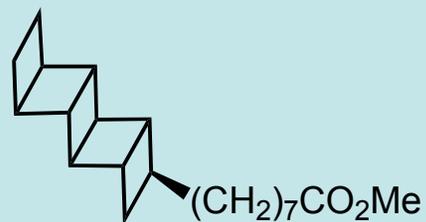
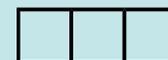


# Ladderanes: Uses and Synthesis

Nicholas Anderson  
Denmark Group Meeting  
October 28, 2008

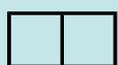
# Outline

- Ladderane types and classifications
- Potential applications of ladderanes
- Synthesis of ladderanes
- Ladderane natural products



# Types of Ladderanes - Linear

Length



[2]-Ladderane



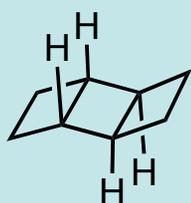
[3]-Ladderane



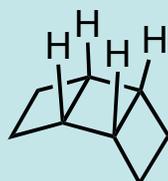
[4]-Ladderane

Increasing the number of fused cyclobutanes increases the heat of formation by ~20-23kcal/mol per ring

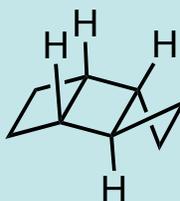
Stereochemistry



cis, anti, cis  
Lowest energy



cis, syn, cis



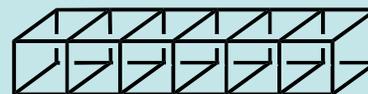
cis, trans  
Highest energy

# Types of Ladderanes - Cyclic

Named based on Size



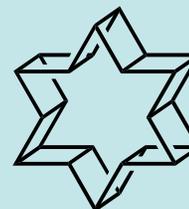
[3]-Prismane



Hexa[4]prismane



[4]-Prismane  
(cubane)



Israelane



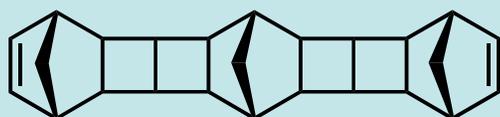
[5]-Prismane

Currently these structures are curiosities.

