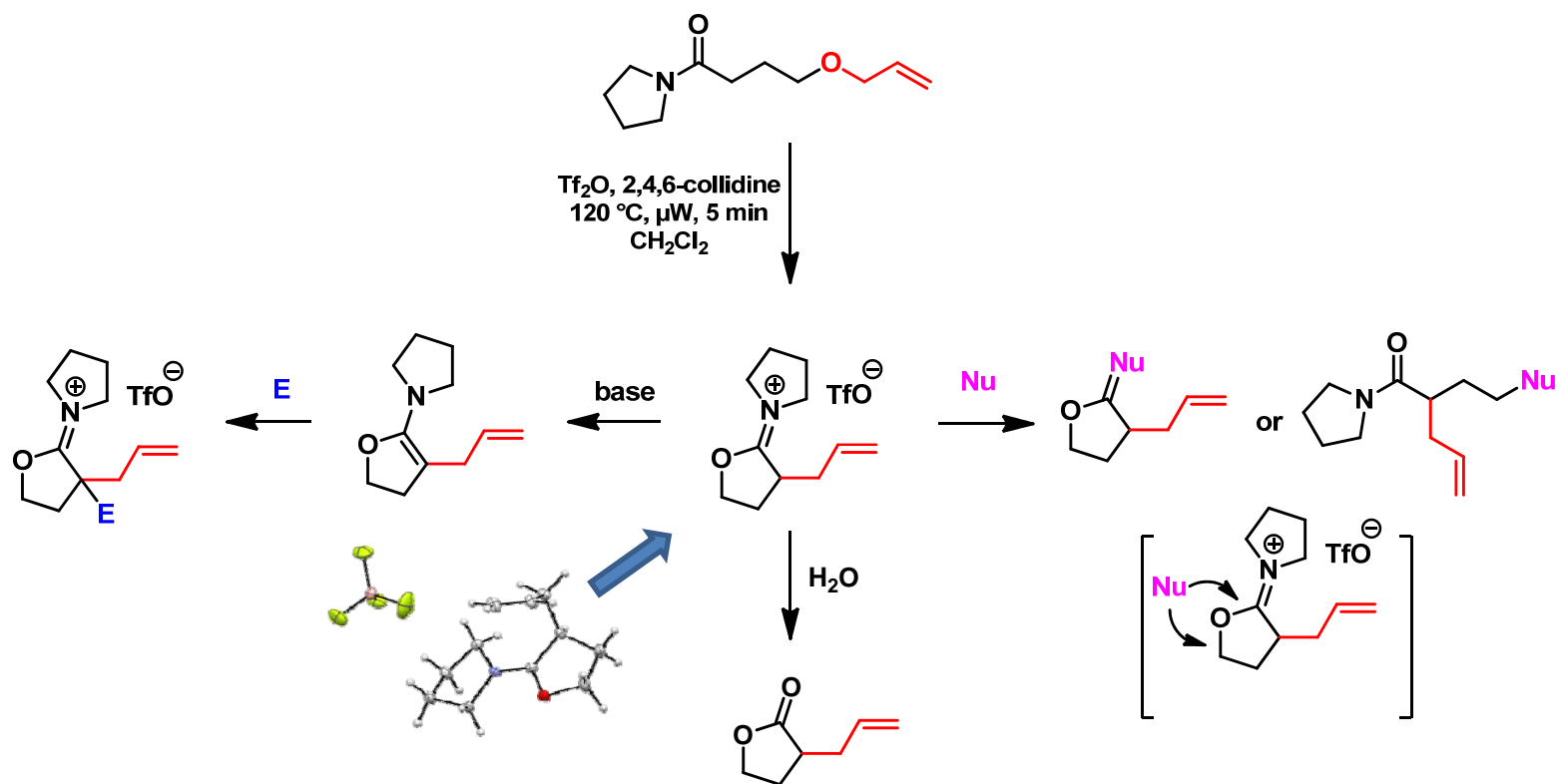


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# Part 1. Nucleophilic/electrophilic capture of in-situ generated iminium ethers

