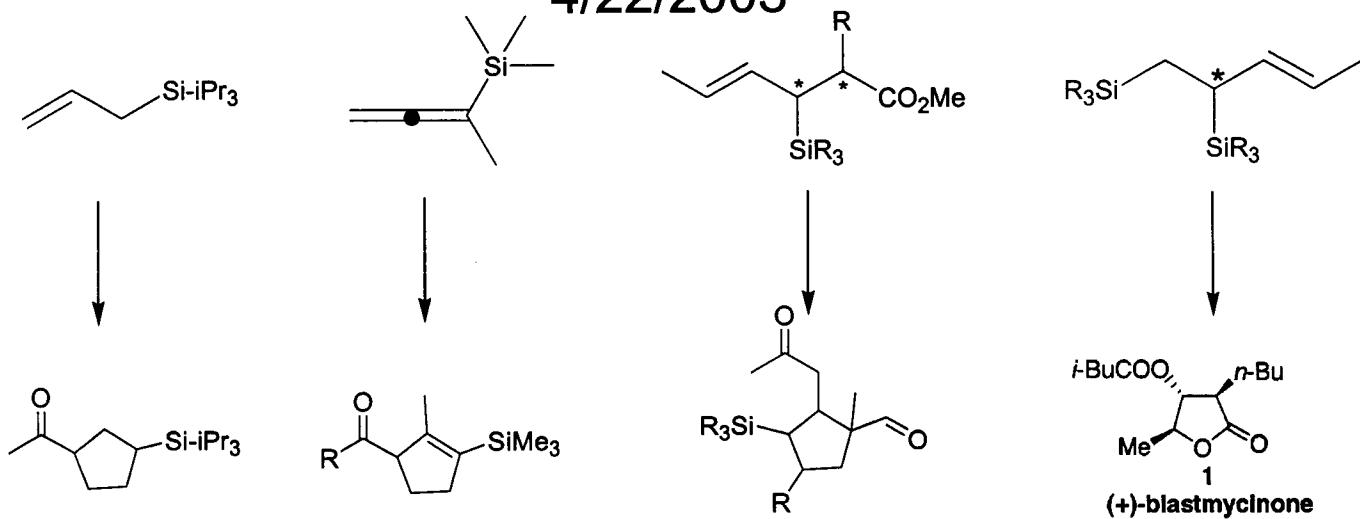


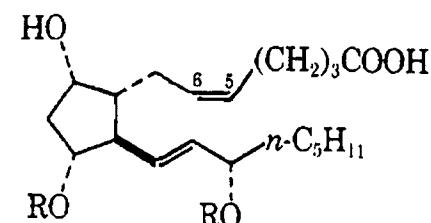
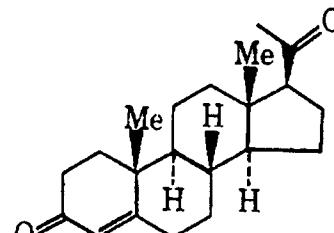
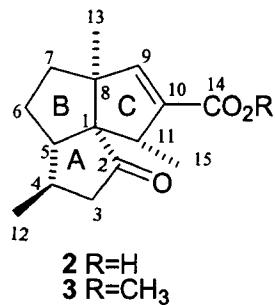
# Silicon Assisted [3+2] Cycloadditions

Adam M. Kiefer

4/22/2003



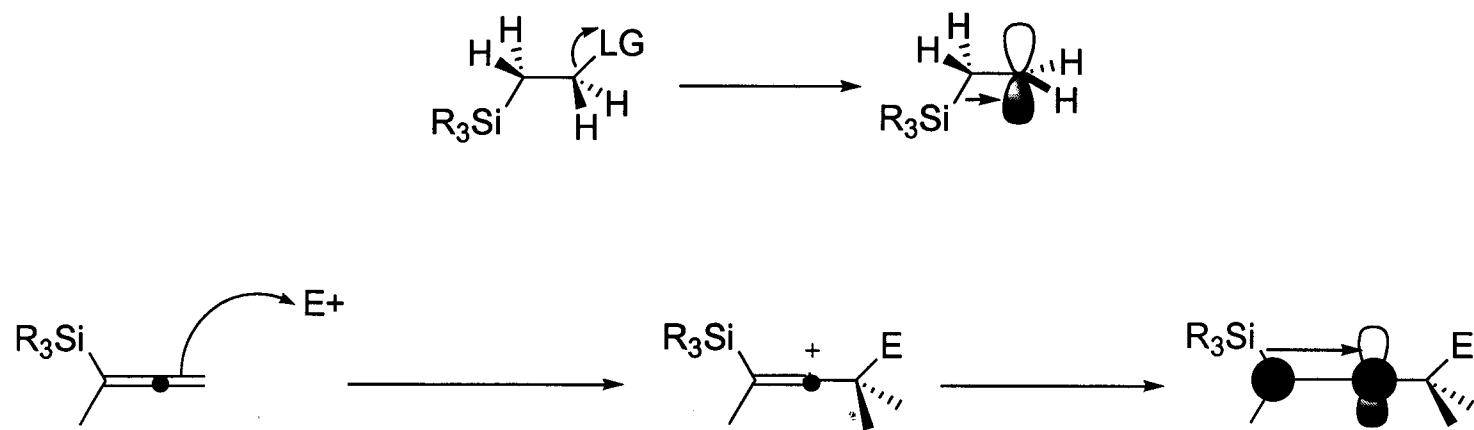
# Total Synthesis Drives New Methodology



- Discovery of biologically active natural products such as the polyquinanes, steroids, and prostaglandins has prompted many new synthetic methodologies
- By the early 1980's, much research had gone into the development of new methods to form 5-membered rings
- There was no synthetic equivalent of the Diels-Alder type reaction for the formation of 5-membered carbocycles
- In 1981 Danheiser presented a novel silicon assisted [3+2] cyclization to form cyclopentenes
- This initial discovery has led to the development of many new silicon assisted cyclizations employing silicon, all of which rely on Silicon's ability to stabilize beta cations

Dragojlovic, V. *Molecules* 2000, 5, 674  
Johnson, W. et al. *J. Am. Chem. Soc.* 1964, 93, 4332  
Corey, E.J. et al. *J. Am. Chem. Soc.* 1969, 91, 5675