# Ladderanes - Molecular Spacers



[10]-Ladderane



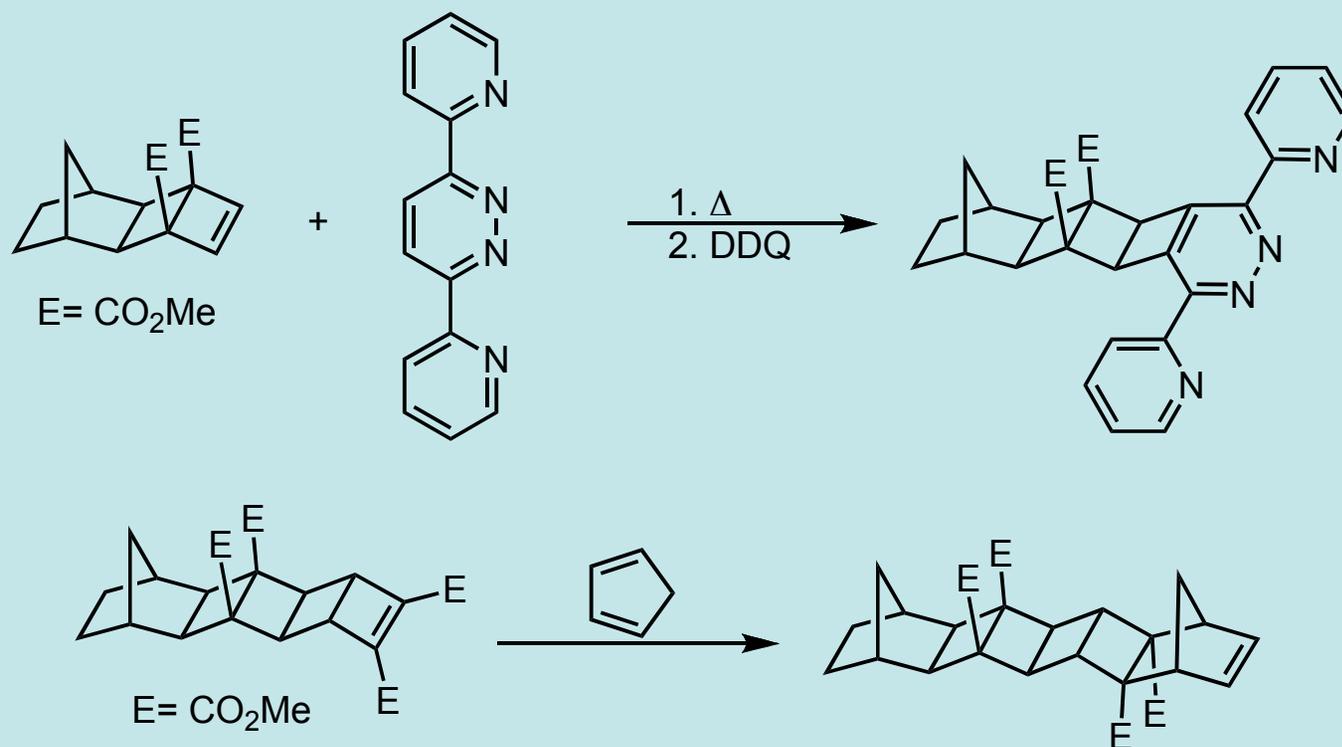
10 $\sigma$  binane



[5]polynorbornane

# Ladderanes - Molecular Spacers

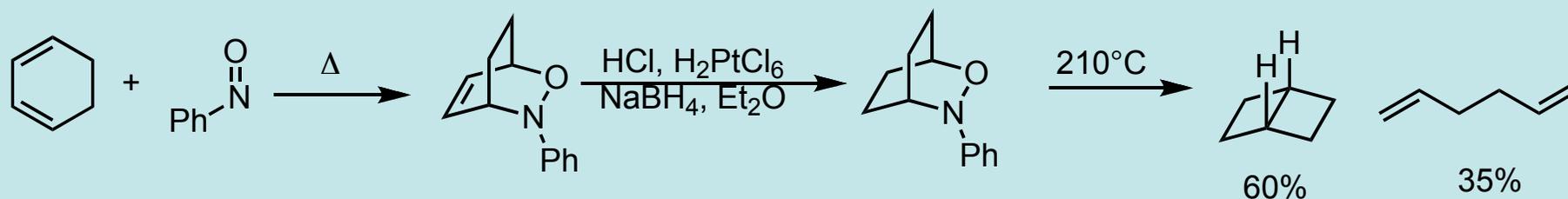
Examples of capping to yield end functionalized materials



Possible applications to molecular electronics

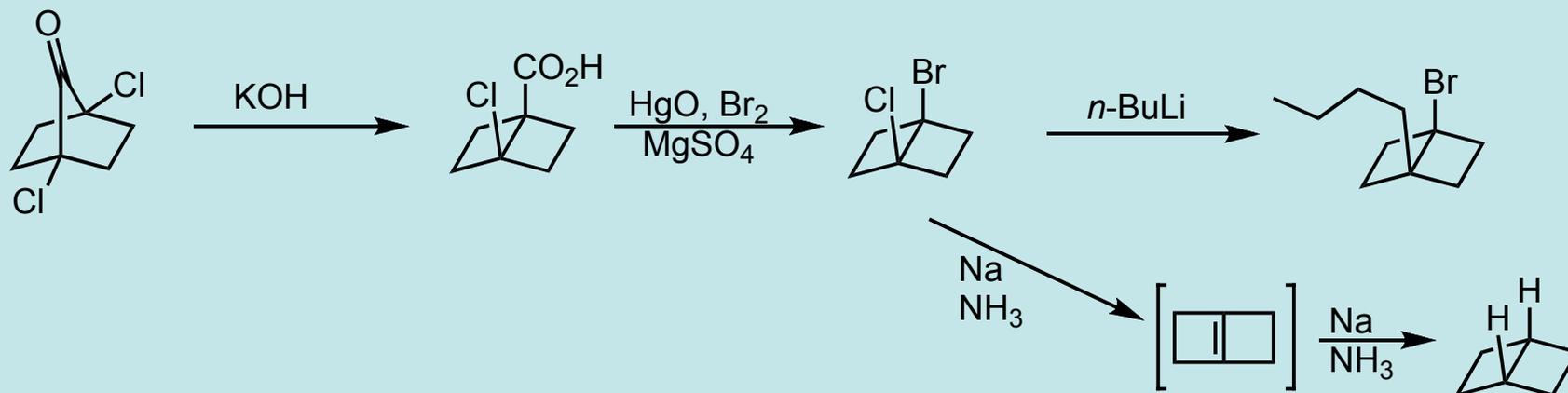
# Synthesis of [2]-Ladderane

First optimized synthesis (1963):



