

Catalysis of 1,3-Dipolar Cycloadditions of Nitrones

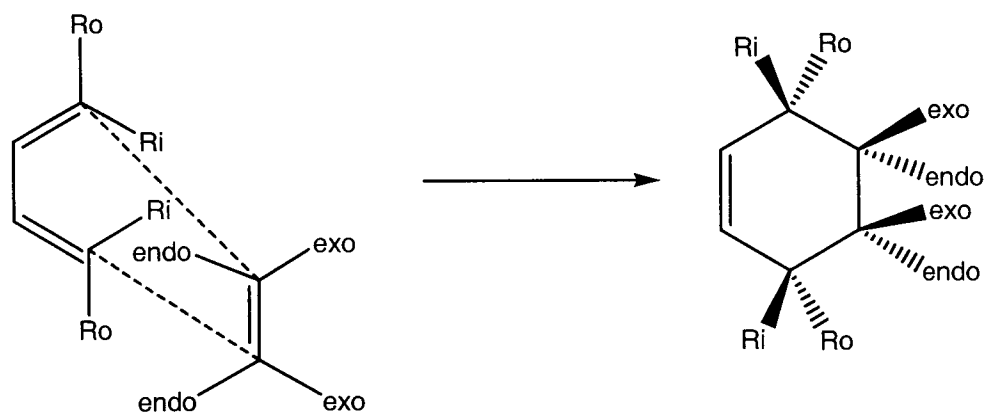
SED Group Meeting

April 6, 2004

Monica Jo Patten

Cycloadditions:

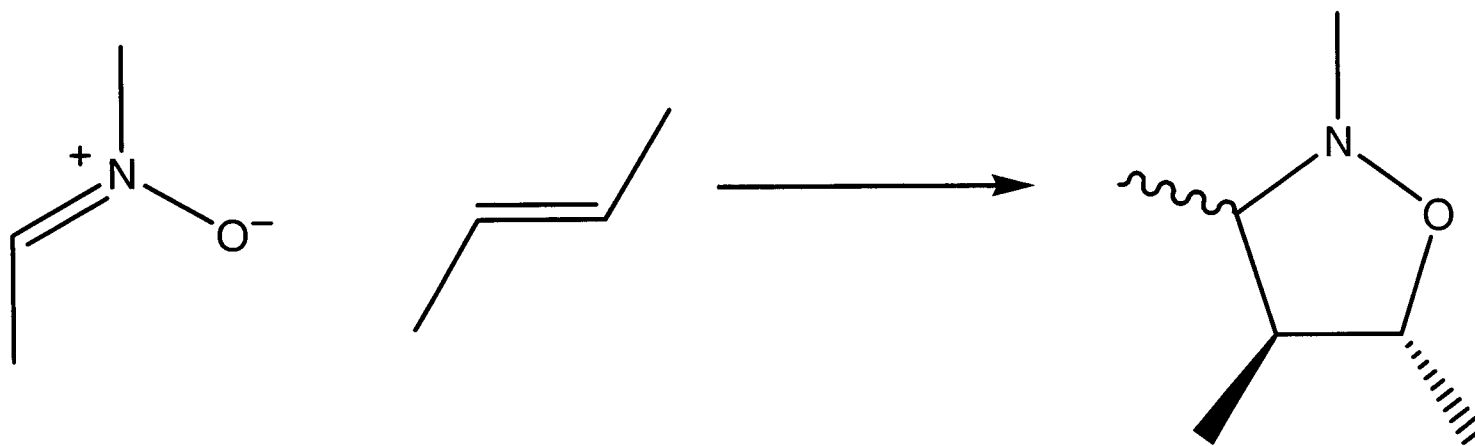
- Multiple stereocenters
- Multiple bond formation
- Functionality



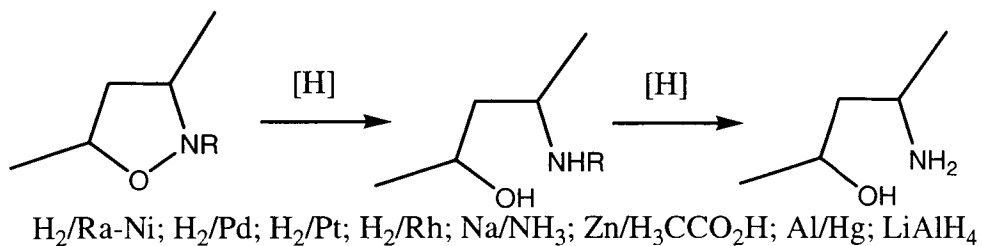
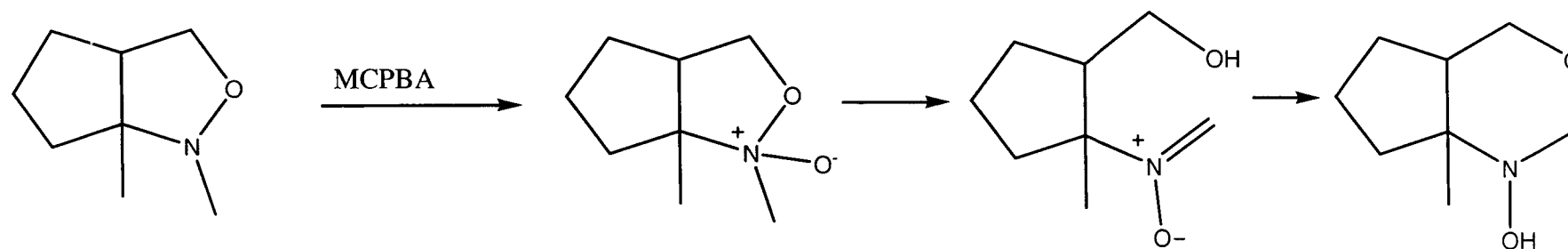
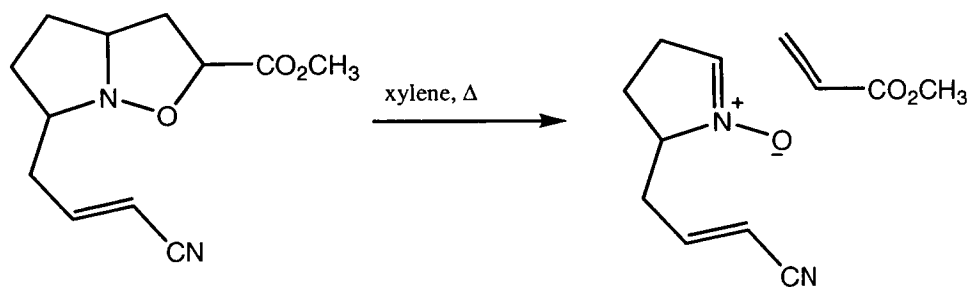
Applications of Nitron

