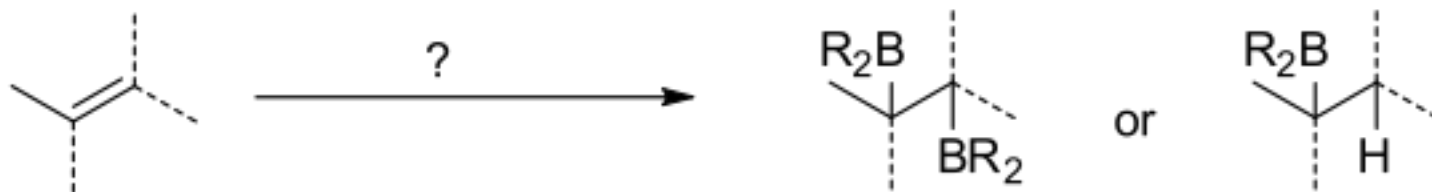


Enantioselective Borylations

David Kornfilt
Denmark Group Meeting
Sept. 14th 2010

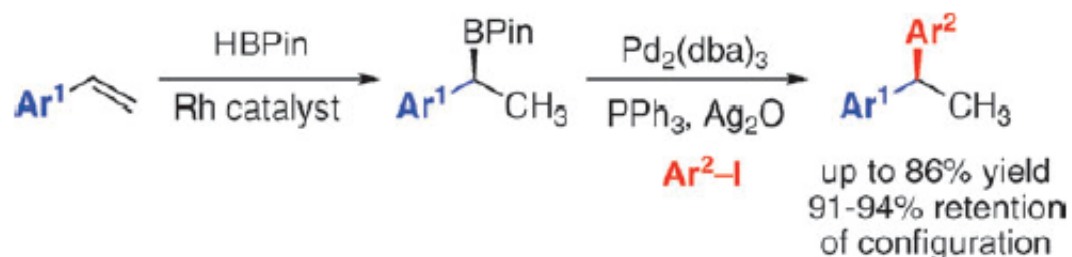
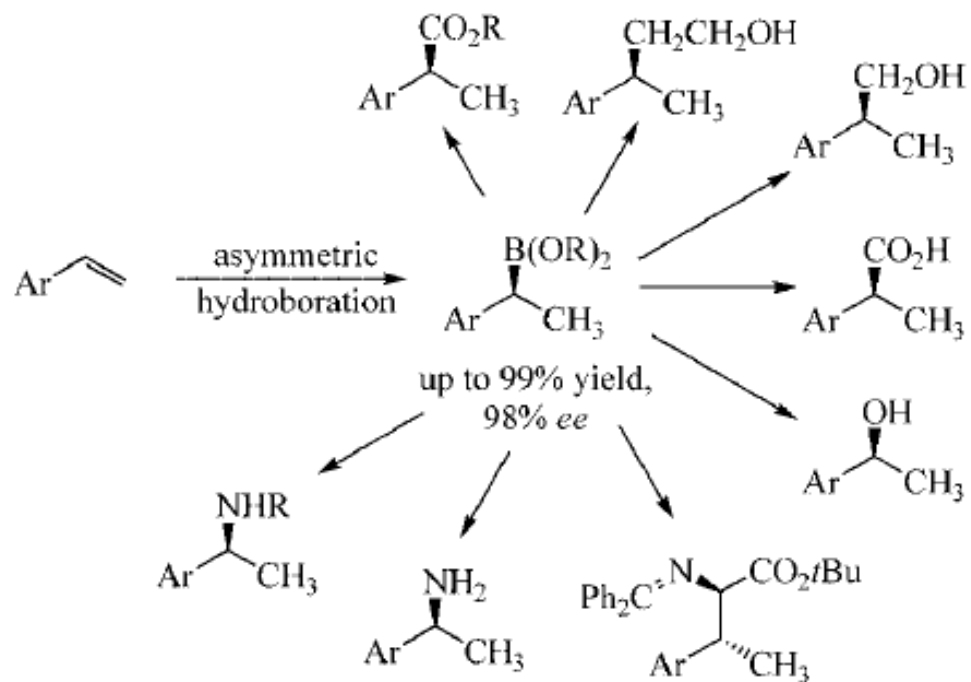
30.000-foot View



Good yield
Good dr
Good er
Good chemoselectivity

Enantioenriched Organoboranes

What to do with them...

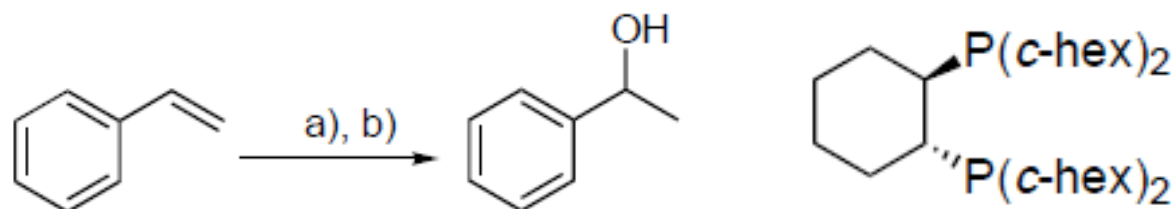


Crudden C. M. et. al., Eur. J. Org. Chem. 2003, 4647-4712

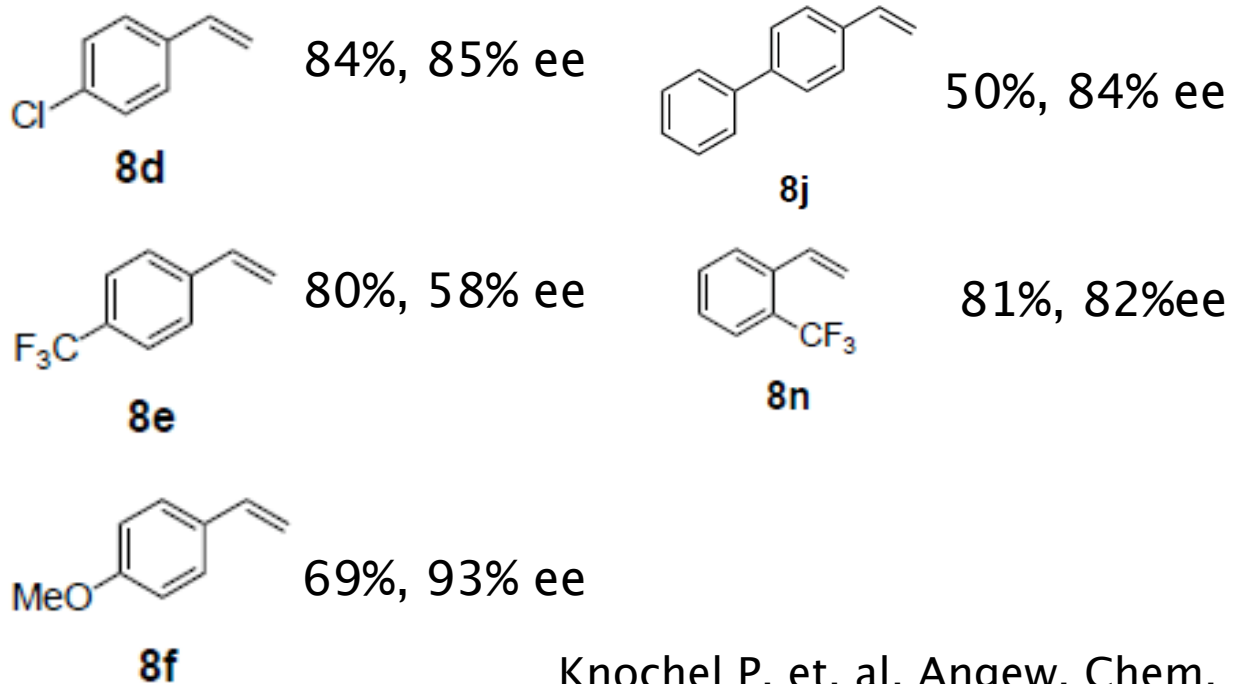
Enantioselective Borylations

- Addition of $\text{HB}(\text{OR}_2)$ to styrenes
- Addition of $\text{HB}(\text{OR}_2)$ to olefins with electron withdrawing groups
- Addition of $\text{HB}(\text{OR}_2)$ to other olefins
- Desymmetrizations and Cyclopropanations
- Addition of $\text{B}_2(\text{OR}_2)_2$ to simple alkenes
- Addition of $\text{B}_2(\text{OR}_2)_2$ to higher olefins

Alkene Monoborylation

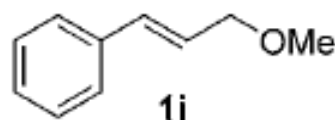
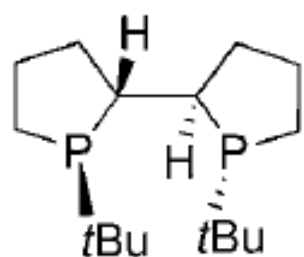
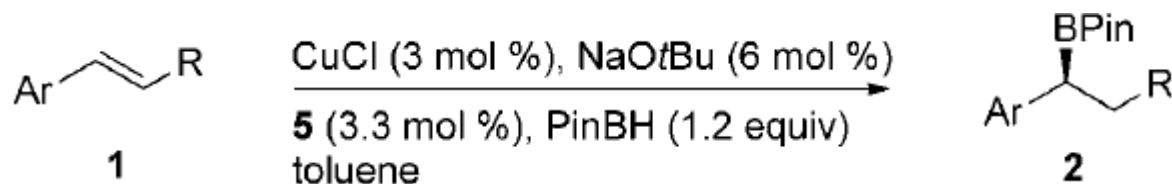


Scheme 4. Rhodium-catalyzed asymmetric hydroboration of styrene with ligands **7a–f**. a) $[\text{Rh}(\text{cod})_2]\text{BF}_4$ (1.0 mol %), **7a–f** (1.2 mol %), catecholborane (1.2 equiv), solvent; b) 3M KOH, H_2O_2 .

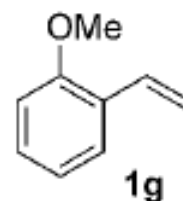


Knochel P. et. al. Angew. Chem. Int. Ed. 2001, 40,

Alkene Monoborylation



61%, 88% ee

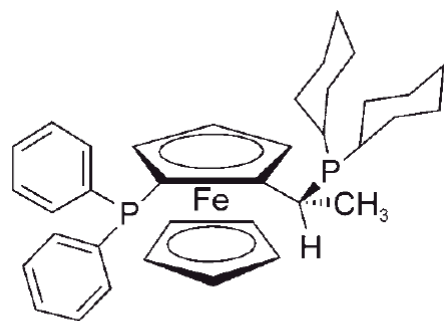


90%, 91% ee

- Very good ee (82–93%) even with *o*-substituents
- Naphthyl groups gave poor ee
- 61–95% yield

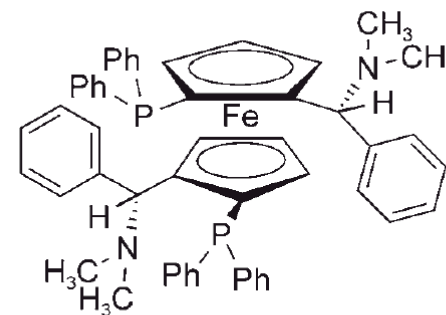
Enantioselective Borylations

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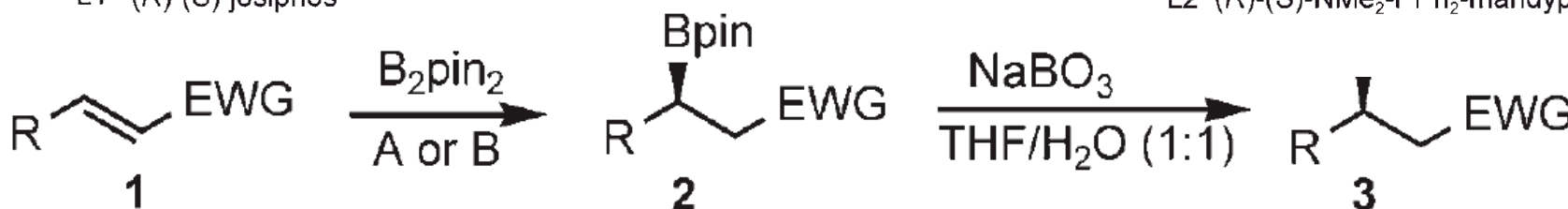


L1 (R)-(S)-josiphos

β -borylation



L2 (R)-(S)-NMe₂-PPh₂-mandyphos



Condition A: 2% CuCl, 3% NaOtBu, 4% L1, 2 equiv. MeOH, 1.1 equiv B₂pin₂ in THF

condition B: 3% CuCl, 3% NaOtBu, 3% L2, 2 equiv. MeOH 1.1 equiv B₂pin₂ in THF

Amides: 3% CuCl, 9% NaOtBu, 3% L2, 2 equiv. MeOH 1.1 equiv B₂pin₂ in THF

R = alkyl, aryl, branched alkyl

Enones: 3% CuCl, 3% NaOtBu, 3% L2, 1 equiv. MeOH 1 equiv B₂pin₂ in THF

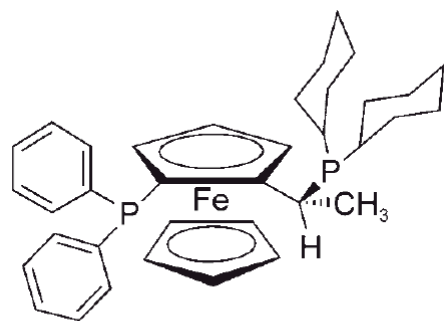
- High yield (89–97%), 82–92 % ee. Insensitive to character of R–group or ketone/ester fragment

- MeOH additive found to be critical for this reaction

Yun J. et. al., Adv. Synth. Catal. 2009, 351, 85

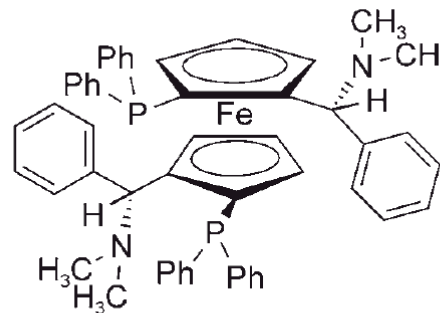
Yun J. et. al., Angew. Chem. Int. Ed. 2008, 47,

Yun J. et. al., Chem. Eur. J. 2009, 15, 1939 – 1

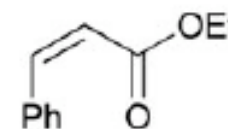
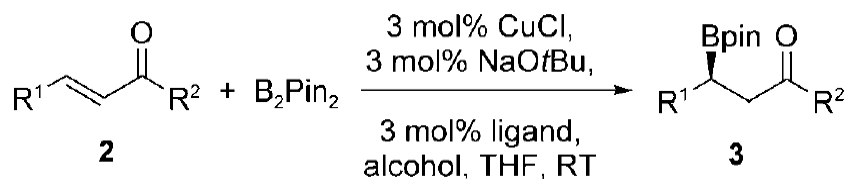


L1 (R)-(S)-josiphos

β -borylation



L2 (R)-(S)-NMe₂-PPh₂-mandyphos



(Z)-2

Investigated solvent effects :