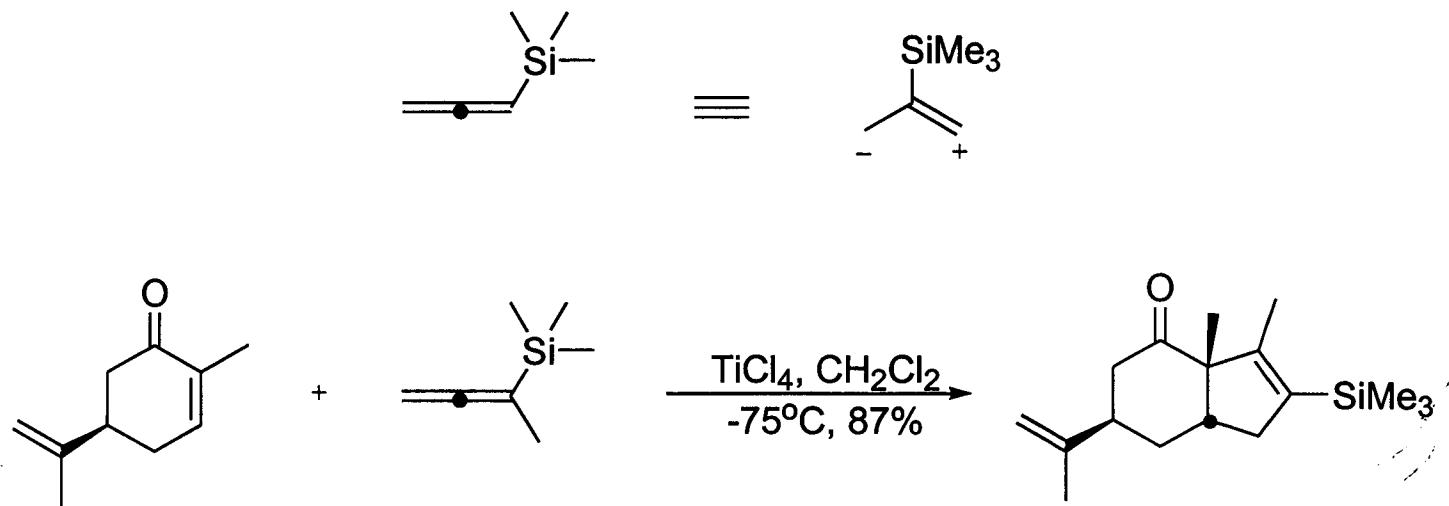
# The Beta Effect



- Silicon is a slightly electropositive atom
- Si-C bond stabilizes formed carbocation via hyperconjugation
- Molecular calculations by Jorgenson show the beta carbocation to be stabilized by 38 Kcal/mol (9Kcal/mol for methyl substituent)
- Likewise a Vinyl cation is shown to be stabilized by 29 Kcal/mol (8 Kcal/mol for methyl)

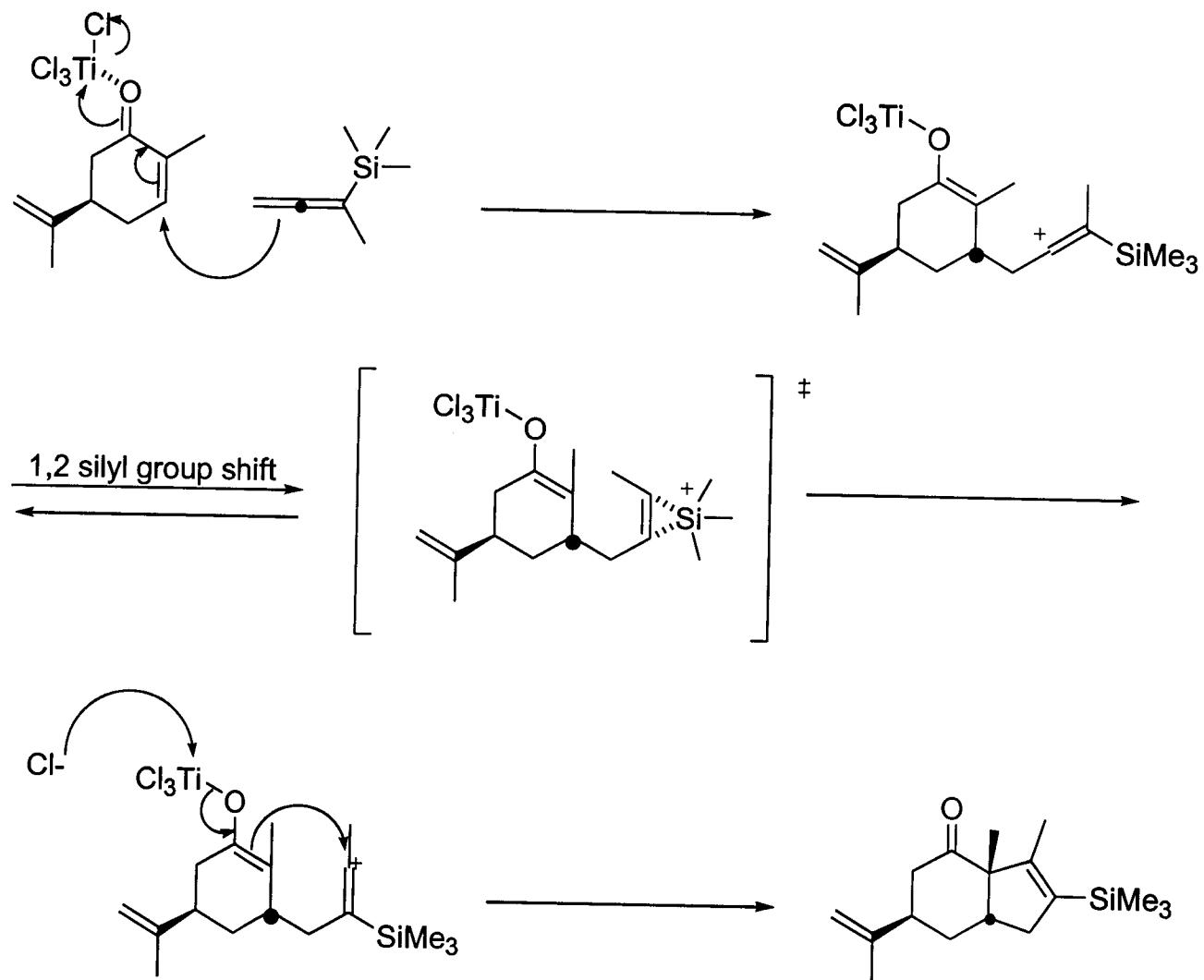
Jorgenson, W. et al. JACS. 1985, 107, 1496