1,3 dipolar additions

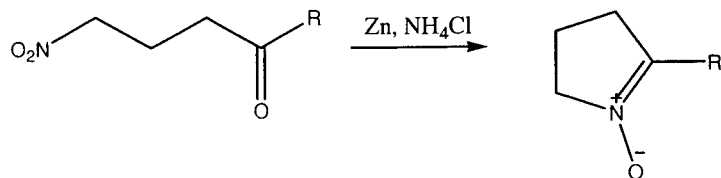
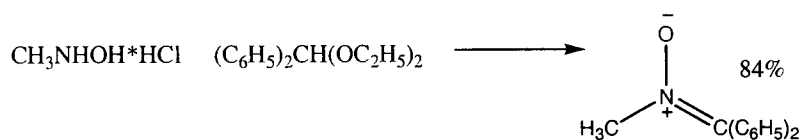
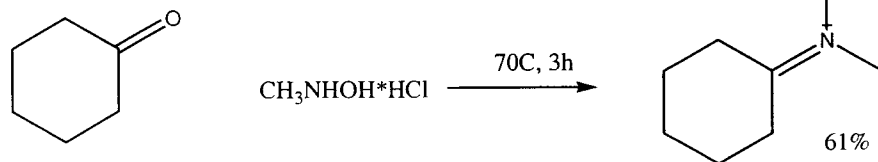
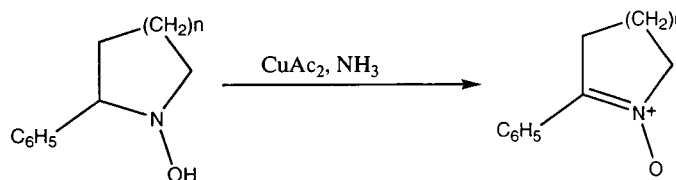
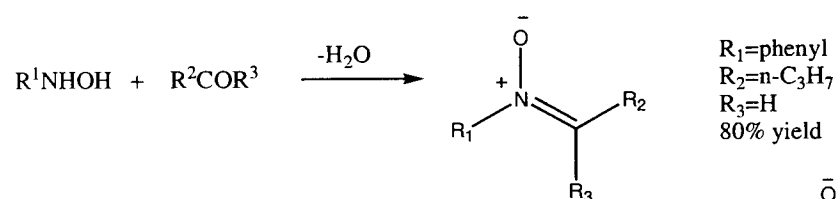
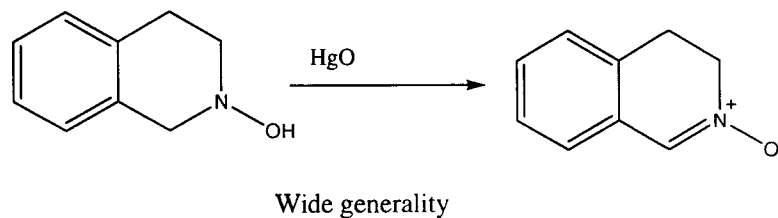
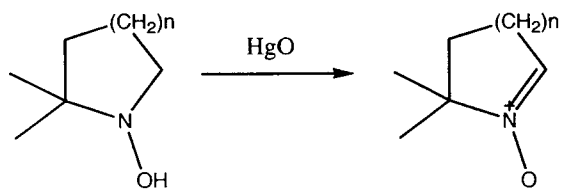
Isoxazolidine:



Masked functionality of isoxazolidines



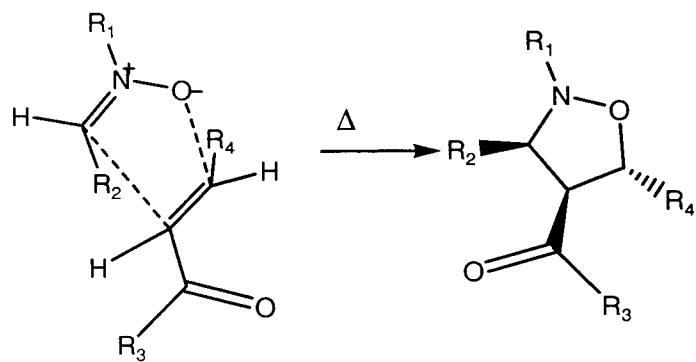
Preparation of Nitrones



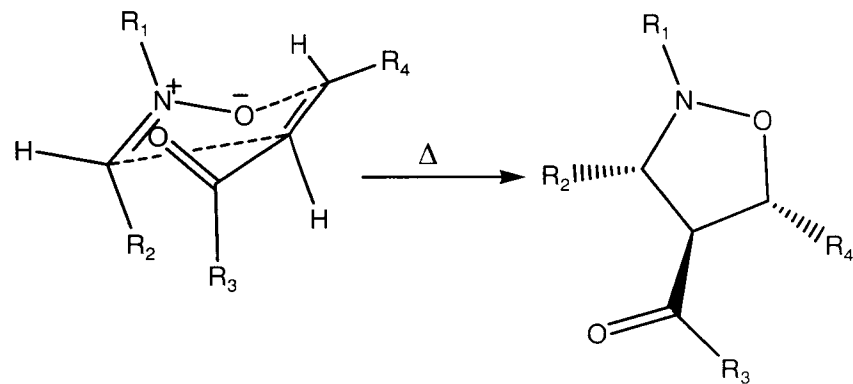
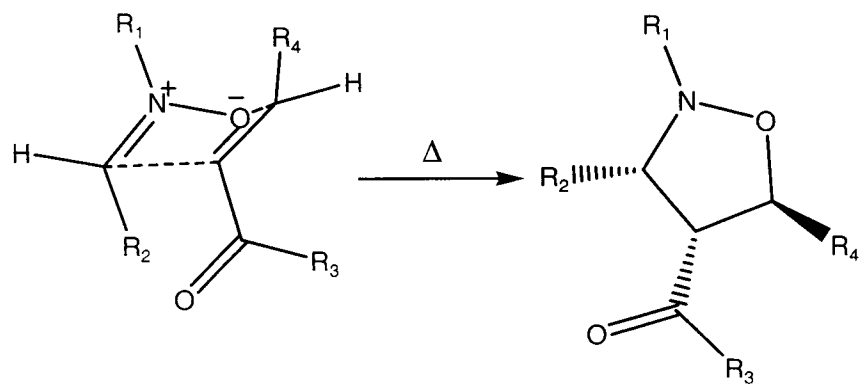
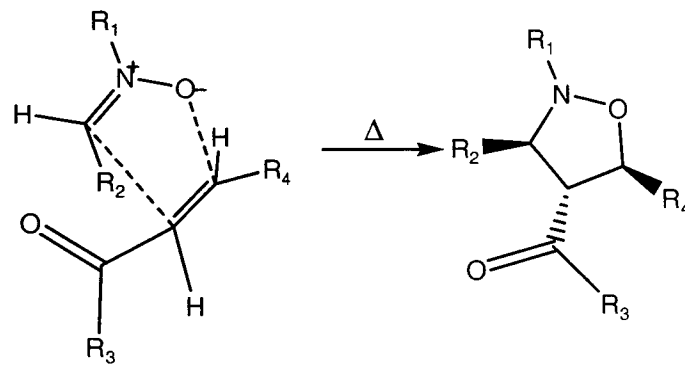
Confalone, P.; Huie, E. "The [3+2] Nitron Olefin Cycloaddition Reaction." *Org. Reactions*. Vol 36.

1,3-Dipolar Cycloaddition Stereochemistry

Exo TS:

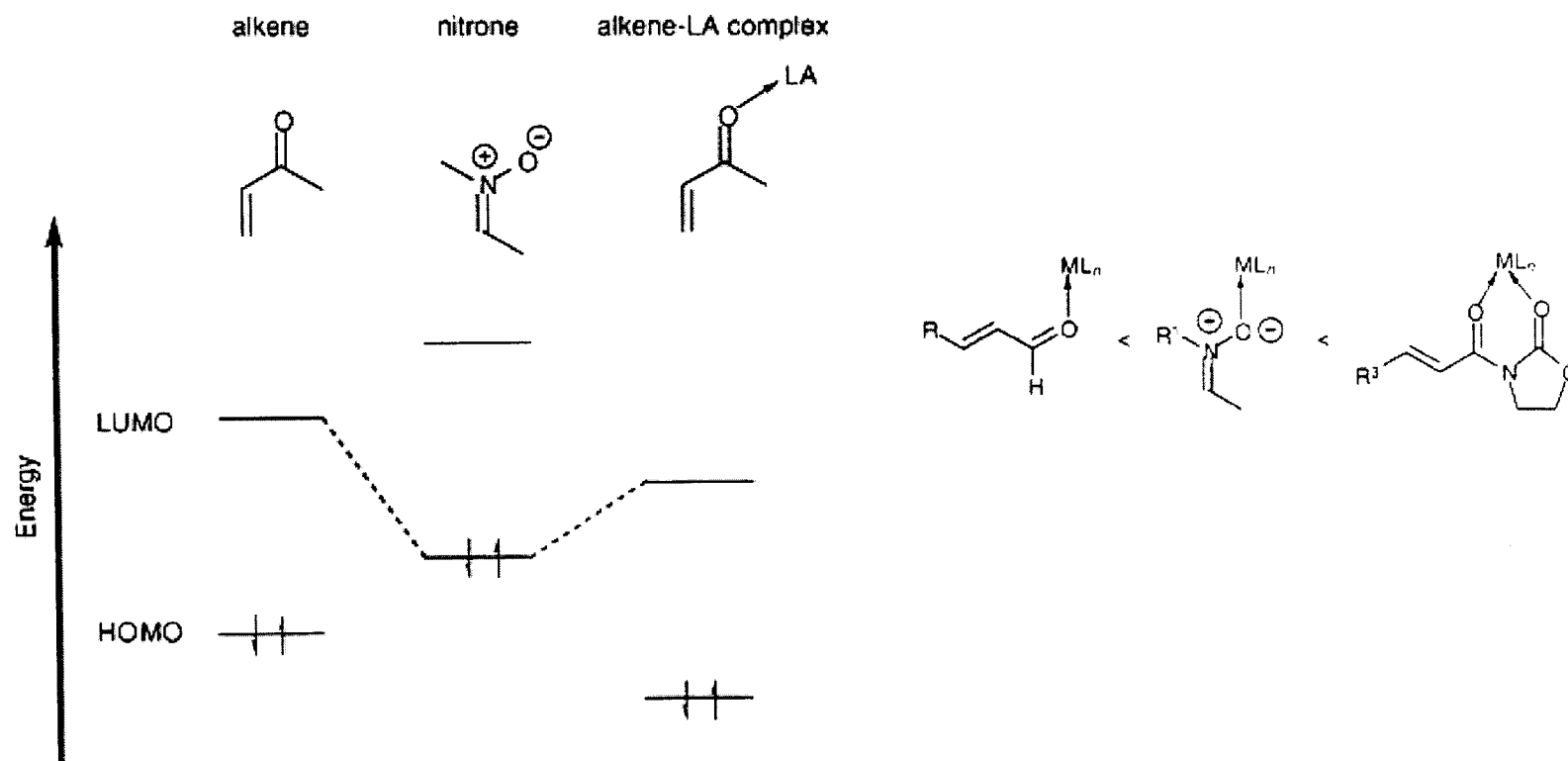


Endo TS:



1,3-Dipolar Nitrono Cycloaddition Reaction: Electronics

Normal electron-demand reactions:



Gothelkf, K.; Jorgensen, K. Chem. Commun., 2000, 1449-1458.; Gothelf, K.; Jorgensen, K. J. Org. Chem, 1994, 59, 5687-5691.

Early usage of chiral catalysts for nitron/alkene cycloadditions

Table 1. Catalytic Effects of Ti(*i*-OPr)₂Cl₂ and Dichlorotitanium Alkoxides 5a-i on the 1,3-Dipolar Cycloaddition Reaction of 1a with 2a^a

entry	catalyst (mol %)	solvent	T (°C)	convn (%)	exo/endo ratio	ee (%) exo-3a (endo-3a)
1		CHCl ₃	50	39	91:9	
2	Ti(<i>i</i> -OPr) ₂ Cl ₂ (100)	CH ₂ Cl ₂	rt	68	87:13	
3	5a (100)	toluene	rt	73	81:19	4
4	5b (100)	toluene	rt	83	90:10	8
5	5c (100)	toluene	rt	91	94:6	19
6	5d (100)	toluene	rt	85	95:5	29
7	5e (100)	CH ₂ Cl ₂	rt	90	89:11	47
8	5e (100)	CH ₂ Cl ₂	0	85	89:11	56
9	5e (100)	CH ₂ Cl ₂	-10	<10		
10	5e (10)	CH ₂ Cl ₂	0	83	89:11	51
11	5e (10)	toluene	0	94	90:10	58
12	5f (100)	CH ₂ Cl ₂	rt	81	77:23	20
13	5g (100)	CH ₂ Cl ₂	rt	78	57:43	27 (45)
14	5h (100)	CH ₂ Cl ₂	rt	72	87:13	34
15	5i (100)	toluene	rt	93	88:12	22
16	5e (10)– ZnCl ₂ (100)	toluene	rt	71	37:63	(43)

^a Reaction conditions: alkene 1a (0.1 mmol) and nitron 2a (0.11 mmol) were dissolved (1 mL solvent). The dissolved catalyst (0.5 mL solvent) was added, and the reaction was stirred for 20 h.

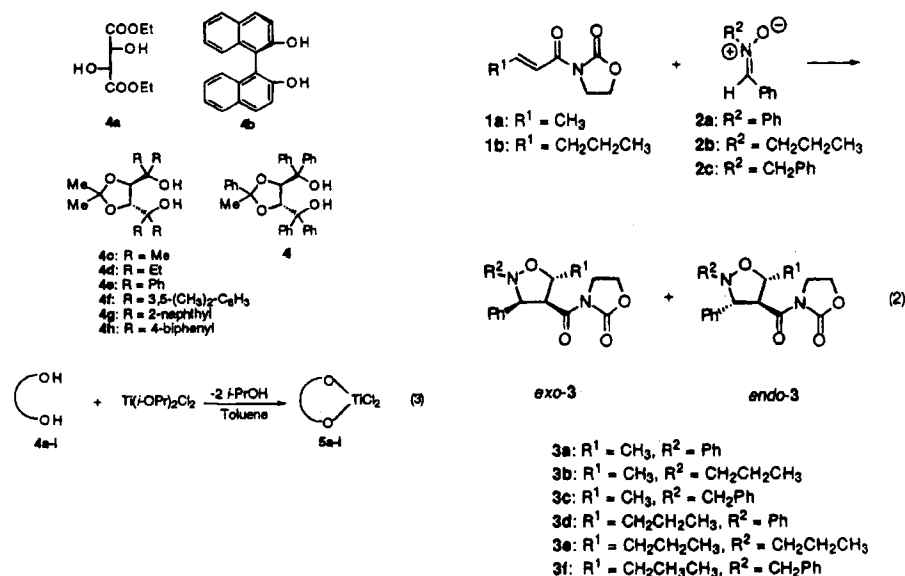
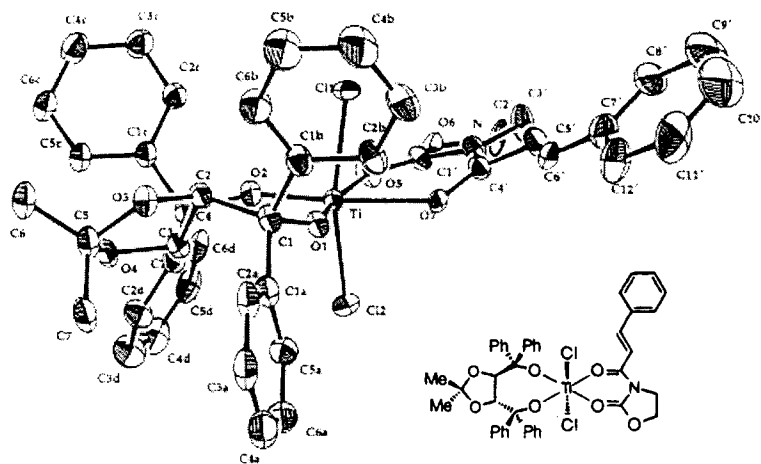


Table 2. Enantioselective 1,3-Dipolar Cycloaddition Reaction for Alkenes 1a,b with Nitrones 2a-c Catalyzed by 5e^a

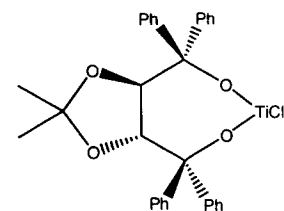
entry	alkene	nitron	T (°C)	product	yield (%) (exo/endo) ^b	ee ^c exo-3/endo-3 (%)
1	1a	2a	0	3a	94 (85:9)	60/62
2	1a	2b	rt	3b	84 (51:33)	15/62
3	1a	2c	rt	3c	71 (38:33)	27/59
4	1b	2a	rt	3d	71 (62:9)	39/48
5	1b	2b	rt	3e	80 (40:40)	10/35 ^d
6	1b	2c	rt	3f	85 (44:41)	24/53

^a Reaction conditions: see Experimental Section. ^b Isolated yields. ^c The ee was determined by ¹H NMR using Eu(hfc)₃ as chiral shift reagent. ^d Ee determined with some uncertainty.

