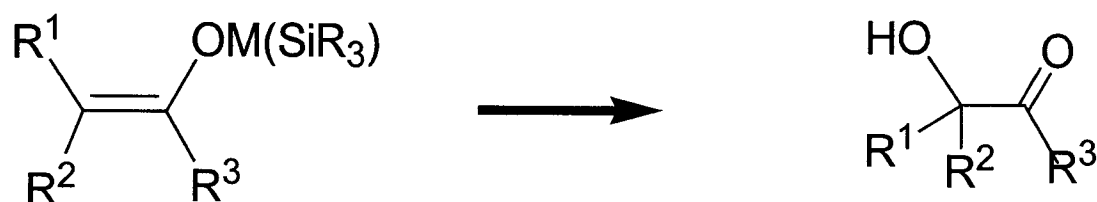


Catalytic Enantioselective α - Hydroxylation of Ketones Using Nitrosobenzene

Min Xie

April 20, 2004

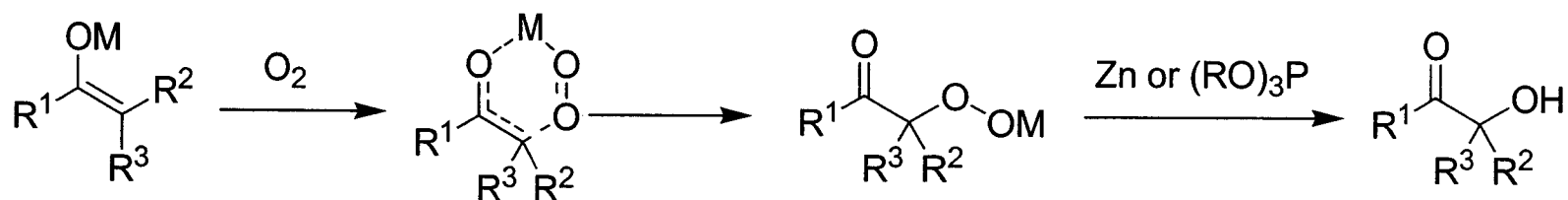
Stoichiometric oxidation of enolates: general synthesis of α -hydroxy carbonyl compound



Oxidants:

- O_2
- Peroxy Reagents
- Metal Oxides
- N-Sulfonyloxaziridines

Oxygen

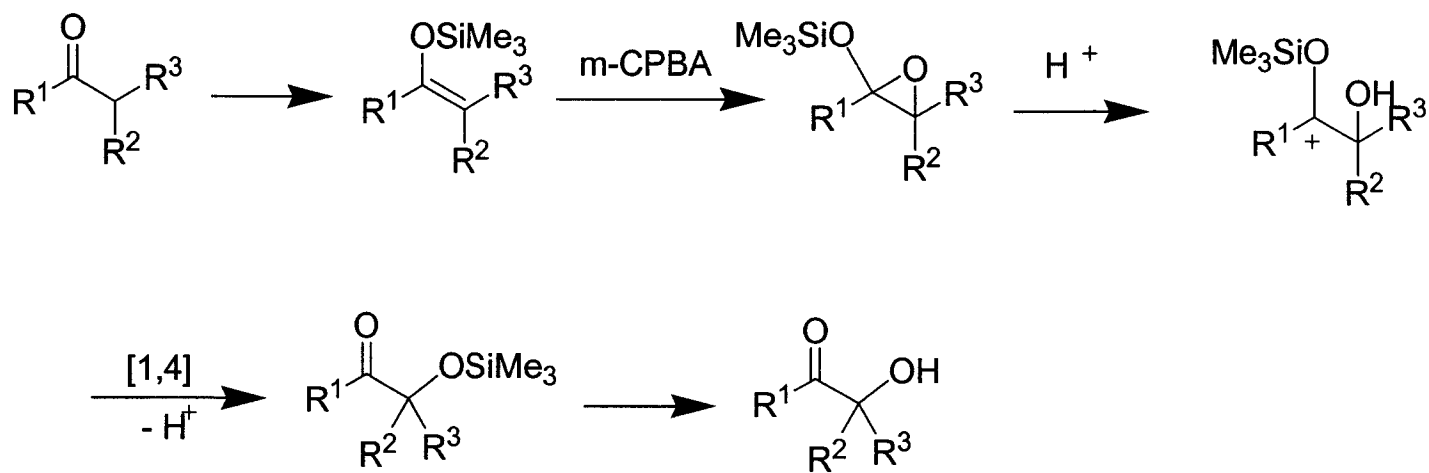


Alternative mechanism: radical chain mechanism.

Over-oxidation is possible when α -C is monosubstituted.

Chen, B.; Zhou, P.; Davis, F. A.; Ciganek, E. Organic Reactions. Vol.62.

