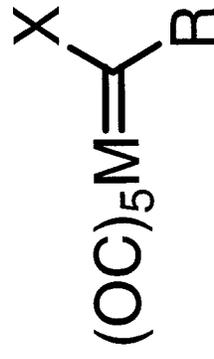


Reaction of Alkynes with Amino Chromium Fischer Carbene Complexes and Synthesis of N-Heterocycles



X = OR, NR₂

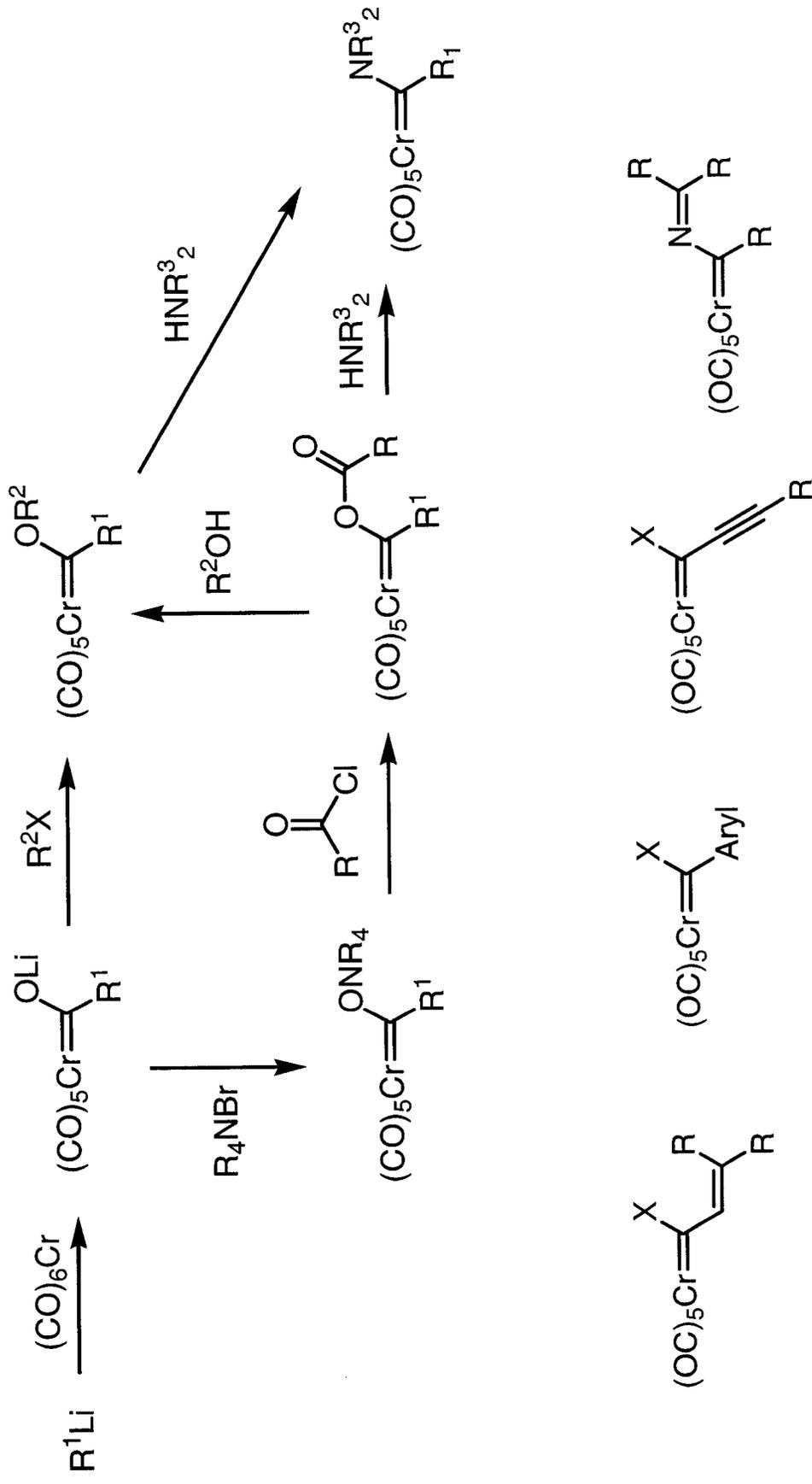
M = Cr, Mo, W

R = Aryl, Alkenyl, Alkynyl

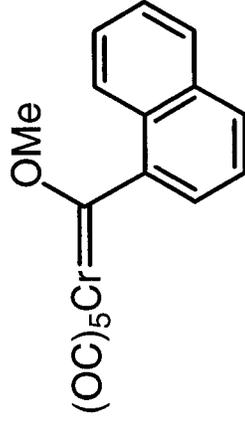
Min Xie

11/29/2005

General Preparation of FCC

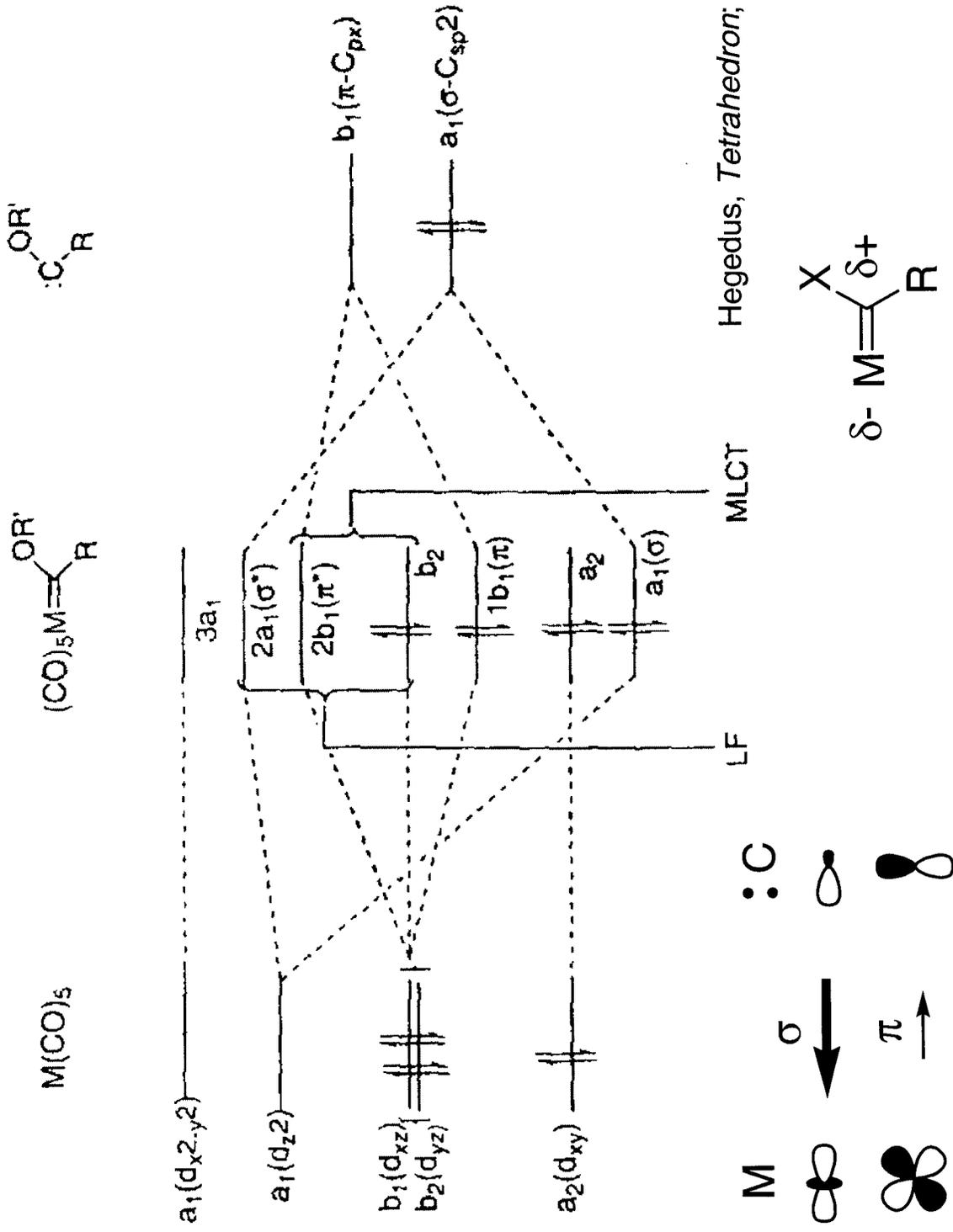