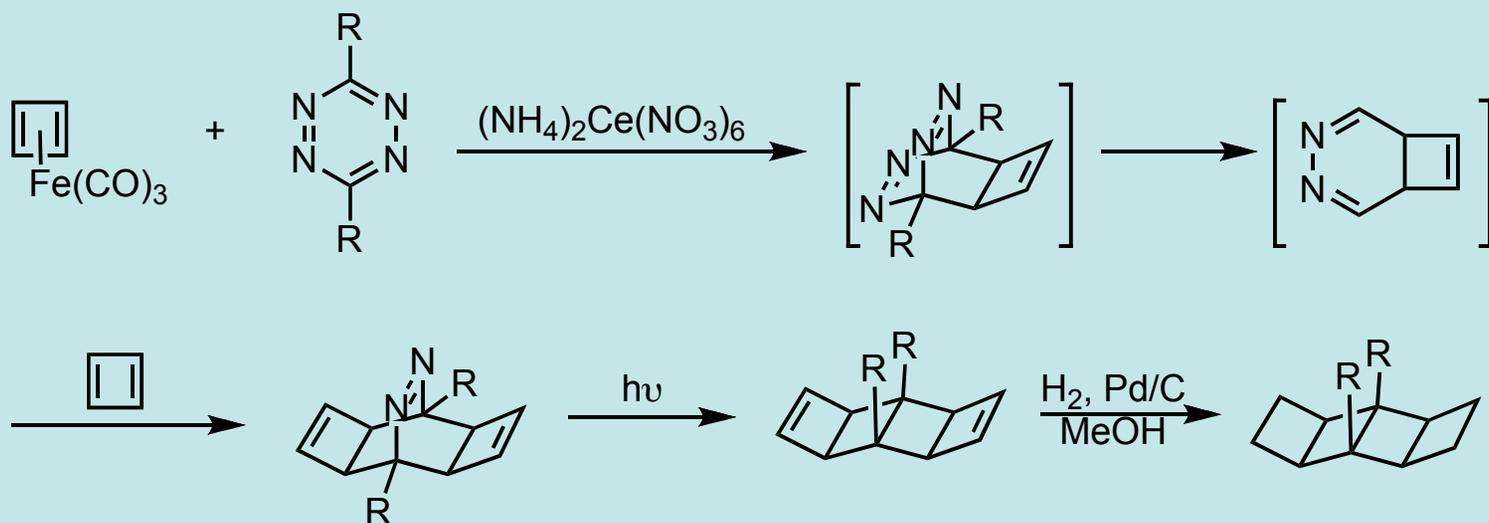
Thermally stable!

Alternative synthesis (1986):



Griffin, C. E.; Hepfinger, N. F.; Shapiro, B. L. *J. Am. Chem. Soc.* **1963**, *85*, 2683-2684.  
Wiberg, K. B.; et al. *Tetrahedron*, **1986**, *42*, 1895-1902.

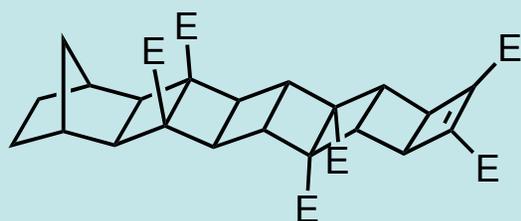
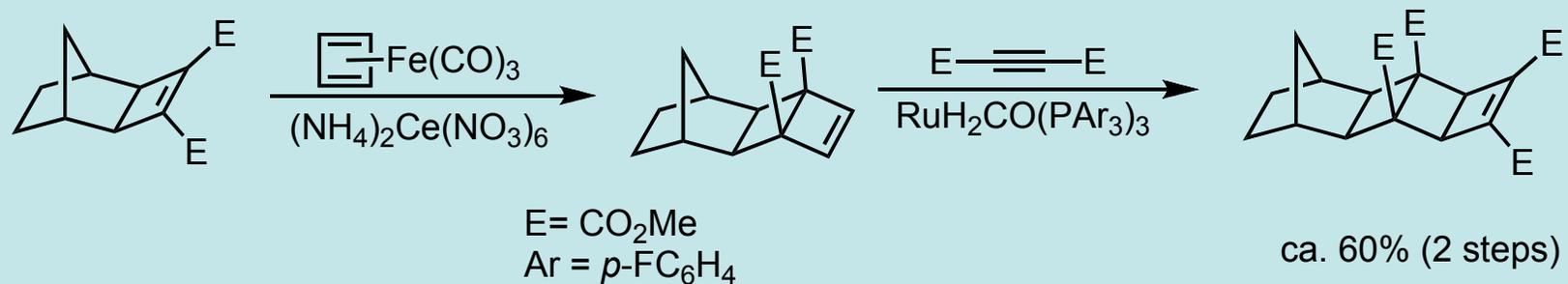
# Synthesis of a [5]-Ladderane



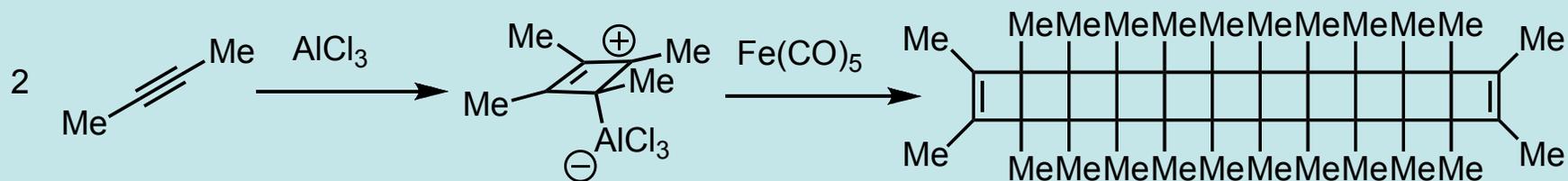
R=  $\text{CO}_2\text{Me}$

- Synthesis of ladderanes has been accomplished
- Ladderanes are kinetically stable
- Cis-trans-cis isomers are typically obtained