- MeOH, iPrOH and tBuOH all effective
- Solvent affects product enantiomer ratio

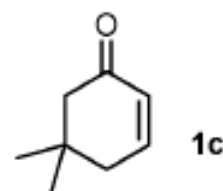
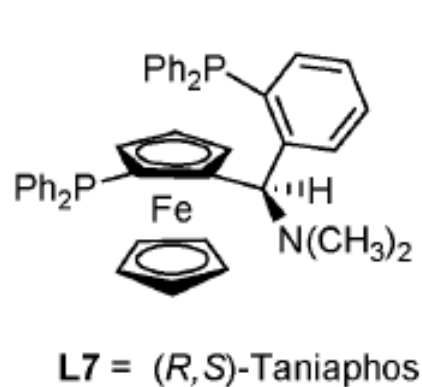
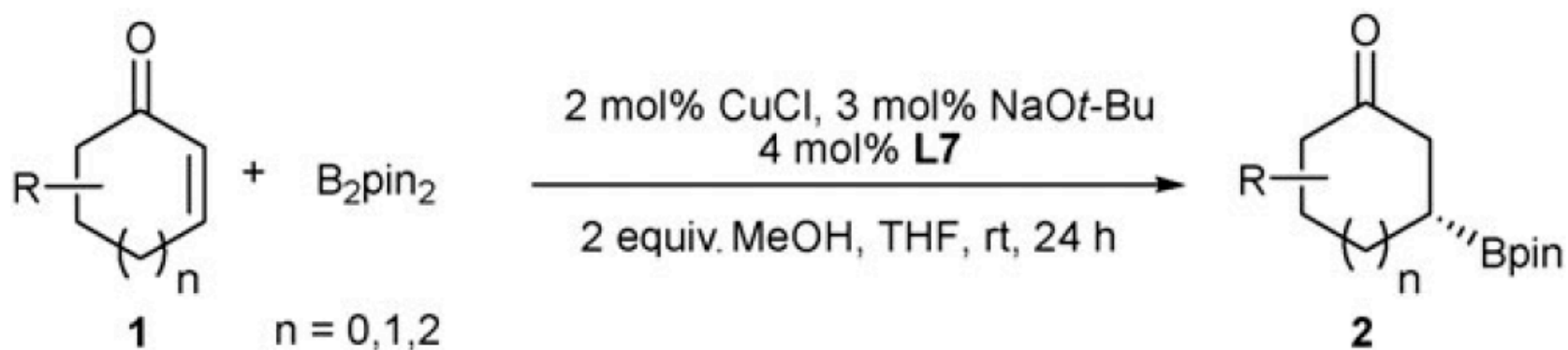
With Rh(phebox)

77%, 93% ee

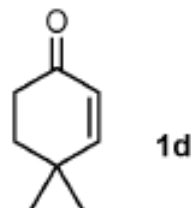
Results :

- Steric factors at R¹ are minimal for all classes.
- Mainly controlled by R² for ketones.
- Can add into extended dienones.

Yun's Addition into Cyclic Enones

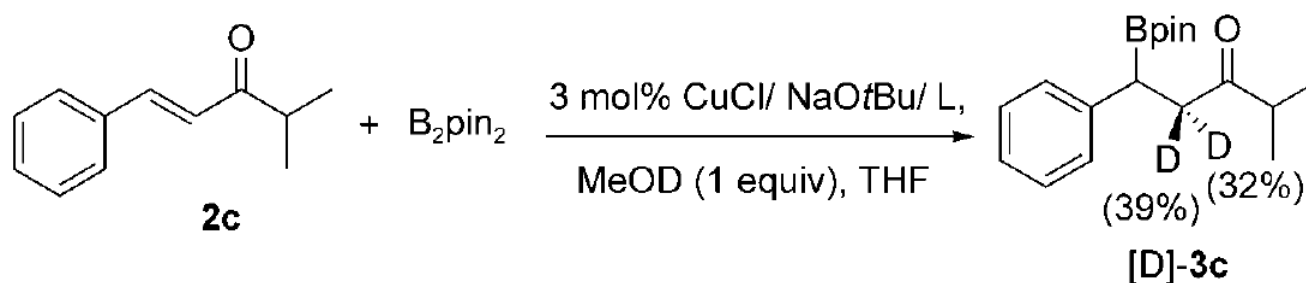
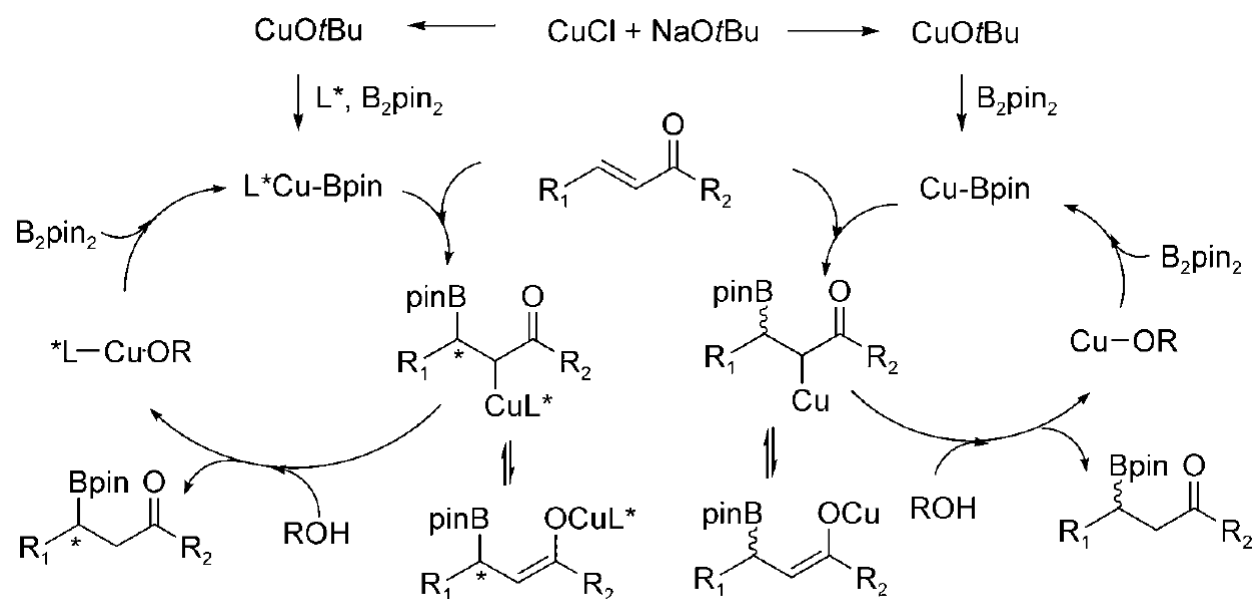


92% yield,
> 99% ee

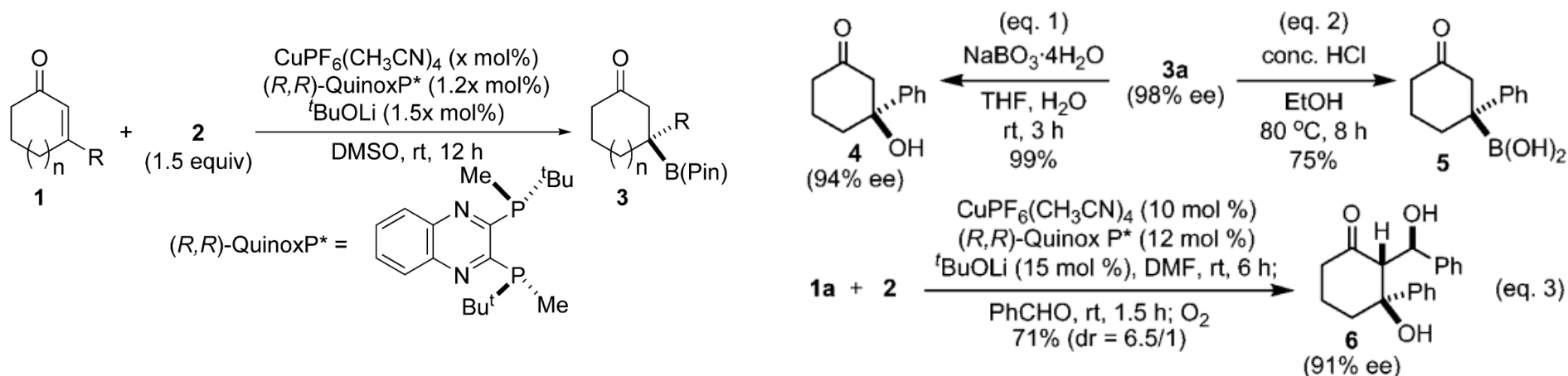


0% yield

β -borylation : Yun's Mechanism

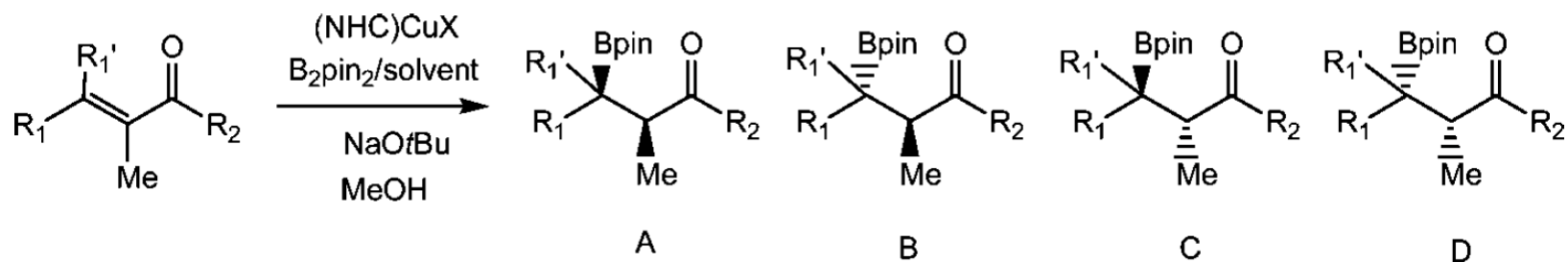
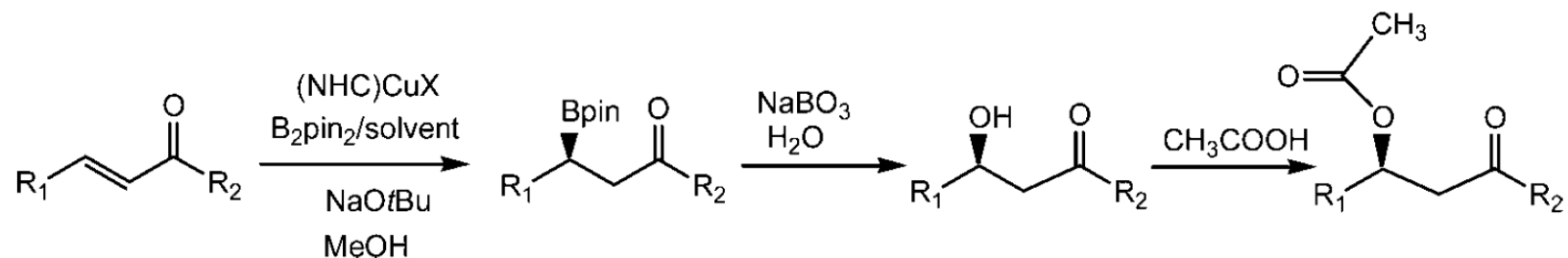


β -borylation



- PF_6 salt hypothesized to accelerate catalyst regeneration by formation of LiPF_6
- No protic additive necessary
- Wide scope in R for $n=0,1,2$: various aryl, unbranched and branched alkyl (80–99% yield, 70–98% ee)

Fernandez' NHC borylation

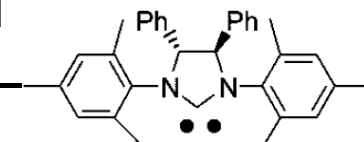


- ee better for syn than anti

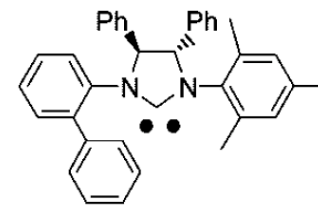
- Syn ee 35–74%, anti ee 5–42%

- Poor dr (70:30 syn best case)

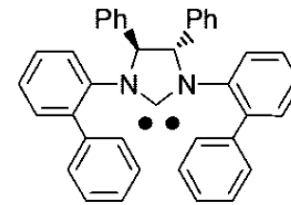
- R_2 is an alkoxy group for all substrates