Scalability and Easy Handling



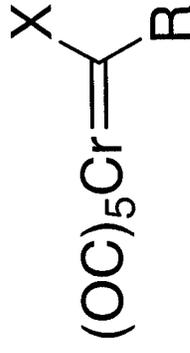
This complex was prepared from 1-bromonaphthalene on a **285 gram scale** in **77 % yield (3 crops)** ... The red pile is **250 grams** of the naphthalene complex ...

Electronic Structure of FCC



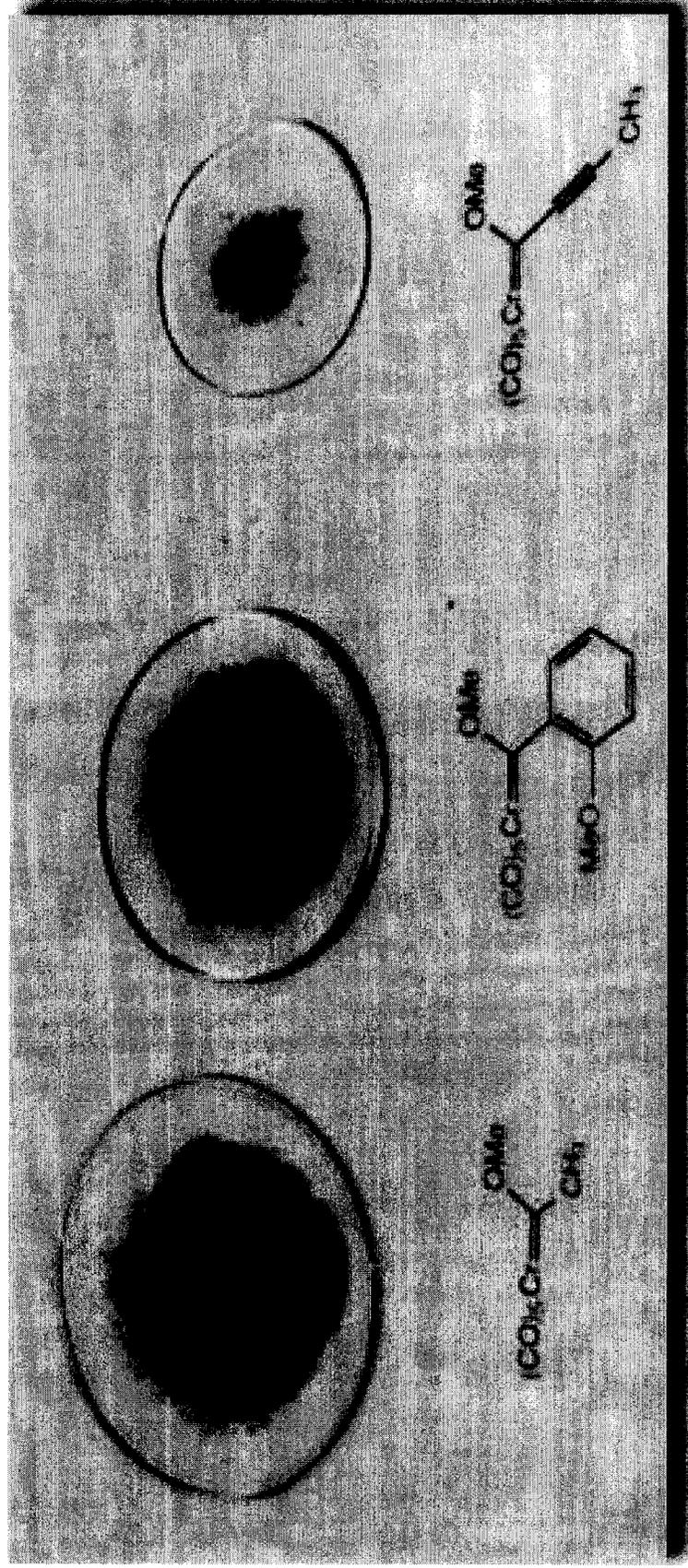
Carbene carbon in FCC is typically electrophilic.