# Higher Order Ladderanes



[7]-Ladderane

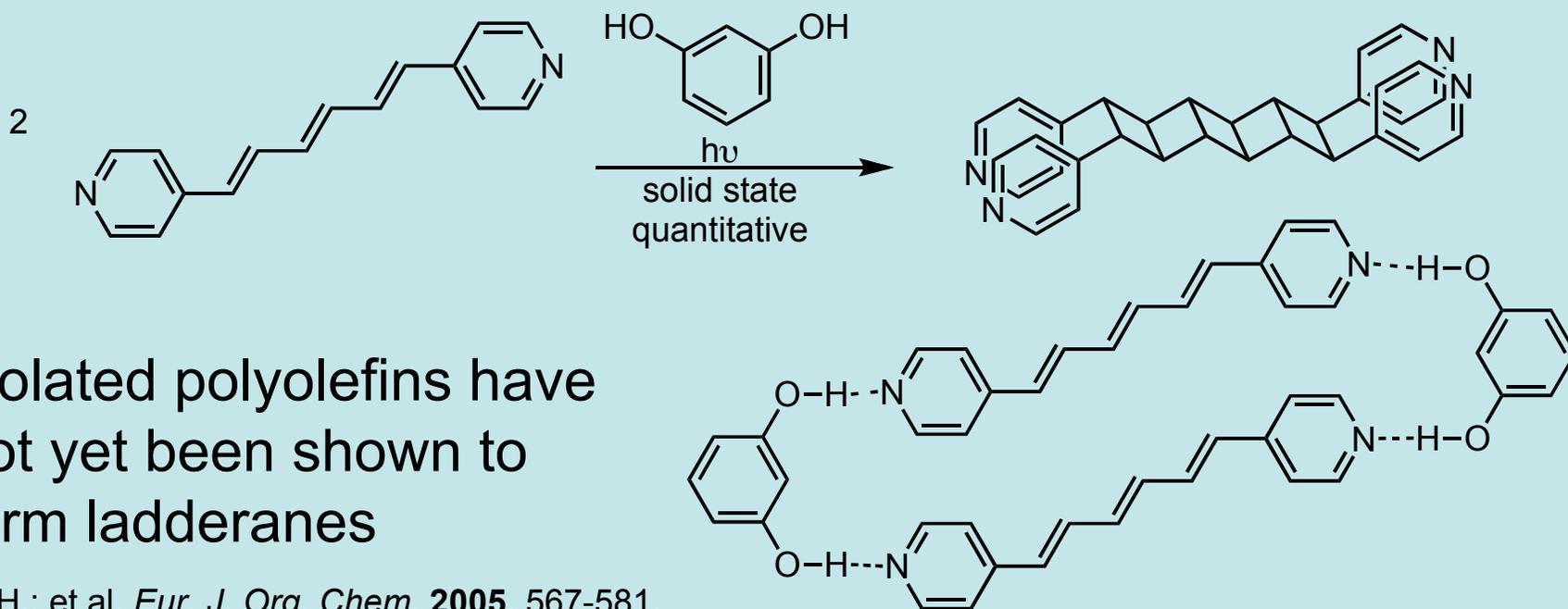
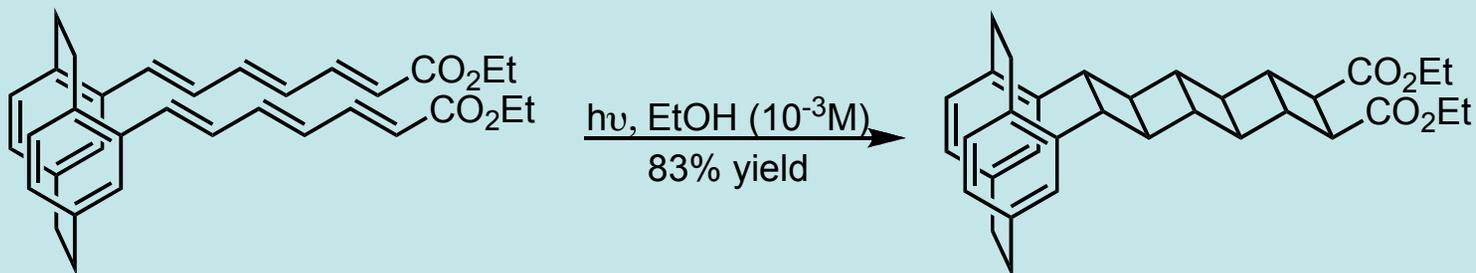


Formed a statistical mixture of up to [11]-Ladderane

Warrener, R. N.; Abbenante, G. *J. Am. Chem. Soc.* **1994**, *116*, 3645-3646.

Marsella, M. J.; et al. *Synlett* **2004**, 192-194.

# One Step Synthesis



Isolated polyolefins have not yet been shown to form ladderanes

Hopf, H.; et al. *Eur. J. Org. Chem.* **2005**, 567-581.

Frisic, T.; MacGillivray, L. R. *Supramol. Chem.* **2005**, 17, 47-51.