L_3

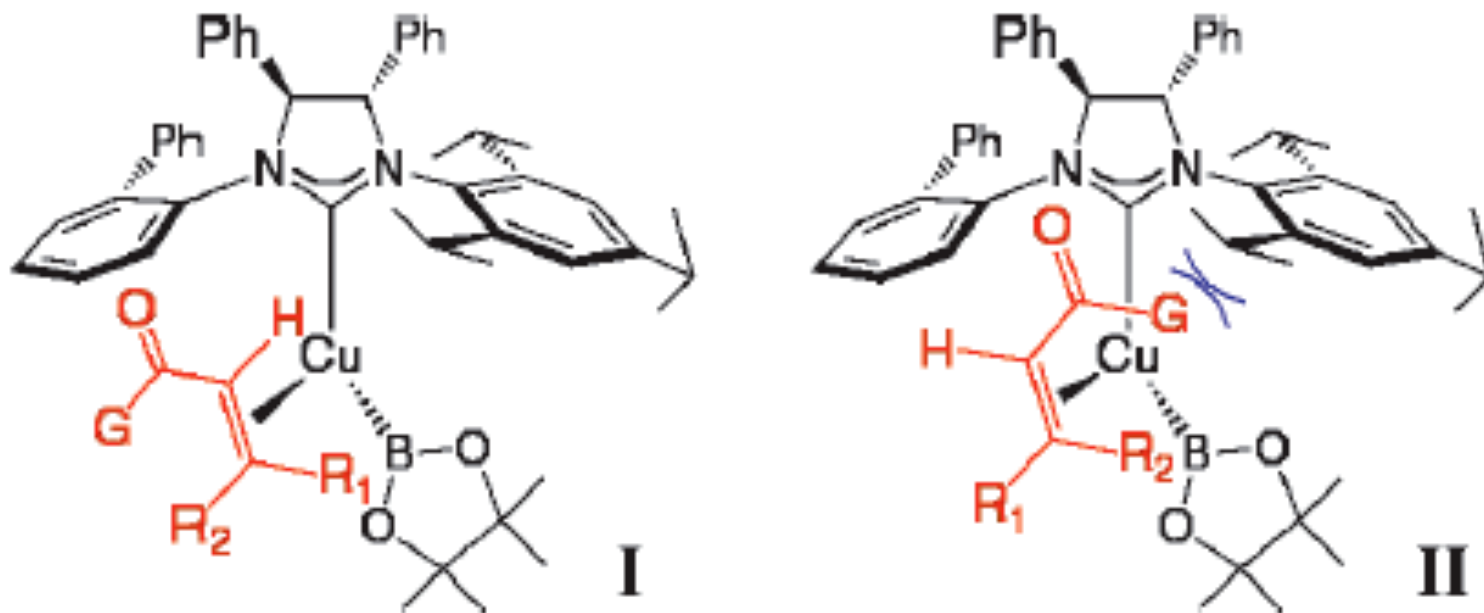


L_4



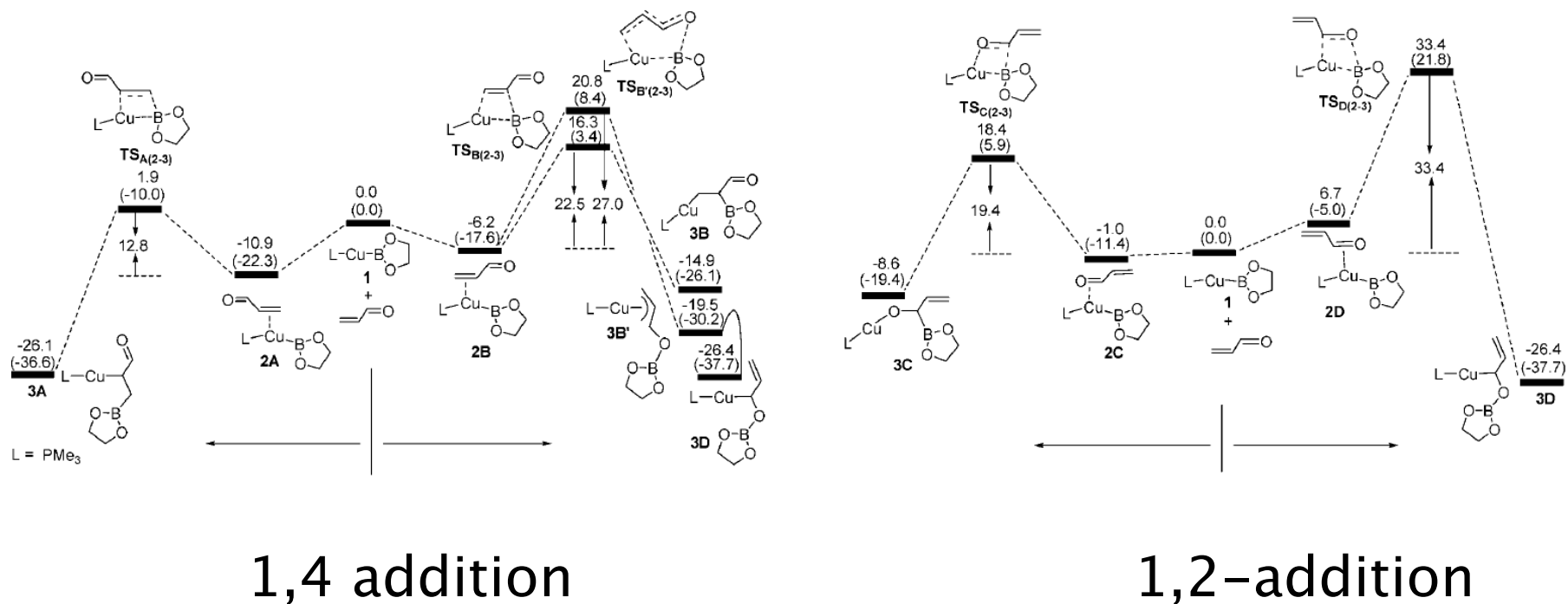
L_5

Stereochemical Model

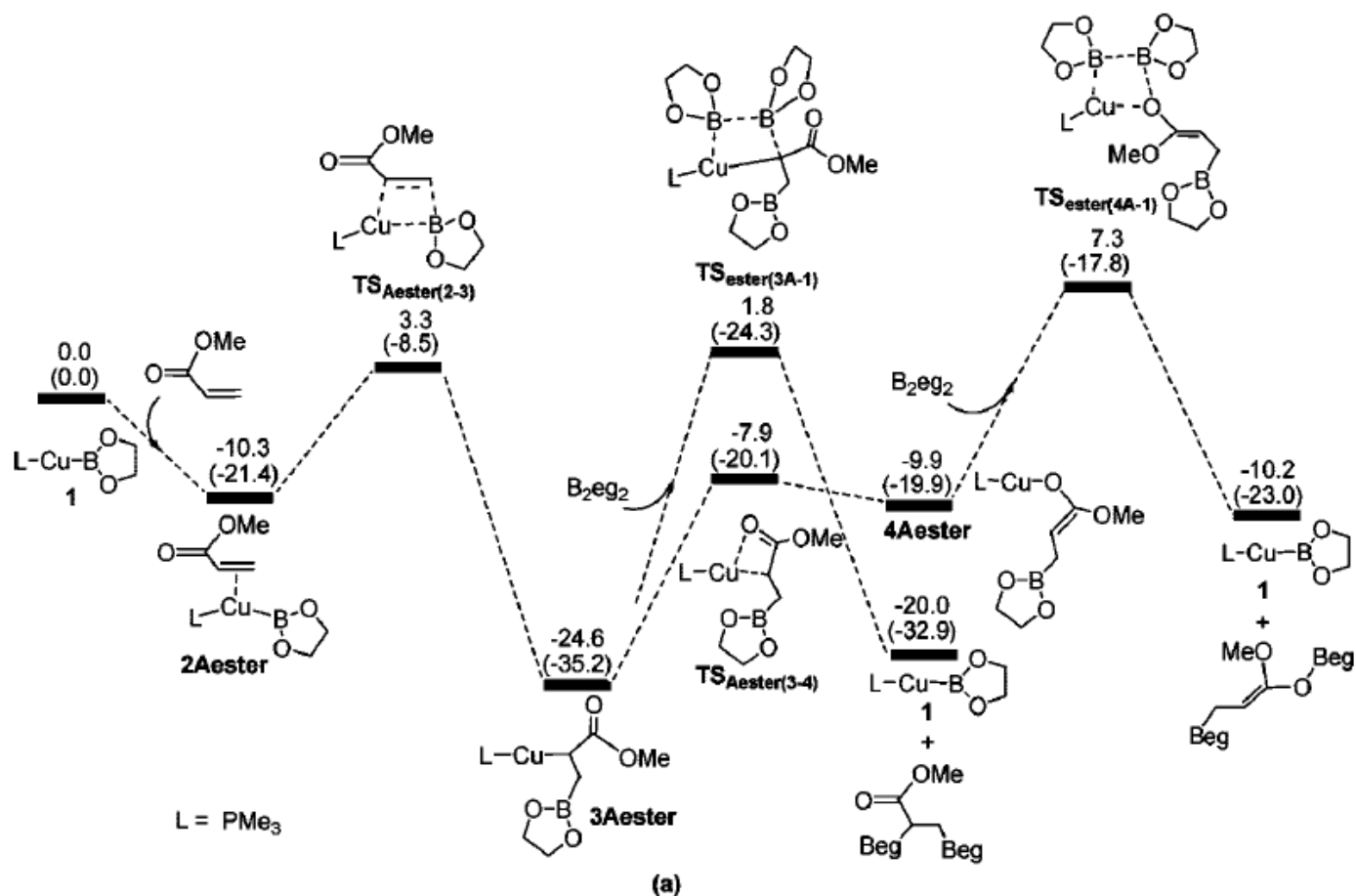


No α -substituted α,β -unsaturated esters tried to check proposal

DFT Calculations



DFT Calculations with Esters

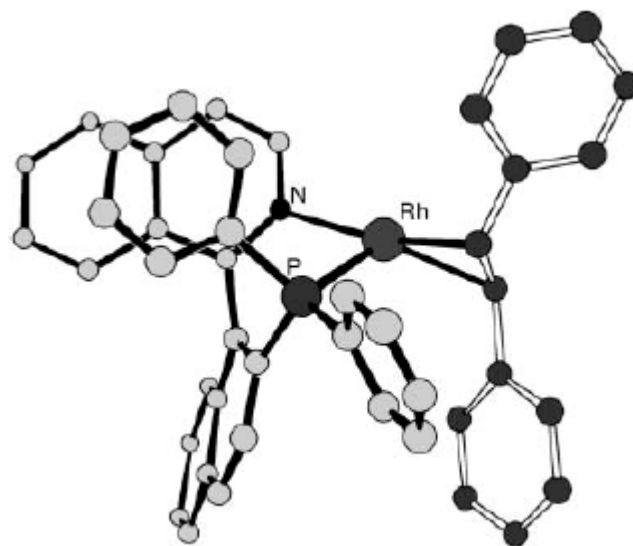
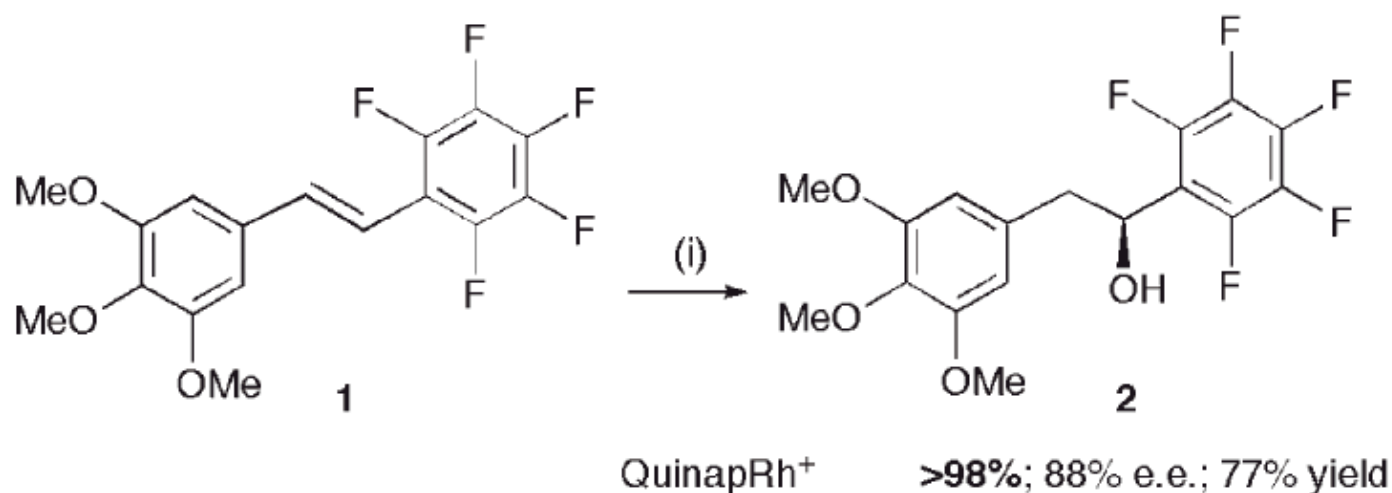


1,4-addition

Enantioselective Borylations

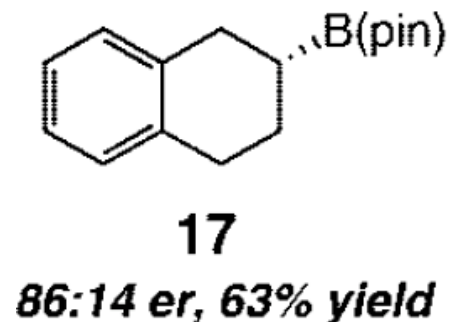
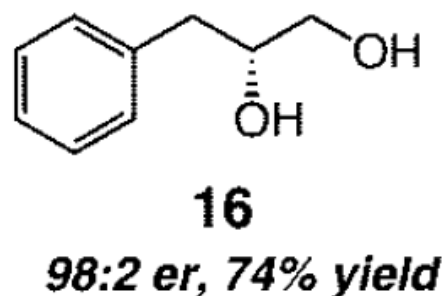
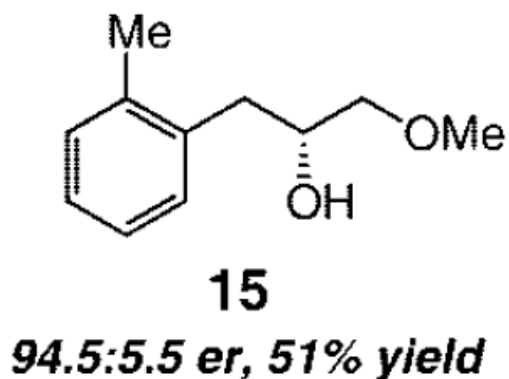
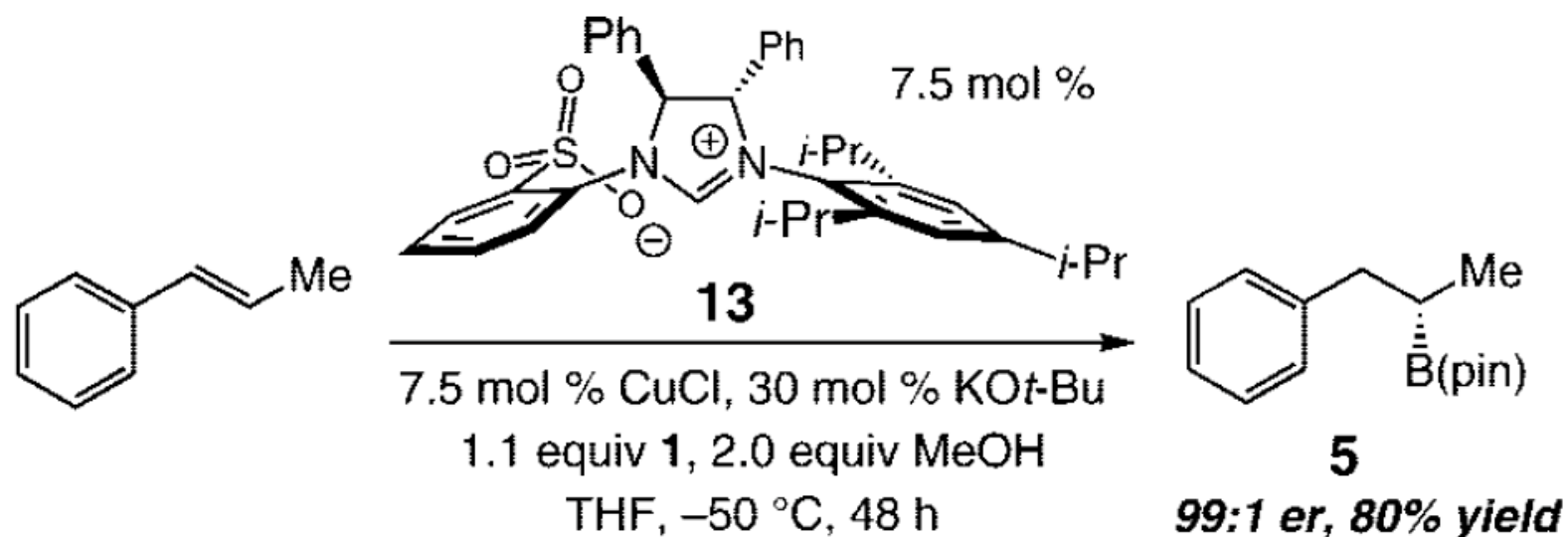
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trans-Alkene Monoborylation

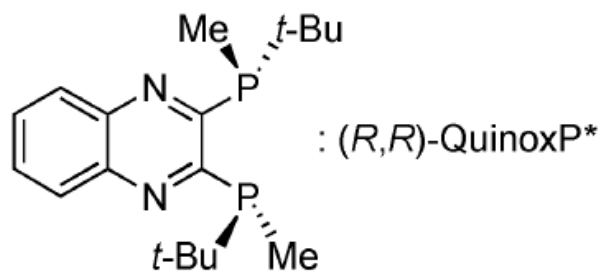
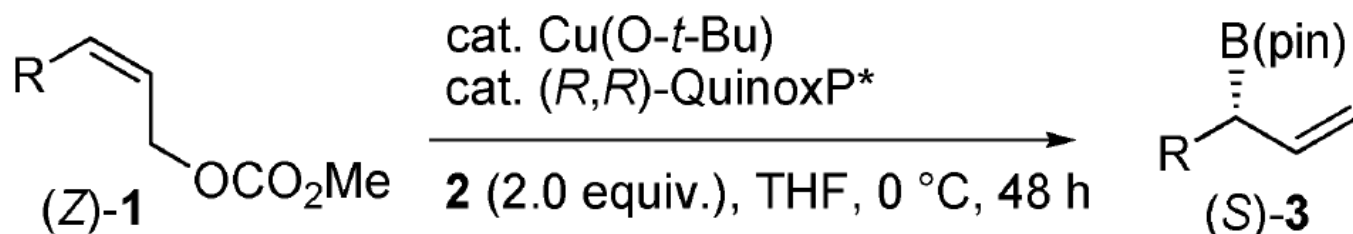