With Daniel and Igor

## Dual Nucleophilic/Electrophilic Capture of Iminium ether:



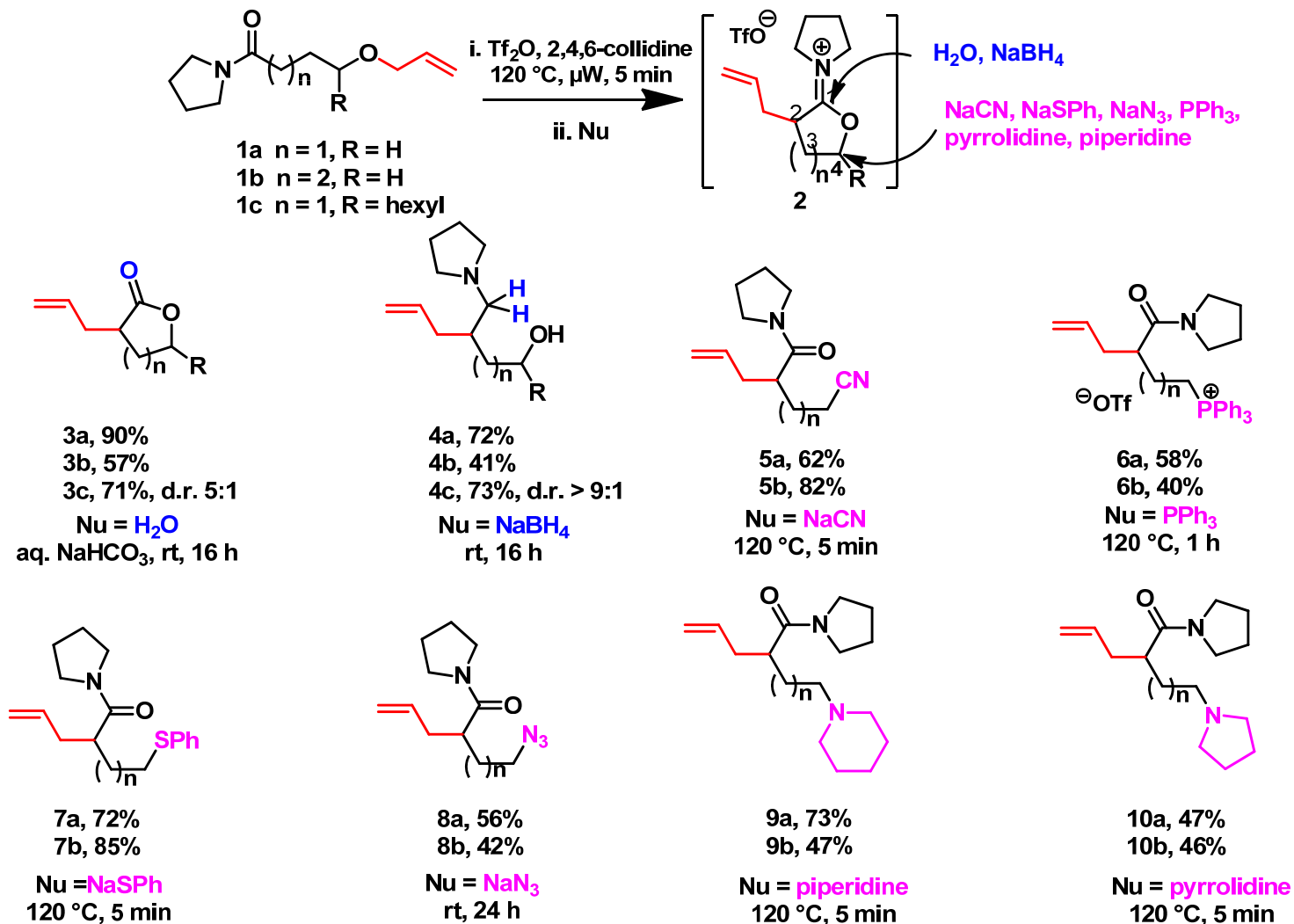


MAX-PLANCK-GESellschaft

# Part 1. Nucleophilic/electrophilic capture of in-situ generated iminium ethers

With Daniel and Igor

## scope of substrates

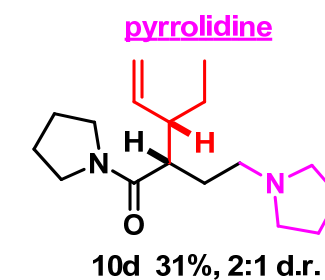
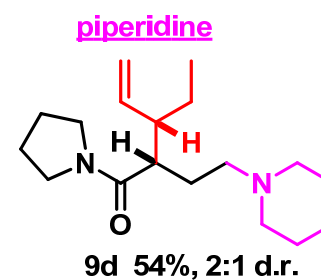
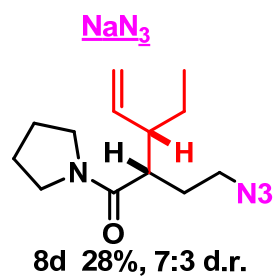
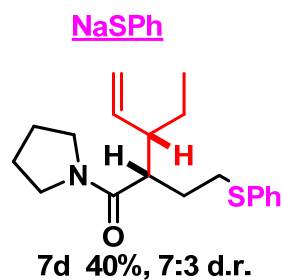
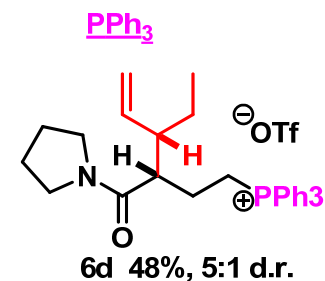
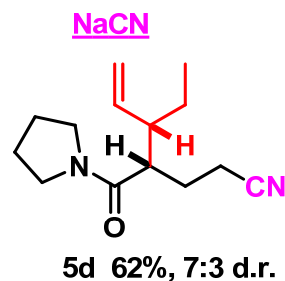
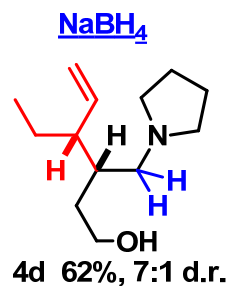
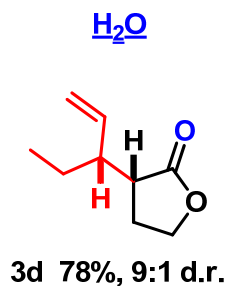
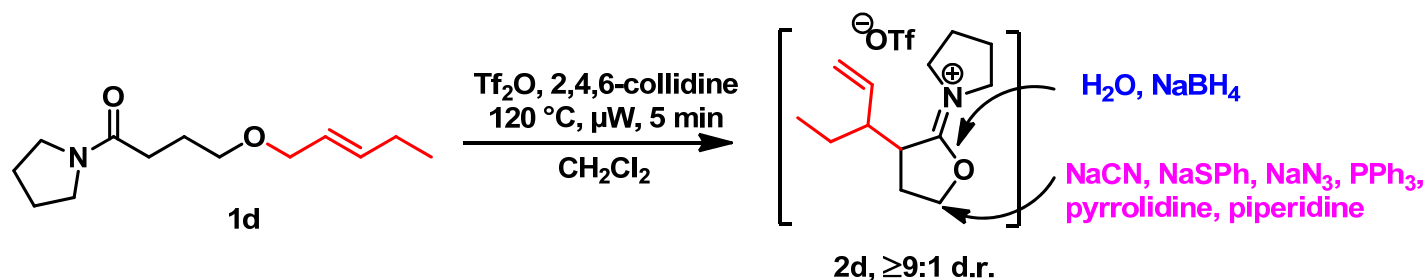