Effect of Heteratom Substituents



- X a better donor (N vs O)
- Carbene carbon a weaker π -acceptor
- Decreased $\text{M} \rightarrow$ carbene backbonding
- Increased $\text{M} \rightarrow \text{CO}$ backbonding
- shortening of the M-CO bond
- lower C=O frequency
- Blue shift of visible spectra
- More stable, less reactive

Effect of Carbon Substituents



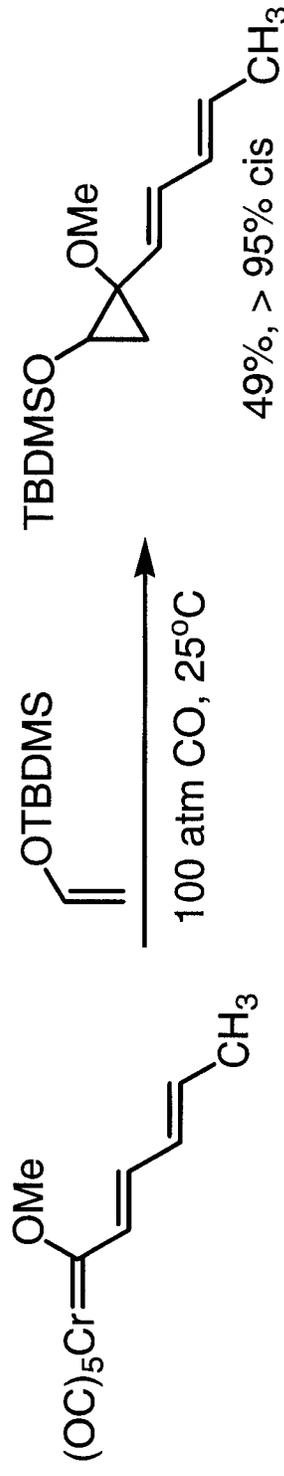
Better π acceptor, smaller HOMO-LUMO gap, red shift.

Typical Reactions of FCC

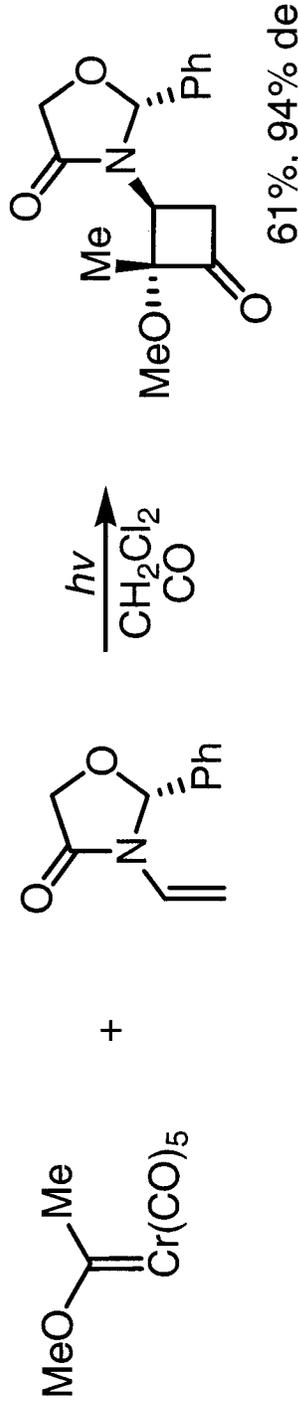


pKa ~12
 methyl acetate pKa ~ 25
 diethyl malonate pKa ~13

Wulff, *Tetrahedron Lett.*, **1998**

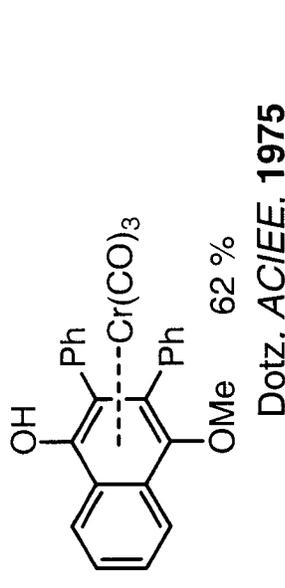


Wulff, *JACS.*, **1988**

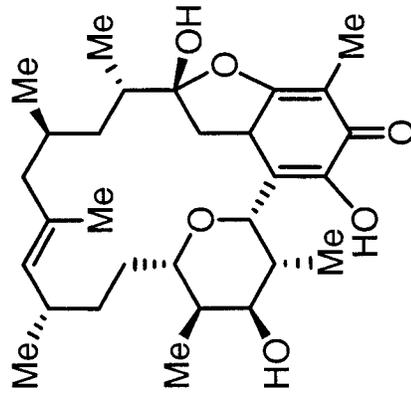
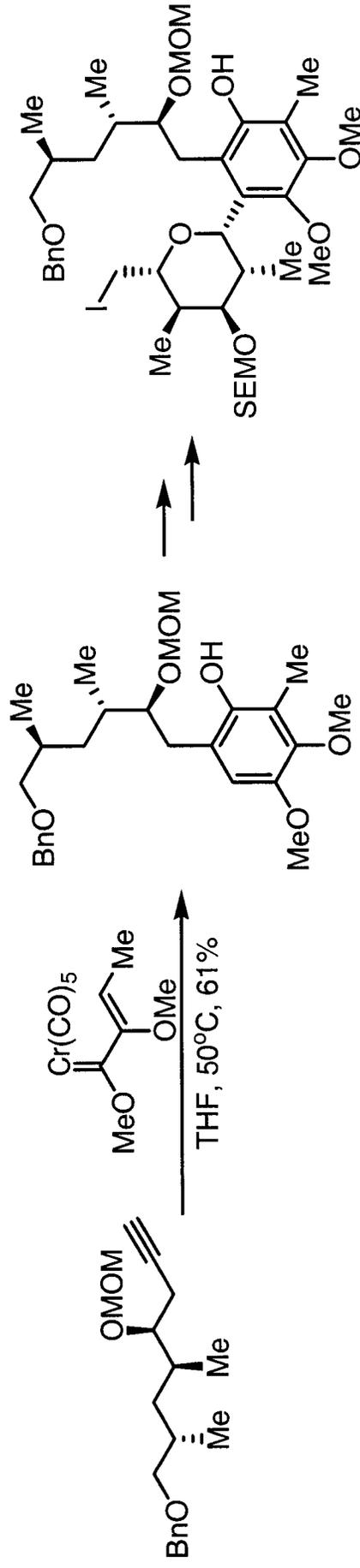