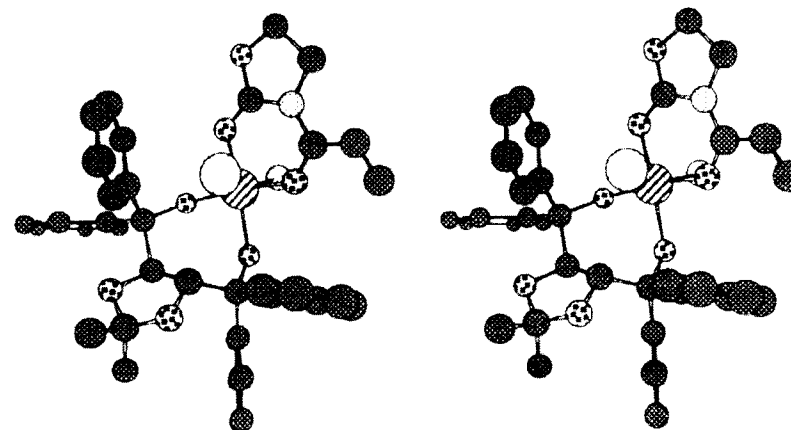
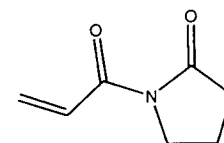
Ti-X₂ Catalysis: Catalyst Dipolarophile Complex



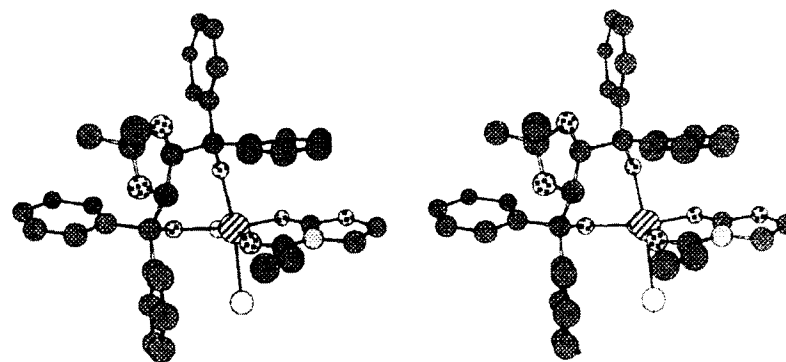
Jorgensen 1995



DiMare 1995



Symmetrical adduct



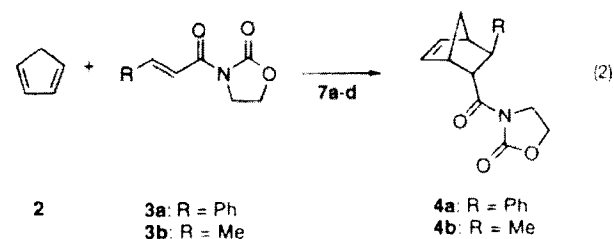
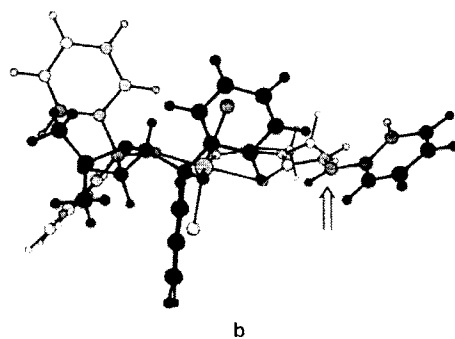
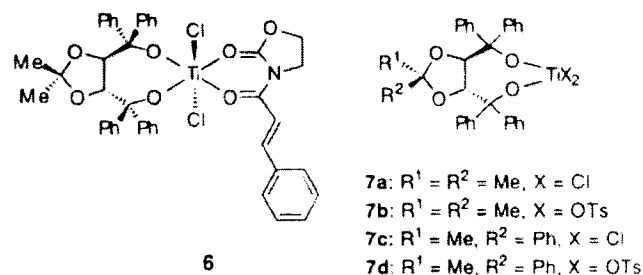
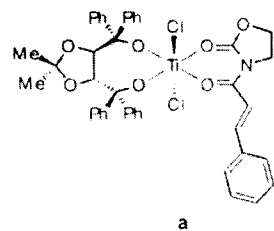
Asymmetrical adduct

Ti-X₂ Catalysis: Diels Alder

Table 1. Asymmetric Ti-TADDOLate-Catalyzed Diels-Alder Reactions of Cyclopentadiene (2) with N-Acyloxazolidinones 3a,b^a

entry	alkene	catalyst/ amount (%)	product (endo:exo) ^b	ee (%) (endo) ^c	reaction temp (°C)	endo:exo ^d	ee (%) (endo) ^d
1		6 (100)	4a (79:21)	45	0		
2	3a	6 (10)	4a (79:21)	54	-20		
3	3a	7a (10)	4a (78:22)	49	0		
4	3a	7b (50)	4a (42:58)	37	-20		
5	3a	7c (10)	4a (83:17)	60	-20	88:12 ^{3g}	64 ^{3g}
6	3a	7d (50)	4a (53:47)	31	-20		
7	3b	6 (10)	4b (86:14)	51	-20	93:7 ^e	55 ^e
8	3b	7a (10)	4b (85:15)	42	-20	83:17 ⁸	44 ⁸
9	3b	7b (50)	4b (76:34)	43	-20		
10	3b	7c (10)	4b (89:11)	86	-20	92:8 ^{3g}	91 ^{3g}
11	3b	7d (50)	4b (67:33)	21	-20	87:13 ⁸	86 ⁸

^a The reactions were run on a 0.1 mmol scale in toluene for 48–72 h. For further details, see the Experimental Section. ^b The *endo:exo* ratio was determined by ¹H NMR spectroscopy on the crude product. ^c The ee was determined by HPLC on a Diacel Chiralpak AD column. ^d Results obtained by others for identical reactions. ^e The catalyst was applied in 100 mol %, see ref 3g.



Ti-X₂ catalysis-1,3 Dipolar addition to nitrones



1a: R¹ = CH₃

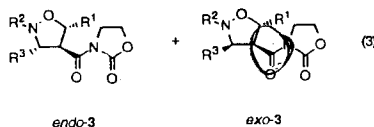
1b: R¹ = CH₂CH₂CH₃

2a: R² = R³ = Ph

2b: R² = CH₂CH₂CH₃, R³ = Ph

2c: R² = CH₂Ph, R³ = Ph

2d: R² = Ph, R³ = *p*-tolyl



3a: R¹ = CH₃, R² = Ph, R³ = Ph

3b: R¹ = CH₃, R² = CH₂CH₂CH₃, R³ = Ph

3c: R¹ = CH₃, R² = CH₂Ph, R³ = Ph

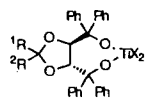
3d: R¹ = CH₃, R² = Ph, R³ = *p*-tolyl

3e: R¹ = CH₂CH₂CH₃, R² = Ph, R³ = Ph

3f: R¹ = CH₂CH₂CH₃, R² = CH₂CH₂CH₃, R³ = Ph

3g: R¹ = CH₂CH₂CH₃, R² = CH₂Ph, R³ = Ph

3h: R¹ = CH₂CH₂CH₃, R² = Ph, R³ = *p*-tolyl



5a: R¹ = R² = Me, X = Cl
 5b: R¹ = Me, R² = Ph, X = Cl
 5c: R¹ = R² = Me, X = O*i*-Pr
 5d: R¹ = R² = Me, X = Br
 5e: R¹ = R² = Me, X = OTf
 5f: R¹ = R² = Me, X = OTos
 5g: R¹ = Me, R² = Ph, X = OTos

Table 1. Catalytic Effects of TiX₂-TADDOLate Complexes on the 1,3-Dipolar Cycloaddition Reaction of Alkene **1a** with Nitron **2a**

entry	catalyst	amount (%)	conv ^a (time (h))	<i>endo:exo</i> ^d	ee (%), <i>endo</i> ^b (<i>exo</i>) ^c
1	5a	10	98 (48)	10:90	62 (60)
2	5b	100	93 (20)	12:88	— (22)
3	5c	100	0	—	—
4	5d	10	98 (20)	64:36	76 (64)
5	5e	10	73 (20)	79:21	0
6	5f	10	<10 (20)	—	—
7	5f	25	55 (48)	82:18	90
8	5f	50	99 (48)	>95:<5	93
9	5g	50	86 (48)	85:15	85

^a Conversions and *endo:exo* ratios were determined by ¹H NMR spectroscopy of the crude product. ^b The enantiomeric excess of *endo-3a* was determined by HPLC (Daicel Chiralcel OD using hexane/*i*-PrOH, 90:10). ^c The ee was determined by ¹H NMR spectroscopy using Eu(hfc)₃ as the chiral shift reagent.