# Danheiser Annulation

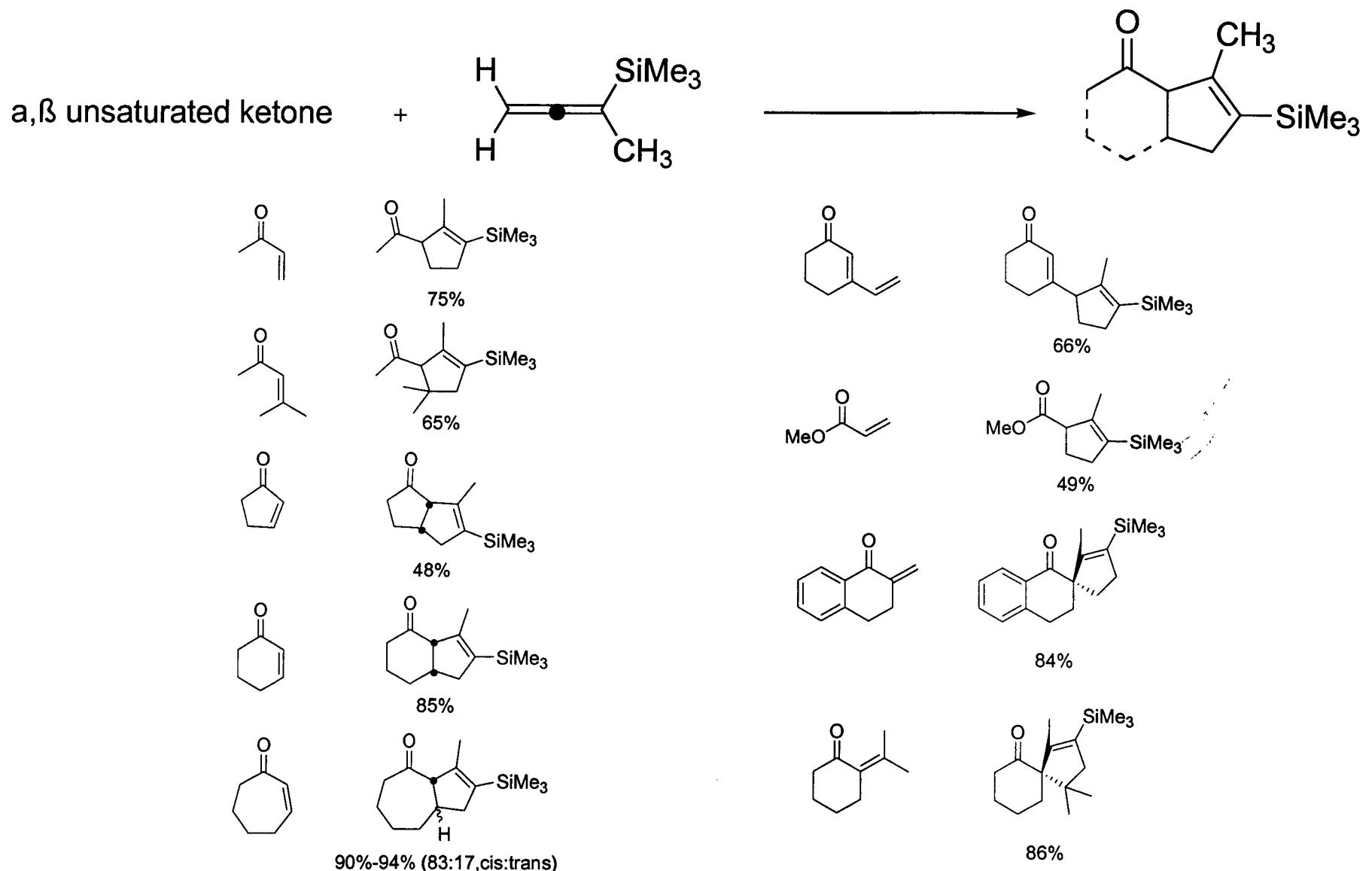


- [3+2] annulation of an  $\alpha,\beta$  unsaturated ketone and a trimethylsilyl allene
- Reaction promoted by 1.5 eq  $\text{TiCl}_4$
- Relies on trimethylsilyl moiety to stabilize  $\beta$  carbocation, as well as a rearrangement which leads to regioselective formation of the olefin

# Proposed Mechanism for Danheiser Annulation

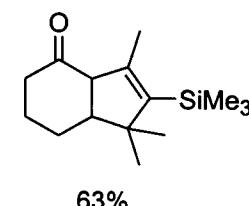
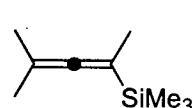
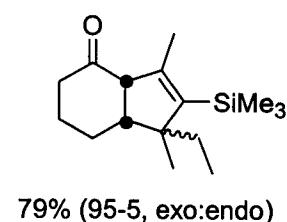
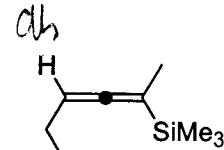
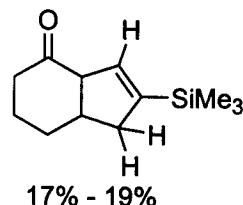
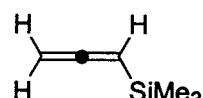
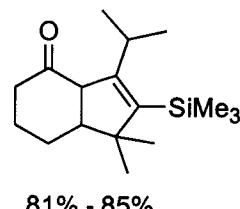
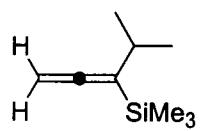
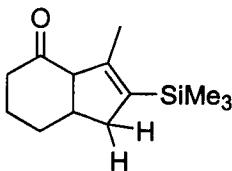
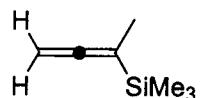
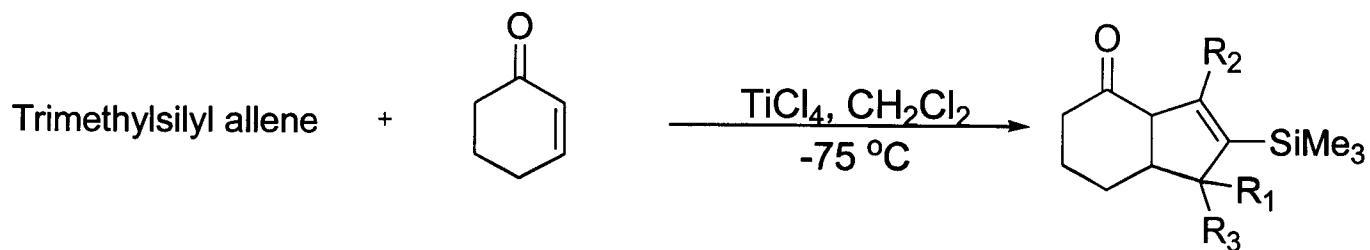


# Reactive Substrates: Allenophiles



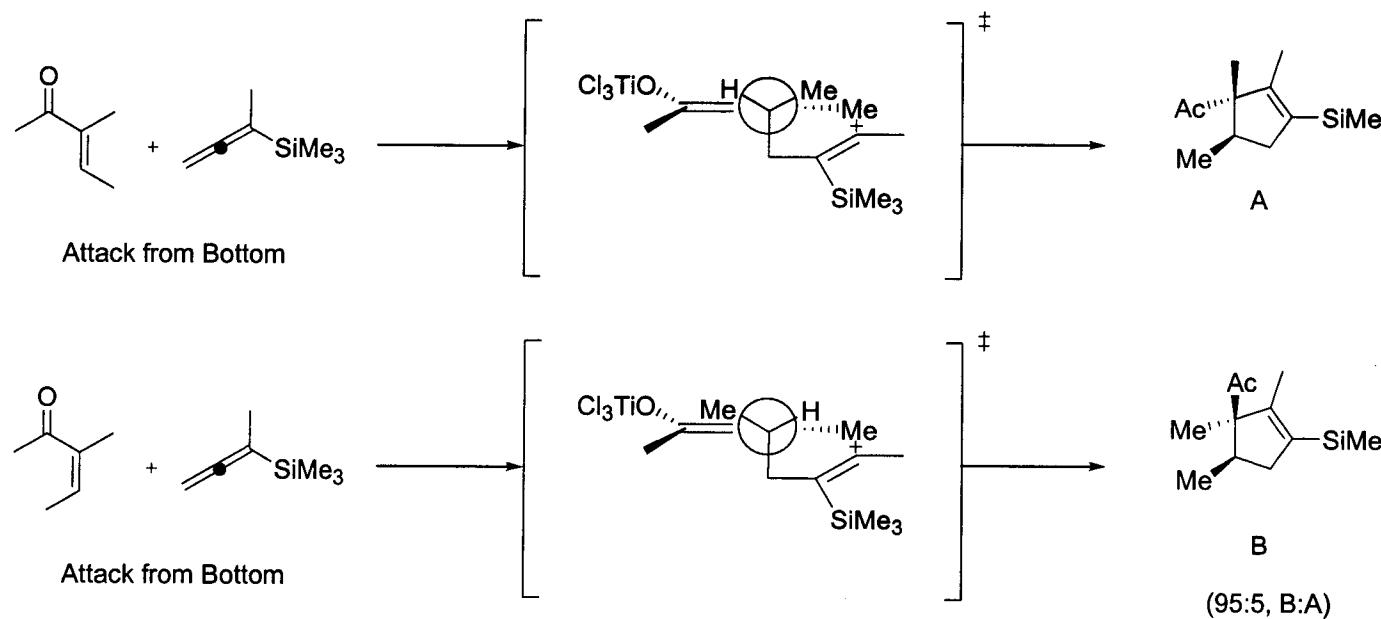
Danheiser, R.L. et al. *JACS*, 1981, 103, 1604  
Danheiser, R.L. et al. *Tetrahedron*, 1983, 39, 935