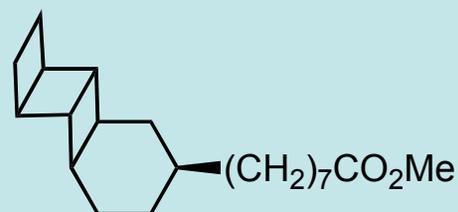
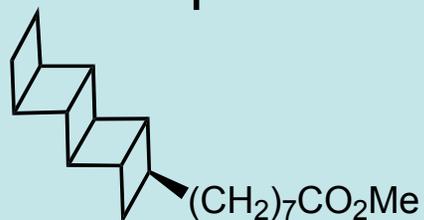
# Ladderanes in Natural Products

## Anammox bacteria

- Deep sea bacteria
- Capable of ammonia oxidation



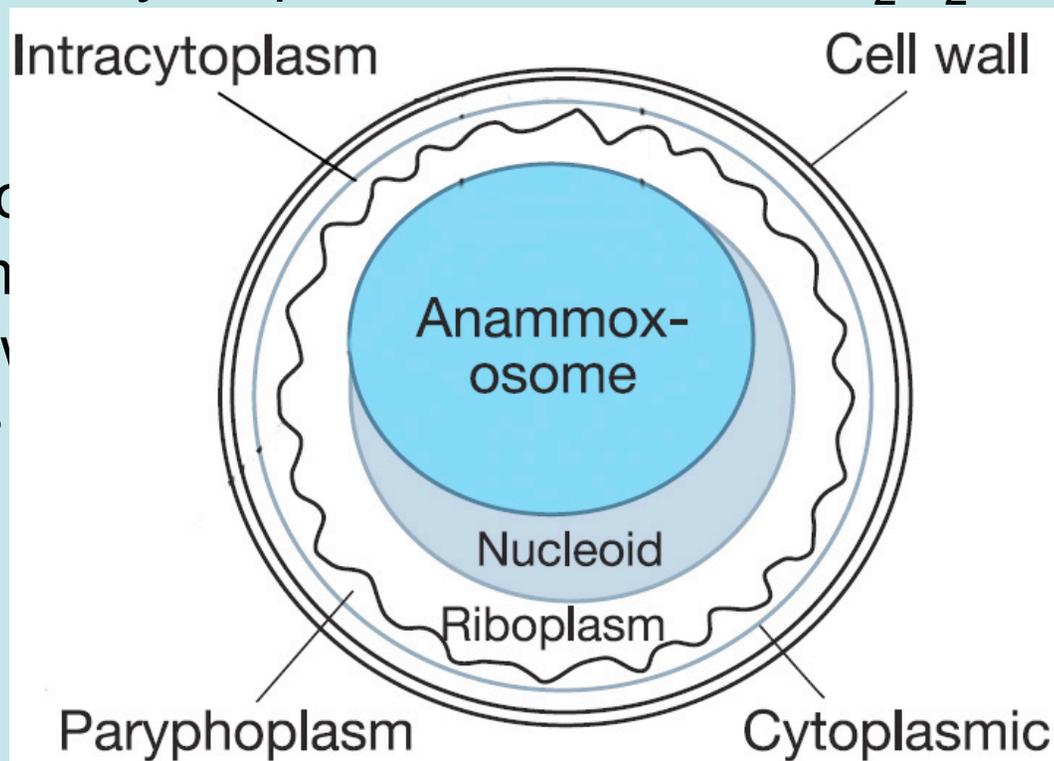
- Produce  $\text{N}_2\text{H}_2$  and  $\text{NH}_2\text{OH}$ 
  - Highly toxic intermediates
  - Membrane permeable
- Ladderane lipids isolated from the bacteria



# Ladderanes in Natural Products

- Ladderane lipids form dense lipid bilayers
  - Ladderane lipid bilayers:  $1.5 \text{ kg/dm}^3$
  - Normal lipid bilayers:  $1.2 \text{ kg/dm}^3$
- Denser bilayers prevent diffusion of  $\text{N}_2\text{H}_2$  and  $\text{NH}_2\text{OH}$

- Study of
- ~3n
- Slow
- 

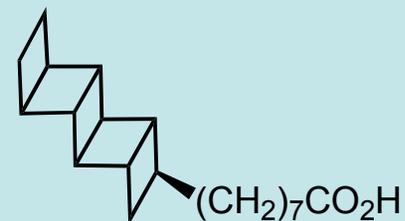


availability  
ox bacteria

Damsté, J. S. S.; et al. *Nature*, **2002**, 419, 708-712.

Damsté, J. S. S.; et al. *FEBS Journal*, **2005**, 4270-4283.

# Pentacycloanammoxic Acid Synthesis



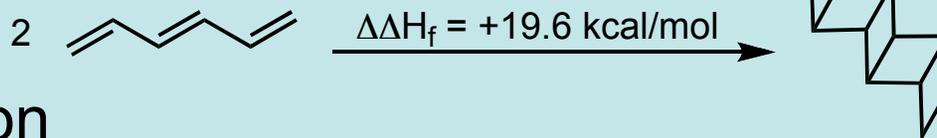
Goals of Total Synthesis:

- Verify structure
- Determine absolute stereochemistry
- Provide a larger amount of material