# Part 1. Nucleophilic/electrophilic capture of in-situ generated iminium ethers

With Daniel and Igor

## reactions of diastereomeric enriched iminium ether 2d

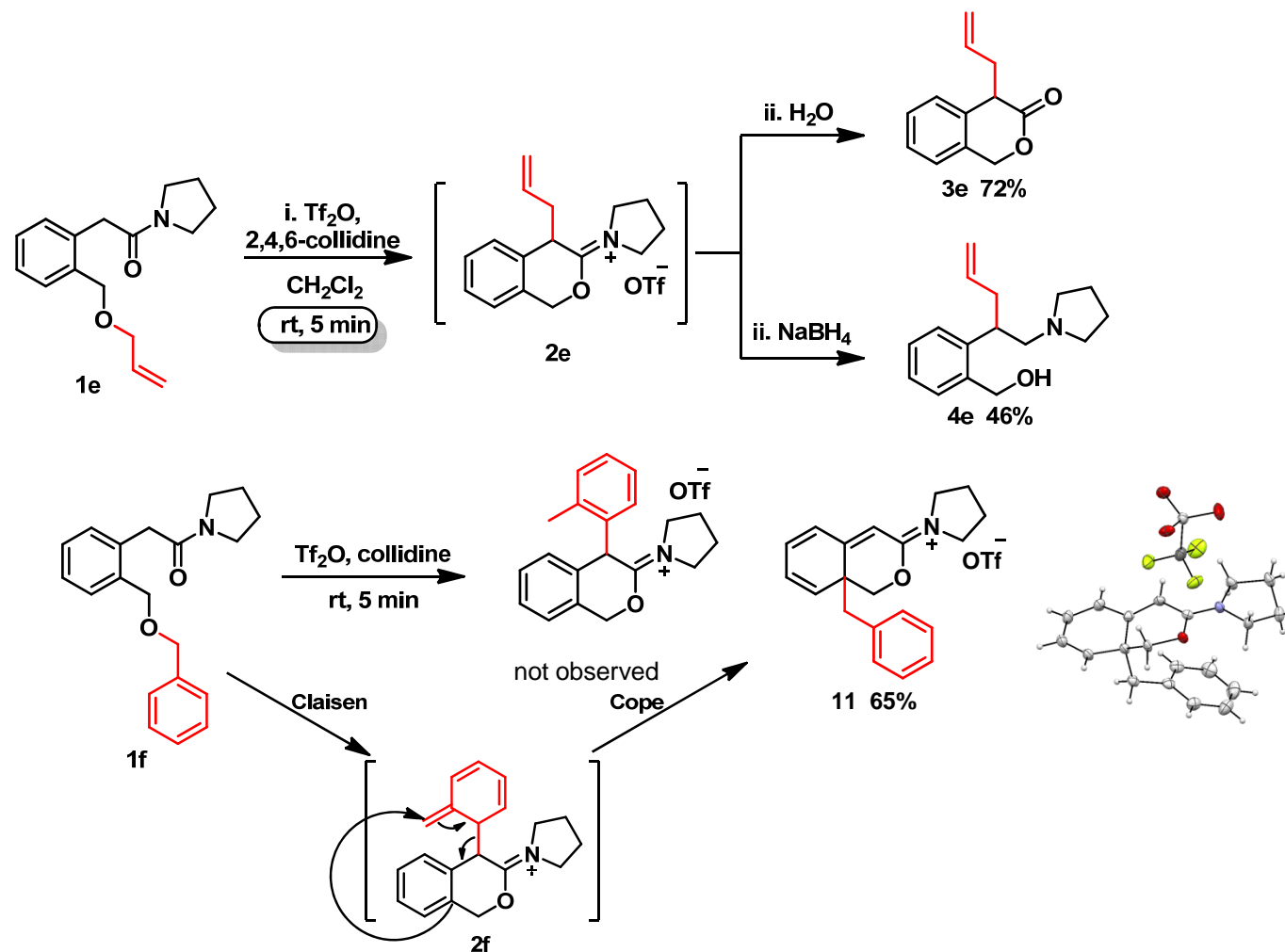




# Part 1. Nucleophilic/electrophilic capture of in-situ generated iminium ethers

With Daniel and Igor

## Dearomatisation inspired by Ji Woong's SM

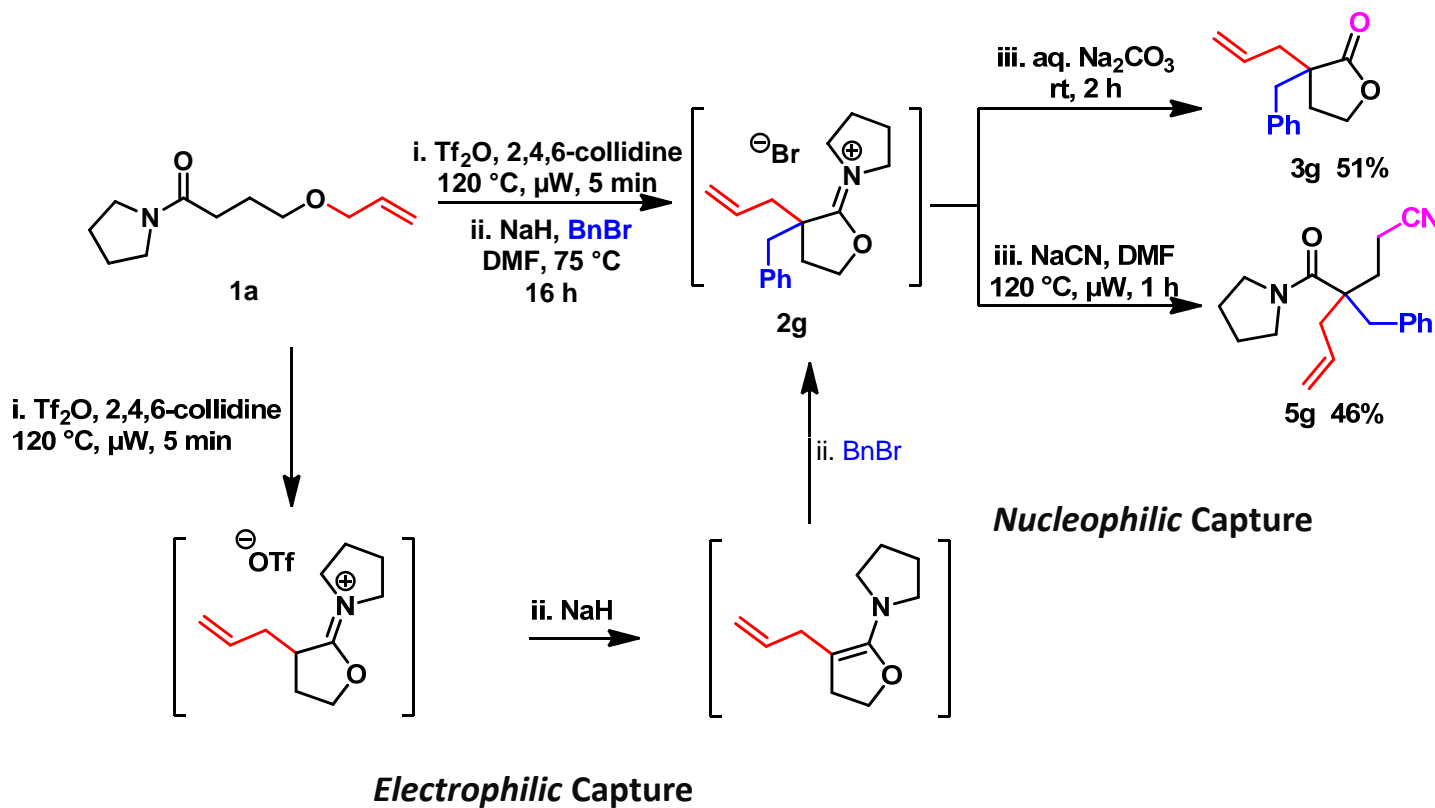




# Part 1. Nucleophilic/electrophilic capture of in-situ generated iminium ethers

With Daniel and Igor

## Tandem Electrophilic/Nucleophilic trap of In-Situ Generated Iminium Ethers



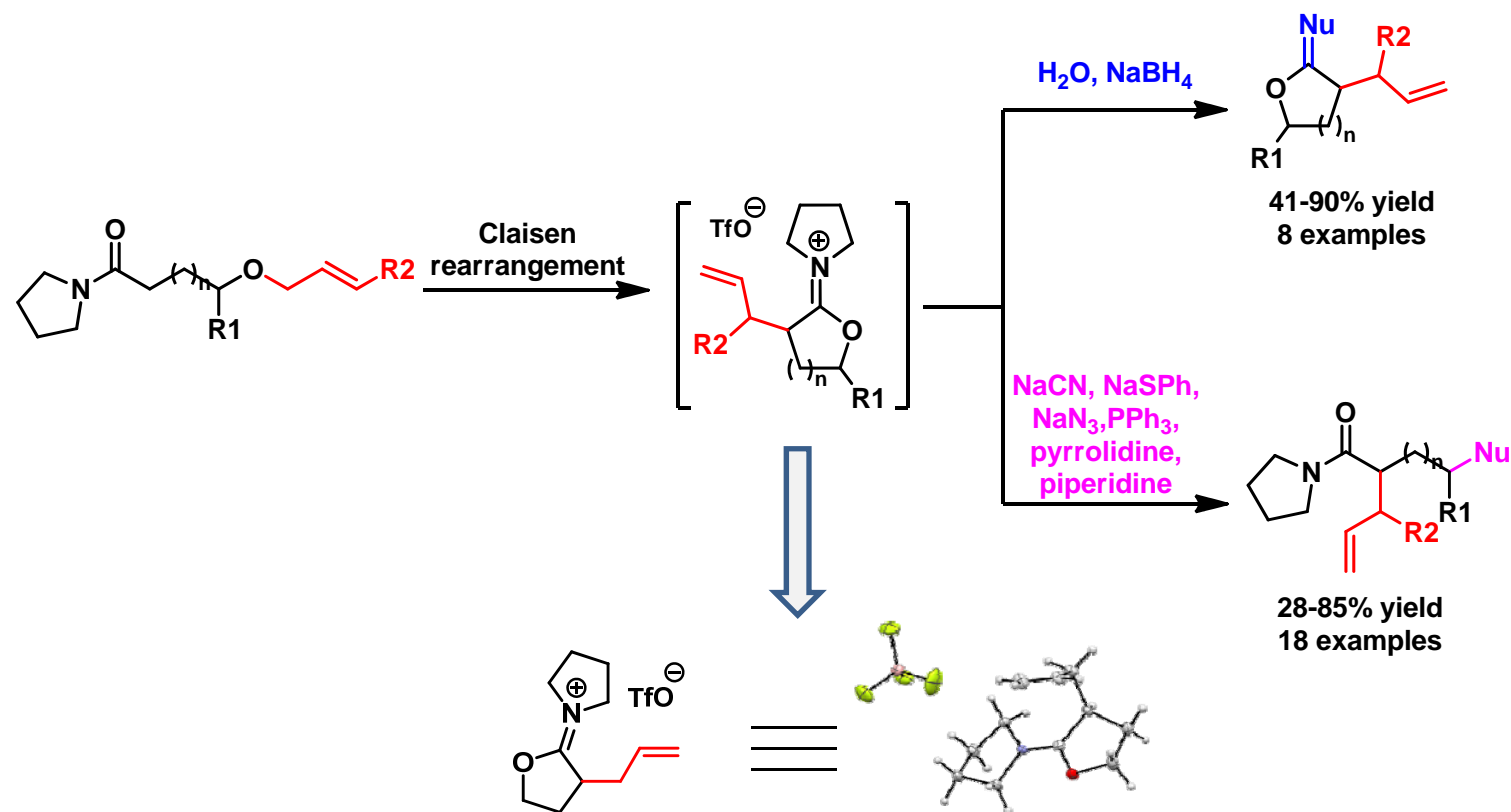


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# Part 1. Nucleophilic/electrophilic capture of in-situ generated iminium ethers

With Daniel and Igor

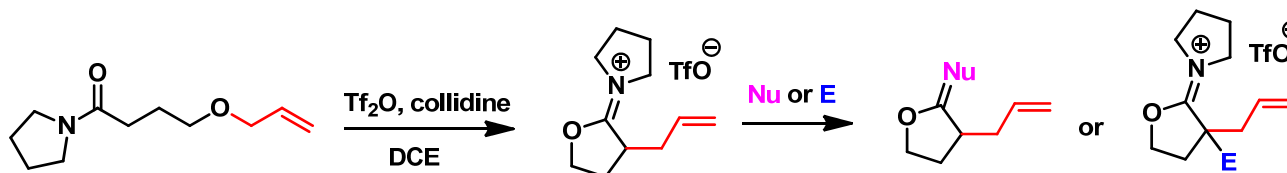
## summary



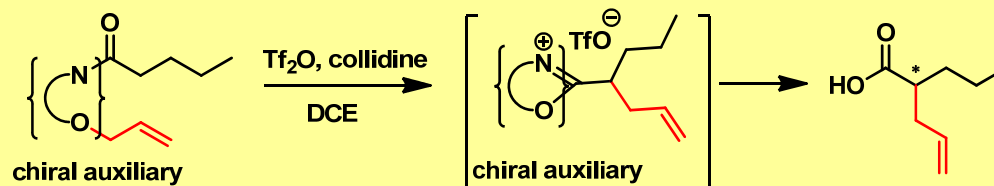


# Electrophilic rearrangement of amides: content

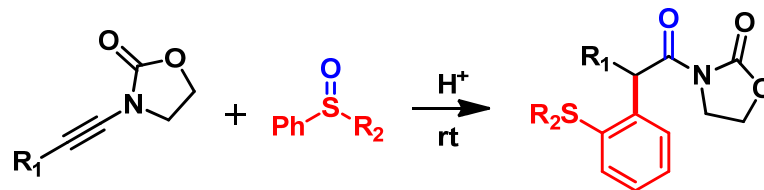
## Part 1. Nucleophilic/electrophilic capture of in-situ generated iminium ethers