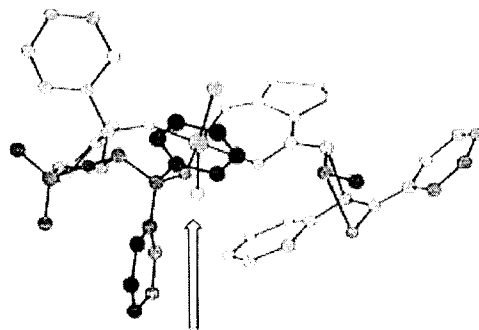


Figure 1. Proposed approach of nitron **2a** to alkene **1a** coordinated to the TiCl₂-TADDOLate catalyst **5a**, leading to *exo-3a*. The arrow shows the axial chloride ligand which is exchanged with bulkier ligands.

Table 2. Ti(OTos)₂-TADDOLate-Catalyzed Asymmetric 1,3-Dipolar Cycloaddition Reactions of Alkenes **1a,b** with Nitrones **2a-d** in the Presence of 50 mol % Catalyst

entry ^a	alkene	nitron	product	yield ^b (%)	<i>endo:exo</i> ^c	ee (%) <i>endo</i> ^d
1	1a	2a	3a	61	>95:<5	93
2	1a	2b	3b	56	>95:<5	40
3	1a	2c	3c	54	>95:<5	51
4	1a	2d	3d	71	>95:<5	91
5	1b	2a	3e	63	>95:<5	93
6	1b	2b	3f	66	>95:<5	53
7	1b	2c	3g	58	>95:<5	56
8	1b	2d	3h	55	>95:<5	92

^a The reactions were performed on a 0.5 mmol scale in toluene at 0 °C, employing 50 mol% catalyst. For further details see the Experimental Section. ^b Isolated yields. ^c The *endo:exo* ratio was determined by ¹H NMR spectroscopy. ^d The ee of the *endo*-isomer was determined by HPLC using a Daicel Chiralcel OD column.

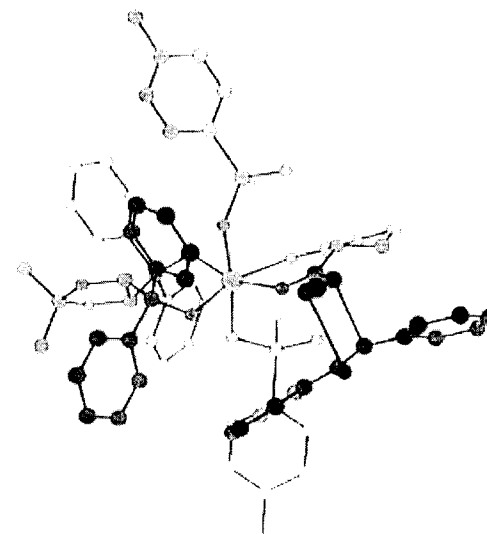


Figure 2. Proposed approach of nitron **2a** to alkene **1a** coordinated to the Ti(OTos)₂-TADDOLate catalyst **5f**, leading to *endo-3a*.

Magnesium catalysts

Table 1. Influence of Different Metal–Ligand Complexes on the Diastereoselectivity of the Reaction of 3-Crotonoyl-1,3-oxazolidin-2-one (1a) with Benzyldenephénylamine N-Oxide (2a)^a

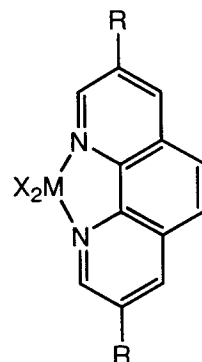
entry	catalyst ^b	conversion ^c (%)	<i>endo</i> -3a: <i>exo</i> -3a
1		<2	
2	Ti(<i>i</i> -OPr) ₂ Cl ₂	68	13:87
3	7	91	5:95
4	5a	63	81:19
5	5b^d	<2	
6	MgI ₂ –I ₂	>90	15:85
7	6a	>90	>95:<5
8	6b^d	>90	>95:<5

^a Reaction conditions: **1a** (0.1 mmol) and **2a** (0.125 mmol) were mixed in CH₂Cl₂ (2 mL) with 50 mg of 4 Å powdered molecular sieves. After the mixture was stirred for 15 min, the catalyst was added. For further details see the Experimental Section. ^b The catalysts were applied in 10 mol %. ^c Reaction time 20 h. ^d 2,9-Dimethylphenanthroline is used as ligand instead of **4a**.

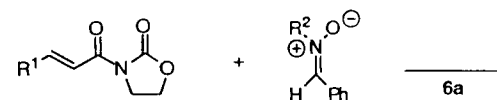
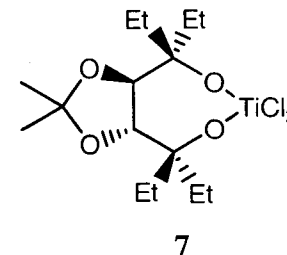
Table 2. Endo-Selective Reactions of Alkenes 1a,b with Nitrones 2a,b in the Presence of 6a as the Catalyst

entry	alkene	nitron	molar scale ^a (mmol)	product	conversion ^b (%)	<i>endo</i> : <i>exo</i> ^c
1	1a	2a	0.1	3a	>90	>95:<5
2	1a	2b	0.1	3b	>90	>95:<5
3	1b	2a	0.1	3c	>90	>95:<5
4	1b	2b	0.1	3d	64	>95:<5
5	1a	2a	0.5	3a	88 ^d	93:7
6	1a	2a	1.0	3a	75 ^d	84:16

^a The molar scale is defined from **1** which is mixed with 1.25 mol equiv of **2** in CH₂Cl₂ after the addition of 50 mg of 4 Å powdered molecular sieves per 0.1 mmol of **1**. After the mixture was stirred for 15 min, 10 mol % **6a** was added. ^b Reaction time 48 h. ^c Determined by ¹H NMR spectroscopy of the crude product. ^d Isolated yields of the *endo*-isomer.



5a M=Cu, R=H, X=OTs
5b M=Cu, R=Me, X=OTs
6a M=Mg, R=H, X=I
6b M=Mg, R=Me, X=I

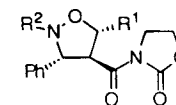


1a: R¹ = CH₃

1b: R¹ = CH₂CH₂CH₃

2a: R² = Ph

2b: R² = CH₂Ph



(4)