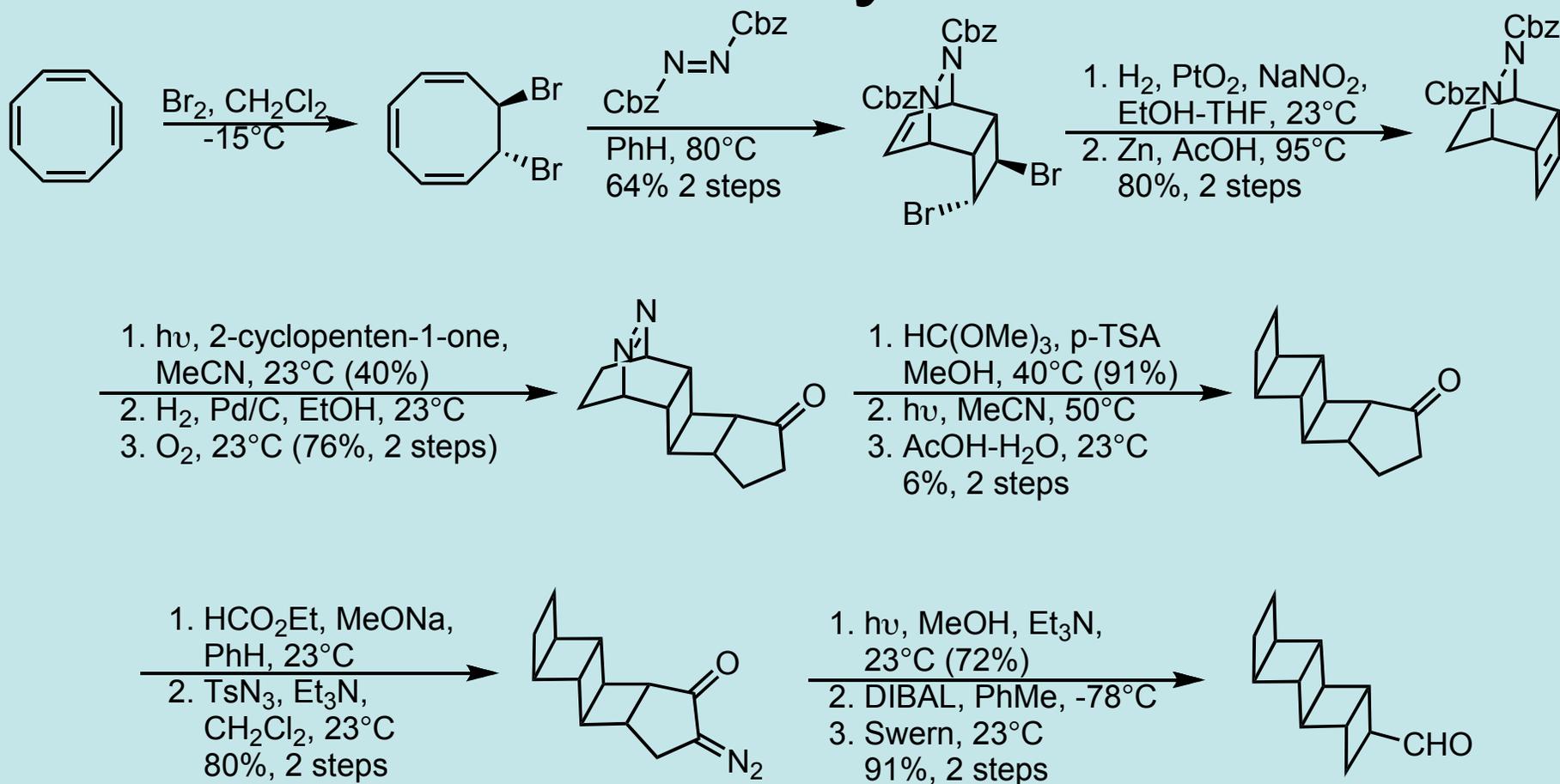
Pentacycloanammoxic Acid

Potential Problems:

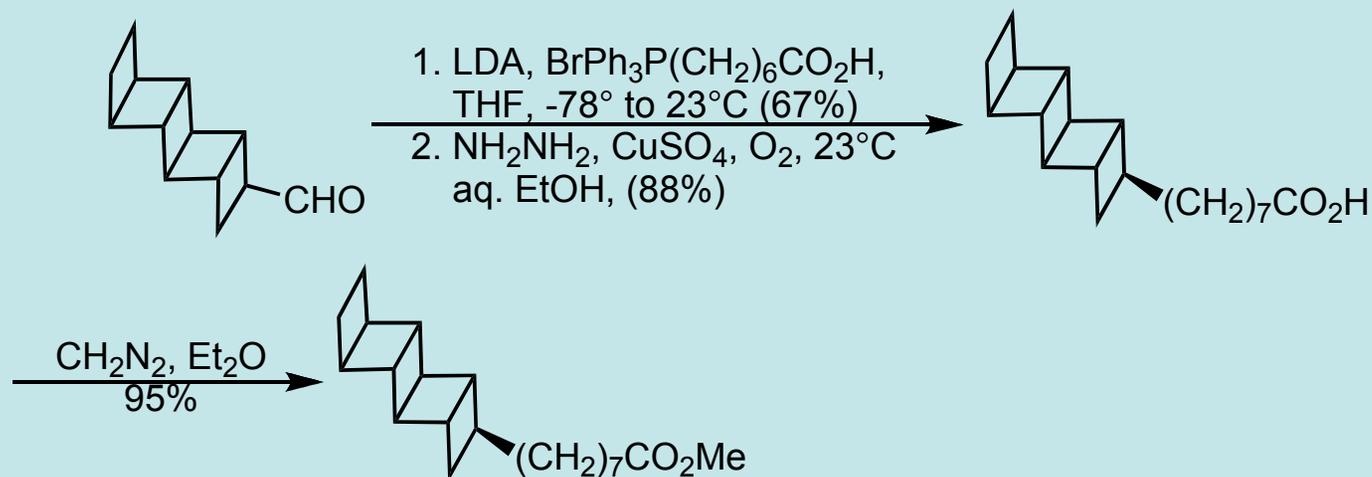
- Highly strained core
- Very little functionalization
- 9 Contiguous stereocenters