Hegedus, *JACS.*, **1991**

Dotz Reaction of Alkoxy Substituted FCC



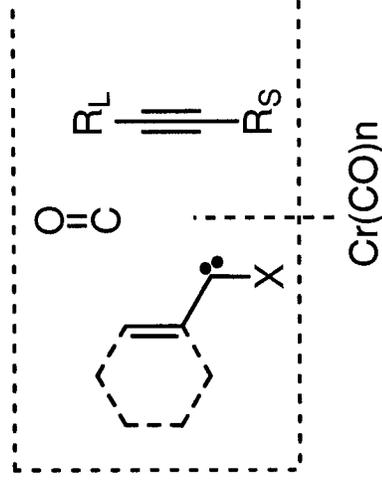
Dotz, *AC/EE*, 1975



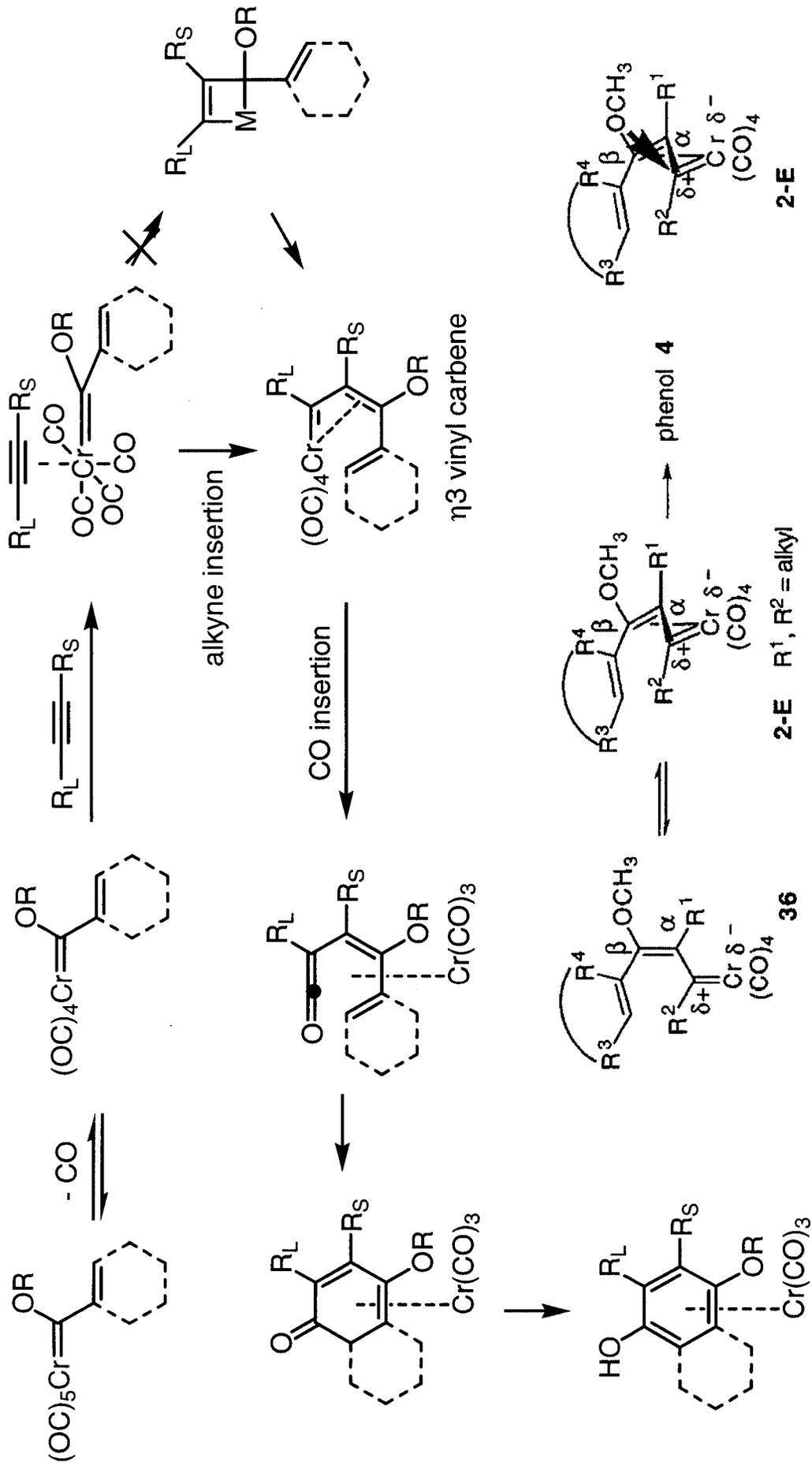
(-)-Kendomycin

White, *OL*, 2004

Chromium-Templated [3+2+1]