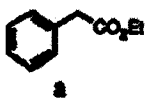
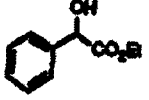
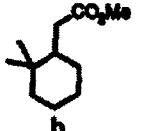
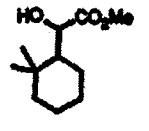
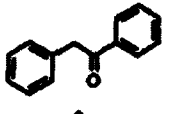
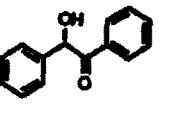


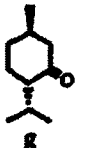
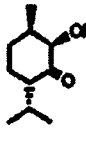
Peroxy Reagents



Chen, B.; Zhou, P.; Davis, F. A.; Ciganek, E. Organic Reactions. Vol.62.

Metal Oxides

Table. Comparison of MoOPH and MoOPD Oxidations of Carbonyl Compounds^a

carbonyl	α -hydroxy carbonyl	Yield % (S.M.)	
		MoOPH	MoOPD
		75 (9)	71 (11)
		63 (7)	72 (9)
		50 (14) +11% benzil	47 (25) +8% benzil
		85 (5) endo:exo 5:1	86 (7) endo:exo 2.4:1
		74 (8)	43 (20)

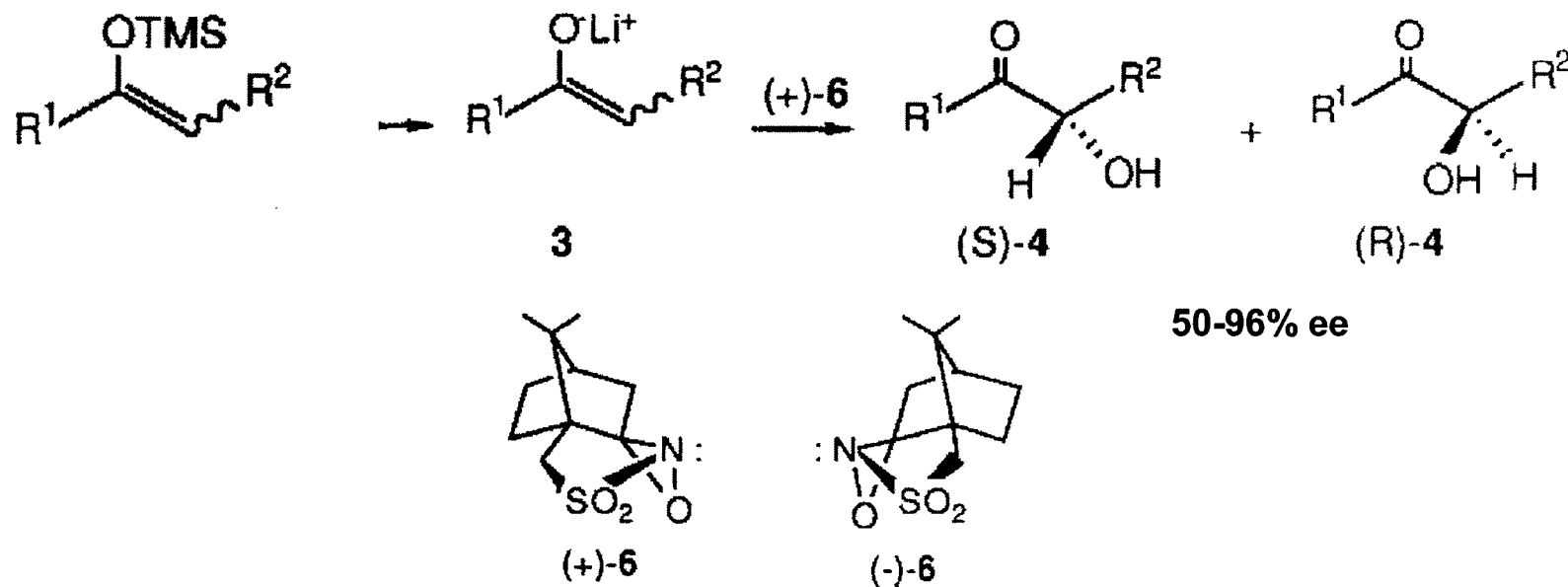
MoO₅·Py·HMPA : MoOPH

MoO₅·DMPU·H₂O: MoOPD

Anderson, J. C.; Smith, S. C. *Synlett* 1990, 107

Diastereoselectivity can be induced by stereo-directing groups or chiral auxiliaries in the enolates.