# Reactive Substrates: Allenes



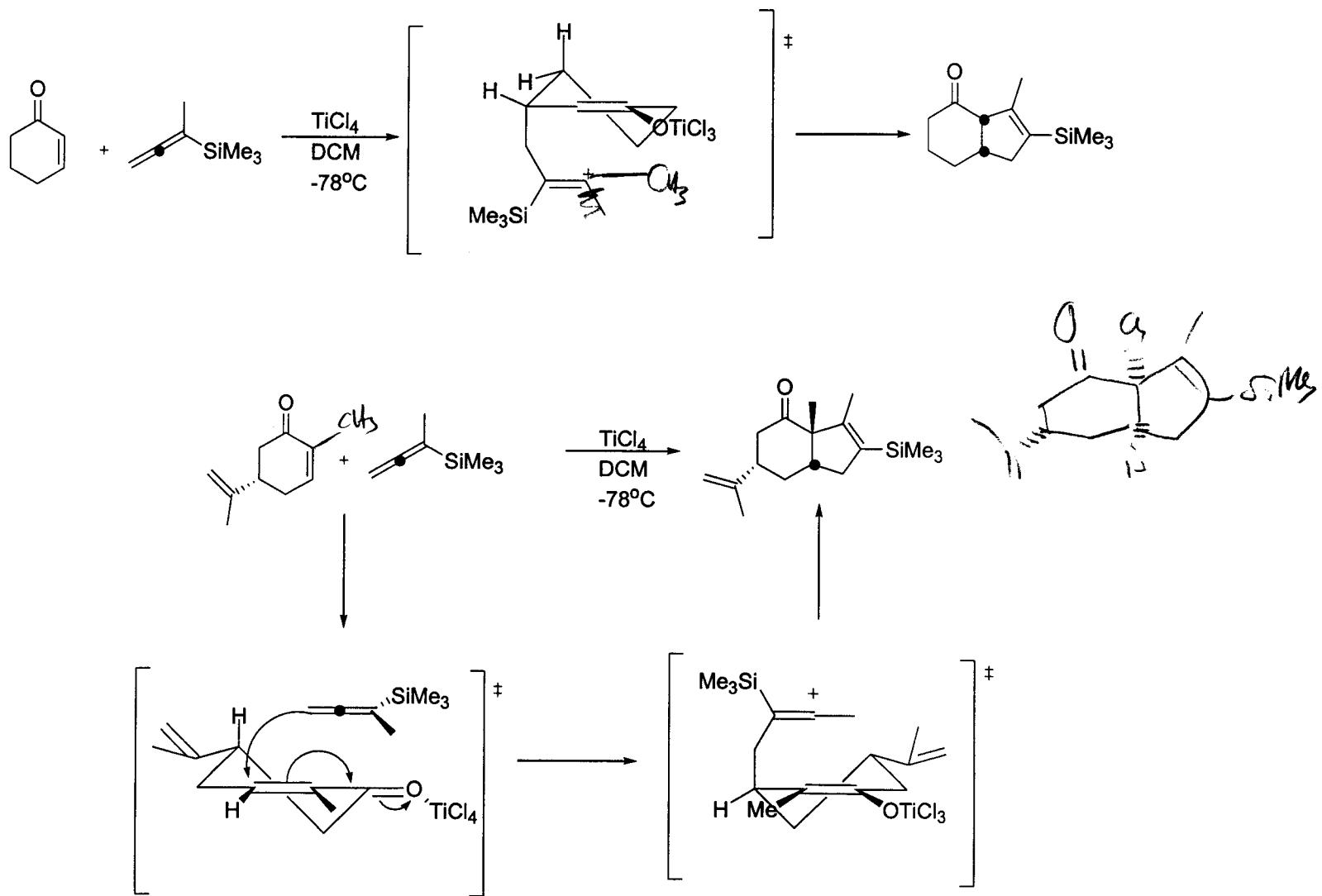
Danheiser, R.L. et al. *JACS*, 1981, 103, 1604  
Danheiser, R.L. et al. *Tetrahedron*, 1983, 39, 935

# Stereochemical Consequences...



Danheiser, R.L. et al. *JACS*, 1981, 103, 1604  
Danheiser, R.L. et al. *Tetrahedron*, 1983, 39, 935

# Stereochemical Consequences

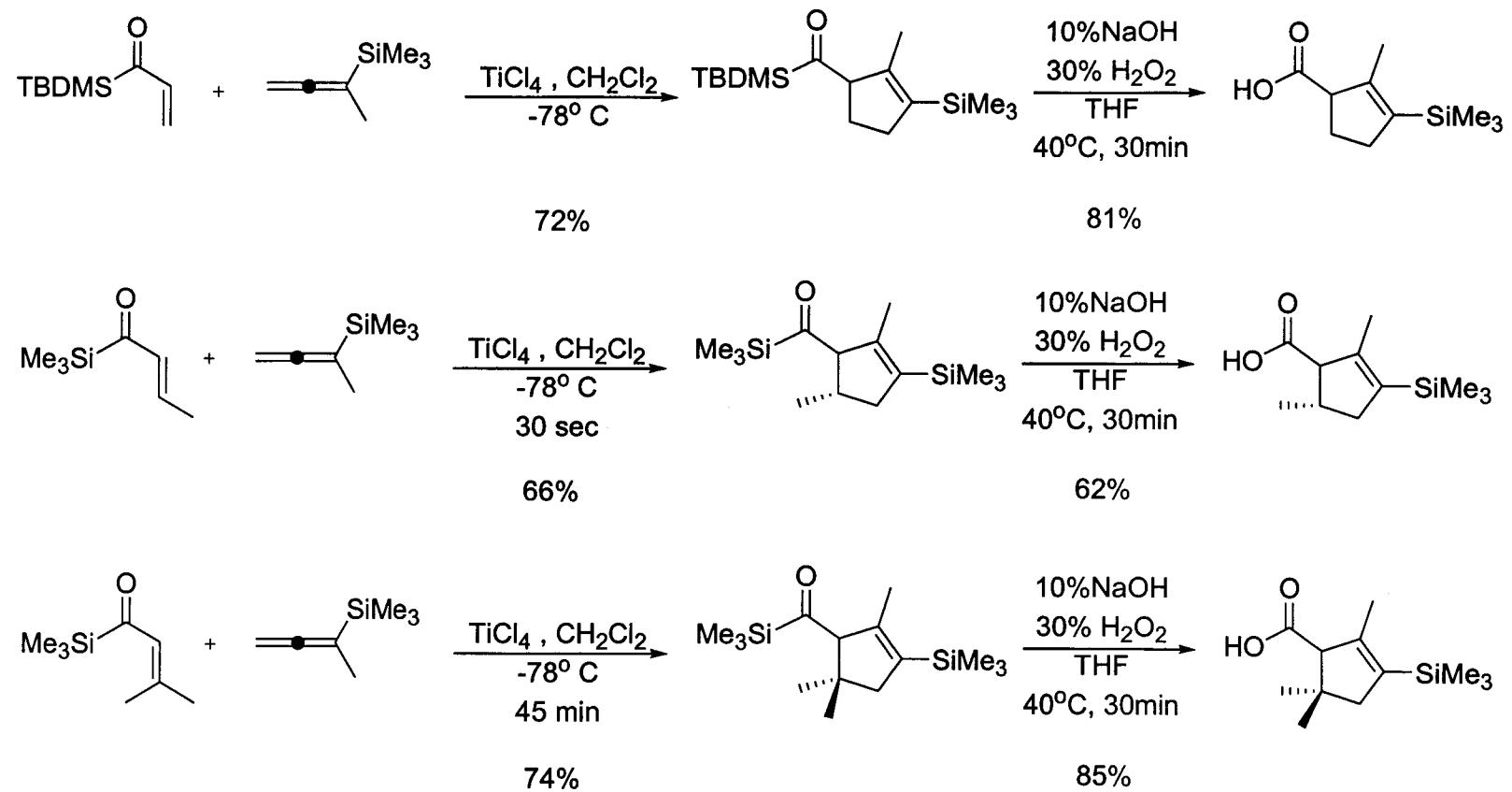


Danheiser, R.L. et al. *JACS*, 1981, 103, 1604  
 Danheiser, R.L. et al. *Tetrahedron*, 1983, 39, 935