# Pentacycloanammoxic Acid Racemic Synthesis



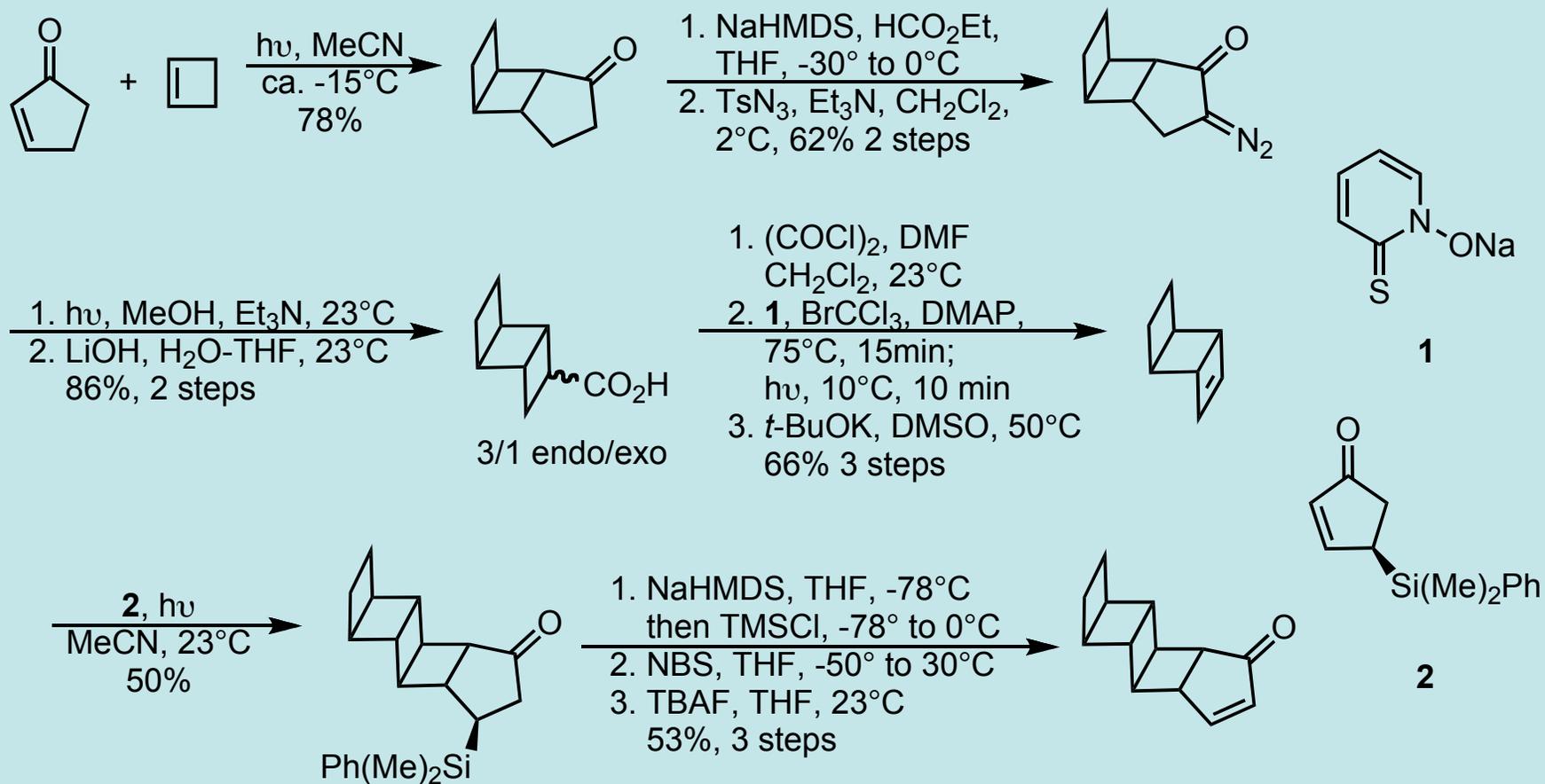
# Pentacycloanammoxic Acid Racemic Synthesis