N-Sulfonyloxaziridines



(+) and (-)-(camphorylsulfonyl)oxaziridines 6

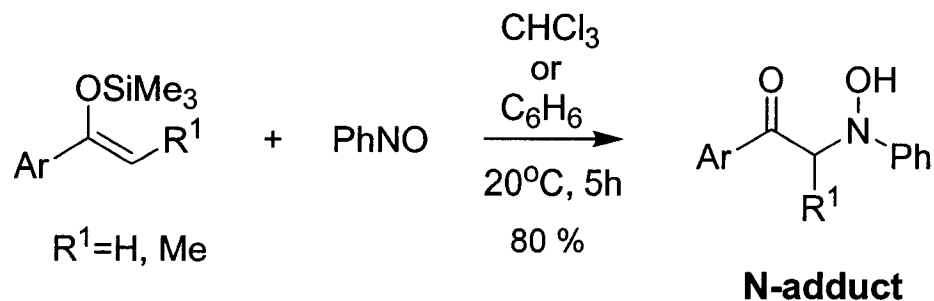
Davis, F. A.; Sheppard, A.C., B. C. ; Lai, G. S. *Tetrahedron Lett.*, **1989**, *30*, 779

Limitation:

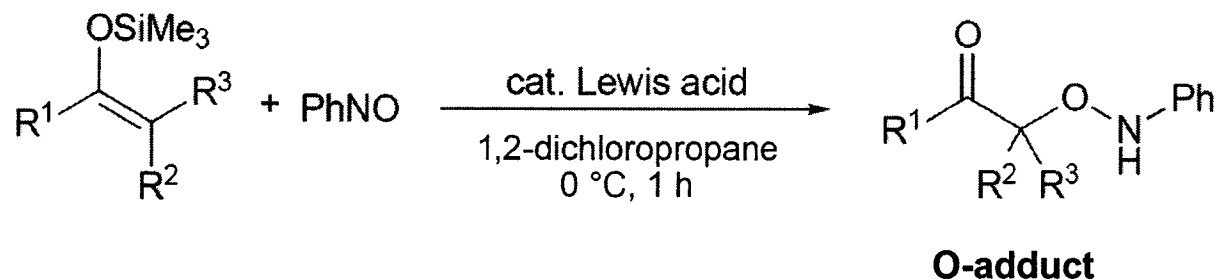
Stoichiometric enantiopure oxigent transfer reagent.

Enantioselectivity depends on structures of oxaziridines and enolates and reaction conditions.

O- vs. N-Nitroso Aldol Reaction



Sasaki, T.; Mori, K.; Ohno, M. *Synthesis* **1985**, 279.

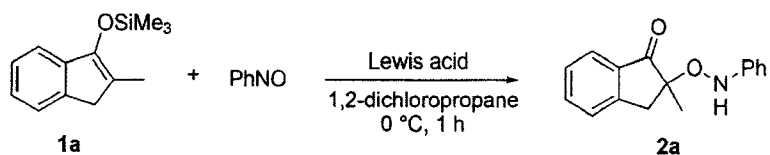


Momiyama, N.; Yamamoto, H. *Angew. Chem., Int. Ed.* **2002**, 41, 2986,3313.

High O-selectivity was obtained w/ LA catalysts and N-selectivity w/o LA.

LA catalyzed O-Nitroso Aldol Reaction

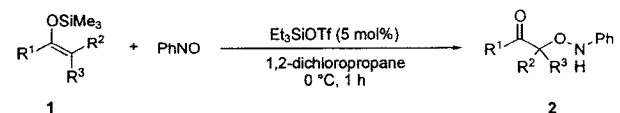
Table 1. O-Alkylation of **1a** catalyzed by various Lewis acids.



Entry	Lewis acid	Equiv [mol %]	Yield [%] ^[a]
1	none		< 1
2	Me ₃ SiOTf	5	86
3	Et ₃ SiOTf	10	94
4	Et ₃ SiOTf	5	88
5	Et ₃ SiOTf	1	74
6	<i>t</i> BuMe ₂ SiOTf	5	83
7	TiCl ₄	5	71
8	FeCl ₃	5	60
9	Me ₃ SiNTf ₂	5	54
10	[AgOTf]	5	52
11	[Cu(OTf) ₂]	5	58
12	[Sn(OTf) ₂]	5	50

Momiyama, N.; Yamamoto, H. *Angew. Chem., Int. Ed.* **2002**, *41*, 2986,3313.

Table 2. O-Alkylation of various silyl enol ethers.