## Part 2. Development of Asymmetric $\alpha$ -Allylation

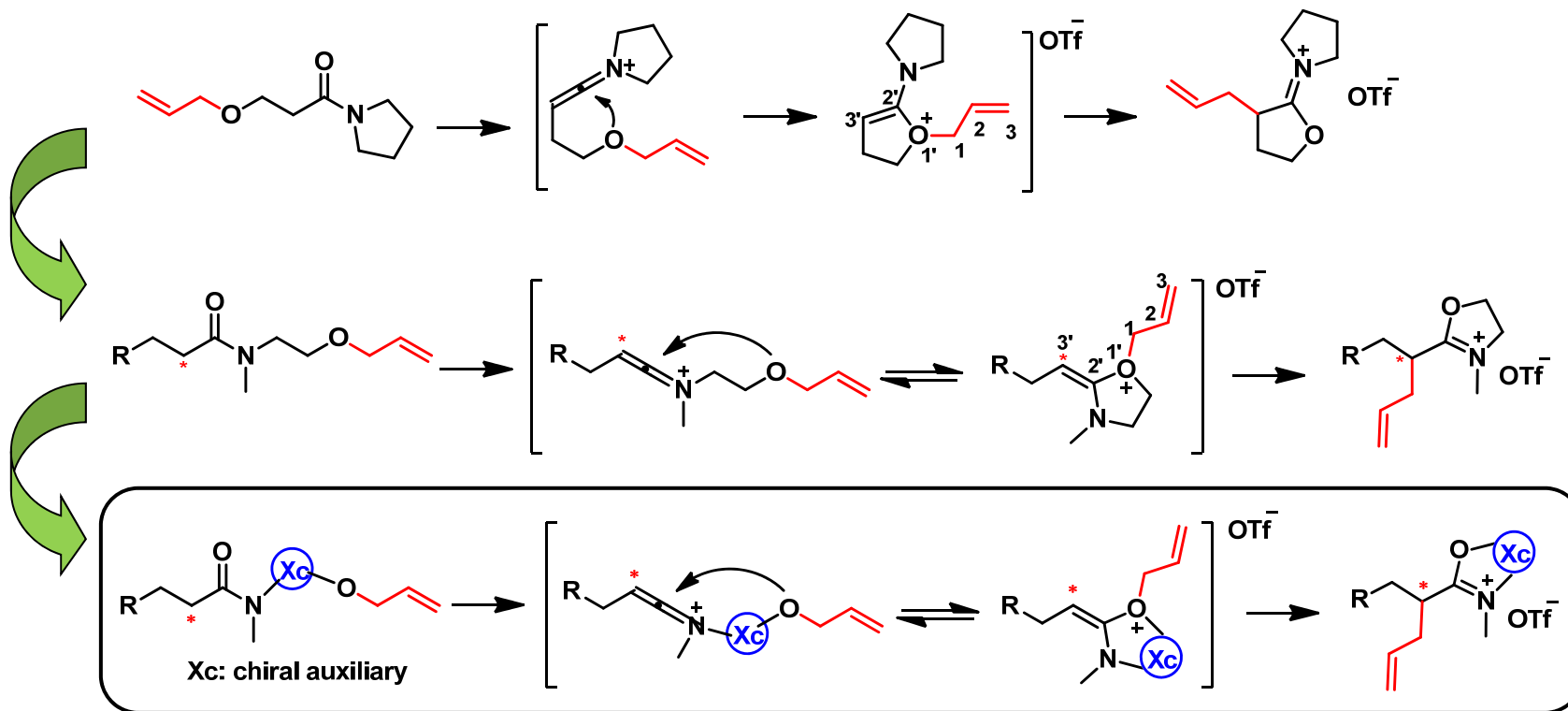


## Part 3. Development of A Brønsted Acid-Catalyzed Redox Arylation





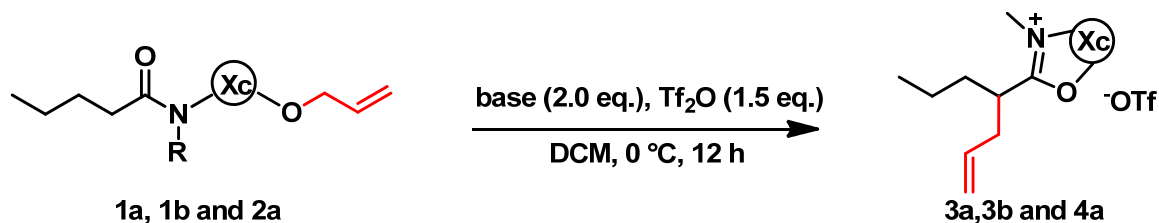
*Previous discovery:*



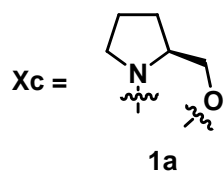
**Asymmetric  $\alpha$ -Allylation**

Part 2. Development of Asymmetric  $\alpha$ -Allylation

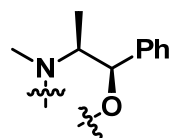
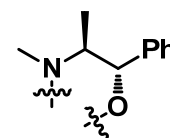
With Danny

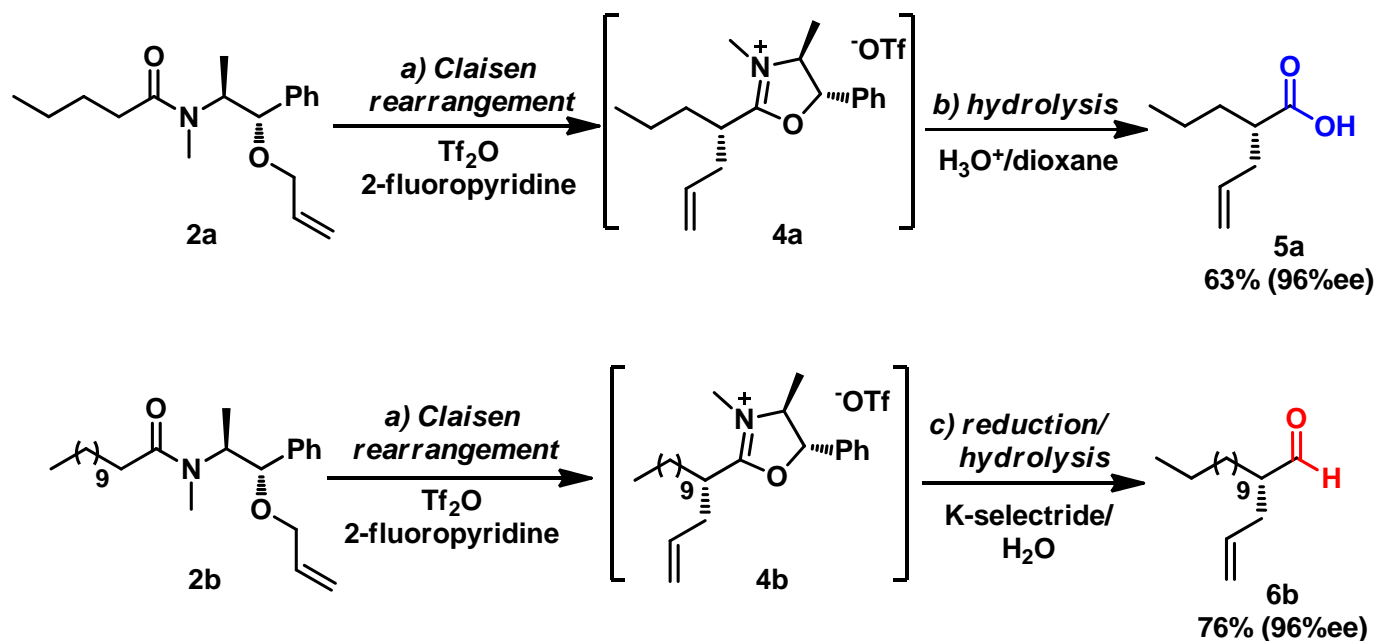


entry	substrate	base (2.0 eq.)	Product (yield)
1	1a	2,4,6-collidine	3a (---)
2	1b	2,4,6-collidine	3b (---)
3	2a	2,4,6-collidine	4a (16%)
4	2a	2-fluoropyridine	4a (> 95%)

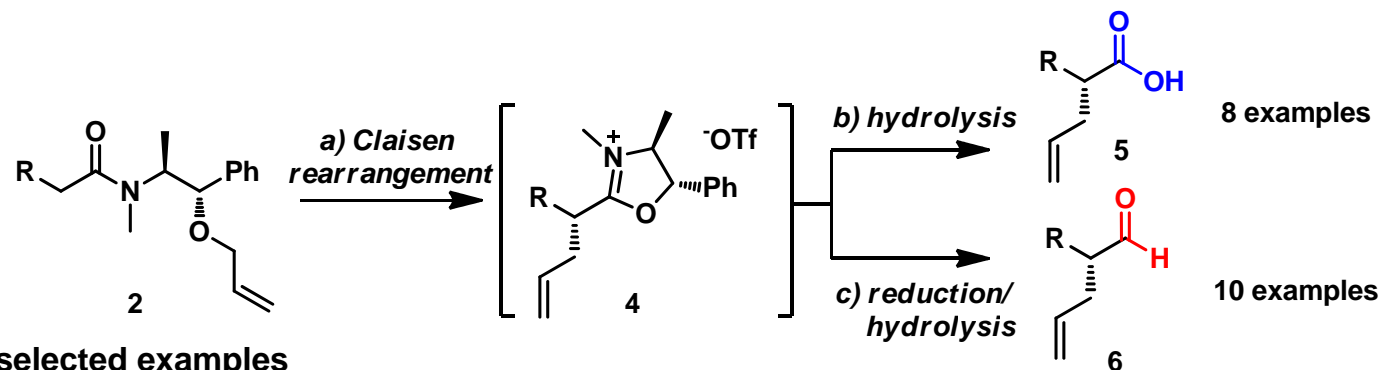


L-prolinol

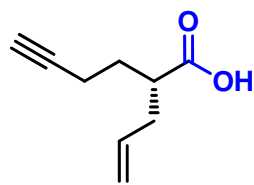
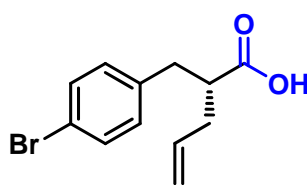
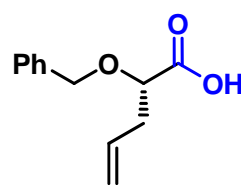
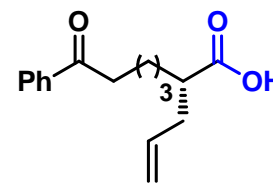
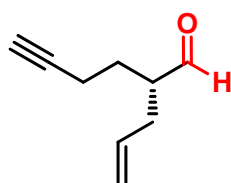
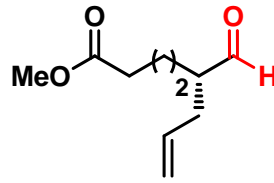
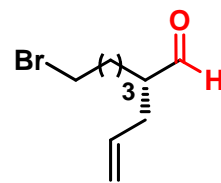
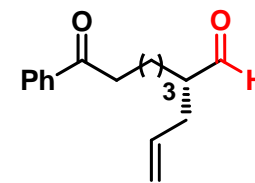
(1*R*,2*S*)-pseudoephedrine(1*S*,2*S*)-pseudoephedrine

**Remove auxiliary towards  $\alpha$ -allylated carboxylic acid and aldehyde**

- a):  $\text{Tf}_2\text{O}$  (1.5 eq), 2-fluoropyridine (2.0 eq), DCM, 0 °C, 12 h;  
b):  $\text{H}_2\text{SO}_4/\text{dioxane}$  (1/1), 100 °C, 24 h;  
c): K-selectride (3.0 eq), -78 °C to rt, 3 h, then silica/DCM/ $\text{H}_2\text{O}$ , rt, 12 h.