Pichon C. et. al. Chem. Commun., 2005, 528

trans-Alkene Monoborylation

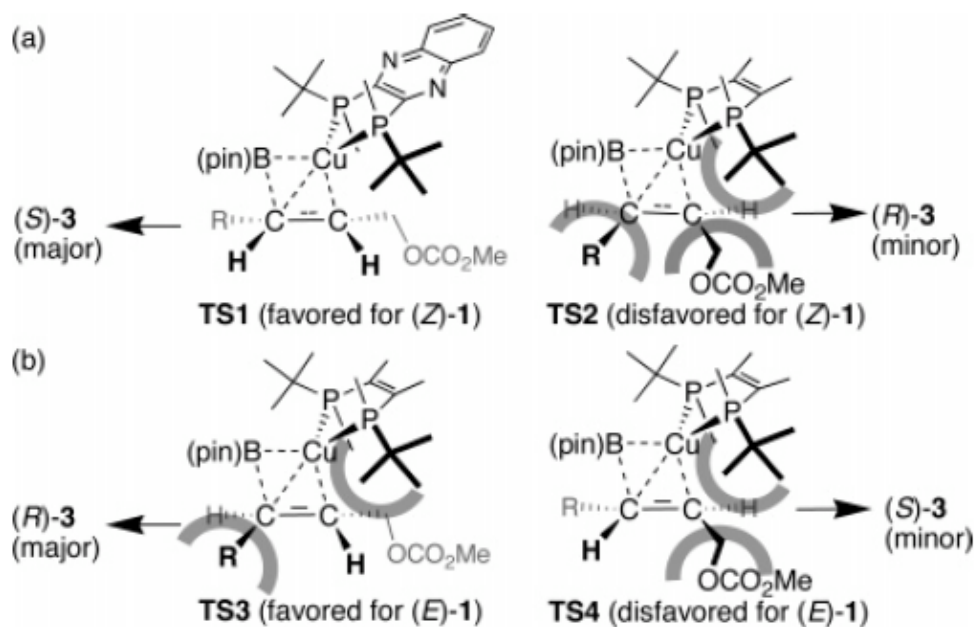
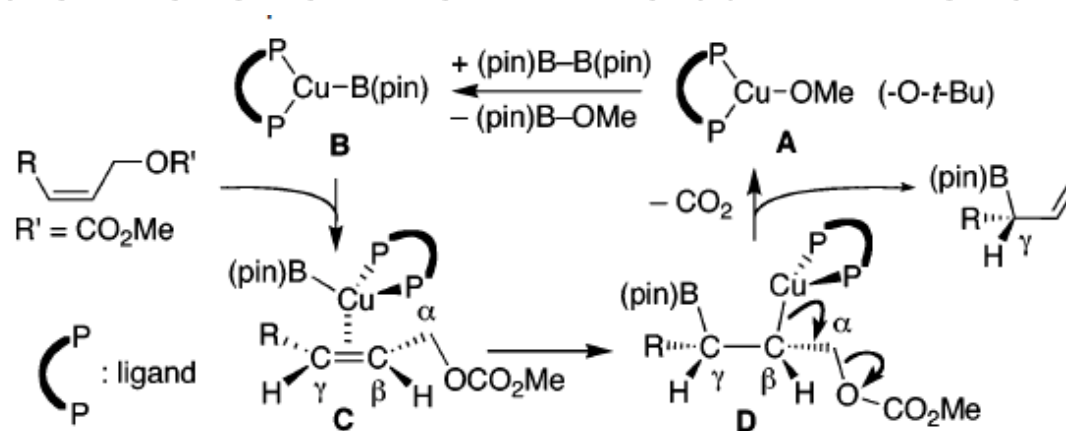


Borylation of (Z)-allylcarbonates

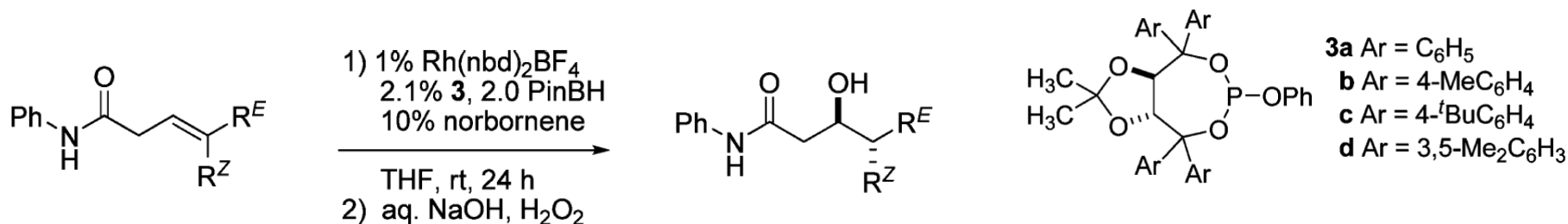


- 64–70%, >90% ee for R=alkyl
- No aryl demonstrated
- No branching at the α -position of R tolerated

Stereochemical Model



Hindered Alkene borylation



- Scope only demonstrated for alkyl substrates.
- Norbornene additive increases ee
- Amide required for 2-point binding.
- Stereospecific

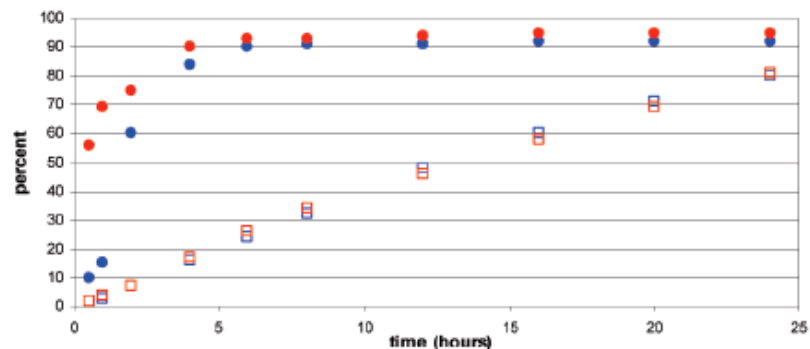
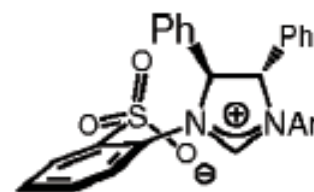
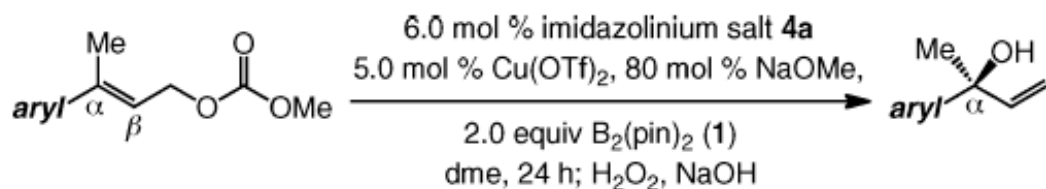
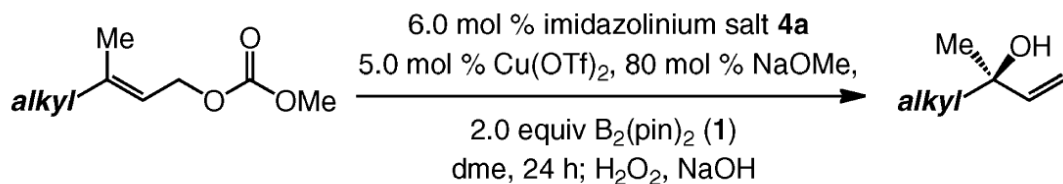


Figure 2. Comparing the yield of *anti*-6 (□) and its enantiomeric excess (●) over time with (red) and without (blue) added norbornene.

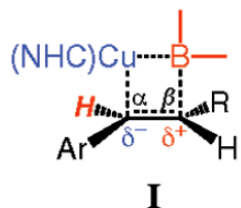
Addition into Trisubstituted olefins



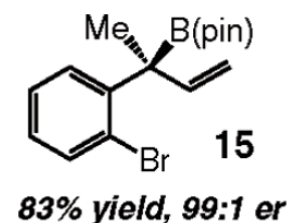
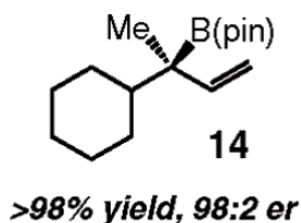
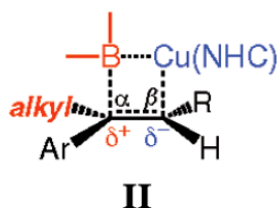
4a Ar = 2,4,6-(Me)₃C₆H₂ (Mes)

4b Ar = 2,6-Et₂C₆H₃

disubstituted aryl alkene



trisubstituted aryl alkene



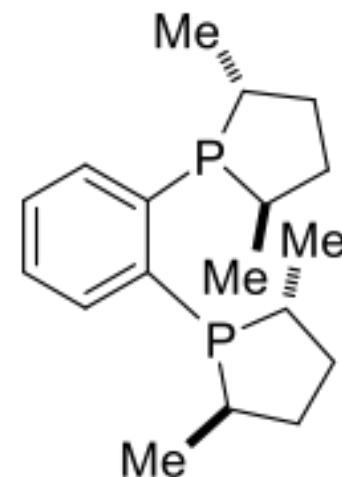
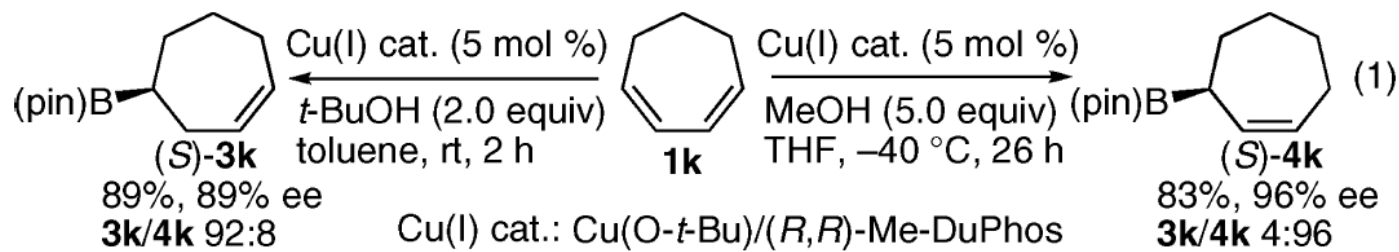
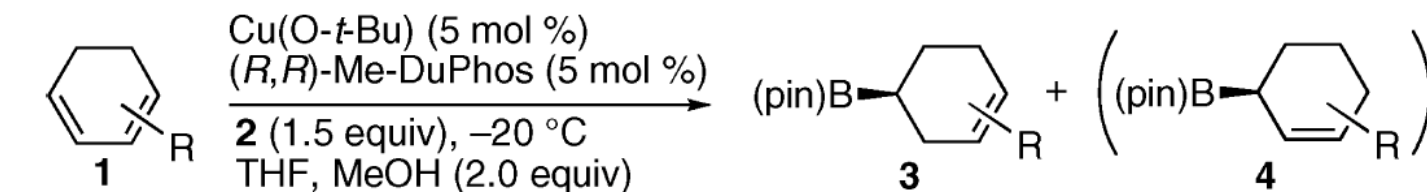
- Stereospecific S_N2' addition, no S_N2 products detected.

- Also works with disubstituted alkenes (er

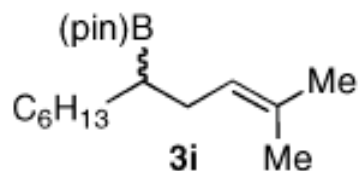
> 90:10)

Hoveyda A. H. et. al., J. AM. CHEM. SOC. 2010, 132, 106:

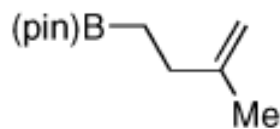
Monoborylation of Dienes



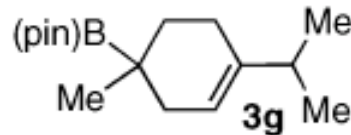
(R,R)-Me-DuPhos



78% ,42% ee



97%

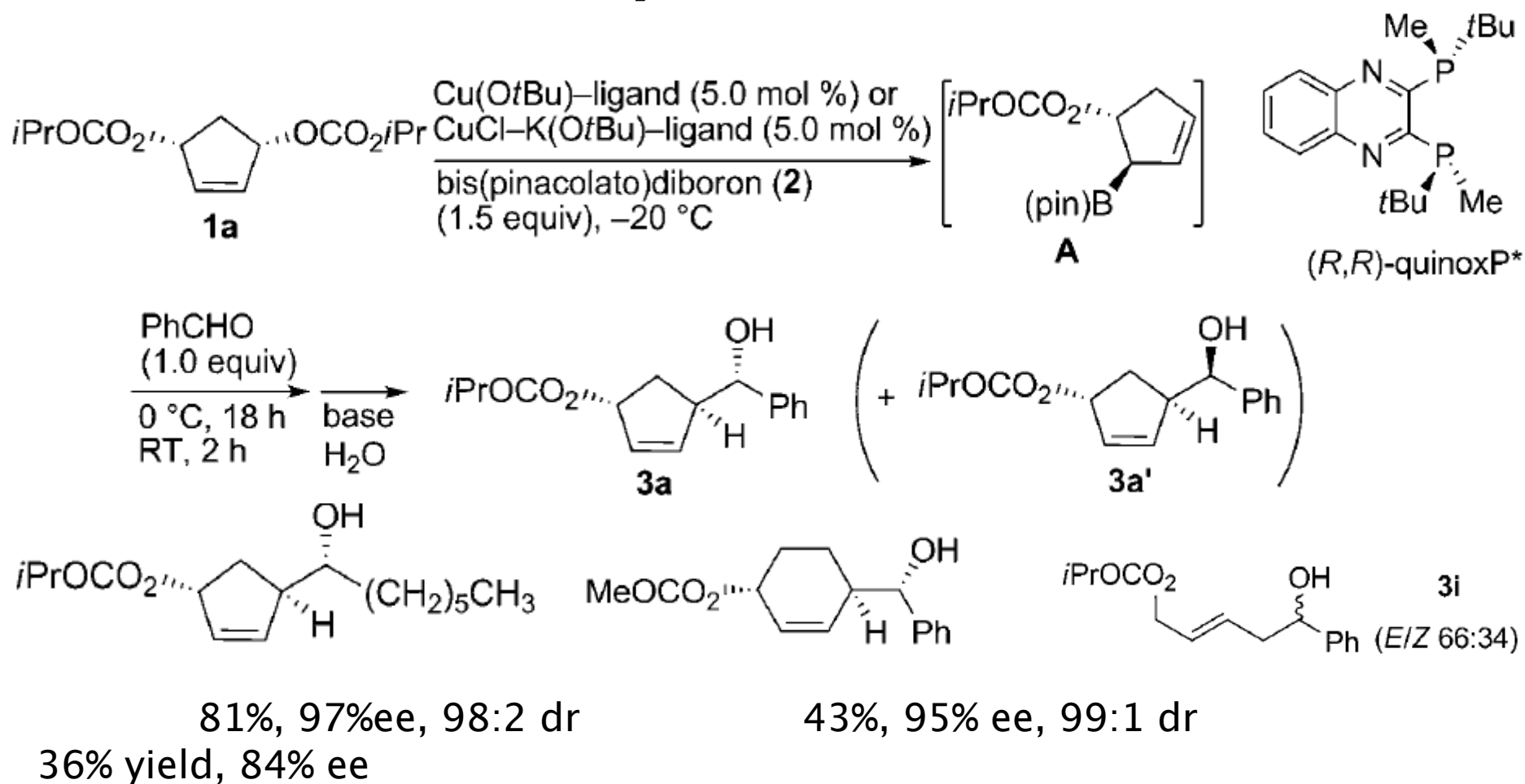


NR

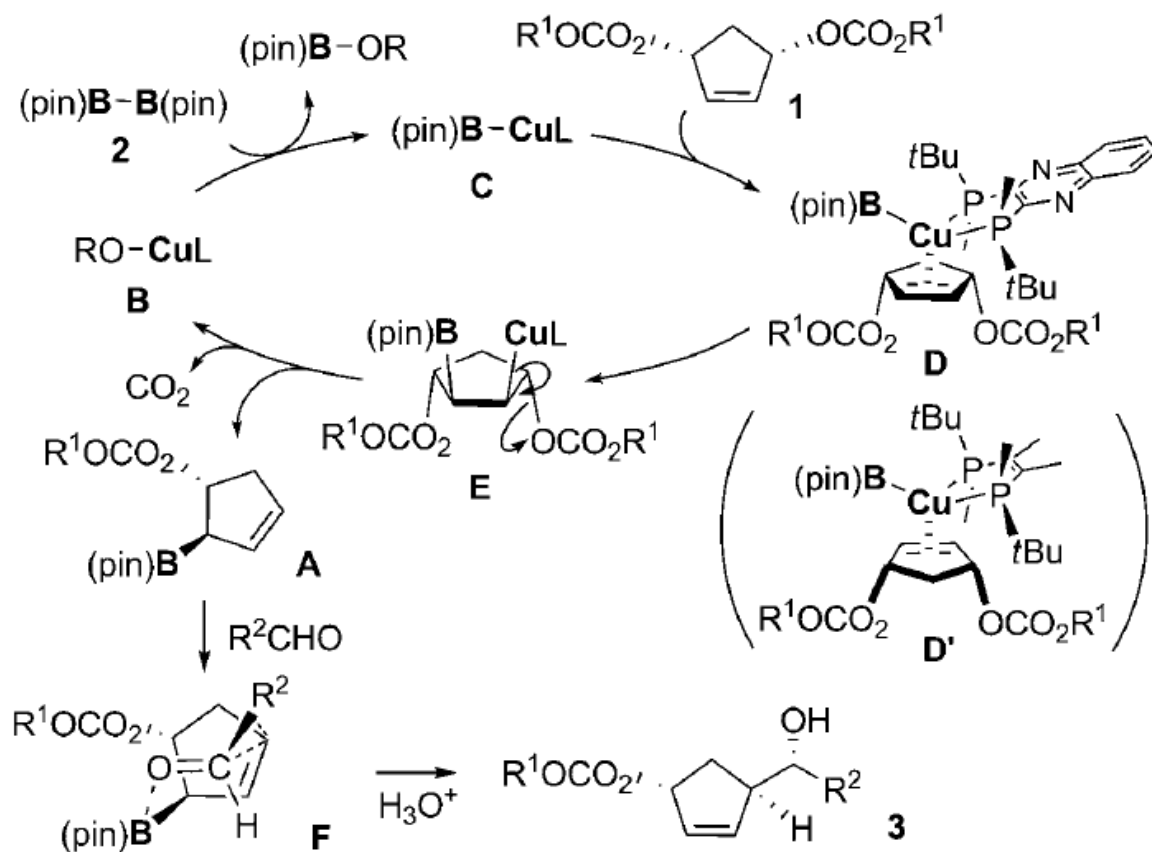
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- Addition of $\text{HB}(\text{OR}_2)$ to olefins with electron withdrawing groups
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- Addition of $\text{B}_2(\text{OR}_2)_2$ to simple alkenes
- Addition of $\text{B}_2(\text{OR}_2)_2$ to higher olefins

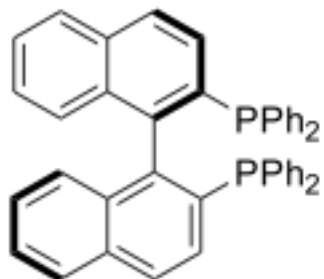
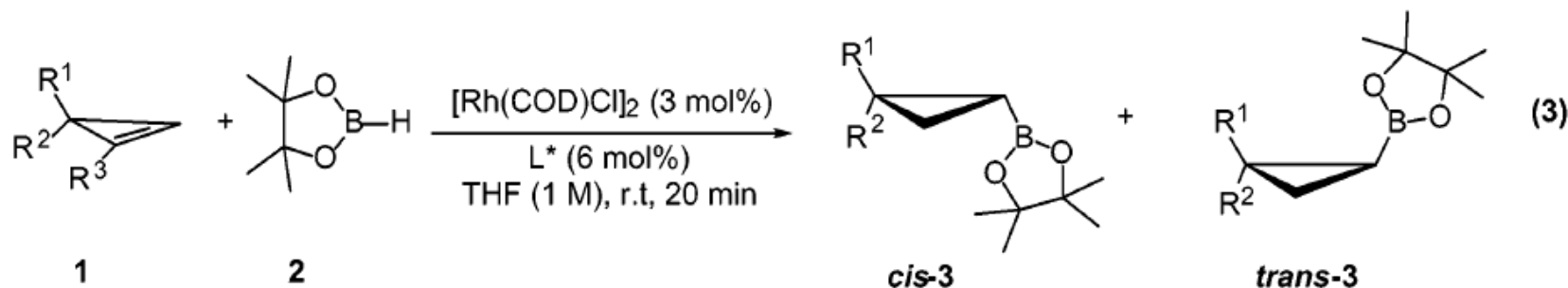
Borodesymmetrization



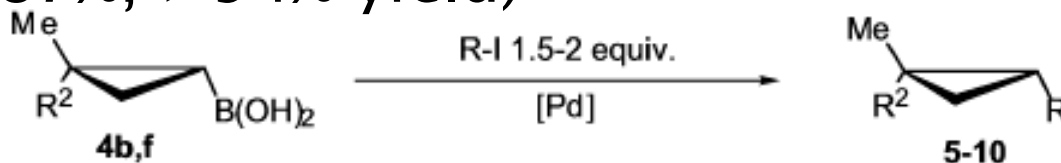
Stereochemical Course



Gevorgyan's Cyclopropanes

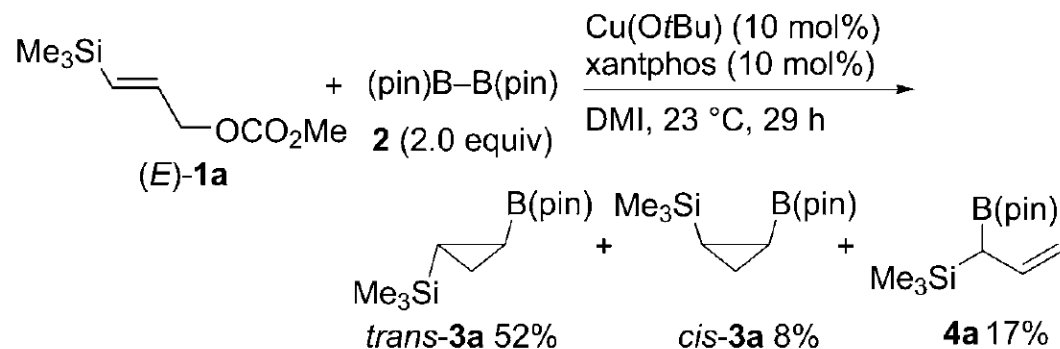
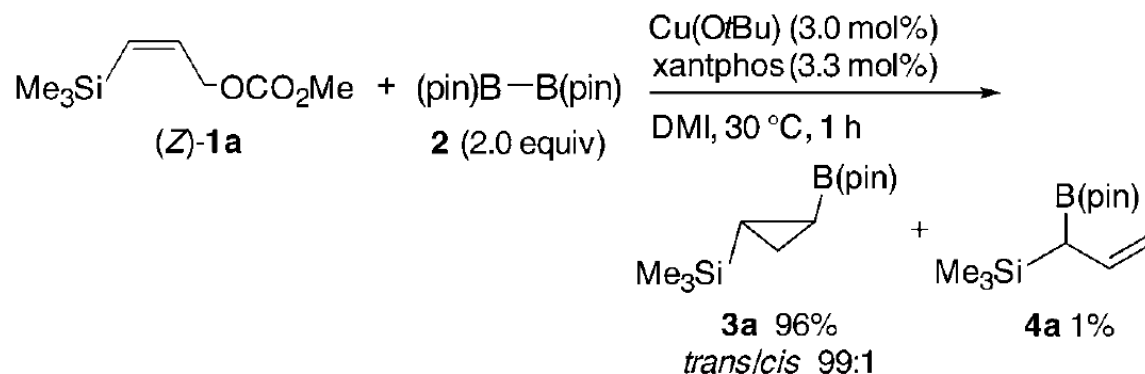


- R^3 is H, otherwise rearrangement to furans observed
- Esters, Alkyl, Aryl groups tolerated. Excellent ee ($> 87\%$, $> 94\%$ yield)



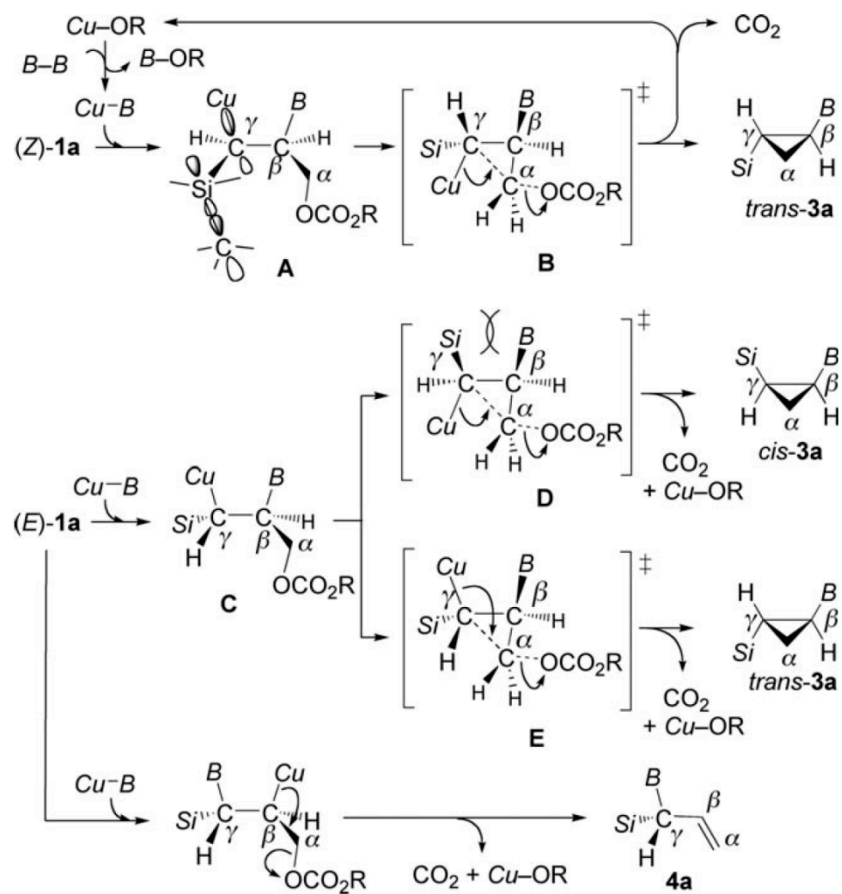
- Further Suzuki coupling possible.

Group Problem

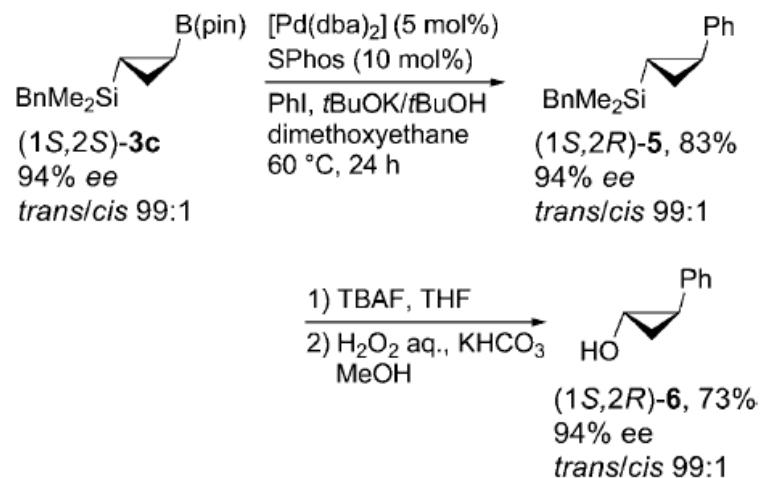
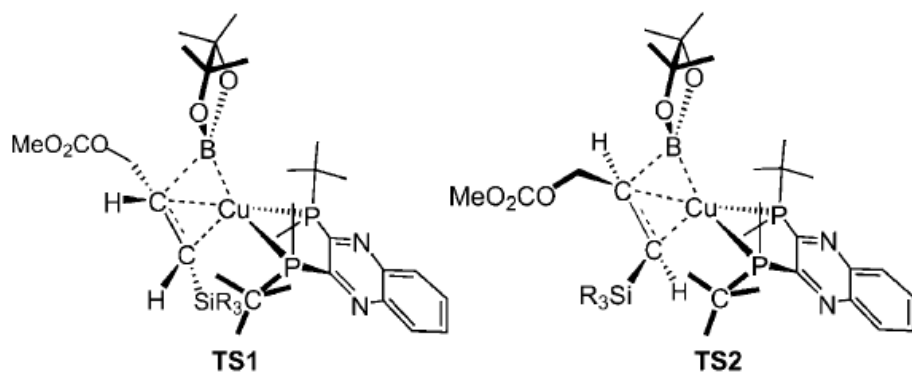
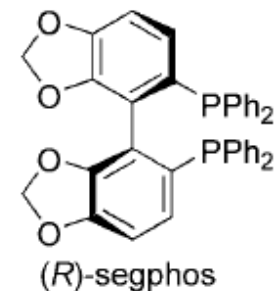
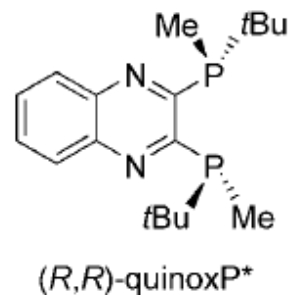
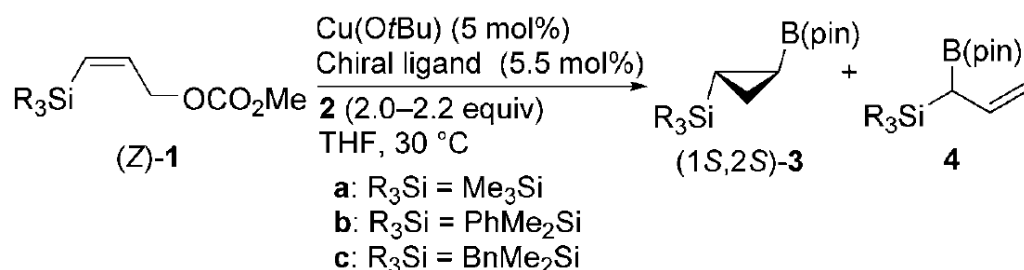