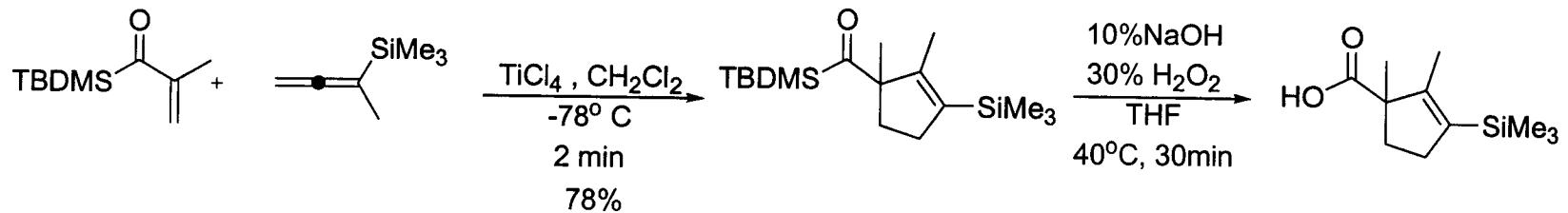
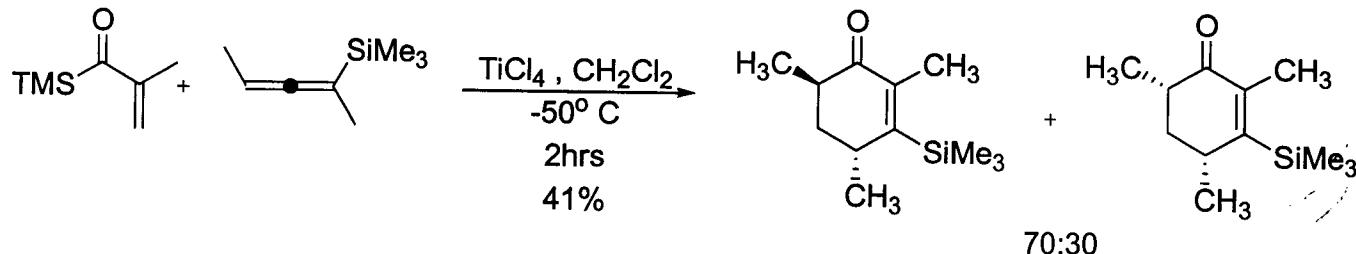
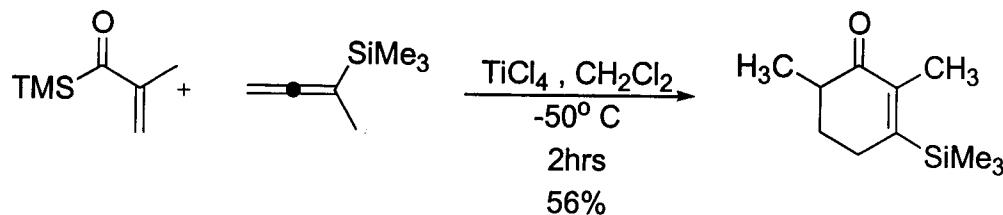
# Limitations of the Danheiser Annulation

- Works Primarily with  $\alpha,\beta$  unsaturated ketones
- Does not work with  $\alpha,\beta$  unsaturated aldehydes, methacrolein under reaction conditions yields a “complex mixture of products”
- Does not work with nitroolefins
- Reaction of  $\alpha,\beta$  unsaturated esters is considerably slow in comparison to the reaction of  $\alpha,\beta$  unsaturated ketones (13.5 hrs : 1 hr)
- ~~$\beta,\beta$  disubstituted allenes are unreactive at  $-78^\circ\text{C}$ , but do react within reasonable times.~~
- Highly sterically encumbered ketones, such as isophorone, are unreactive
- Substitution at the  $\alpha$  position of the allene is a necessity, due to the relative instability of the terminal vinyl cation

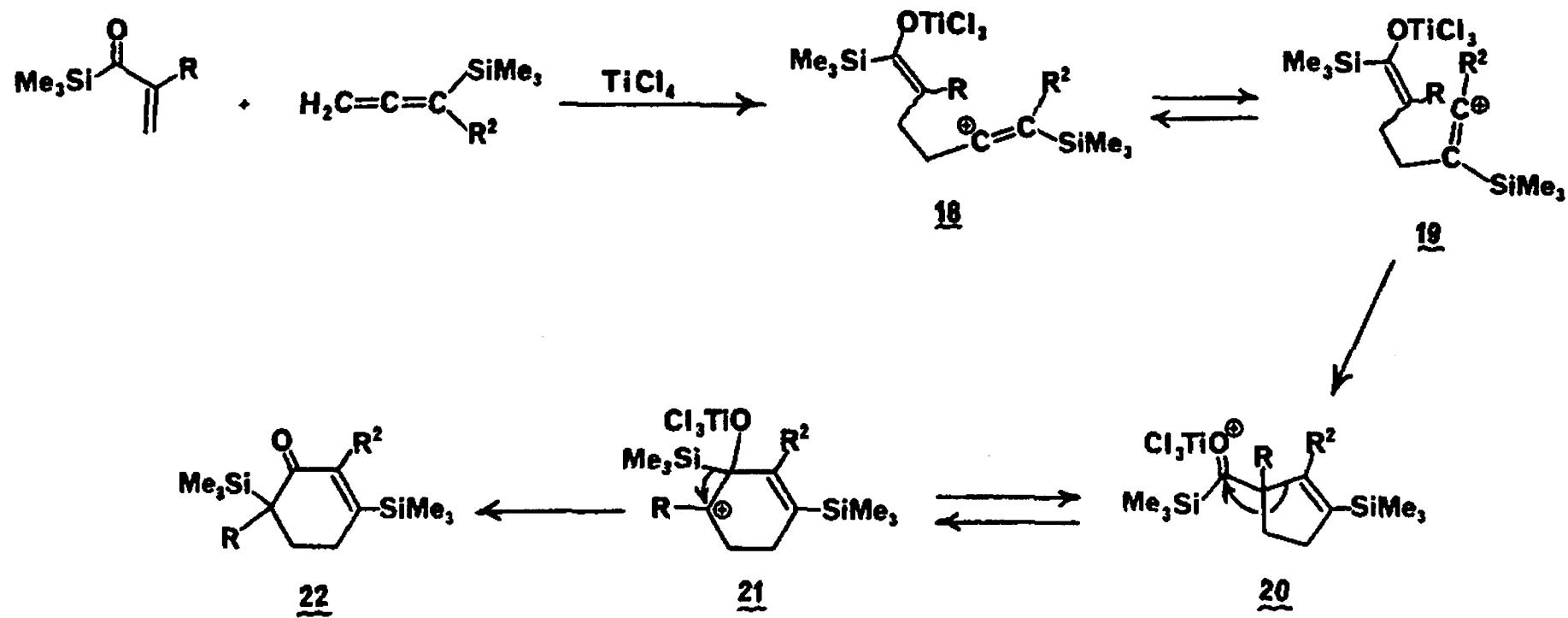
# Reaction of Allenylsilanes with Acylsilanes