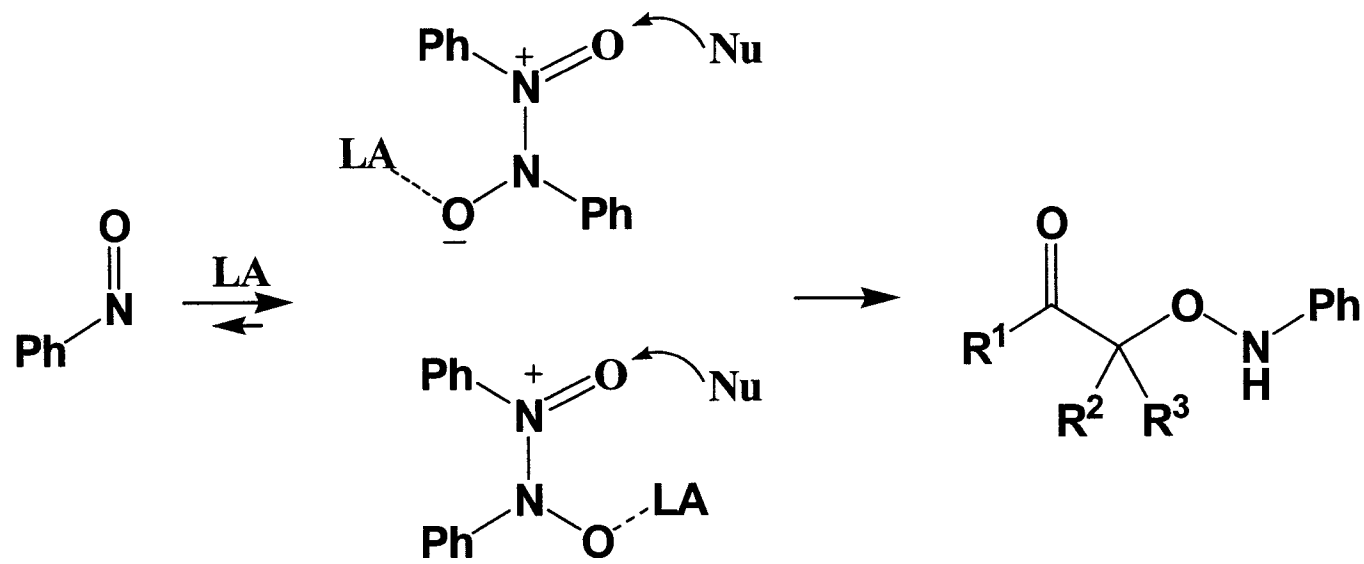


Entry	1	Yield [%] ^[a]
1		94
2		51
3		62
4		44
5		54
6		62

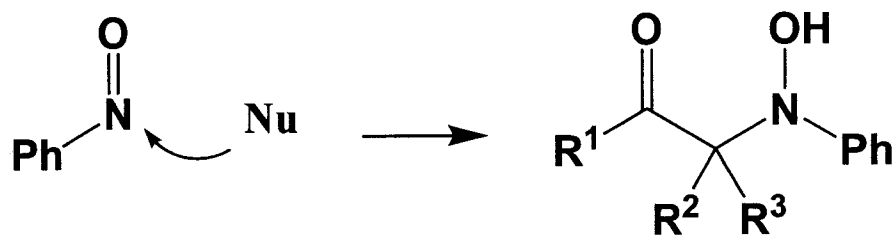
O-selectivity remained through significant structural variation.

Roles of LA in Regioselectivity

Pathway with LA Cat. : dimer



Pathway without LA Cat. : monomer



Momiyama, N.; Yamamoto, H. *Org. Lett.* **2002**, *4*, 3579.

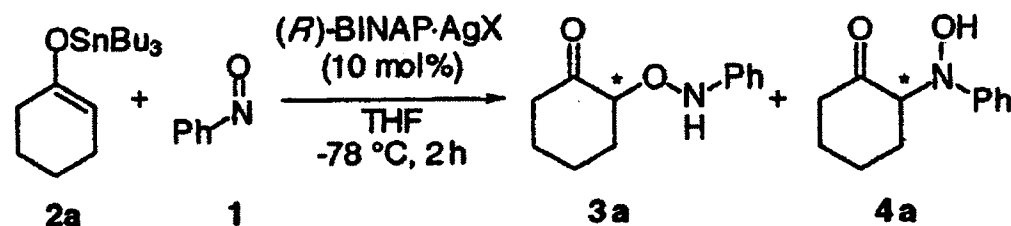
Roles of LA in Regioselectivity

Promotion of the dimerization of PhNO:

- ★ In situ IR spectra were recorded on a ReactIR 1000 instrument from ASI Applied Systems. Me_3SiOTf (1 equiv) was added to a solution of PhNO (1 equiv) in 1,2-dichloropropane; as Me_3SiOTf was added, the absorption at 1505 cm^{-1} (PhNO N=O stretch) decreased in intensity while the absorption at 1264 cm^{-1} (trans azoxybenzene N-O stretch) simultaneously increased in intensity.
- ★ ESI mass spectra of the triethylsilyl triflate-nitrosobenzene complex showed peaks at m/z 214 ($\text{Ph}_2\text{N}_2\text{O}_2^+$) and m/z 329 [$\text{Et}_3\text{Si}(\text{PhNO})_2^+$], thus confirming the presence of the dimer.