General Mechanism of Dotz Reaction



Meijere, ACIE, 2000

Wulff, Organometallics, 1998

Dissociation of CO

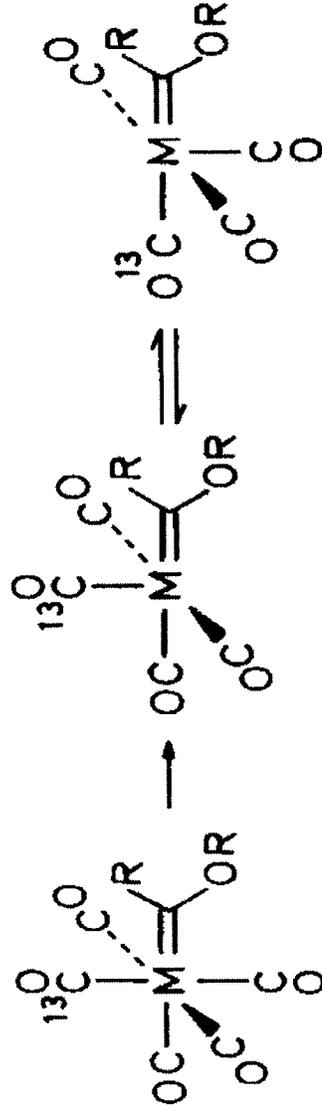
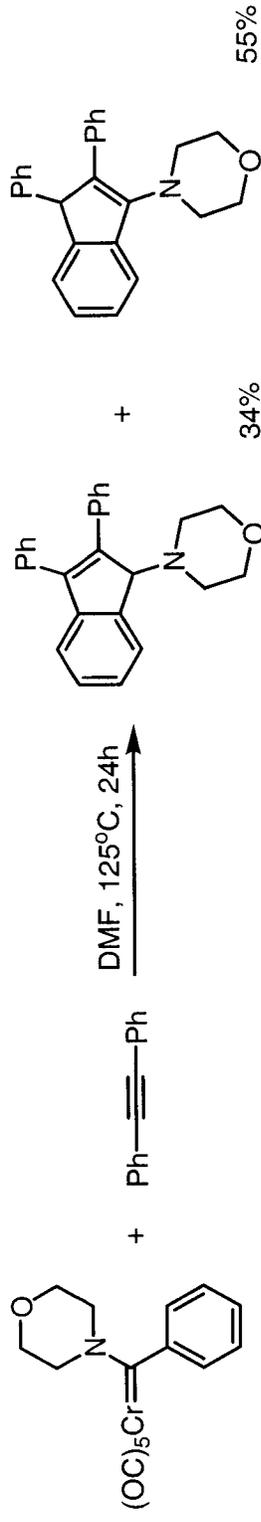


Table III. Rates of CO Dissociation from $(\text{CO})_3\text{M}=\text{CRR}'$

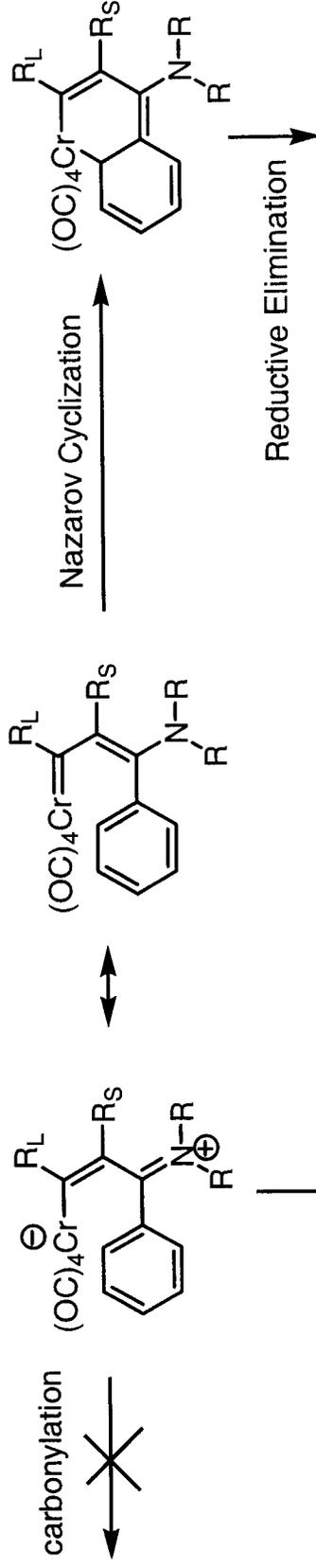
compd	temp, °C	rate, ^a $\text{s}^{-1} \times 10^5$	ΔG^\ddagger , kcal mol^{-1}
3Cr	44 ^b	18.0	24.0
3Mo	27 ^b	4.9	23.4
3W	77 ^c	2.3	28.0