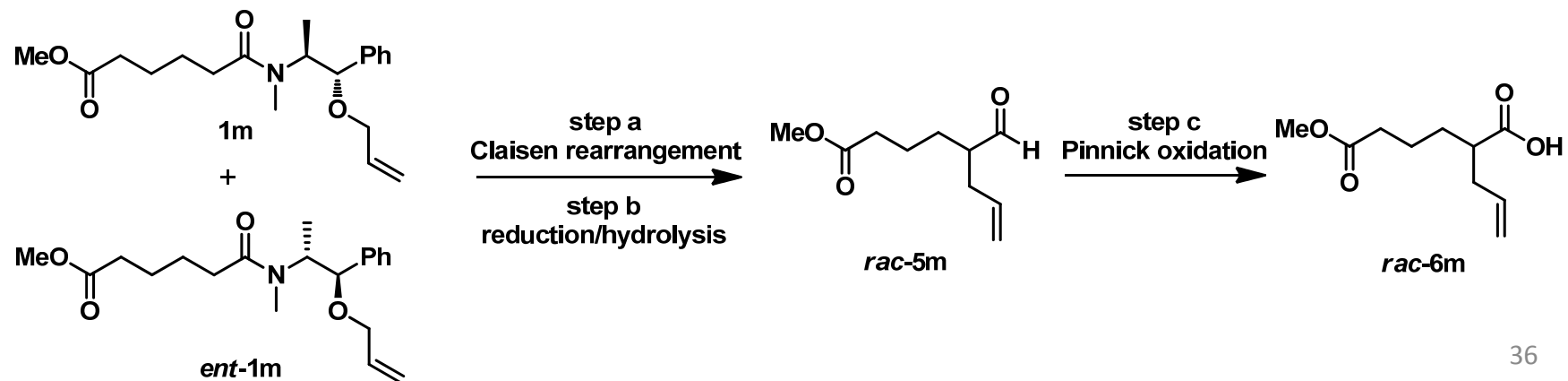
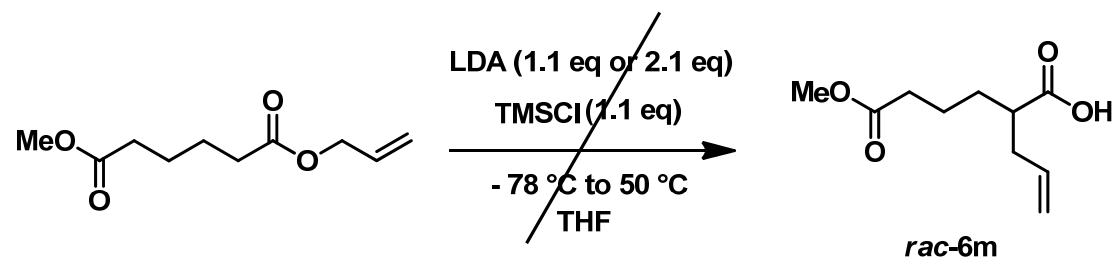
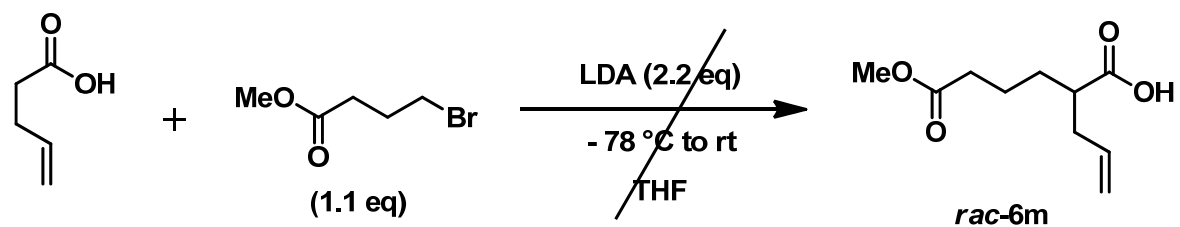
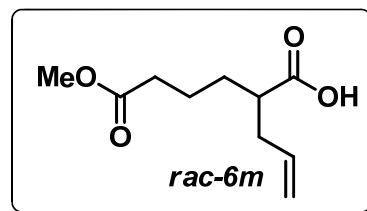


selected examples

*in situ* hydrolysis to carboxylic acids**5c**  
75%(88%ee)**5d**  
43%(88%ee)<sup>b,c</sup>**5g**  
47%(94%ee)**5h**  
80%(95%ee)*in situ* reduction/hydrolysis to aldehydes**6c**  
68%(90%ee)**6m**  
87% (90%ee)<sup>b,e</sup>**6k**  
55% (90%ee)<sup>b,e</sup>**6h**  
54%(91%ee)<sup>e</sup>

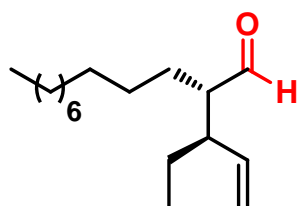
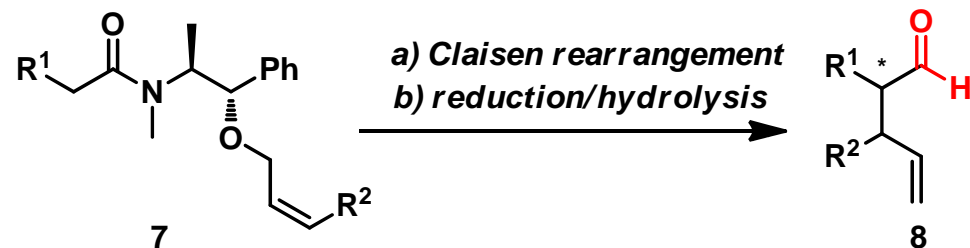
Part 2. Development of Asymmetric  $\alpha$ -Allylation

With Danny

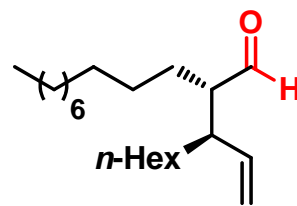


Part 2. Development of Asymmetric  $\alpha$ -Allylation

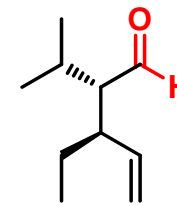
With Danny



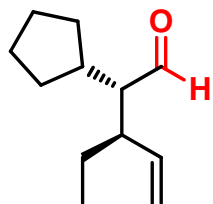
8a, 72%  
(d.r. 3.2/1, 94% ee)



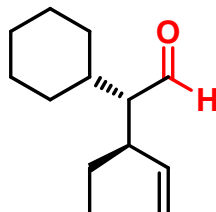
8b, 67%  
(d.r. 2.6/1, 93% ee)



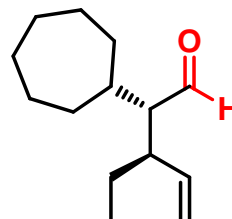
8c, 42%  
(d.r. 3.7/1, 94% ee)



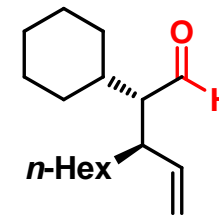
8d, 65%<sup>b</sup>  
(d.r. 4.3/1, 94% ee)



8e, 82%<sup>c</sup>  
(d.r. 5.3/1, 93% ee)



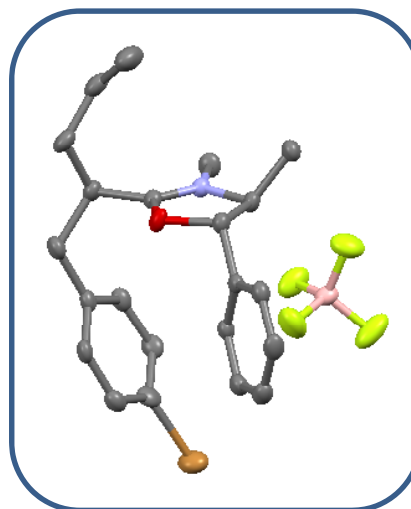
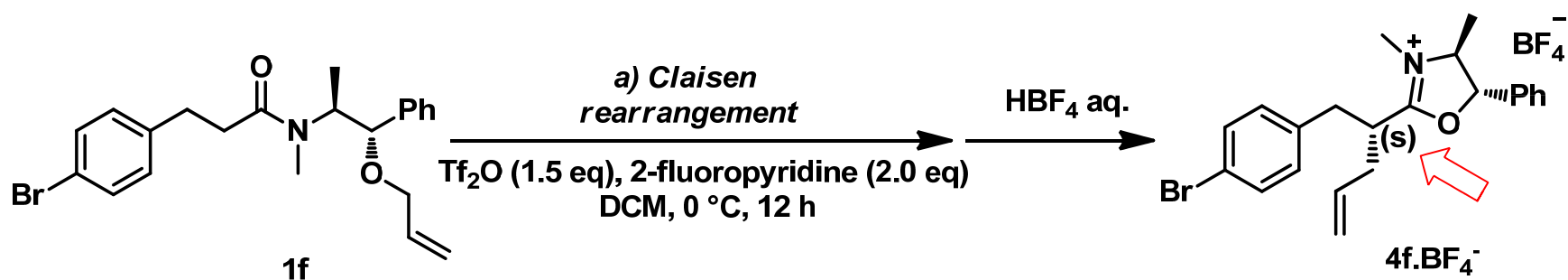
8f, 53%  
(d.r. 5.7/1, 92% ee)