Enantioselective O-Nitroso Aldol Reaction : BINAP·AgX Catalysts

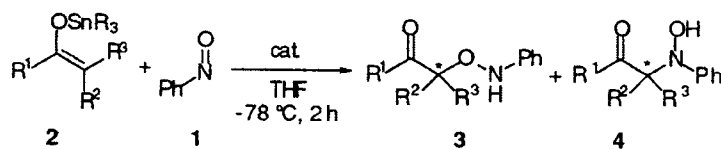
Table 1. (*R*)-BINAP–Silver Complex-Catalyzed Nitroso Aldol Reaction of Tin Enolate **2a** with Nitrosobenzene^a



entry	AgX	yield, % ^b	3a/4a	ee, % ^c	
1	AgOTf	88	>99/1	3a: 91	4a: n.d. ^d
2	AgClO ₄	95	>99/1	3a: 95	4a: n.d.
3	AgBF ₄	64	>99/1	3a: 54	4a: n.d.
4	AgNTf ₂	78	88/12	3a: 63	4a: 3
5	AgSbF ₆	85	81/19	3a: 87	4a: <1
6	AgPF ₆	86	80/20	3a: 76	4a: <1
7	AgOAc	82	72/28	3a: 87	4a: 23

BINAP·AgOTf & BINAP·AgClO₄ provide best diastereo- and enantioselectivities.