Cyclopentannulation of Amino FCC

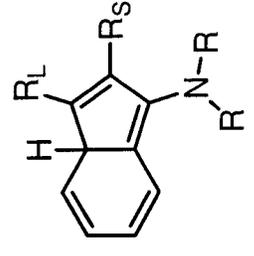


Yamashita, *TL*, 1986

95%



Reductive Elimination

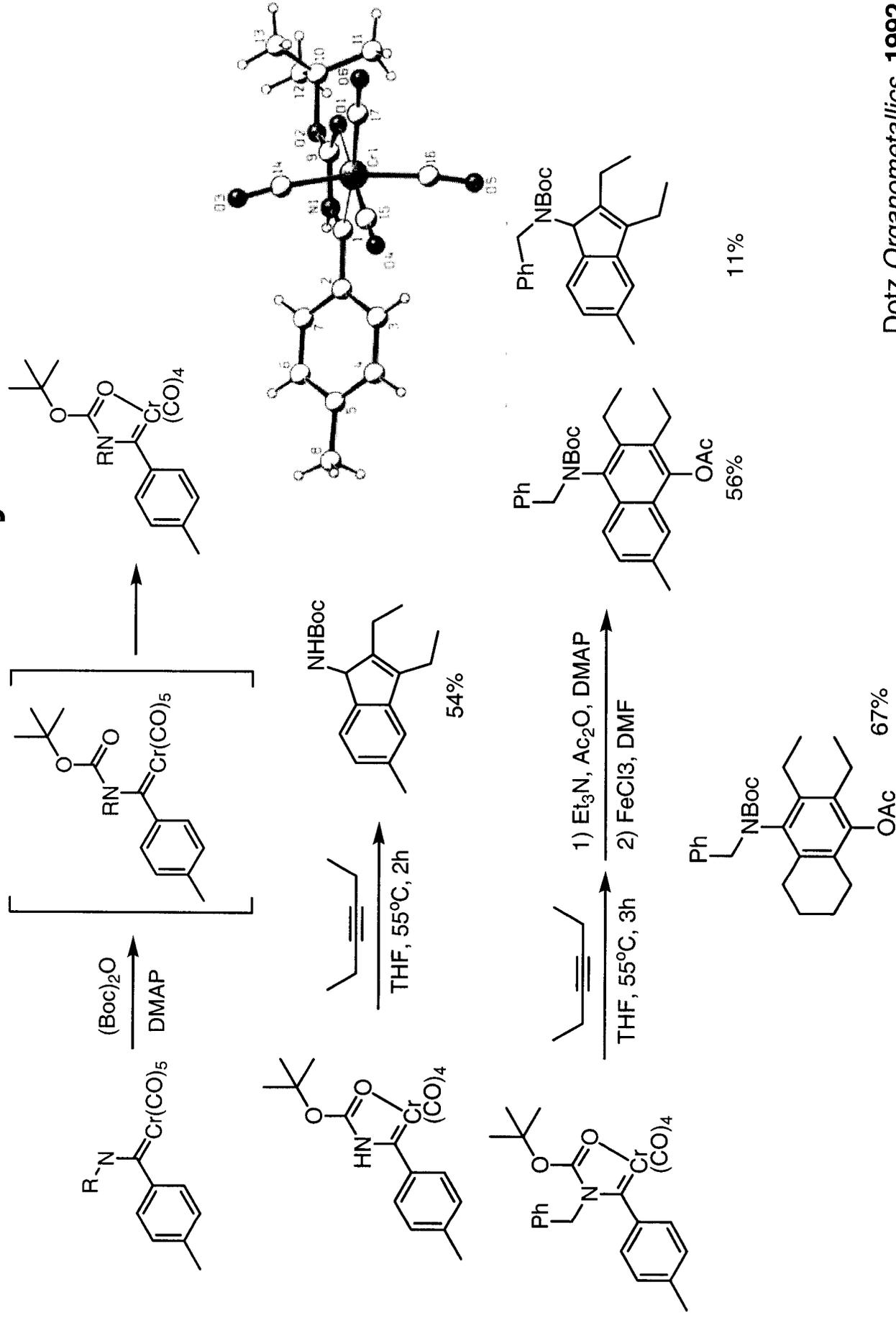


- Cr(CO)₄

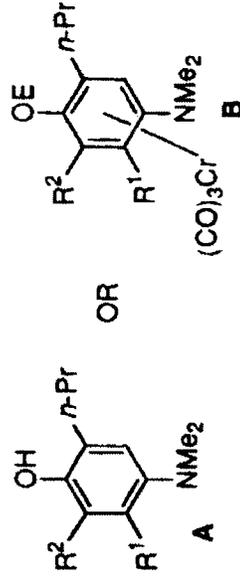
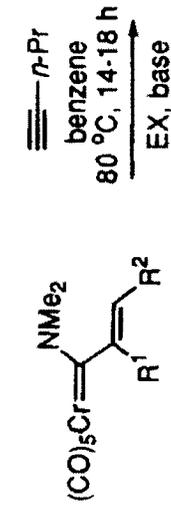
Wulff, *J Organomet Chem*, 1987

Amino Aryl FCC generally favors the formation of indenes.

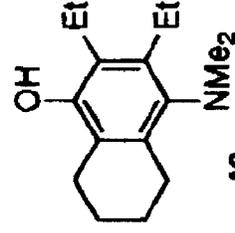
Benzannulation of Acylamino FCC



Benzannulation of Amino Alkenyl FCC



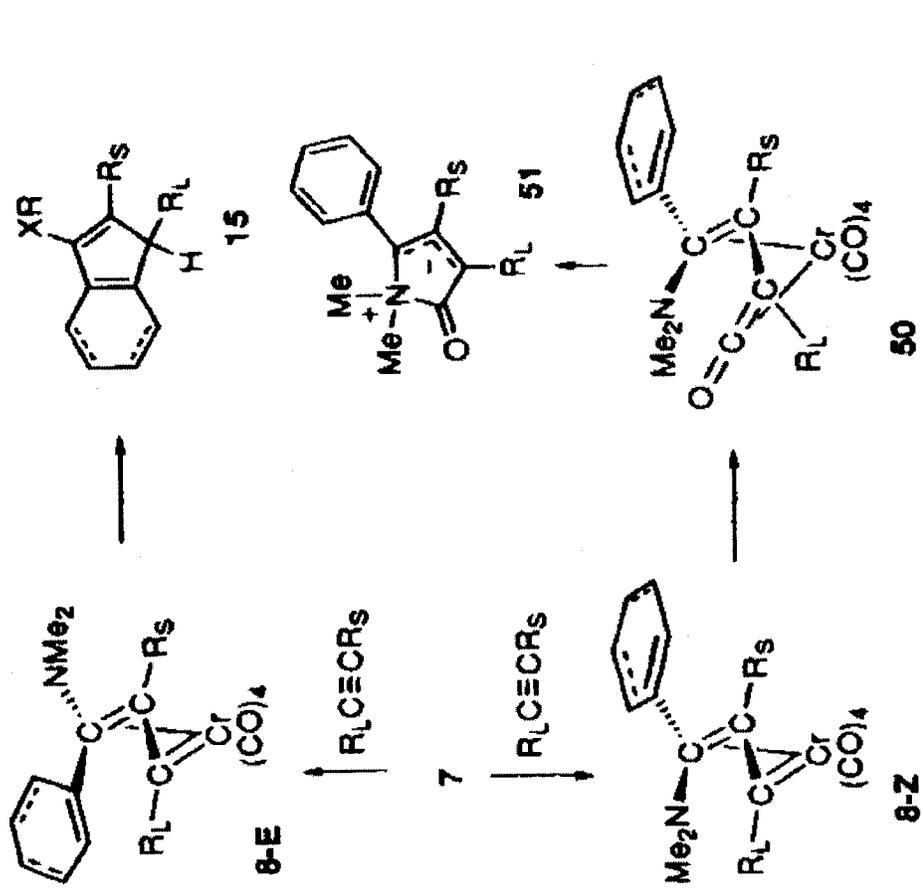
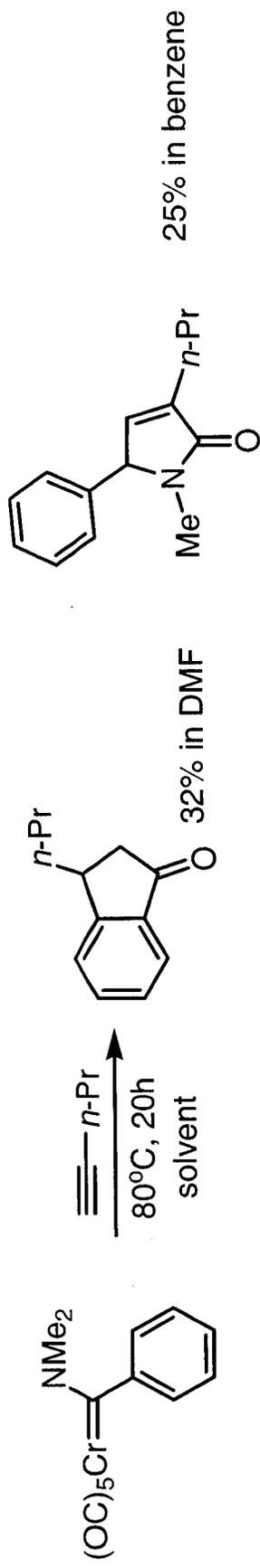
entry	complex	R^1	R^2	Ex (equiv)	base (equiv)	phenol A (% yield) ^b	complex B (% yield) ^b
1	30	Me	H	none	none	38 (57)	36 (65)
2	30	Me	H	TBSCl (2.0)	<i>i</i> -Pr ₂ EtN (3.0)	34 (57)	37 (42)
3	26	H	Me	none	none	35 (39)	
4	26	H	Me	TBSCl (2.0)	<i>i</i> -Pr ₂ EtN (3.0)	31 (64) ^{c,d}	
5	27	H	Ph	none	none		
6	19	$-(CH_2)_4-$		none	none		38 (50) ^{c,e,f,g}
7	19	$-(CH_2)_4-$		TBSOTf (1.5)	2,6-lutidine (2.5)		