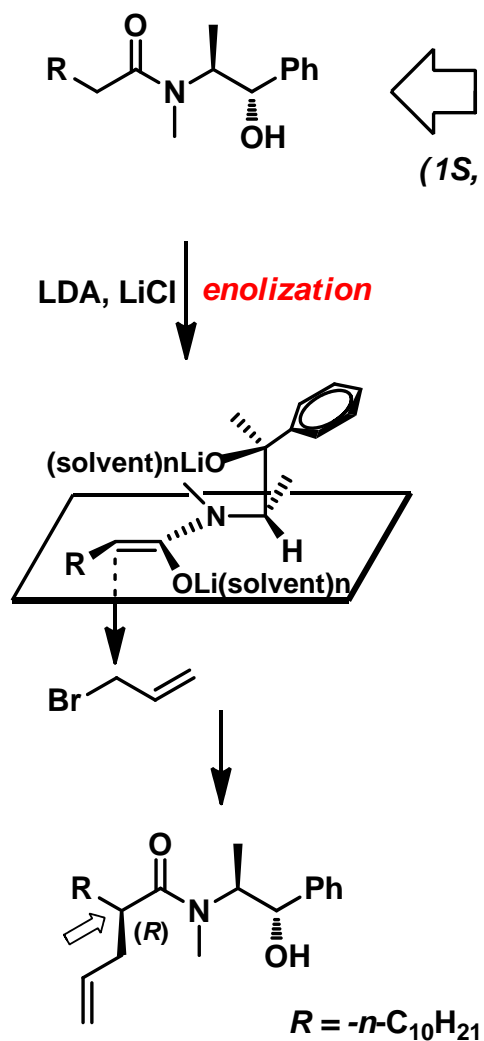
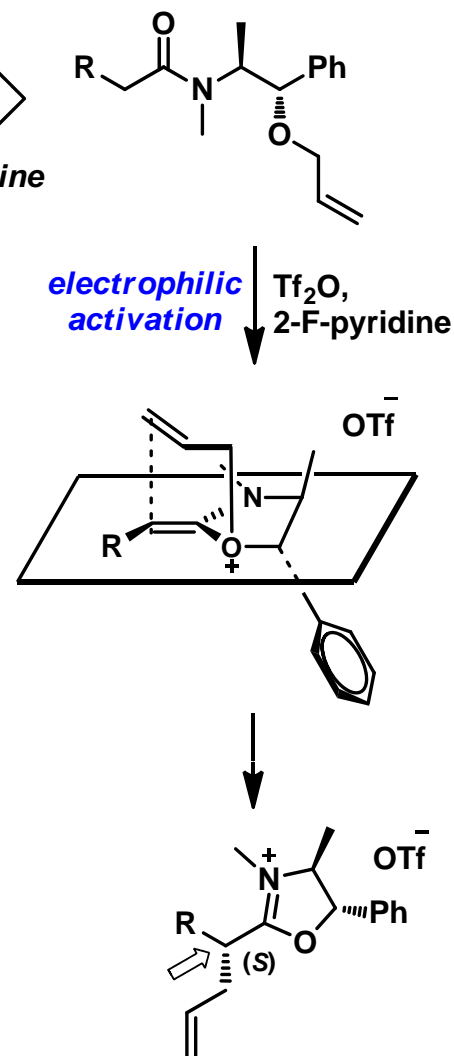


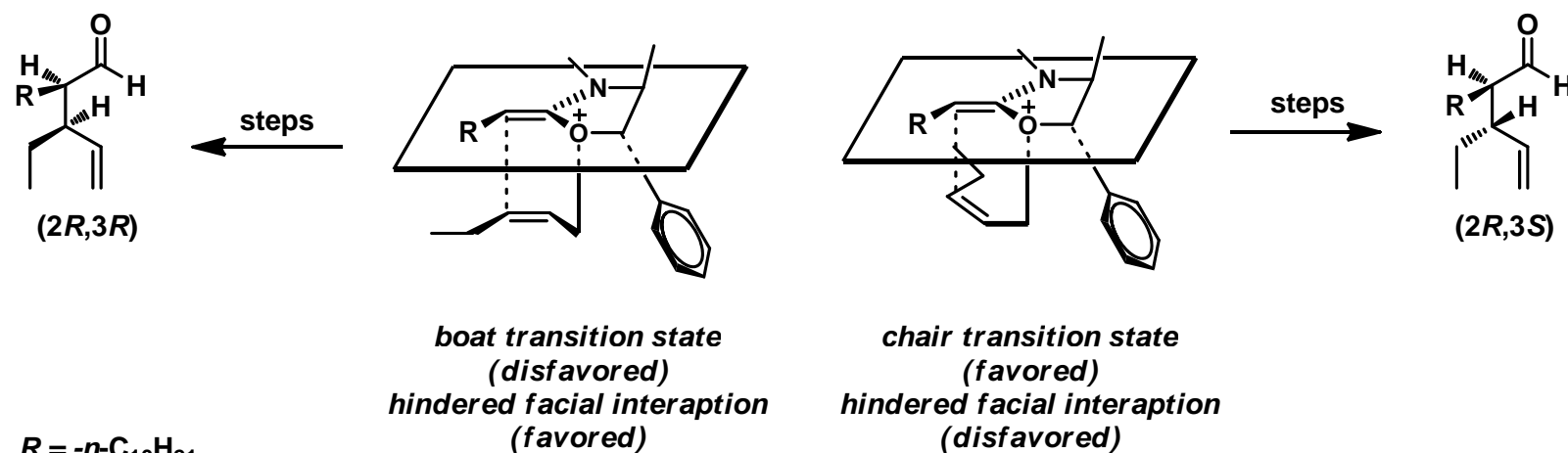
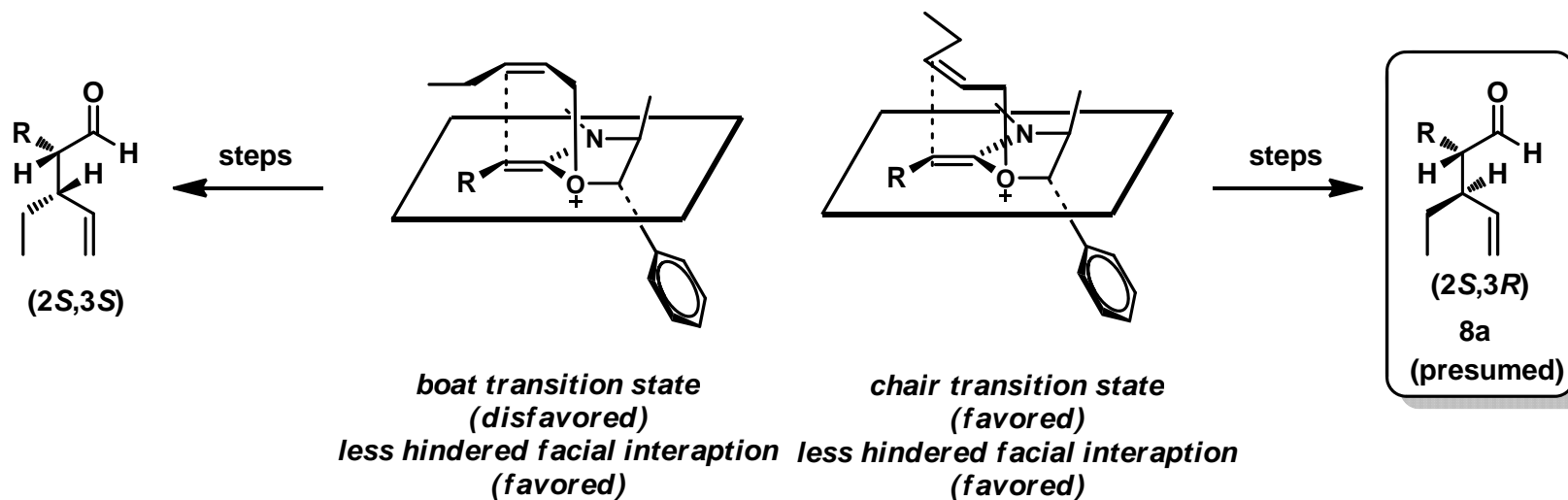
8g, 69%<sup>c</sup>  
(d.r. 3.2/1, 87% ee)