Enantioselective O-Nitroso Aldol Reaction : Tin enolate

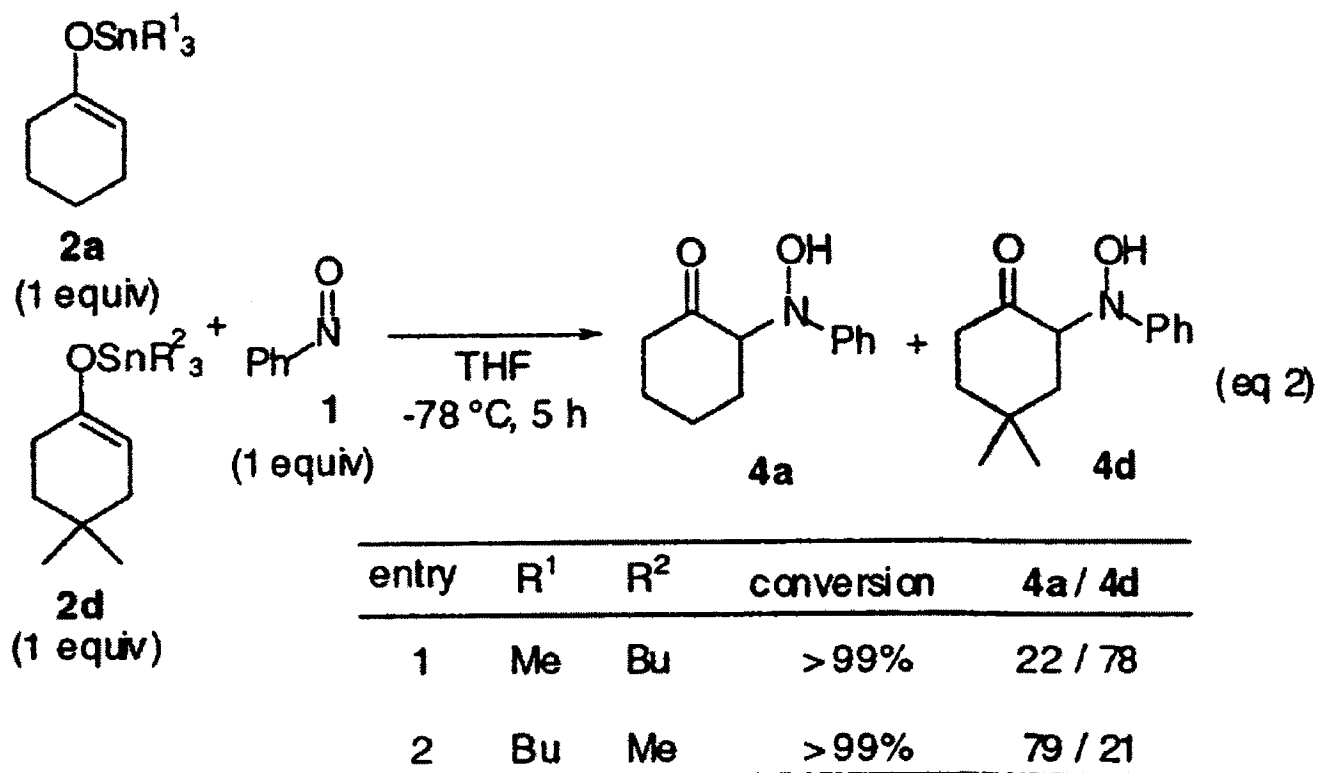


entry	enolate	cat. (mol%) ⁱ	R	yield,% ^j	3 / 4	ee,% of 3 ^k
1 ^{h, s}		A (10)	Bu	95	>99 / 1	95
2 ^{h, s}		B (10)	Bu	92	>99 / 1	91
3 ^{h, s}		C (10)	Bu	93	>99 / 1	92
4 ^{h, h}		B (10)	Me	95	>99 / 1	97
5 ^{h, h}	2a	C (10)	Me	94	>99 / 1	94
6 ^{h, h}		B (2)	Me	78	>99 / 1	96
7 ^{h, s}		A (10)	Bu	97	85 / 15	91
8 ^{h, s}		C (10)	Bu	97	83 / 17	95
9 ^{h, h}		B (10)	Bu	96	>99 / 1	95
10 ^{h, h}	2b	C (10)	Me	97	>99 / 1	88
11 ^{h, h}		B (10)	Me	94	>99 / 1	87
12 ^{h, s}		C (10)	Bu	90	66 / 34	85
	2c					
13 ^{h, h}		B (10)	Me	92	>99 / 1	90
	2d					

Catalyst A: (*R*)-BINAP·AgClO₄
Catalyst B: (*R*)-ToIBINAP·AgOTf
Catalyst C: (*R*)-ToIBINAP·AgClO₄

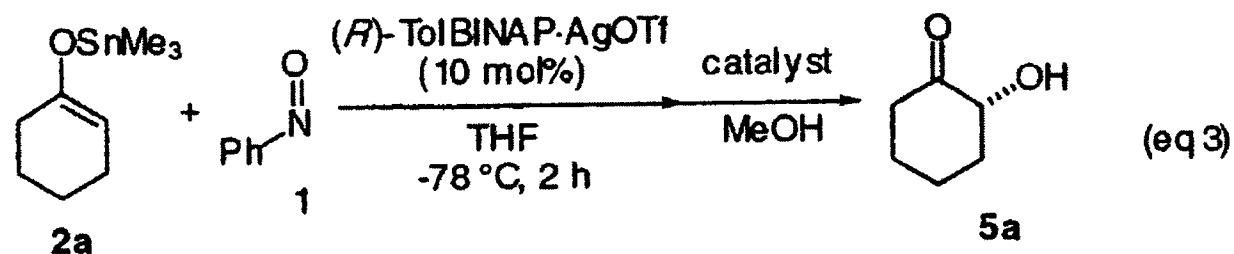
Bu₃Sn enolates have slightly increased N-selectivity.

Bu₃Sn vs. Me₃Sn Enolates



Higher reactivity of Bu₃Sn enolates resulted in more significant uncatalyzed process.

Synthesis of Enantioenriched α -Hydroxy ketones



catalyst	conditions	yield, %	ee, %
CuSO ₄ (0.3 eq)	0 °C, 12 h	93	97

Transformation to α -hydroxy ketone is smooth and facile.

Momiyama, N.; Yamamoto, H. *J. Am. Chem. Soc.* **2003**, *125*, 6038.

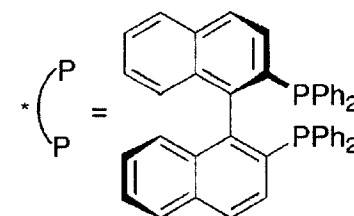
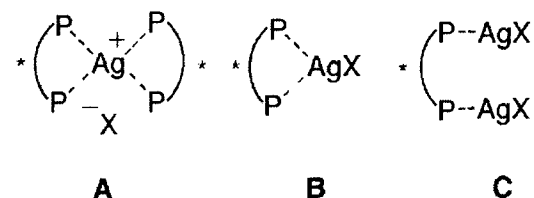
Role of LA in Enantioselectivity: Structure of BINAP·AgX Complexes

BINAP·AgX Catalysts:

1:2 (AgOTf·(R)-BINAP) complex (A)

1:1 (AgOTf·(R)-BINAP) complex (B)