Propose a mechanism, and explain the selectivities

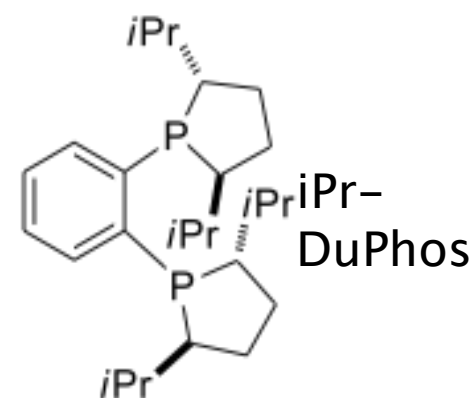
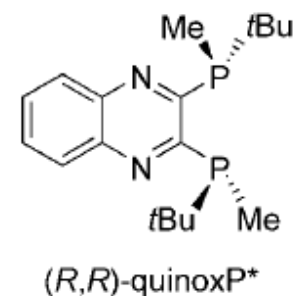
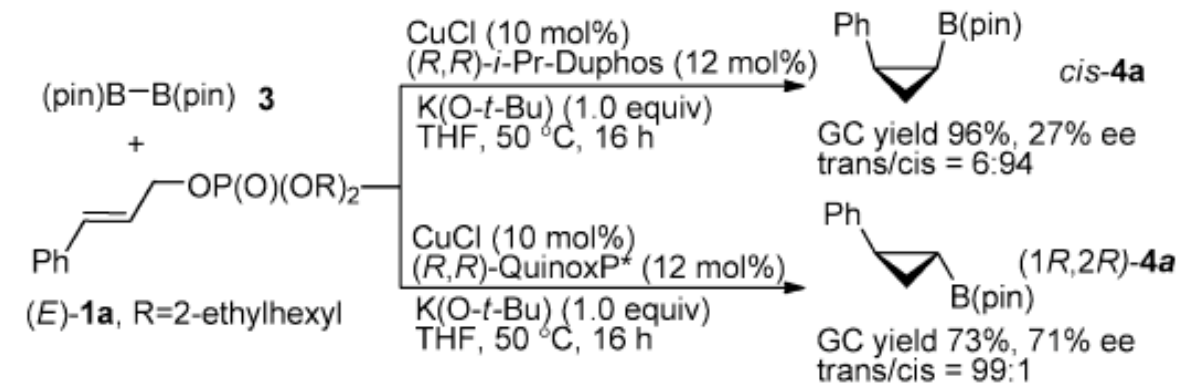
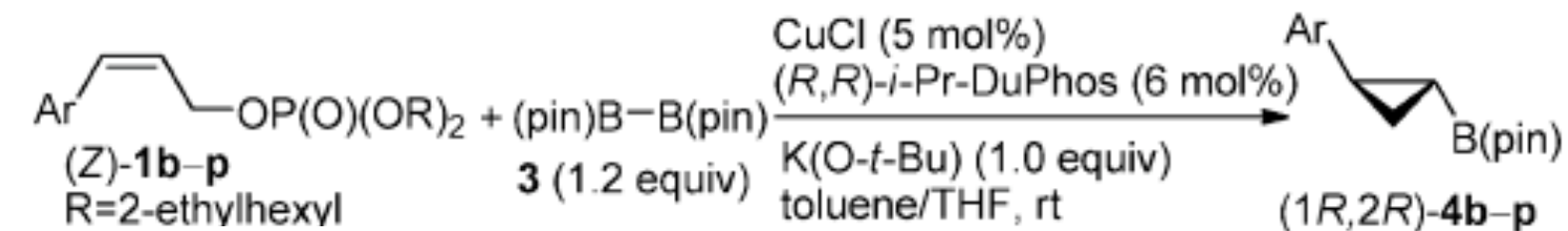
Group Answer



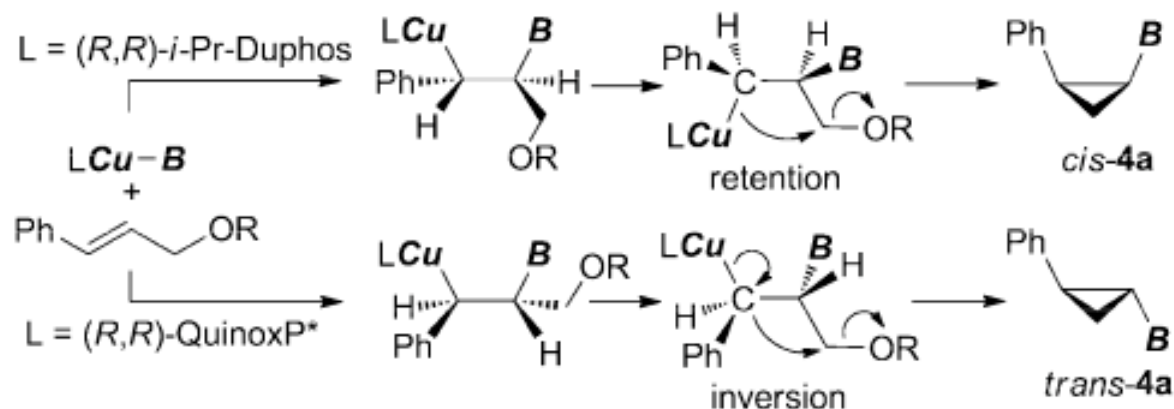
Synthesis of bifunctionalized cyclopropanes from allylcarbonate borylation



Synthesis of Cyclopropanes



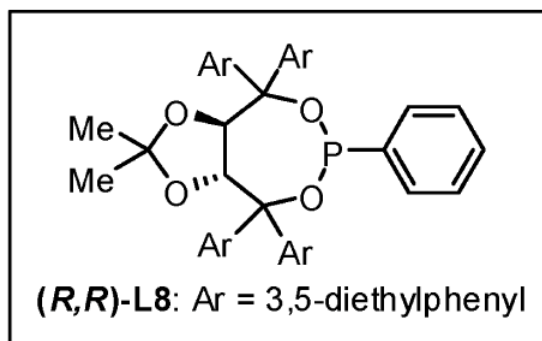
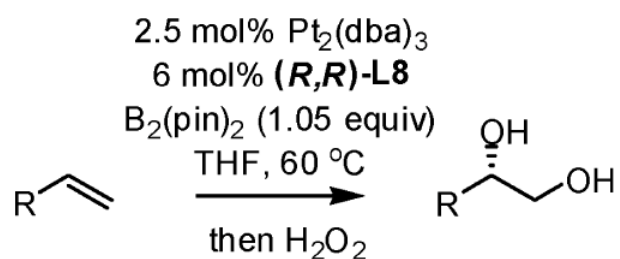
Ito et. al., J. AM. CHEM. SOC. 2010, 132,



Enantioselective Borylations

- Addition of $\text{HB}(\text{OR}_2)$ to styrenes
- Addition of $\text{HB}(\text{OR}_2)$ to olefins with electron withdrawing groups
- Addition of $\text{HB}(\text{OR}_2)$ to other olefins
- Desymmetrizations and Cyclopropanations
- Addition of $\text{B}_2(\text{OR}_2)_2$ to simple alkenes
- Addition of $\text{B}_2(\text{OR}_2)_2$ to higher olefins

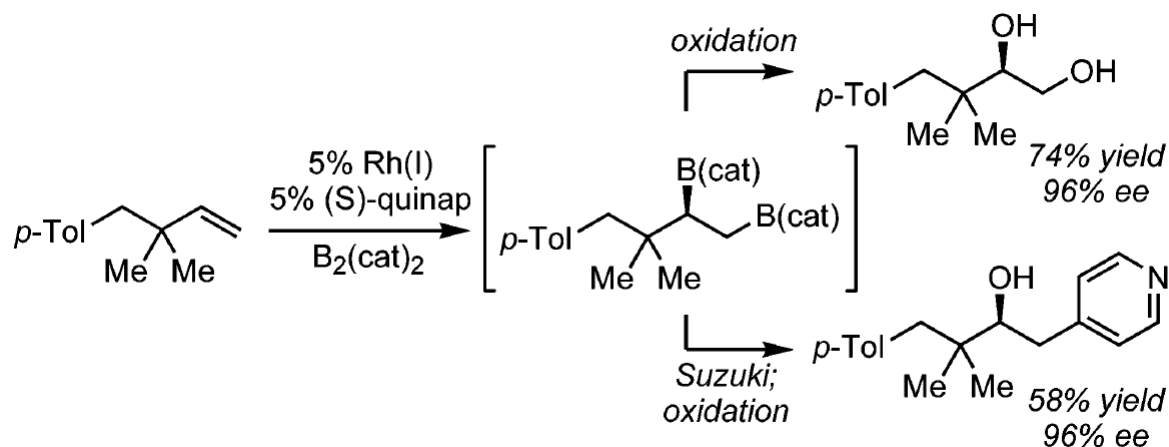
Styrene diborylation



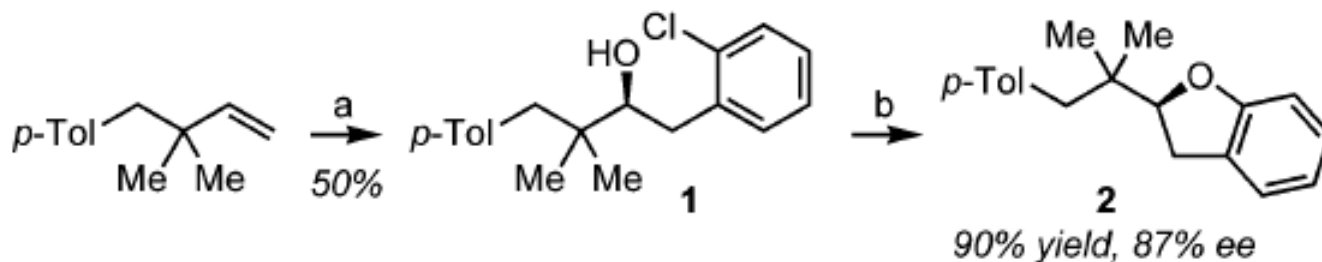
substrate	yield (%) ^b	ee (%) ^c
	83	92
	77	94
	80	94
	87	94
	46	90
	52	87
	86	93
	84	86
	93	90 ^d
	92	90

Morken J. P. et.
 al. J. AM.
 CHEM. SOC.
 2009, 131,

Diborylation of Terminal Olefins

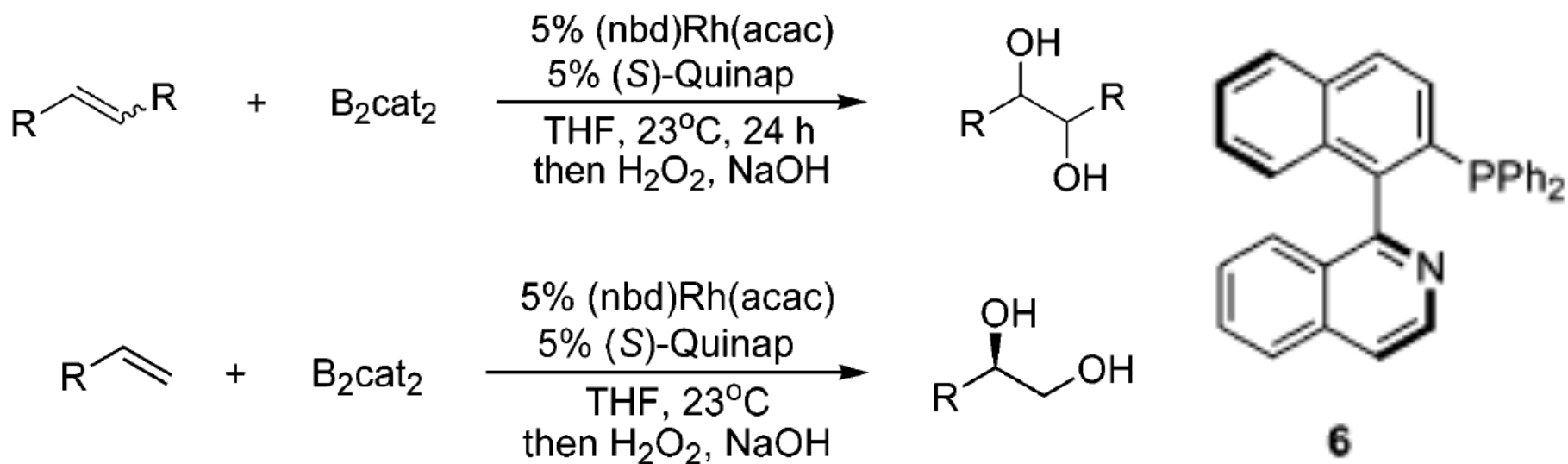


- Best ee's (>93%) when allylic position is a quaternary cent
Highly sensitive to substitution
- 41–82% yields, 48–77 % over two steps for terminal olefin
no allylic hydrogens
- Can couple with numerous bromides and triflates



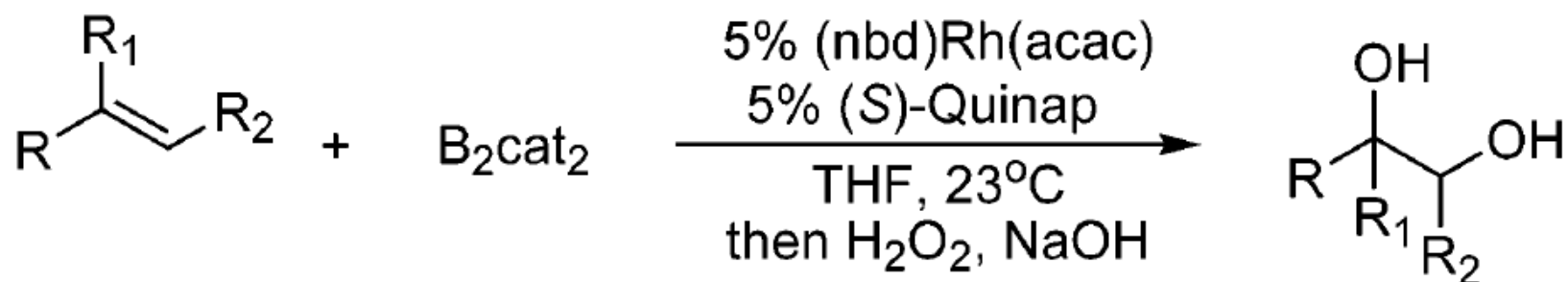
Morken J. P.
et.al., ORGANIC
LETTERS 2004,

Extensions by Morken



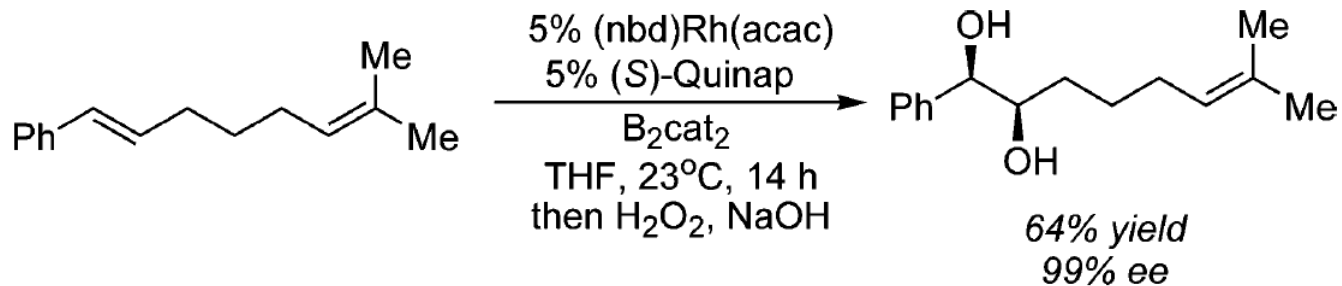
- Best results with trans alkenes (>97% ee)
- Poor selectivity observed without steric hindrance at allylic position for terminal alkenes.