**Single-crystal X-ray determination of the absolute configuration**

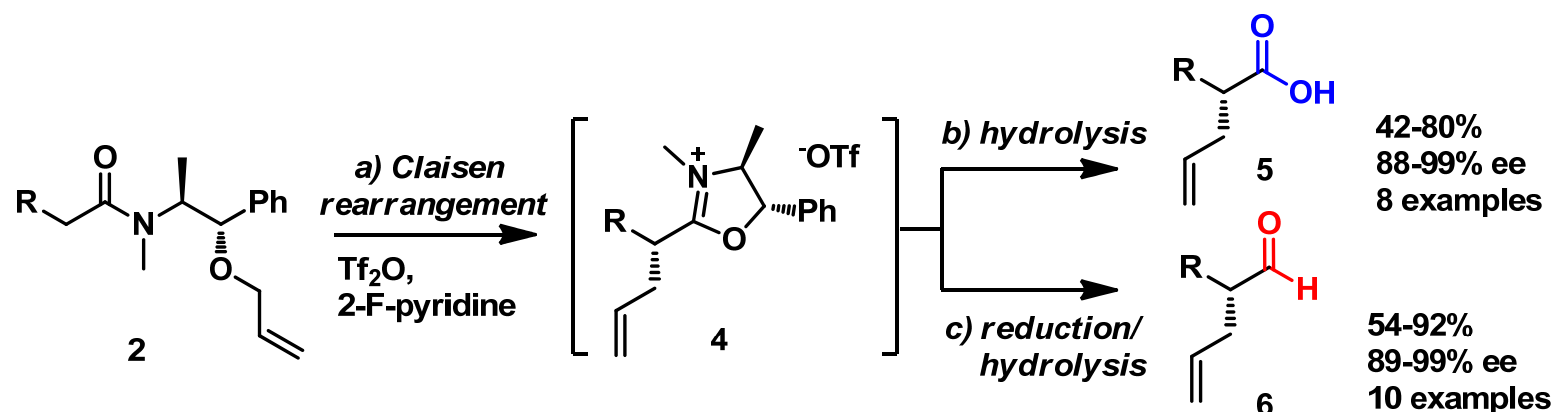


**Myers alkylation****Claisen rearrangement strategy**

**Mechanistic proposal for the observed diastereo- and enantioselectivity** $R = -n-C_{10}H_{21}$



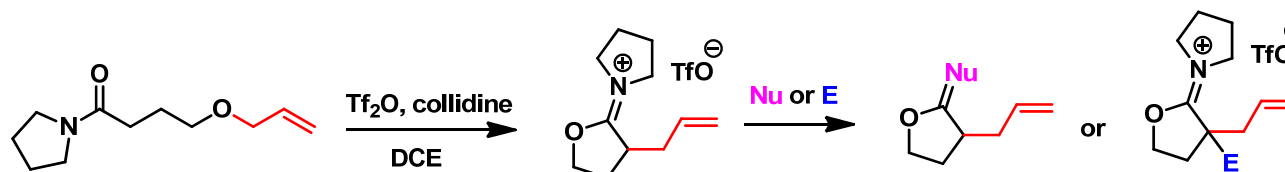
## summary



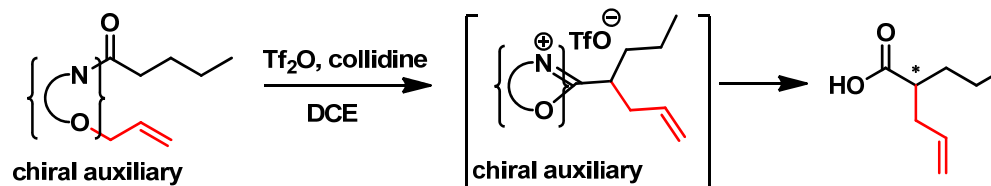


# Electrophilic rearrangement of amides: content

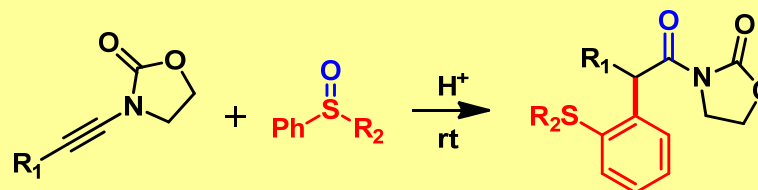
## Part 1. Nucleophilic/electrophilic capture of in-situ generated iminium ethers



## Part 2. Development of Asymmetric $\alpha$ -Allylation



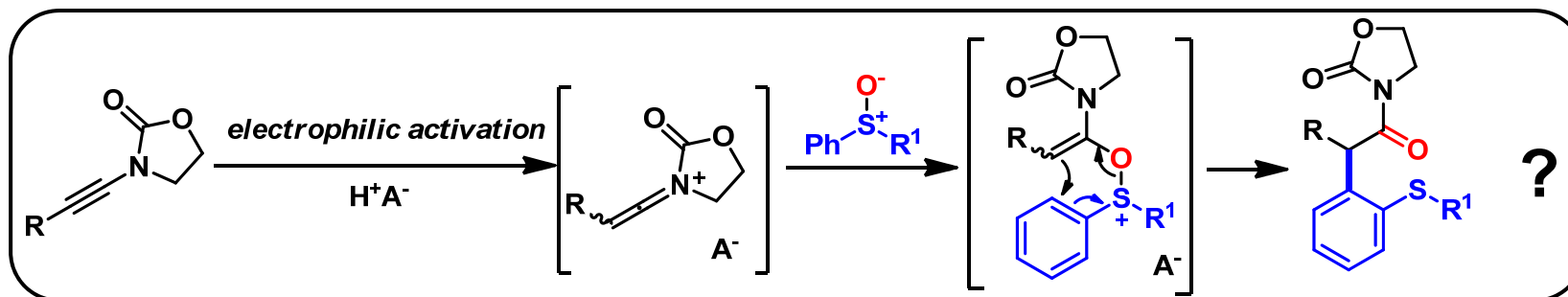
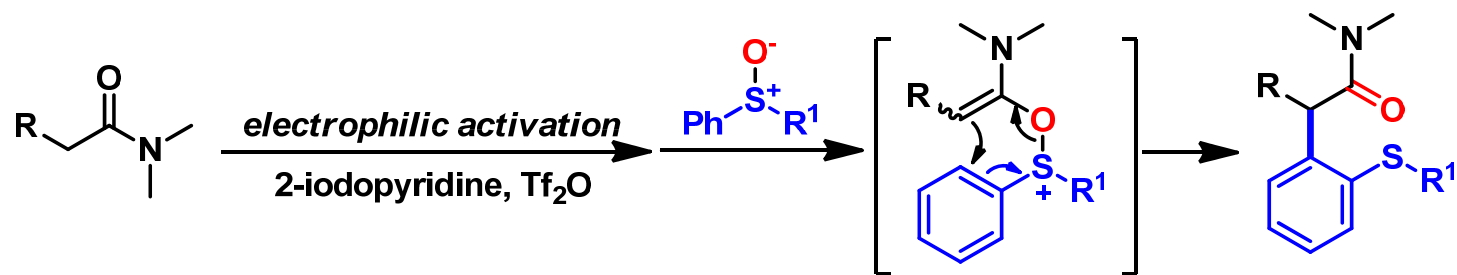
## Part 3. Development of A Brønsted Acid-Catalyzed Redox Arylation





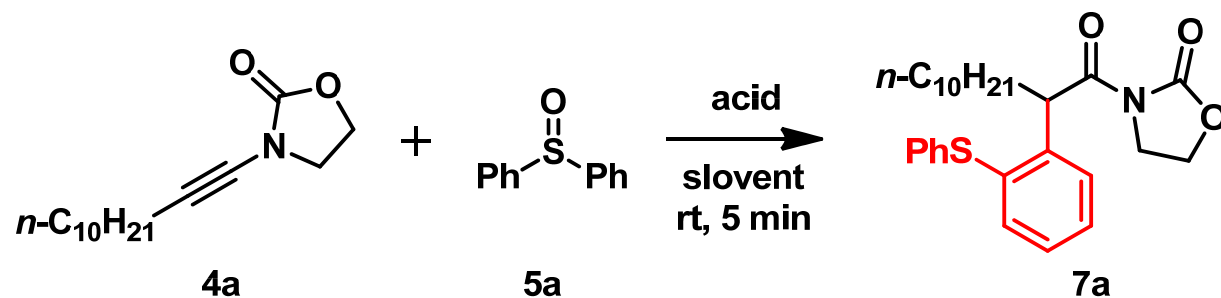
## Part 4. Development of A Brønsted Acid-Catalyzed Redox Arylation

**Previous work:**





## Part 4. Development of A Brønsted Acid-Catalyzed Redox Arylation



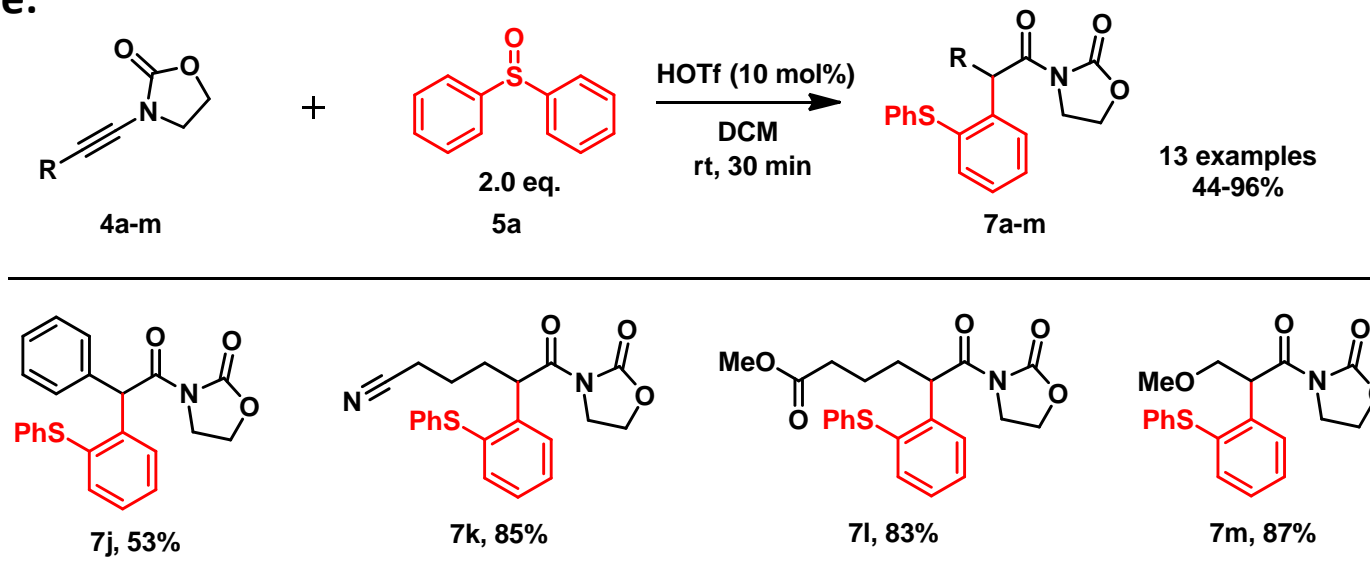
entry	acid(eq)	amount of nucleophile	solvent	yield
1	HOTf(1.0 eq)	1.0 eq	DCM	89%
2	TFA(1.0 eq)	1.0 eq	DCM	NR
3	HOTf(0.2 eq)	1.0 eq	DCM	90%
4	HOTf(1.0 eq)	2.0 eq	DCM	> 95%
5	HOTf(1.0 eq)	1.0 eq	toluene	74%
6	HOTf(0.1 eq)	2.0 eq	DCM	> 95%