endo-3

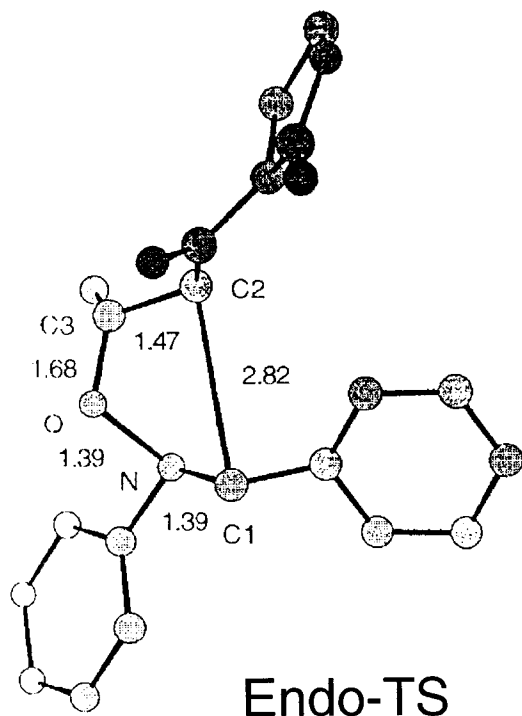
3a: R¹ = CH₃, R² = Ph

3b: R¹ = CH₃, R² = CH₂Ph

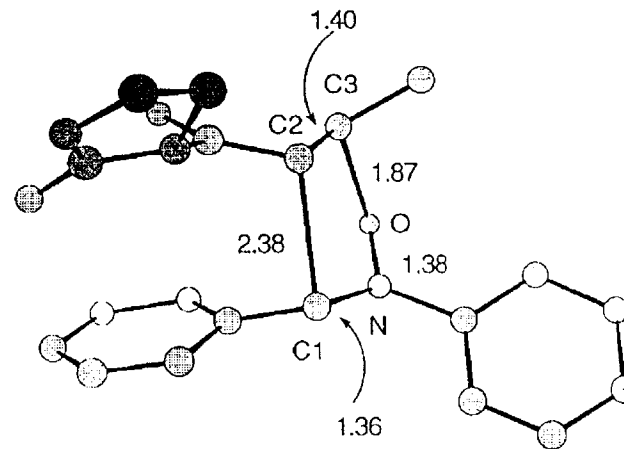
3c: R¹ = CH₂CH₂CH₃, R² = Ph

3d: R¹ = CH₂CH₂CH₃, R² = CH₂Ph

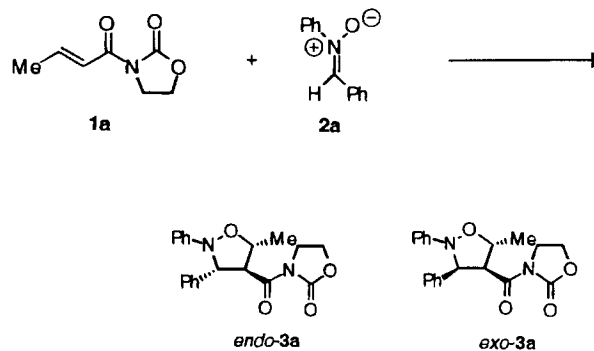
Uncatalyzed TS:



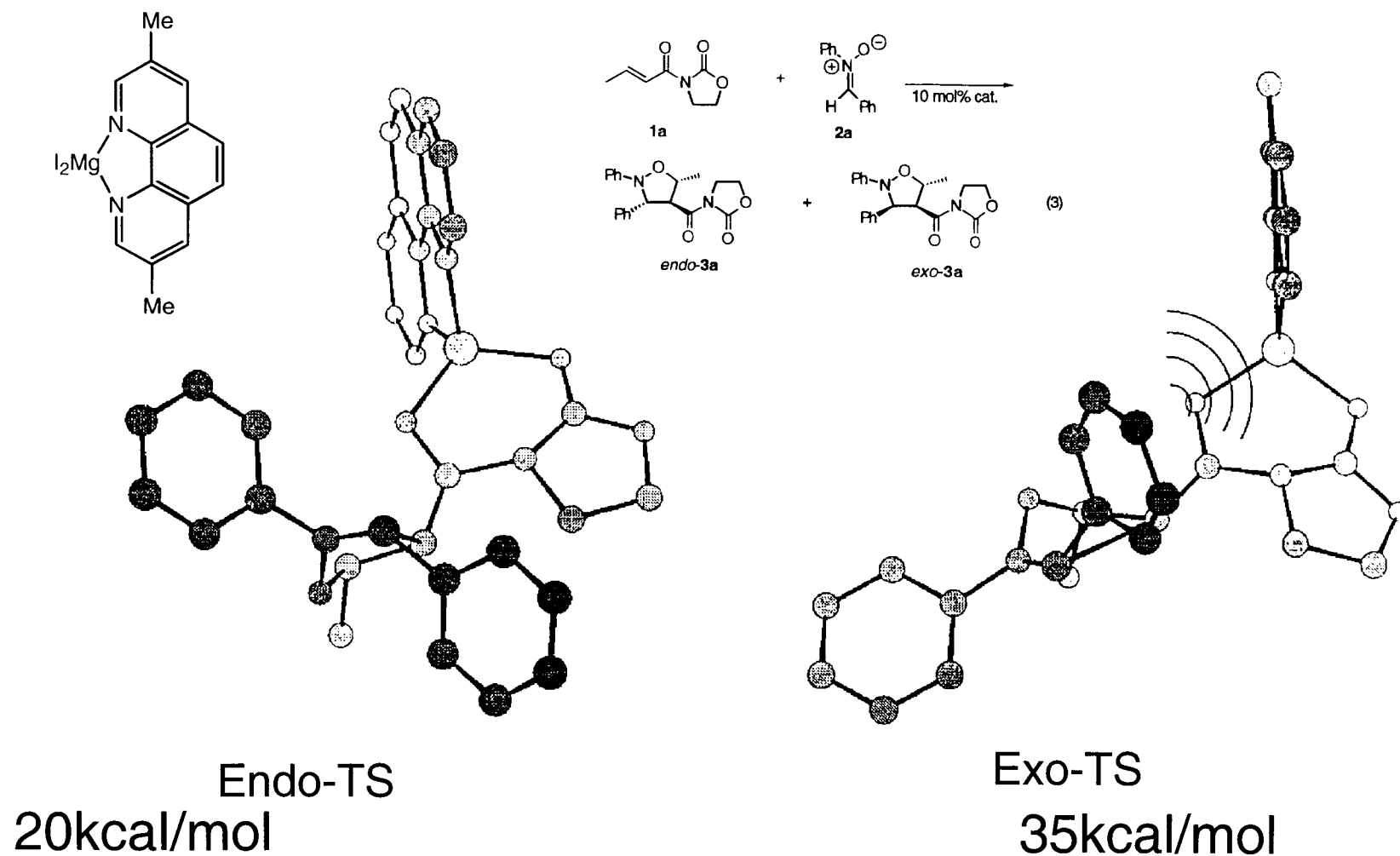
65kcal/mol



40 kcal/mol



Transition state for Magnesium Catalyst 6a



Enantioselective Mg catalyst

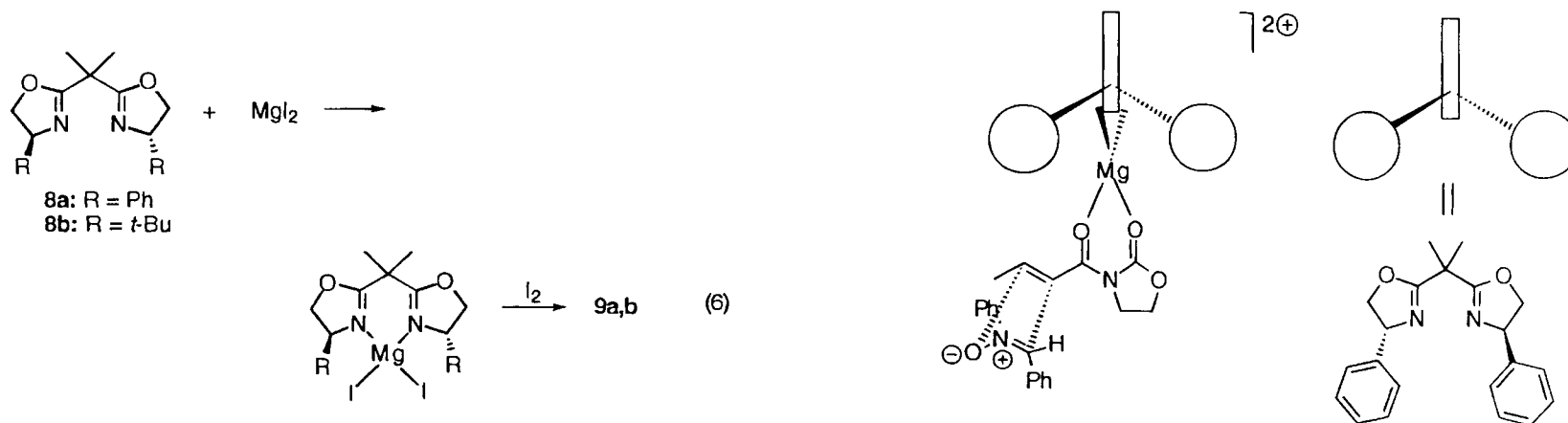


Figure 4. Proposed α -*Re* approach of nitron 2a to alkene 1a coordinated to 9a.