# Reaction of Allenylsilanes with Acylsilanes

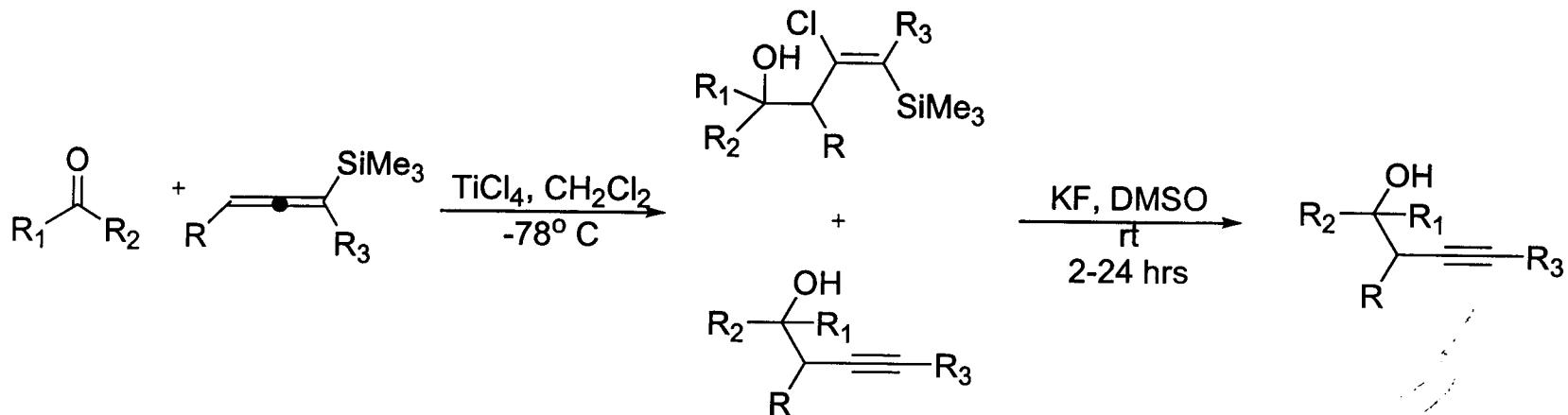


# Proposed Mechanism for [3+2] Cyclopentenannulation with Acylsilanes



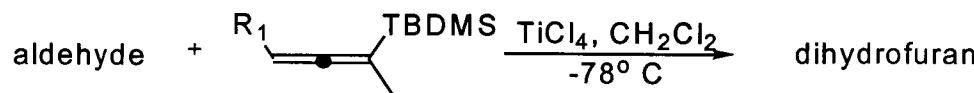
- Reaction pathway dependent on formation of tertiary carbocation 21
- If bulkier Silyl group is used, vinyl shift disfavored due to sterics
- Using acyl silanes, cyclopentenyl carboxylic acids and  $\alpha,\beta$  unsaturated

# Adaptation of the Danheiser Annulation to the Synthesis of Heterocycles

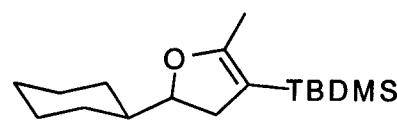


- Early studies demonstrate annulation reaction conditions with aldehydes and ketones produces homopropargyl alcohols in moderate to high yield
- Chloride anion either assists in desilylation, or is trapped by vinyl cation
- These reactions occur preferentially over cyclization to the dihydofuran
- Solution: Increase steric bulk around the silicon to depress the rate of desilylation

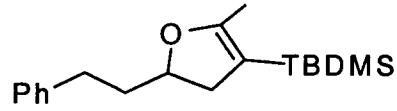
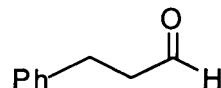
# Synthesis of Substituted Dihydrofurans and Alkaloids



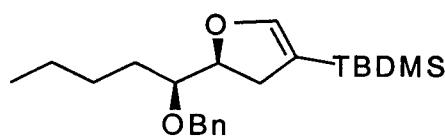
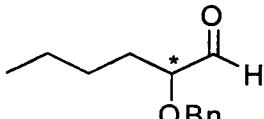
R1=H



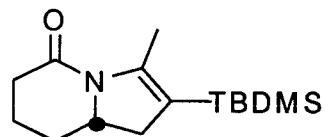
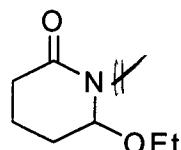
76%



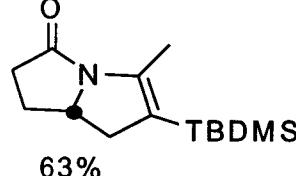
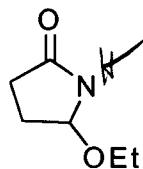
70%



88%

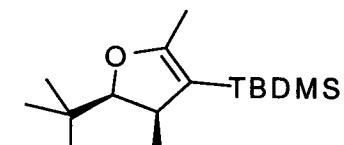
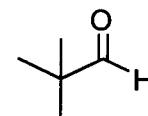


(67%)

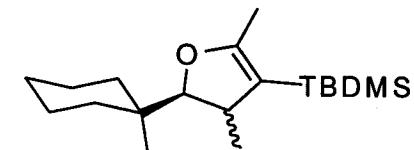
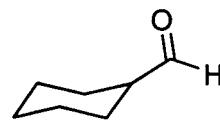


63%

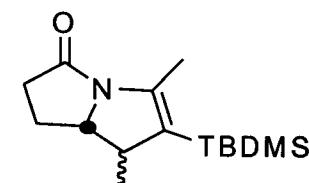
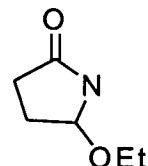
R1=Me



(92%)



97% (7:1, cis:trans)



60% (1.2:1, cis:trans)

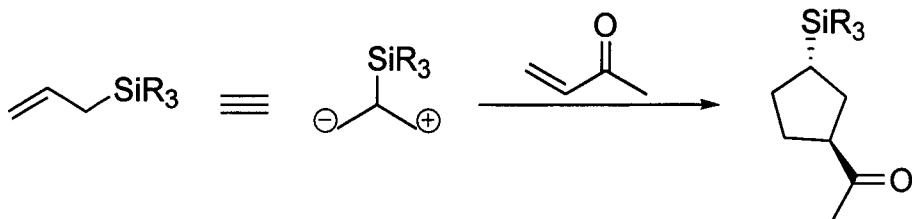
TBDMS group is difficult to functionalize!