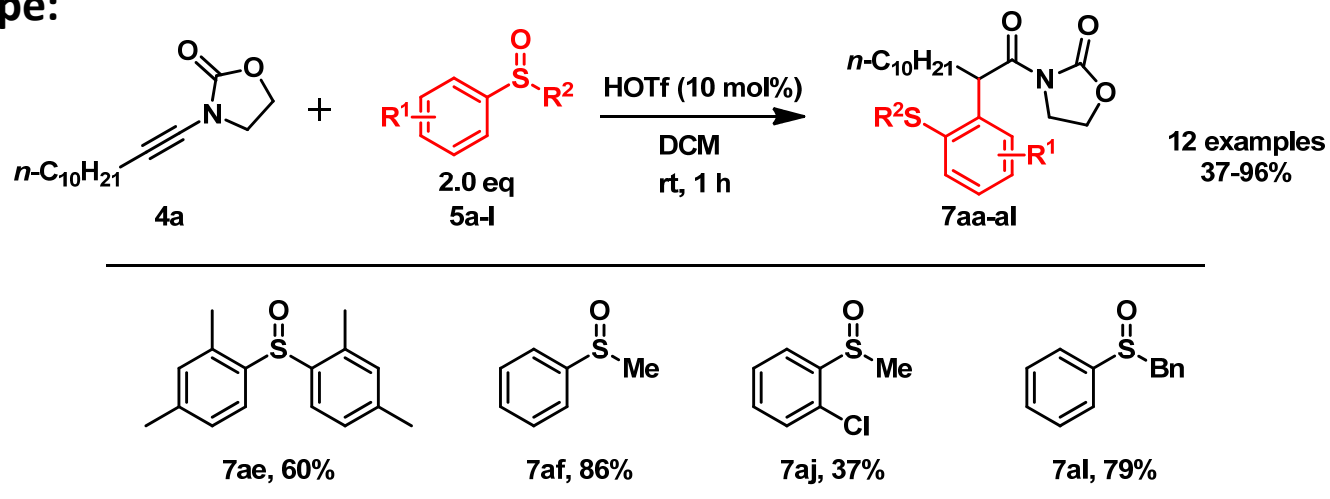
# Part 4. Development of A Brønsted Acid-Catalyzed Redox Arylation

With Huang and Xie

## Substrate scope:



## Nucleophile scope:

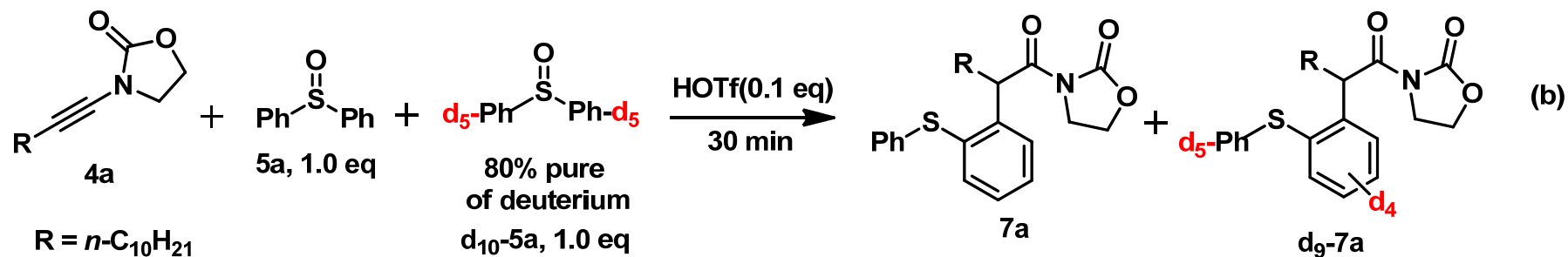
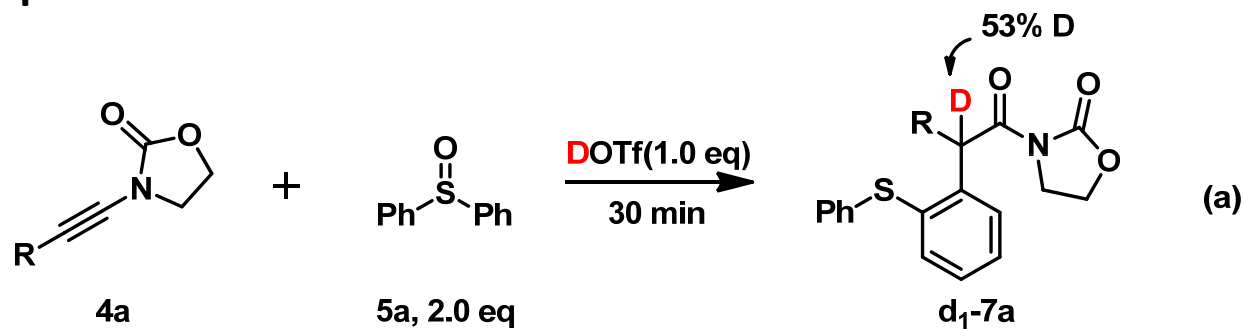




## Part 4. Development of A Brønsted Acid-Catalyzed Redox Arylation

With Huang and Xie

labelling experiments:



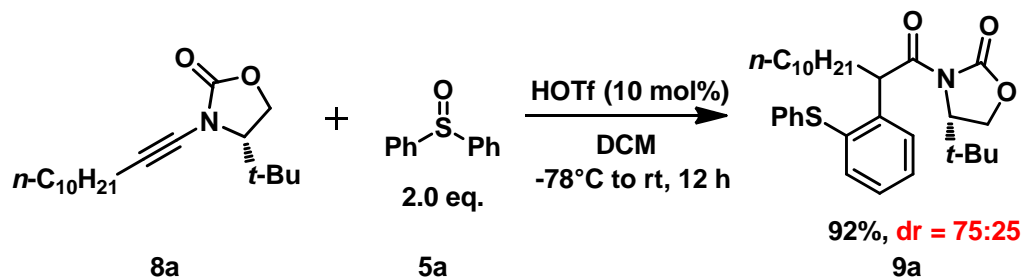
**7a** : **d<sub>9</sub>-7a** = 63 : 37



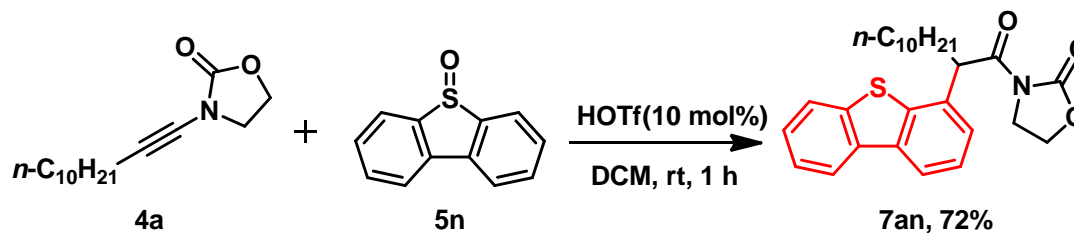
## Part 4. Development of A Brønsted Acid-Catalyzed Redox Arylation

With Huang and Xie

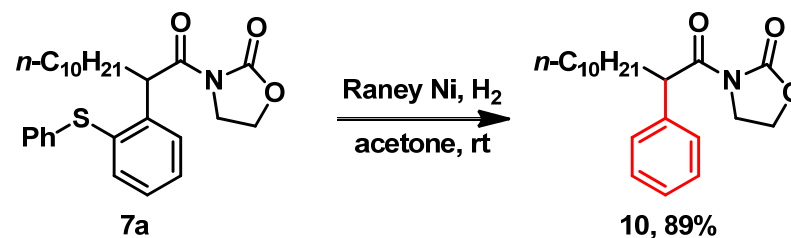
### Asymmetric arylation reaction



### Synthesis of a dibenzothiophene:



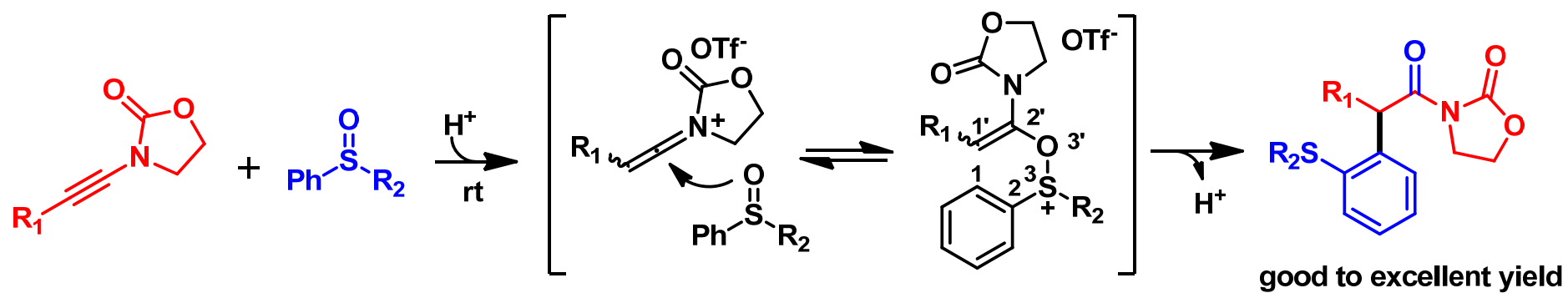
### Elaboration of products:





## Part 4. Development of A Brønsted Acid-Catalyzed Redox Arylation

### summary



# Acknowledgements



**Prof. Ming Bao**



**Prof. Nuno Maulide**