Alkene Diborylation

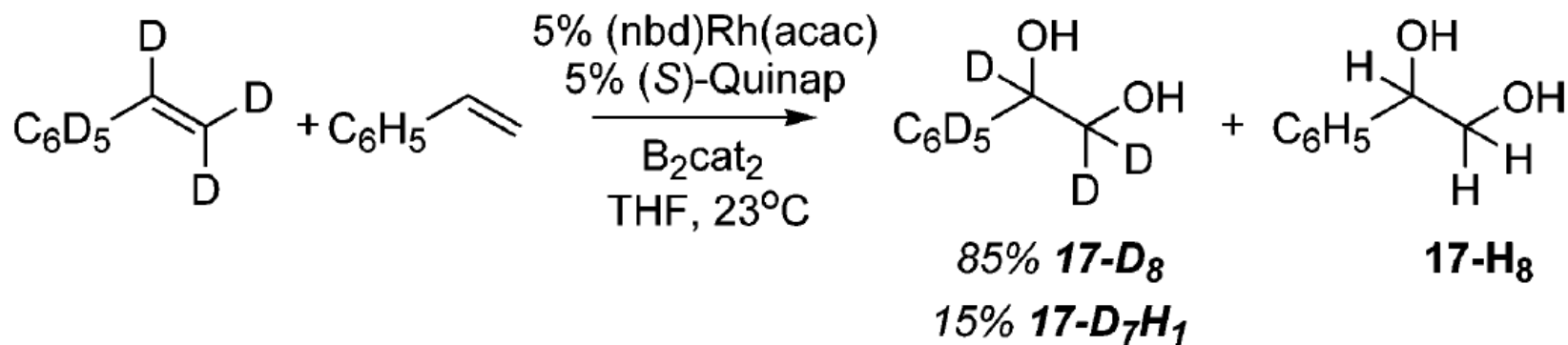
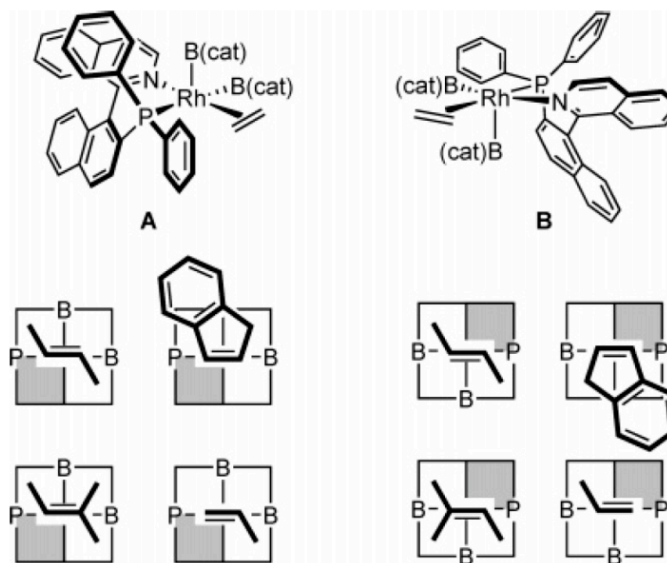
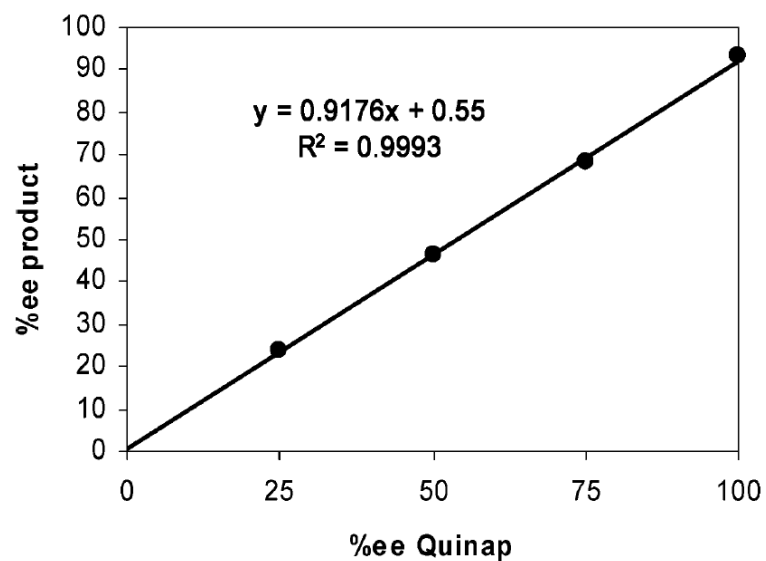


1,1-disubstituted gives poor ee (25–41%), moderate yield (67–79%)

Good ee (>90%) for trisubstituted, poor yield (8–17%)



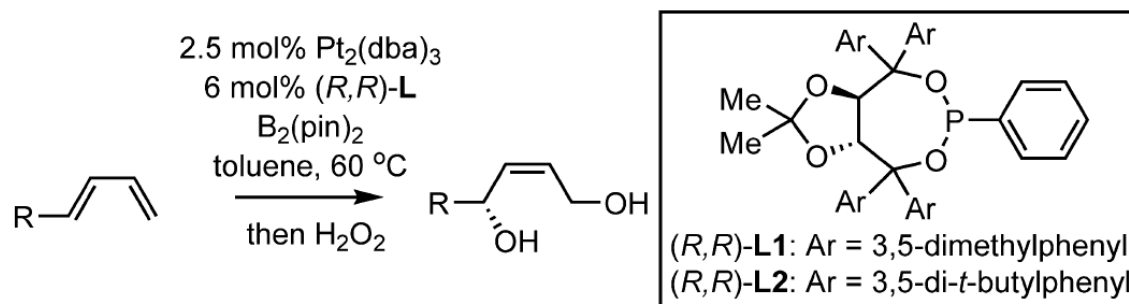
Alkene Diborylation



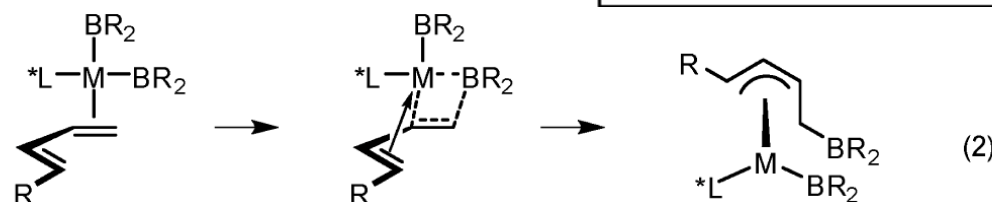
Enantioselective Borylations

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- Addition of $\text{B}_2(\text{OR}_2)_2$ to higher olefins

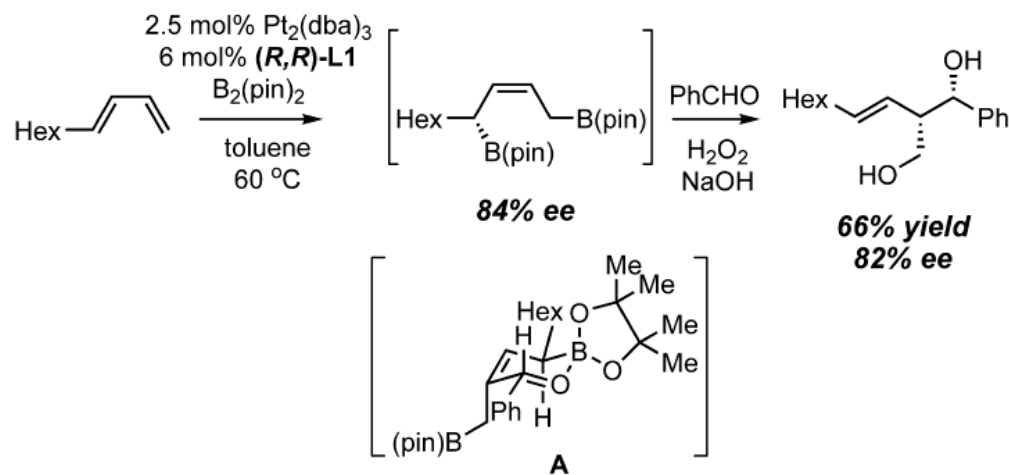
Diborylation of Dienes



- 70–92% yield,
> 84% er



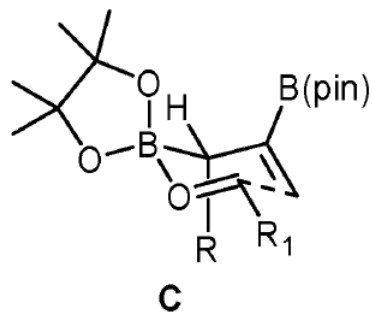
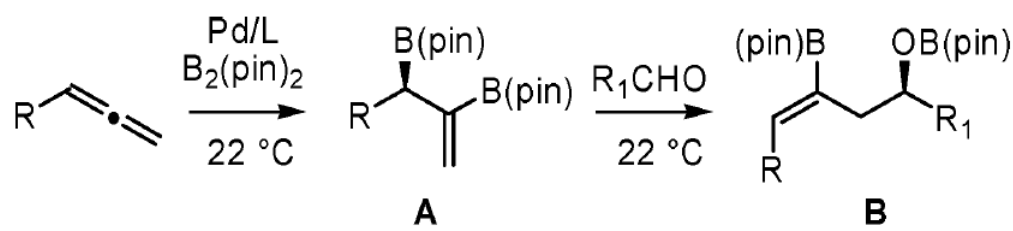
- Cyclic, acyclic dienes



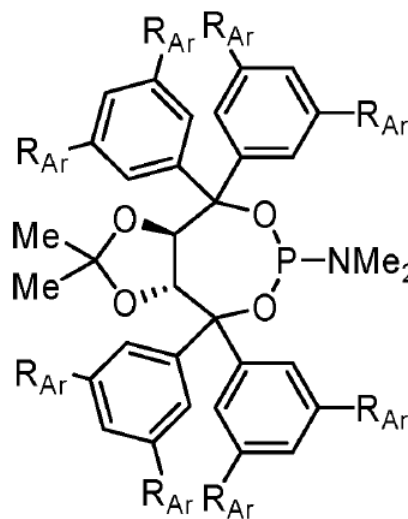
- Low yield when α -carbon is a quaternary center

- NR with Z-diene

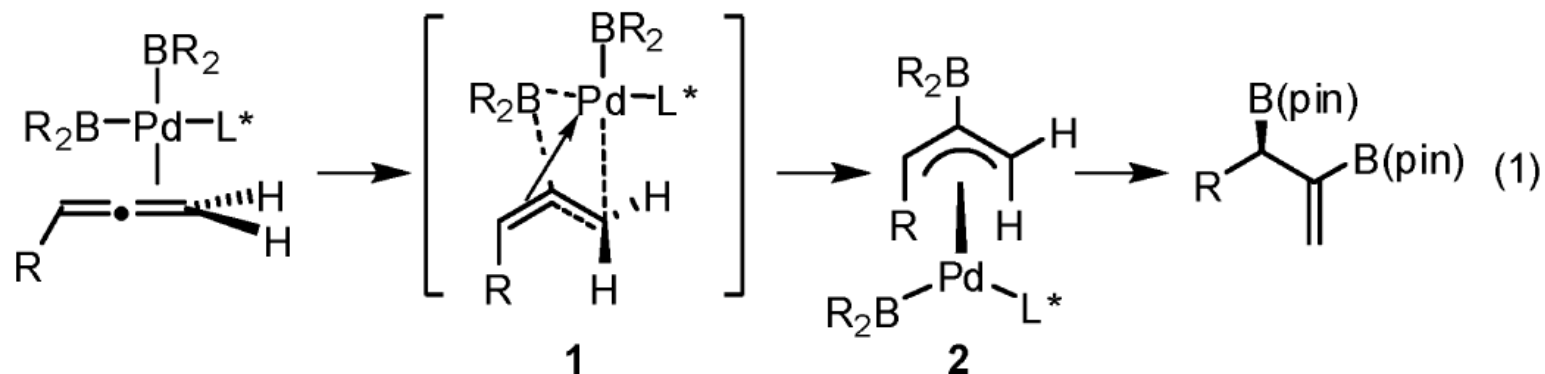
Allene borylation



1. 6 mol % (*R,R*)-**2**
 2.5 mol % Pd₂(dba)₃
 1.2 equiv B₂(pin)₂
 22 °C, 10 h
 2. R²CHO (0.6 equiv)
 22 °C, 14 h
 3. NaOH, H₂O₂



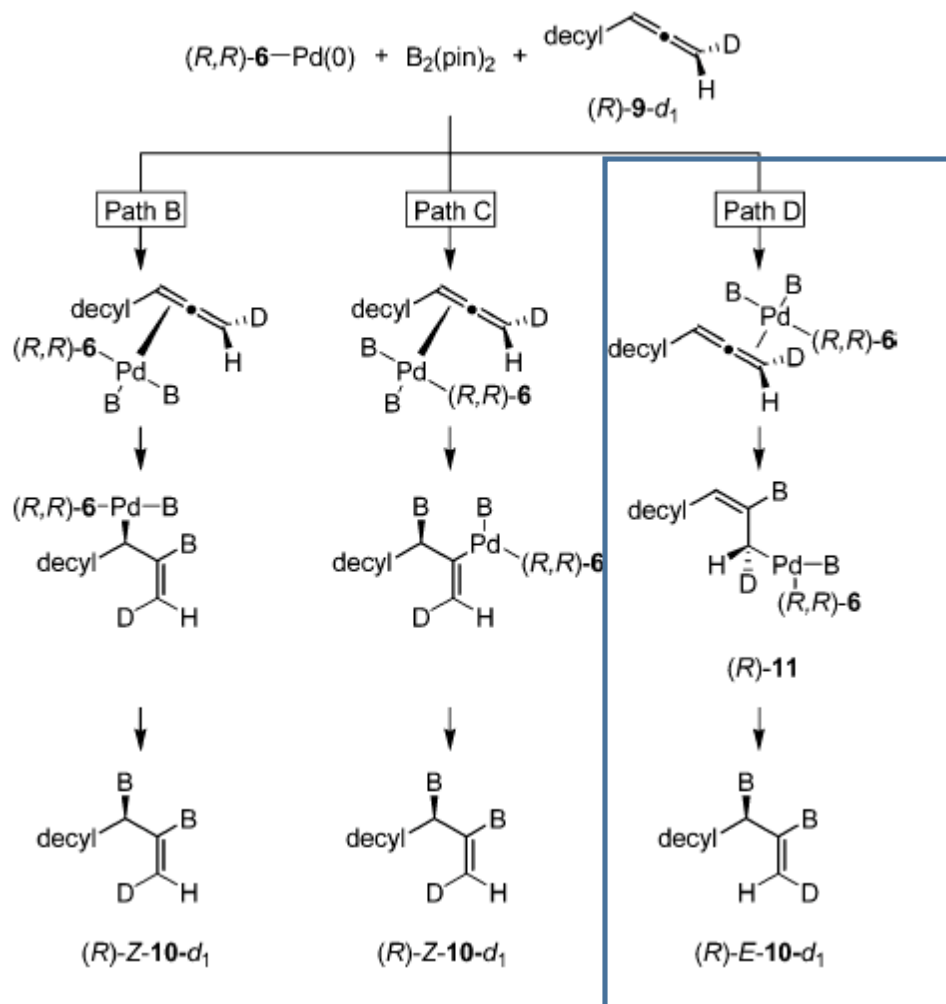
1: R_{Ar} = H
 2: R_{Ar} = Me
 3: R_{Ar} = tBu