Table 4. Metal-Catalyzed Asymmetric 1,3-Dipolar Cycloaddition Reactions of 1a,b with 2a,b in the Presence of (*S*)-9a (10 mol %) as the Catalyst

entry	alkene	nitron	molar scale ^a (mmol)	product	conversion (%)	<i>endo:exo</i>	<i>ee endo</i> ^f (%)
1	1a	2a	0.1	3a	72/24 h	92:8	79
2	1a	2a	0.1 ^b	3a	73/24 h	65:35	<2
3	1a	2a	0.5	(+)- 3a	>95/48 h	84:16	71
4	1a	2a	0.5 ^d	(-)- 3a	>95 ^e /48 h	84:16	75 (72) ^f
5	1a	2b	0.1	3b	82/14 d	89:11	0
6	1b	2a	0.1 ^d	(-)- 3c	>95/14 d	>95:<5	82
7	1b	2a	0.5 ^d	(-)- 3c	>95 ^g /14 d	53:47	82 (52) ^f

^a The molar scale is defined from **1**. ^b Without 4 Å activated powdered molecular sieves. ^c *ee* of the *endo*-isomer was determined by HPLC using a Daicel Chiralcel OD column. ^d (*R*)-**9a** 10 mol % was applied as the catalyst. ^e Isolated yield: 81%. ^f *ee* of the *exo*-isomer. The *ee* was determined by ¹H NMR spectroscopy using Eu(hfc)₃ as chiral shift reagent. ^g Isolated yield 66%.

Inverse electron-demand reactions

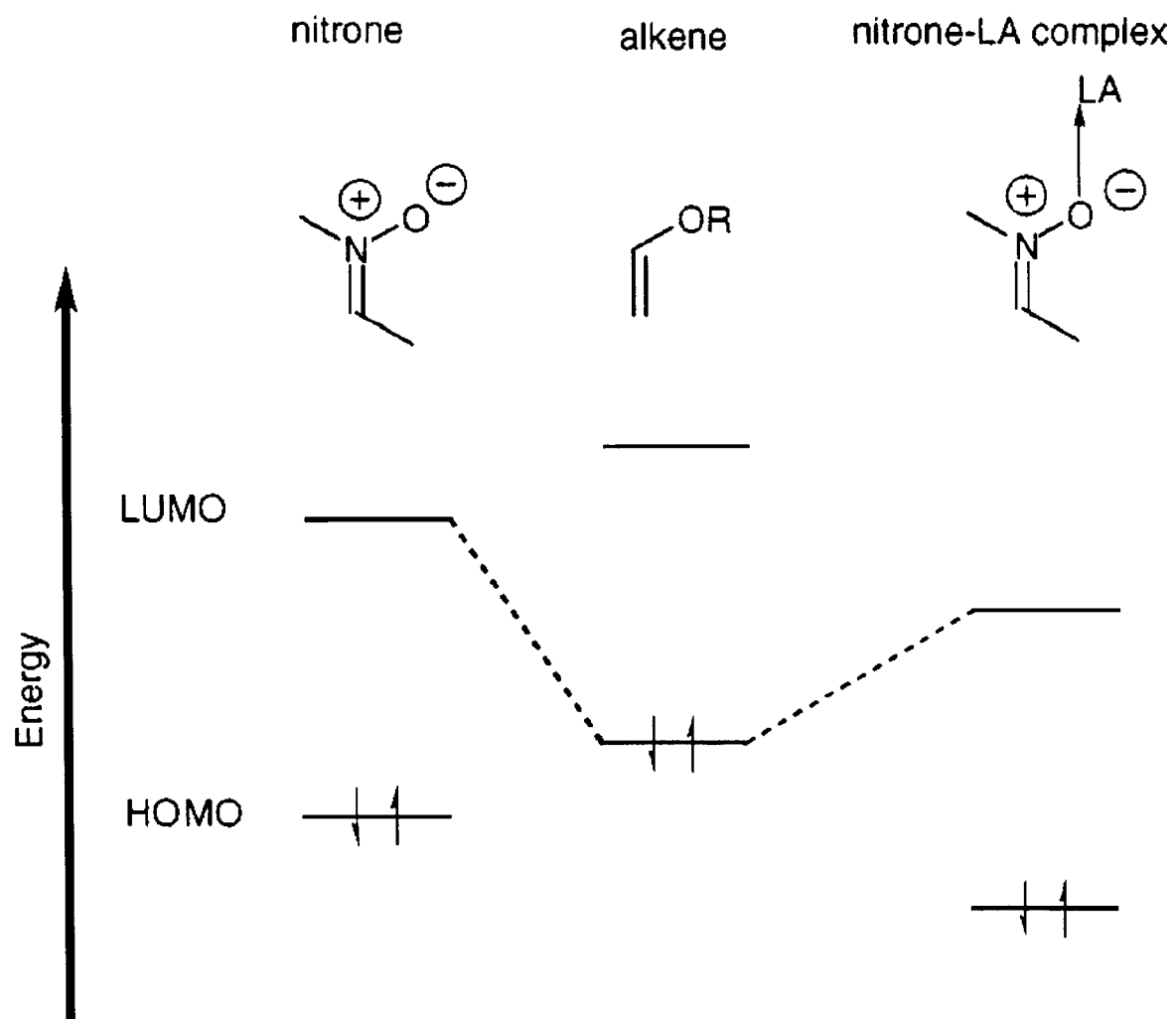
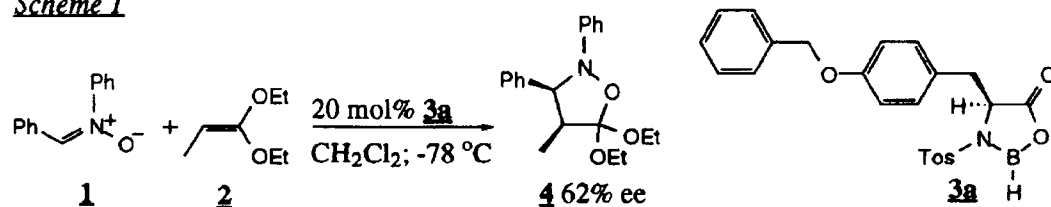


Fig. 2 The catalytic alteration of the nitron FMO's in the inverse electron-demand 1,3-dipolar cycloaddition reaction.

Boron catalysis

Scheme 1

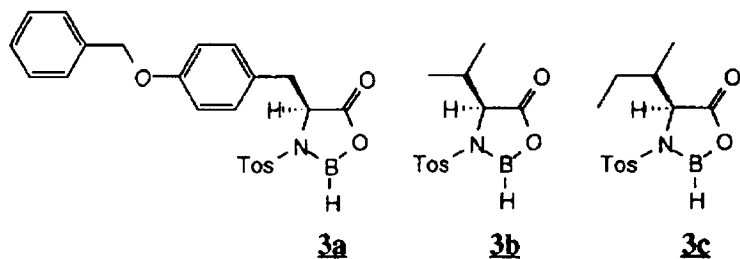


Influence of Co-solvent on Enantioselectivity of the 1,3-Dipolar Cycloaddition of Nitron **1** and Ketene Acetal **2** Catalyzed by Oxazaborolidine **3a** in CH_2Cl_2 ^a.

co-solvent ^b	% ee 4	co-solvent ^b	% ee 4
THF	62 (-) ^{3a,11}	EtCN	16 (+)
<i>t</i> BuOMe	26 (+)	EtNO ₂	43 (+)
<i>n</i> BuO <i>n</i> Bu	14 (+)	DMSO	60 (+) ^c
PhOMe	4 (+)	sulfolan	15 (+)
PhOPh	58 (+)	PhNO ₂	33 (+)
PhCH ₂ OCH ₂ Ph	79 (+) ^c	PhI	8 (+)
	33 (+)		
	71 (+) ^d		

^aReactions were run with 1.0 mmol nitron and 10 mol% oxazaborolidine **3a** (*in situ* prepared from 1M $\text{BH}_3\text{-SMe}_2$ in CH_2Cl_2), 1.5 eq. ketene acetal in 4 ml solvent at $-78\text{ }^\circ\text{C}$; ^b0.1 ml (2.5 vol %) co-solvent; ^c15 vol% co-solvent; ^d7.5 vol% co-solvent; ^e10 vol% co-solvent.