Danheiser, R.L. et al. JACS, 1985, 107, 7233

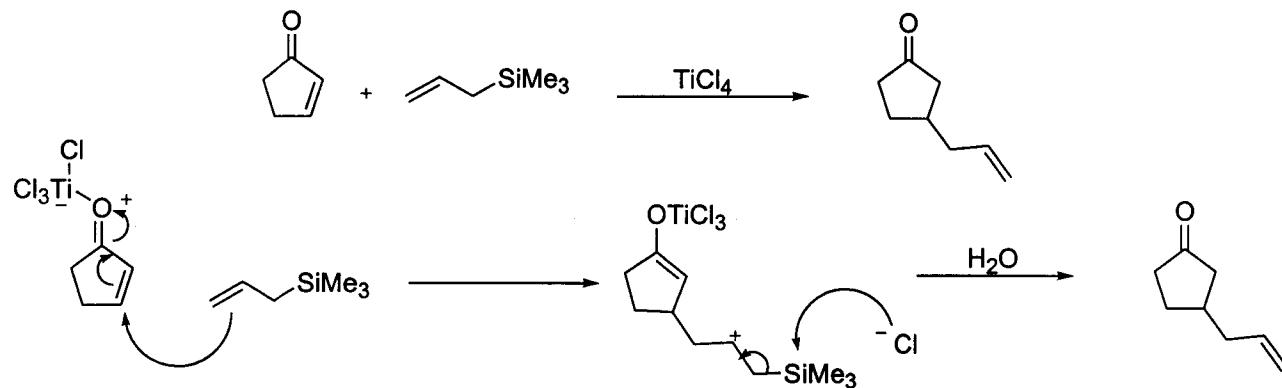
# Summary of Allene Chemistry

- Excellent regiochemistry imparted by Silicon group
- Predictable stereochemistry due to constraint of the 5-membered ring (usually cis ring juncture)
- Established scope and limitations
- Chiral starting materials lead to chiral products in a predictable fashion
- Functionalization of Silicon group difficult
- Harsh conditions to cleave C-Si bond limit functionality of starting materials

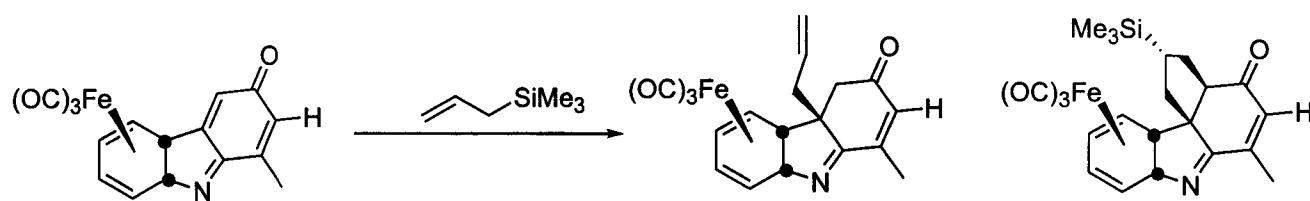
# Allylsilanes as Masked 1,3 Dipoles...



Hosomi-Sakurai Reaction (Sakurai allylation)

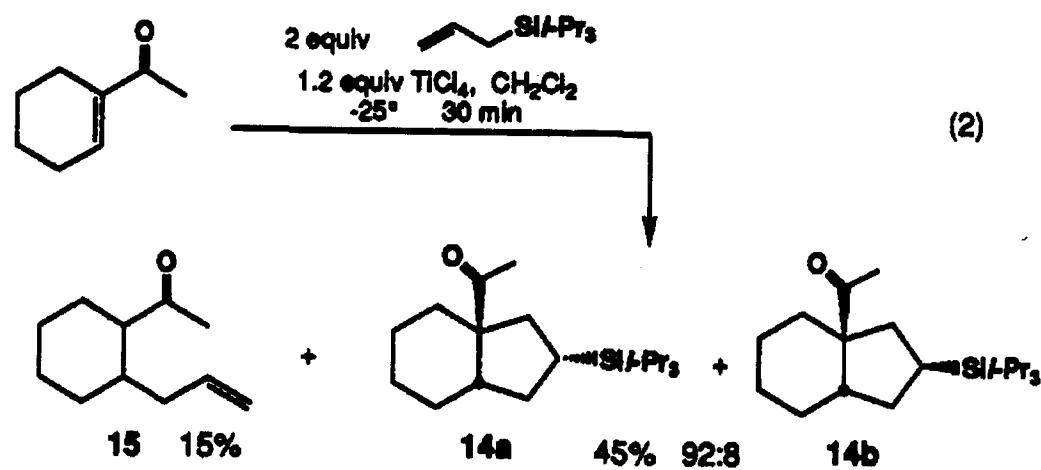
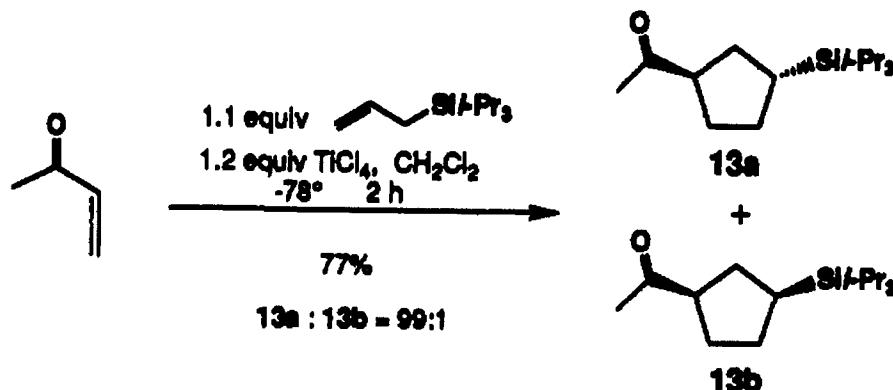


Knolker's Observation

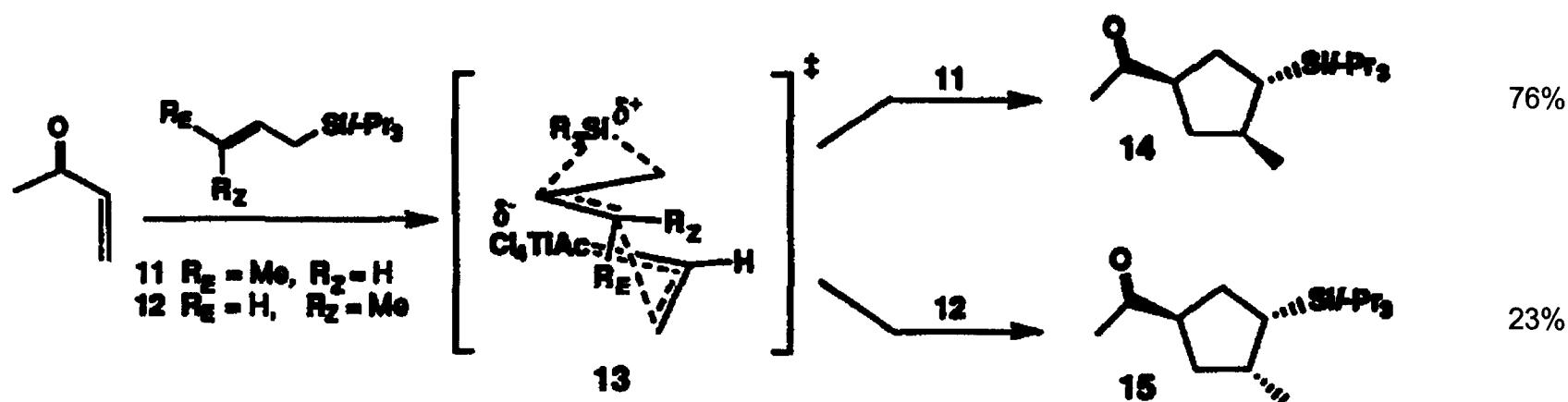
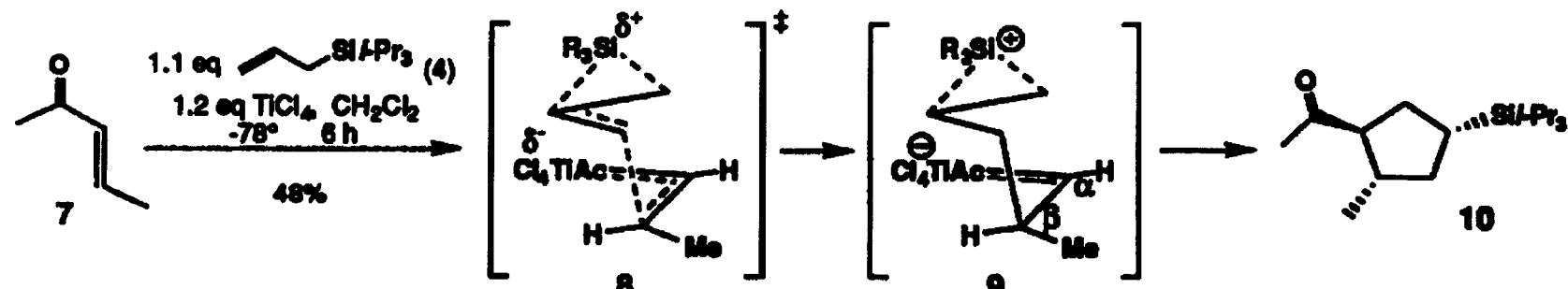