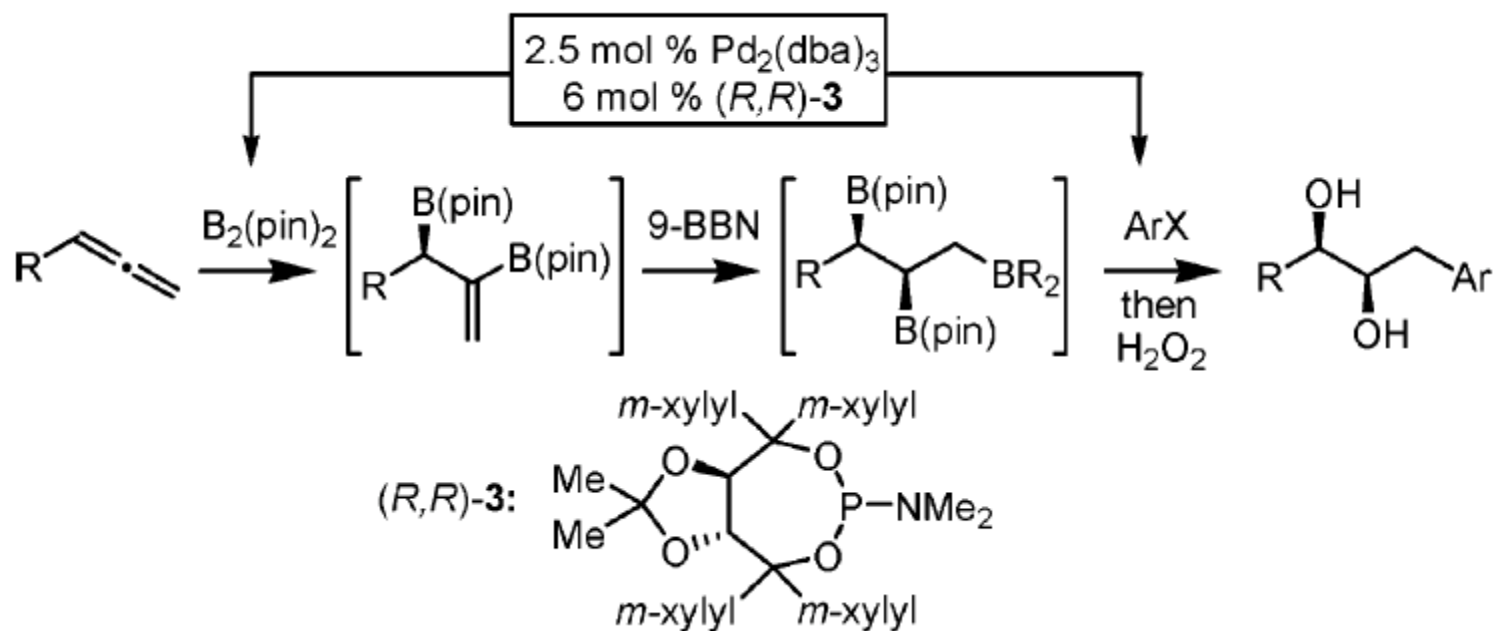
Morken J. et. al., J. AM. CHEM. SOC. 2007, 129, 87
 Morken J. P. et. al., ORG. LETT. 2005, 7, 5505–550

Allene Borylation

Scheme 6

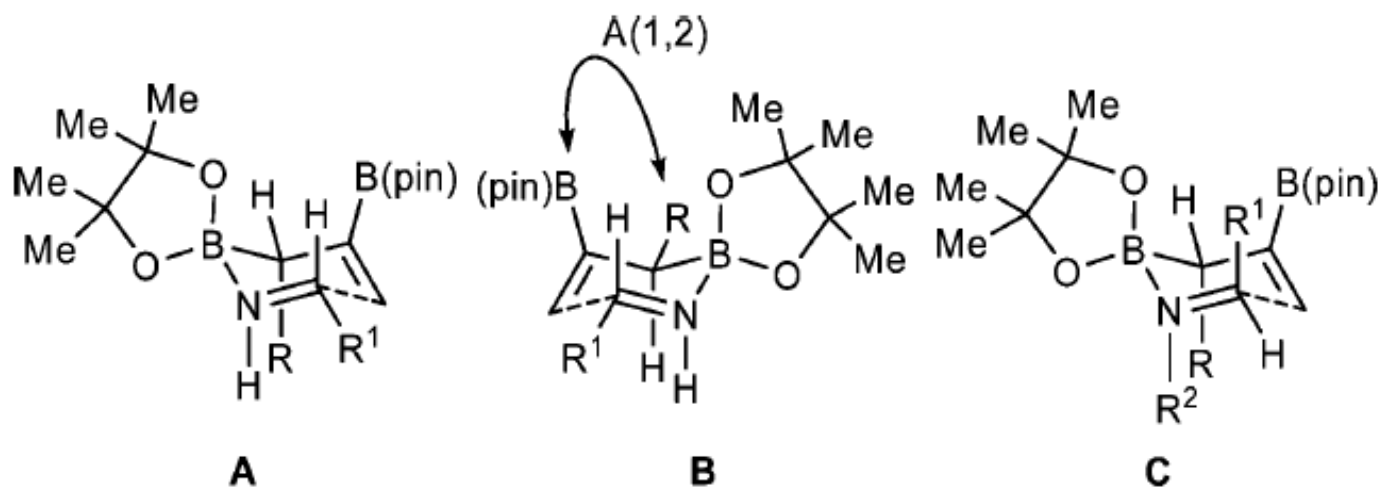
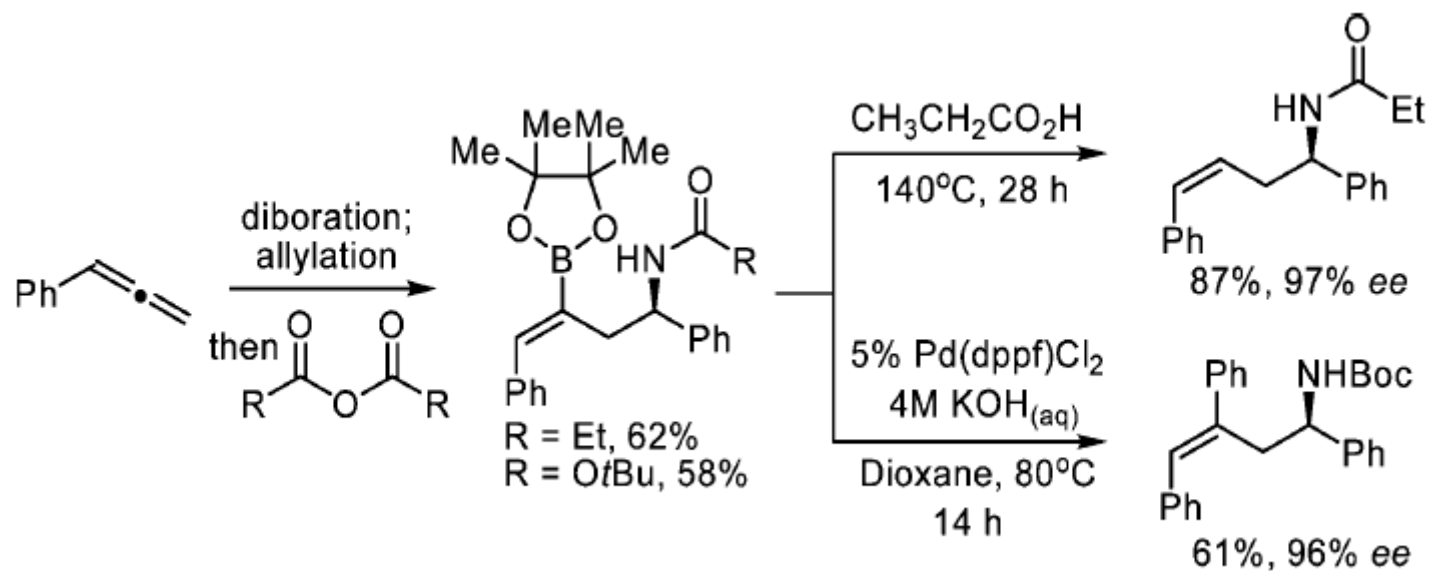


Allene Borylation One-Pot

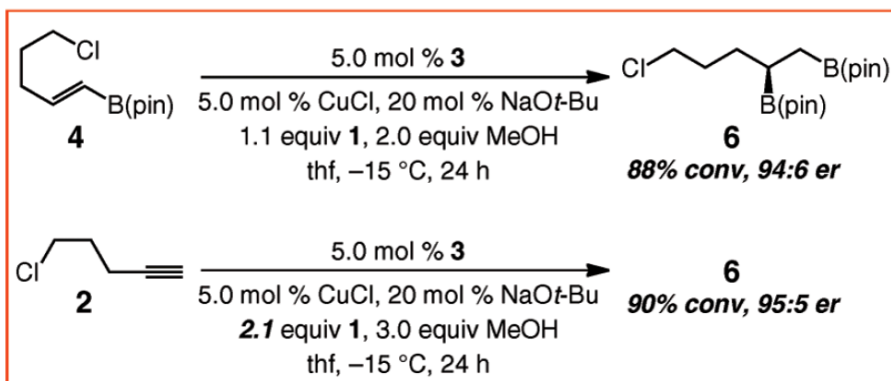
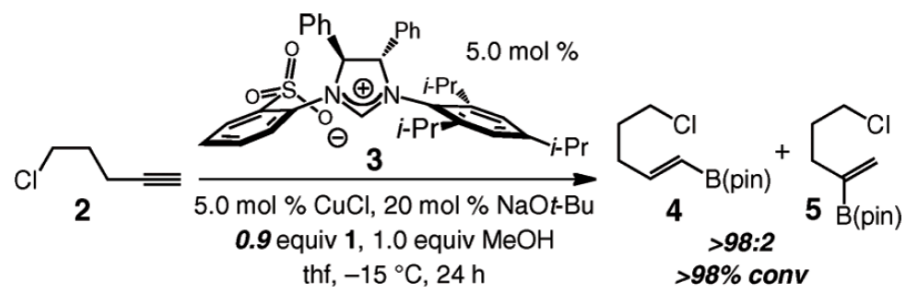


- R alkyl, branched, benzyl, aryl 37–62% yield, 91–94% ee
- Can couple with aryl, vinyl halides, pseudohalides

Allene borylation - Tandem Reactions

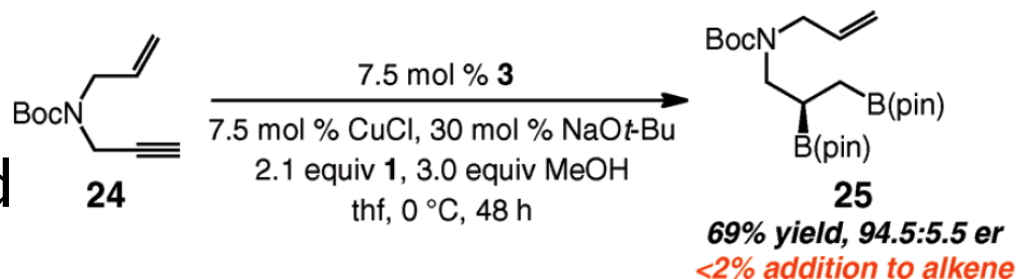


Sequential Alkyne Diboration

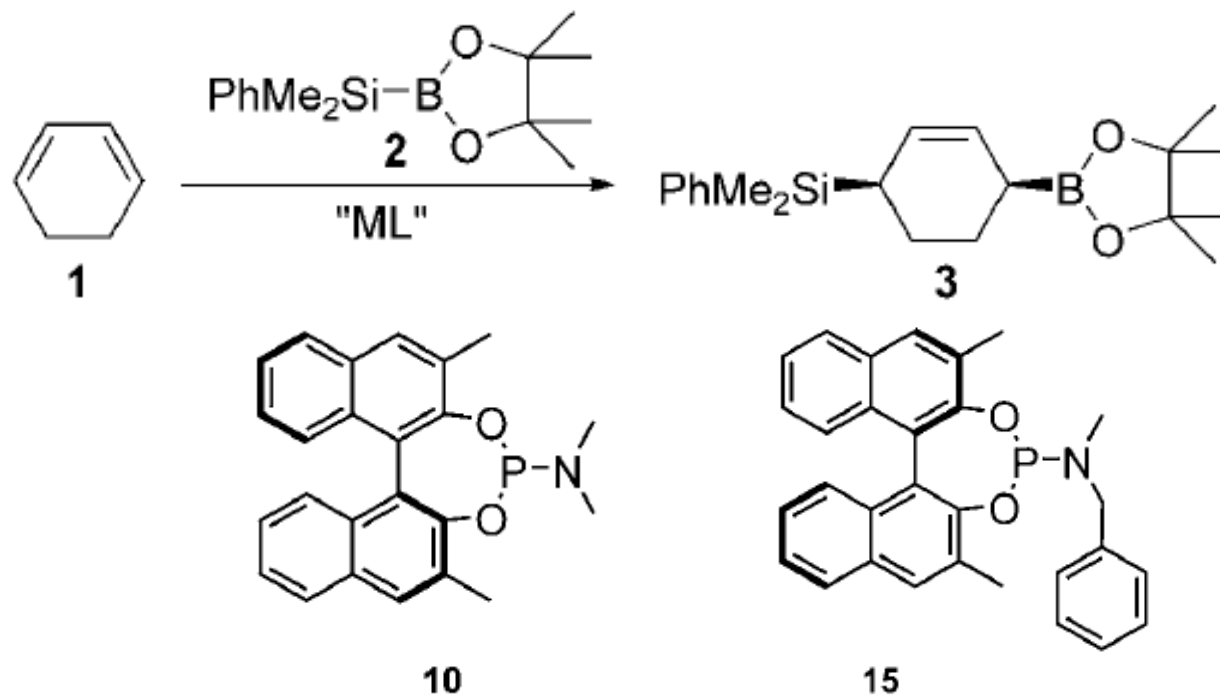


	Yield	er
5	78	96.5:3.5
6	60	96.5:3.5
7	76	97:3
8	61 ^e	97.5:2.5

Can also control regioselectivity based on NHC catalyst



Silaborations



Entry	Metal complex ^[a]	Ligand ^[b]	Time [h]	Conversion [%] ^[c]	ee [%] (product)
1	Pt(acac) ₂	(<i>S</i>)- 10	48	84	57 (3)
2	Pt(acac) ₂	(<i>S</i>)- 11	48	50	27 (3)
3	Pt(acac) ₂	12	48	52	32 (3)
4	Pt(acac) ₂	13	48	26	21 (3)
5	Pt(acac) ₂	14	48	11	13 (3)
6	Pt(acac) ₂	15	48	92	58 (<i>ent</i> - 3)

Conclusions

- Asymmetric Borylation is a highly selective and synthetically useful of installing C–B bonds.
- Products are versatile and can be used in allylations, oxidations, cross-coupling reactions etc.
- Can synthesize tertiary allylic alcohols, and tertiary chiral allylborons.
- Boron is a cheap and non-toxic element.
- Product stability limits the utility of boron chemistry – more stable boron esters will increase use.
- Asymmetric borylation of cis-alkenes continues to be an unsolved problem.