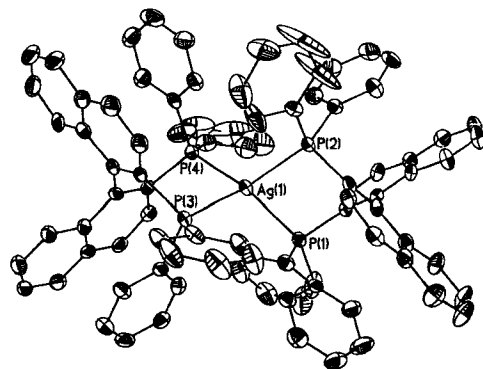
2:1 (AgOTf·(R)-BINAP) complex (C)



entry	Silver salt (eq)	A / B / C, mol.ratio
1	AgOTf (0.4)	<1 / <1 / >99
2	AgOTf (0.5)	<1 / 19 / 81
3	AgOTf (0.6)	3 / 53 / 44
4	AgOTf (0.8)	17 / 65 / 18
5	AgOTf (1.0)	21 / 63 / 16
6	AgOTf (1.5)	69 / 31 / <1
7	AgOTf (2.0)	>99 / <1 / <1
8	AgOAc (1.0)	<1 / 93 / 7
9	AgOCOCF ₃ (1.0)	29 / 71 / <1

3 complexes can be exclusively formed and separated.

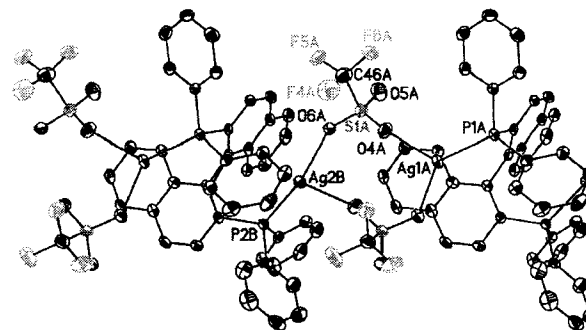
Structure of BINAP·AgX



1. 1:2 (AgOTf·(R)-BINAP) complex

Catalyst A

Figure 2. 1:1 (AgOCOCF₃·BINAP) complex



Catalyst B

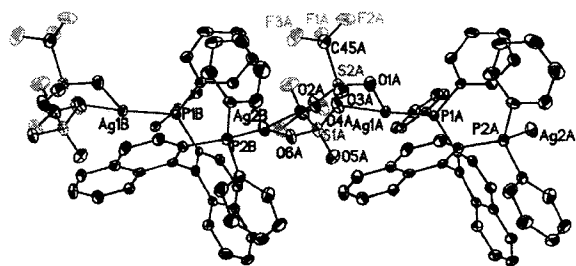
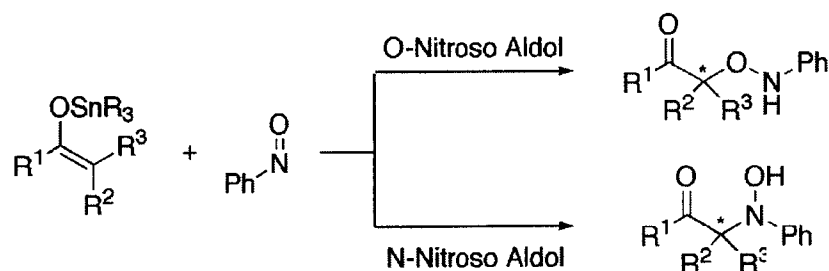


Figure 3. 2:1 (AgOTf·(R)-BINAP) complex

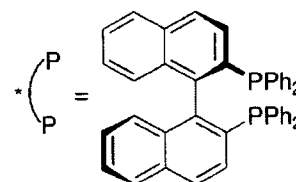
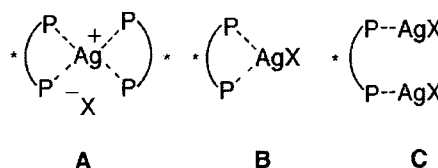
Catalyst C

Role of LA in Enantioselectivity

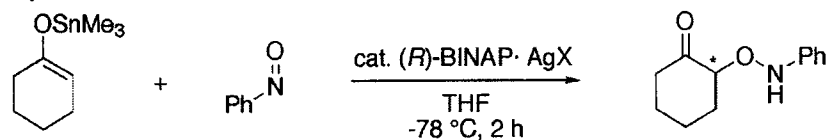


Complex A: >99% N-selective, <1% ee

Complex C: O- / N- = 95 : 5, 9% ee



Complex B: *Table 1.* 1:1 (AgX·(R)-BINAP) Complexes in O-Nitroso Aldol Synthesis^a



entry	mol %	AgX	yield, % ^b	ee, % ^c
1 ^d	10	AgOTf	88	99
2	2	AgOAc	93	97
3	2	AgOCOCF ₃	94	95

1:1 complex **B** is the responsible catalyst for catalyzed O-nitroso aldol process.

Drawback

- Stoichiometric use of toxic trialkyltin compounds.
- Aldehydes are not reported as substrates.