Reversal of Enantioselectivity in the Catalytic Asymmetric 1,3-Dipolar Cycloaddition of Nitron **1** and Ketene Acetal **2** Catalyzed by Oxazaborolidines **3a**.



borane	e.e. 4 (%)	e.e. 4 (%)	e.e. 4 (%)
$\text{BH}_3\text{-THF}$	62 (-)	4 (-)	0
$\text{BH}_3\text{-SMe}_2$	48 (-)	70 (+)	73 (+)

^aAll reactions were run on a 1.0 mmol nitron scale with 10 mol% oxazaborolidine **3** (*in situ* prepared from 1M $\text{BH}_3\text{-THF}$ in THF or 1M $\text{BH}_3\text{-SMe}_2$ in CH_2Cl_2) and 1.5 eq. ketene acetal in 4 ml solvent at $-78\text{ }^\circ\text{C}$.

Boron

Scheme 2

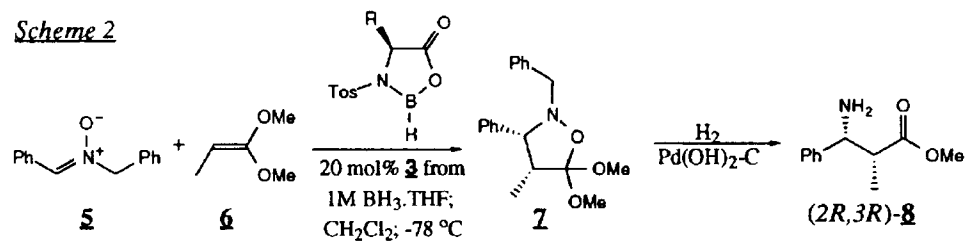


Table 3. Chiral Oxazaborolidine Catalyzed Asymmetric 1,3-Dipolar Cycloaddition of Nitron **5** in CH_2Cl_2 /THF

entry	substituent R in 3	e.e. 8 (%) ^a
1	<i>i</i> -Bu	45
2	Ph	17
3	PhCH_2	11
4	PhCH_2CH_2	59
5	4-(PhCH_2O)- PhCH_2	0
6	indolyl- CH_2	46

^a In all cases the (2*R*,3*R*)-**8** enantiomer was formed in excess

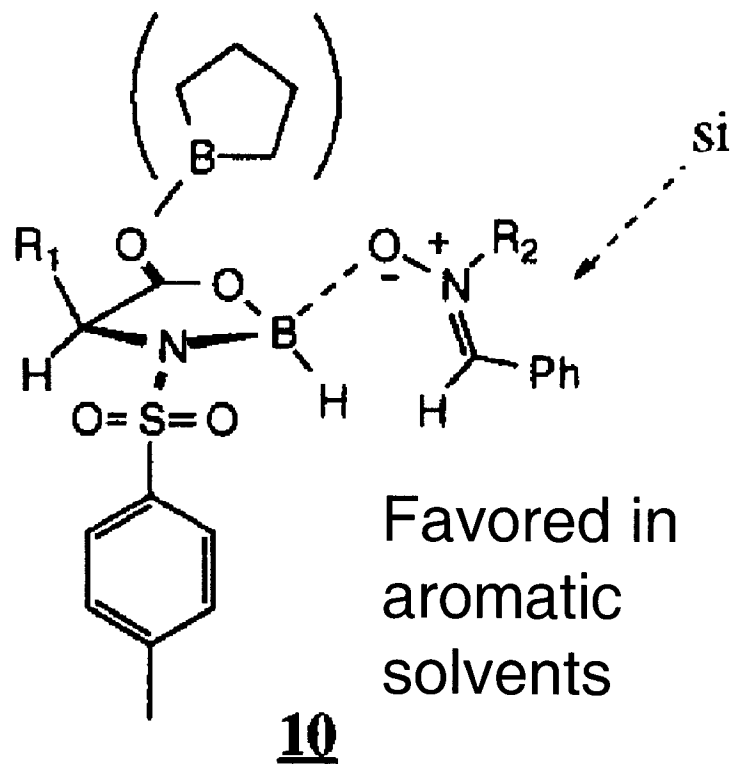
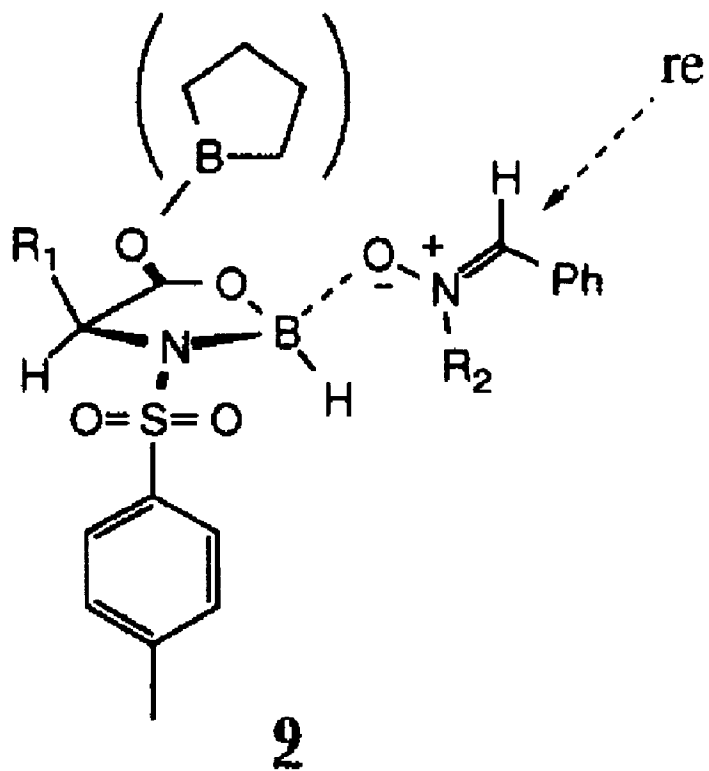
Effect of Co-solvents on Enantioselectivity of 1,3-Dipolar Cycloaddition of Nitron **5** with Ketene Acetal **6** Catalyzed by Chiral Oxazaborolidines **3**^a Prepared from 1M BH_3 - SMe_2

entry	catalyst	R	co-solvent ^a	vol%	e.e. (%) 8
a	3a	4-(PhCH_2O)- PhCH_2	-	-	8 (2 <i>R</i> ,3 <i>R</i>)
b			PhCH_3	50	18 (2 <i>S</i> ,3 <i>S</i>)
c			$\text{PhCH}_2\text{OCH}_2\text{Ph}$	2.5	34 (2 <i>S</i> ,3 <i>S</i>)
d	3b	<i>i</i> -Pr	-	-	47 (2 <i>R</i> ,3 <i>R</i>)
e			THF	100	21 (2 <i>R</i> ,3 <i>R</i>)
f			PhCH_3	50	25 (2 <i>R</i> ,3 <i>R</i>)
g	3d	PhCH_2CH_2	-	-	11 (2 <i>R</i> ,3 <i>R</i>)
h			PhH	50	15 (2 <i>S</i> ,3 <i>S</i>)
i	3e	Ph	-	-	10 (2 <i>S</i> ,3 <i>S</i>)
j			$\text{PhCH}_2\text{OCH}_2\text{Ph}$	2.5	11 (2 <i>S</i> ,3 <i>S</i>)
k			PhH	50	34 (2 <i>S</i> ,3 <i>S</i>)
l			PhCH_3	50	40 (2 <i>S</i> ,3 <i>S</i>)
m			PhCH_3	97.5	74 (2 <i>S</i> ,3 <i>S</i>)
n			PhCH_3	100 ^b	68 (2 <i>S</i> ,3 <i>S</i>)

^a dichloromethane used as standard solvent; catalyst preparation from BH_3 - SMe_2 (1M in CH_2Cl_2);

^b catalyst preparation from BH_3 - SMe_2 (1M in toluene) and reaction in toluene as solvent.

Proposed Boron TS



Summary

- Nitrene/Alkene cycloadditions can either be inverse or normal electron demand

- Enantioselectivity and diastereoselectivity can be controlled using catalysis

- Formation of isoxazolidine important for synthetic purposes:

 - Up to 3 stereocenters can be introduced

 - Masked functionality