complex
mixture of
products

not observed not observed

Reaction of Amino Aryl FCC



Benzannulation of Electron Poor Amino Alkenyl FCC

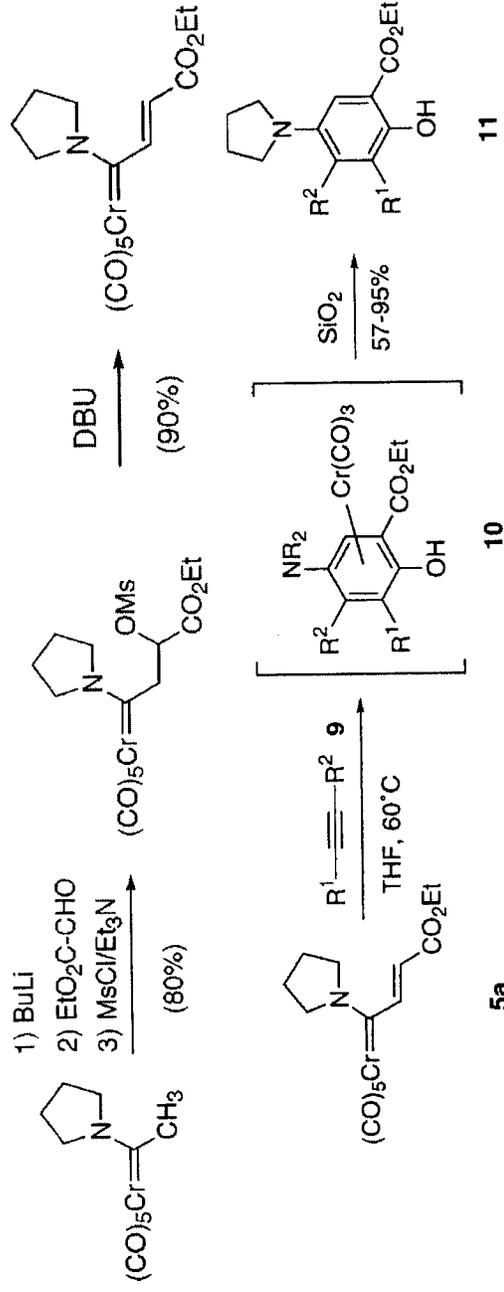
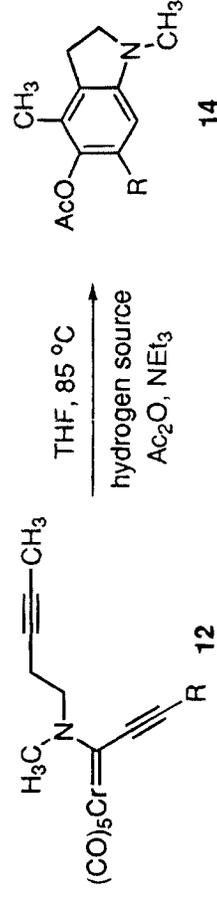
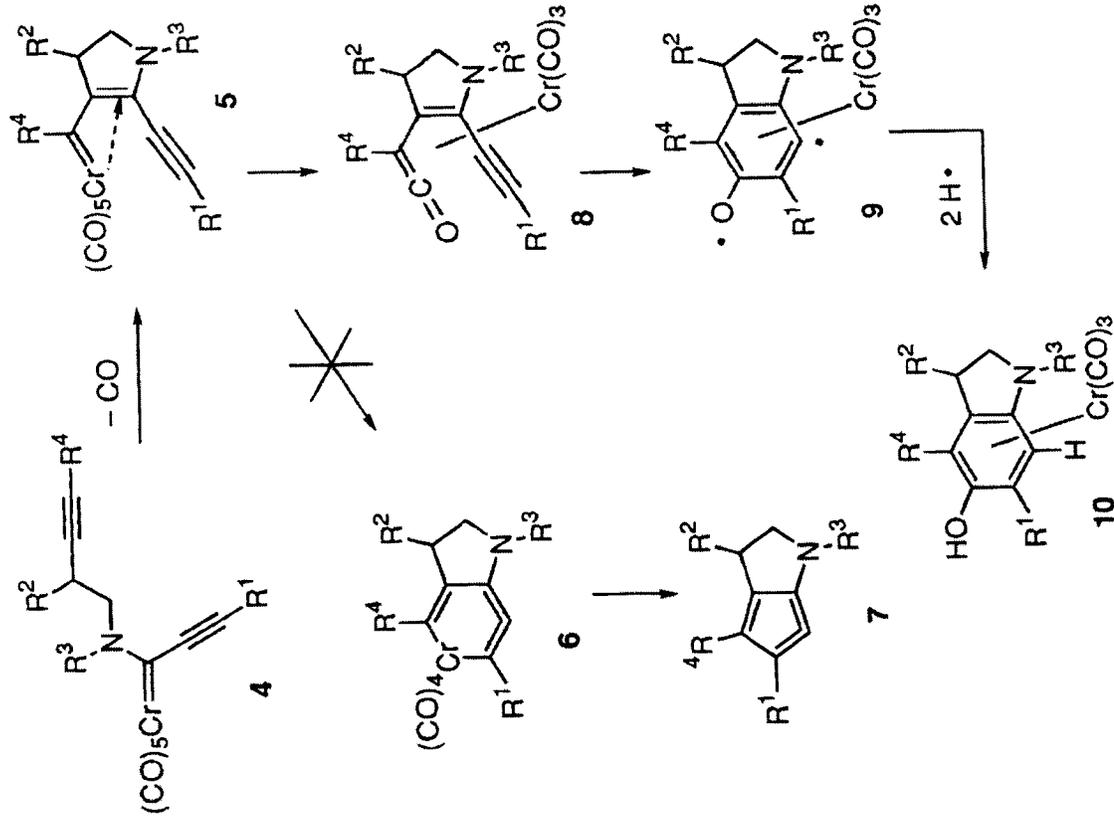