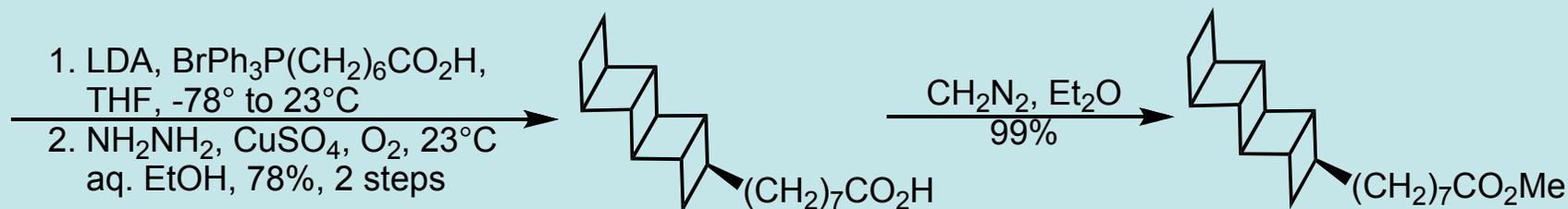
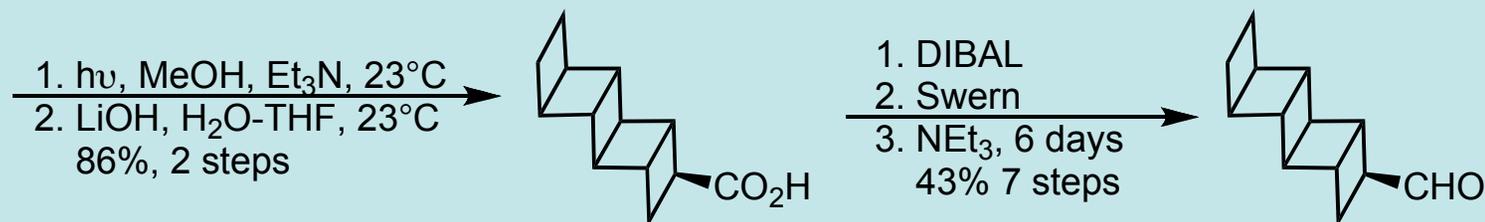
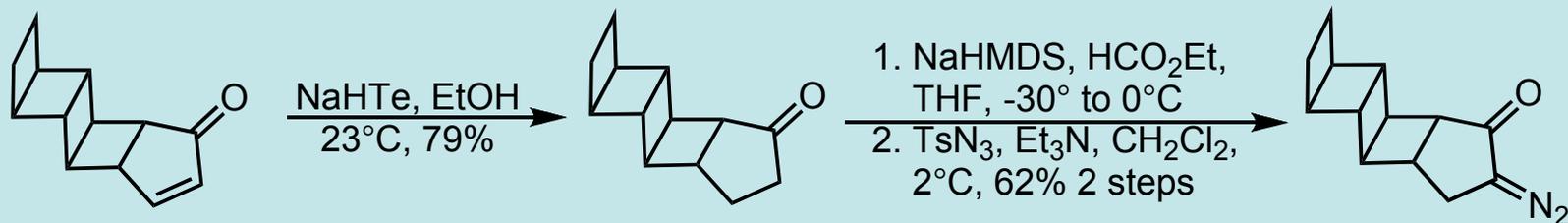
Overall yield: 0.25%

- Successful structure confirmation
- Did not establish the absolute configuration
- Did not provide a convenient route to large quantities of material

# Pentacycloanammoxic Acid Enantioenriched Synthesis



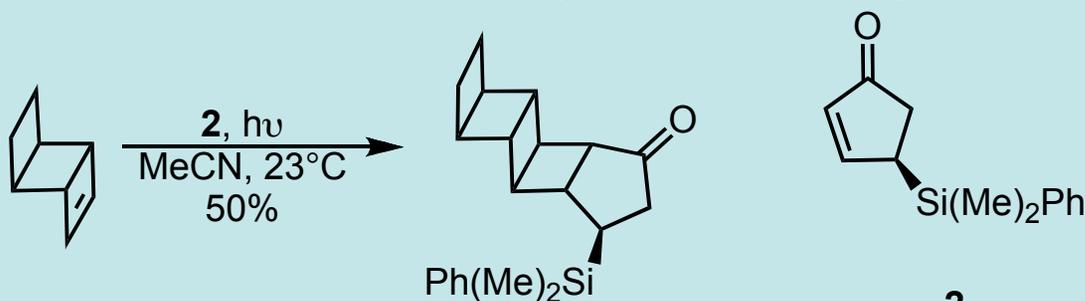
# Pentacycloanammoxic Acid Enantioenriched Synthesis



# Pentacycloanammoxic Acid Enantioenriched Synthesis

Overall yield: 1.9%, 17 steps

- Reconfirms the molecular structure
- Enantiopure product will allow determination of absolute stereochemistry of natural product



- Provides only a slightly more convenient route to the natural product
- Development of an asymmetric [2+2] cycloaddition

# Conclusions

- Synthesis of small ladderanes is well developed
- Synthesis of larger ladderanes and prismanes is less well developed
- While ladderanes have a number of potential uses, they need to be further investigated
- The appearance of ladderane natural products is exciting and the recent development of a reasonable synthesis of the natural product should allow their continued investigation