Proline-Catalyzed Asymmetric α -Aminoxylation of Ketones

Table 1: Proline-catalyzed direct α -aminoxylation of different ketones.¹⁴

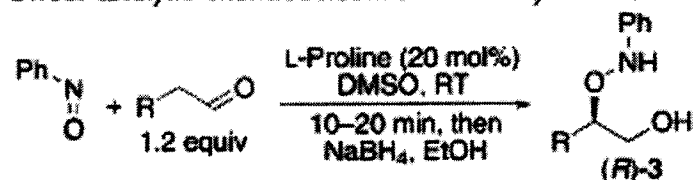
Reaction scheme: Ketone (1) + Oxime (2) $\xrightarrow[\text{DMSO, RT}]{\text{(S)-proline 20 mol\%, 2-3h}}$ Product 3 + Product 5

Entry	Ketone	R ¹	R ²	Yield [%] ¹⁴	3:5	ee [%] of 3 ¹⁴	ee [%] of 5 ¹⁴
1			-(CH ₂) ₅	70	> 100/1	> 99	-
	1a			(91) ¹⁴	(>100/1) ¹⁴	(>99) ¹⁴	-
2		H	Me	93	81:19	> 99	11
3		Et	Me	66	98:2	99	7
4		H	CH ₂ CH=CH ₂	87	8:22	> 99	n.d. ¹⁴
5		H	<i>i</i> Pr	64	90:10	> 99	n.d. ¹⁴
	1e						

A. Bøgevig, H. Sundén, A. Córdova, *Angew. Chem. Int. Ed.* **2004**, 43, 1109

Proline-Catalyzed Asymmetric α -Aminoxylation of Aldehydes

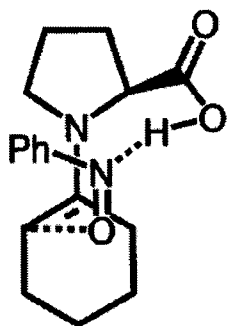
Table 1: Direct catalytic enantioselective α -aminoxylation of aldehydes.



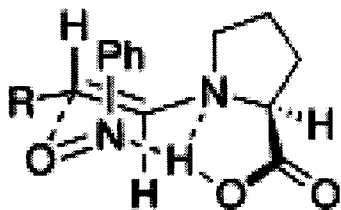
Aldehyde	Product	Yield ^(a) [%]	ee ^(b) [%]
	3a	82	99
	3b	60	97
	3c	71	99
	3d	75	99
	3e	86	99
	3f	73	99
	3g	54	99
	3h	61	94

G. Zhong, *Angew. Chem., Int. Ed.*, 2003, **42**, 4247.

Possible TS in Proline-Catalyzed Asymmetric α -Aminooxylation

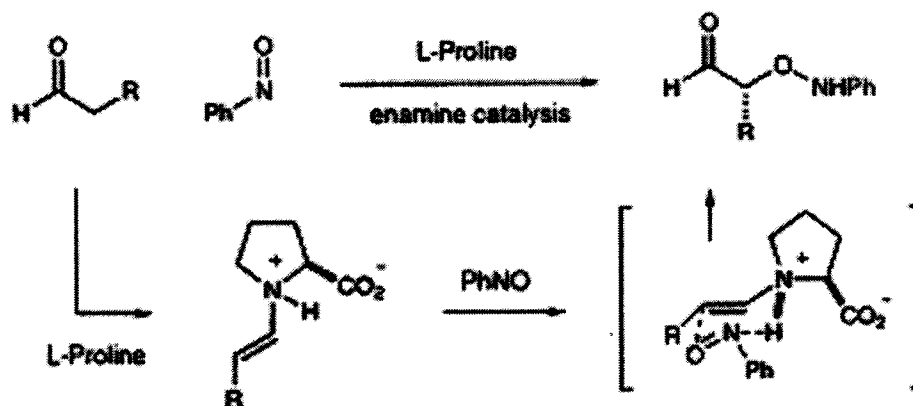


Y. Hayashi, J. Yamaguchi, K. Hibino, M. Shoji,
Tetrahedron Lett. **2003**, *44*, 8293



G. Zhong, *Angew. Chem., Int. Ed.*, **2003**, *42*, 4247.

Organocatalyzed Direct α -Oxyamination



Brown, S.P.; Brochu, M. P.; Sinz, C. J.; MacMillan, D. W. C.
J. Am. Chem. Soc. **2003**, *125*, 19808.

Si face of E-enamine is attacked by PhNO.

Conclusion

- Synthesis of α -hydroxy ketones using nitrosobenzene has been demonstrated.
- Catalytic systems with high regio- and enantioselectivity have been proved promising.