Table 1. Benzannulation of carbene complex **5a** (all the reactions were carried out in THF at 60°C at 0.03 M in carbene complex)

Entry	Compd	R ¹	R ²	Yield ^a (%)
1	11a	<i>n</i> -Pr	H	72
2	11b	<i>n</i> -Bu	H	88
3	11c	<i>n</i> -Hex	H	66
4	11d	Ph	H	95
5	11e	<i>p</i> -Tolyl	H	58
6	11f	Ferrocenyl	H	85
7	11g	1-Cyclopentenyl	H	82
8	11h	CH ₂ OTMS	H	73
9	11i^b	(CH ₂) ₂ OTs	H	68
10	11j	COOEt	H	94
11	11k	COOMe	H	89

Barluenga, *JOC*, 1998
 Barluenga, *Tetrahedron*, 2000

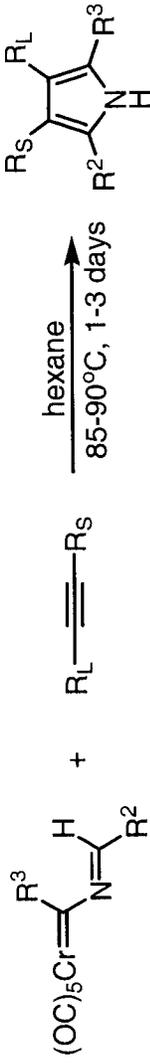
Reaction of Amino Alkynyl FCC: Enynyl Ketene Cycloaromatization



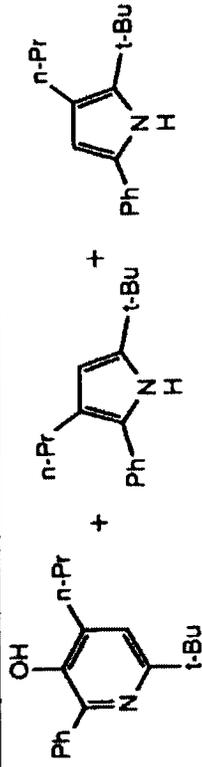
R	carbene complex ^a	hydrogen source (HS)	[12]	[HS]	% yield of 14
Ph	12a	1,4-cyclohexadiene	0.02	2.0	49
Ph	12a	γ -terpinene	0.02	2.0	53
<i>t</i> -Bu	12b	γ -terpinene	0.02	2.0	31
SiMe_3	12c	γ -terpinene	0.02	2.0	ND ^b
1- C_6H_9	12d	γ -terpinene	0.02	2.0	41

Dotz, *Synthesis* **1992**

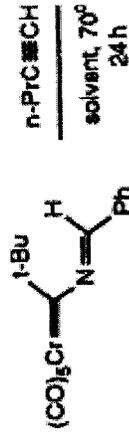
Reaction of Imino FCC



entry	complex	R ₂	R ₃	R _L	R _S	% yield of 13
1	4a	Ph	Ph	Et	Et	97
2	4a	Ph	Ph	<i>n</i> -Pr	OEt	94
3	4a	Ph	Ph	Et	(C=O)CH ₃	98
4	4b	Ph	CH ₃	Et	Et	81
5	4c	Ph	<i>t</i> -Bu	<i>n</i> -Pr	H	94
6	4c	Ph	<i>t</i> -Bu	Et	Et	85
7	4d	CH ₃	Ph	Et	Et	84

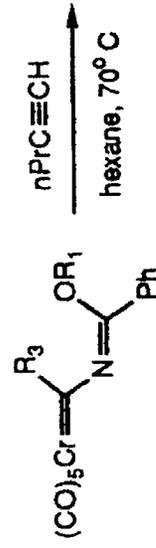


4c



hexane 5%
acetonitrile 14%

<1%
61%



16a R₃ = Ph, R₁ = Me

16b R₃ = Ph, R₁ = C(O)CH₃

complex

16a

16b

5g 51%

5g 31%

5g 4%

5g 9%

Summary on Amino FCC

- Relatively low reactivity.
- Highly substrate and condition dependent.
Tunable Selectivity.
- Mechanistically Complicated.
- Especially effective in construction of highly functionalized 5-membered heterocycles
- Stoichiometric amount of metal