Knolker, H. et al. *SYNLETT*, 1990, 429

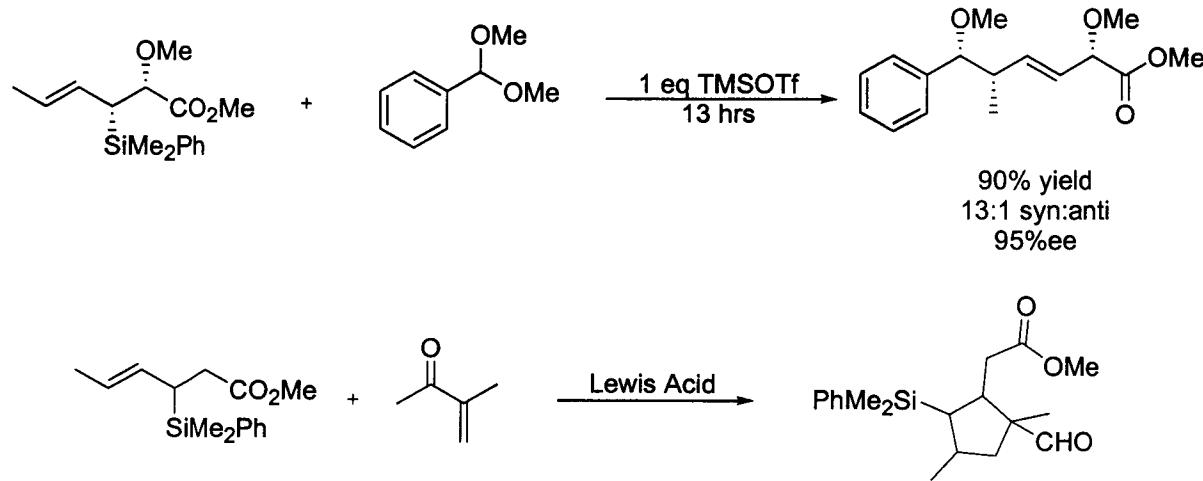
# Danheiser's Contribution



# Rationalization of Stereochemistry

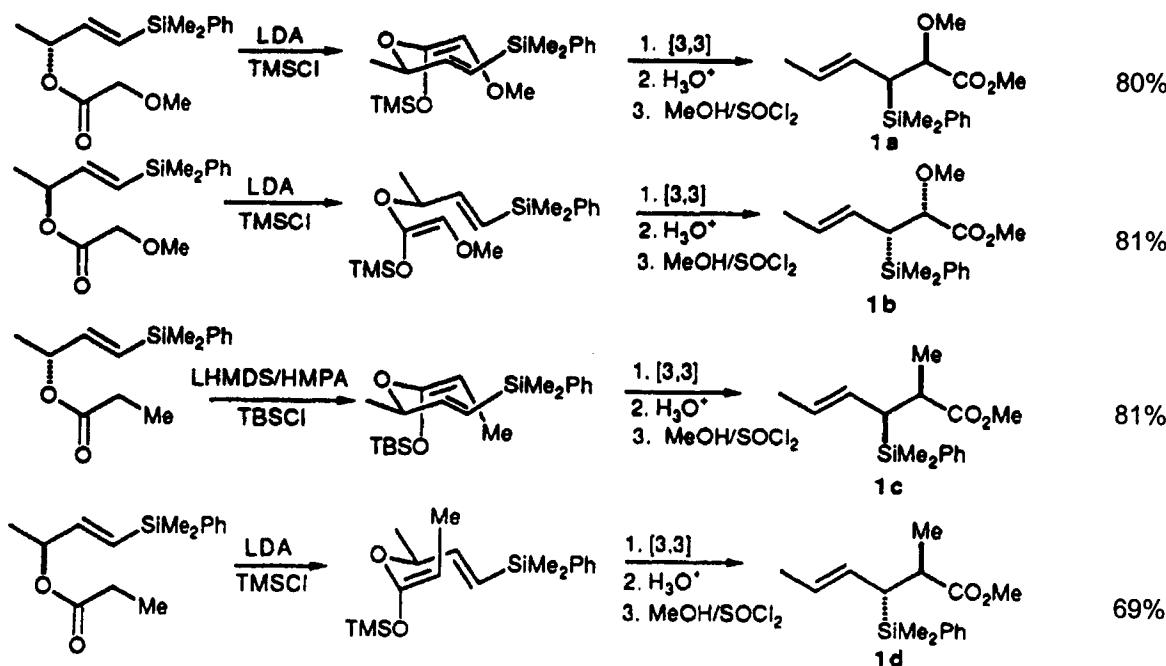
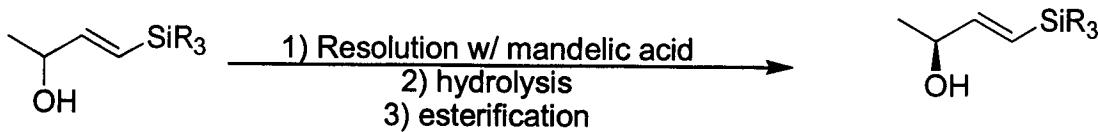


# The Use of Chiral *E*-crotylsilanes Towards Stereoselective Synthesis of Cyclopentanes



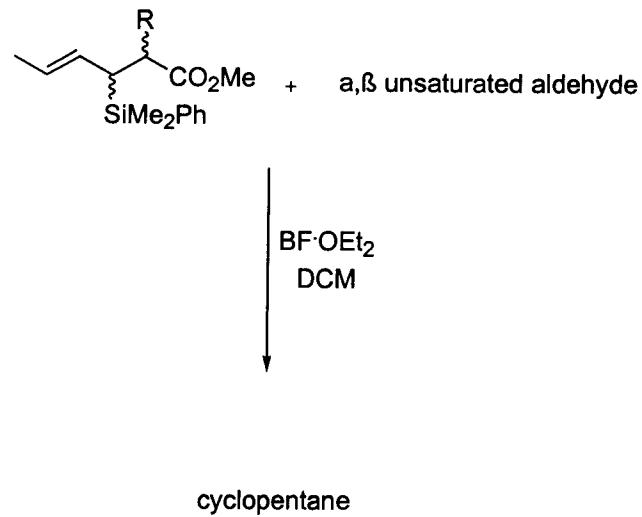
- Addition of chiral crotyl silanes to oxocarbenium ions was studied extensively by Panek
- It was conceived that addition of these nucleophiles to  $\alpha,\beta$  unsaturated ketones would yield the corresponding cyclopentane in high yields and enantiopurity

# Preparation of Chiral E-crotylsilanes



Panek, J.S., Sparks, M.A. *JOC*, 1991, 56, 3431

# Asymmetric Synthesis of Tetrasubstituted Cyclopentanes

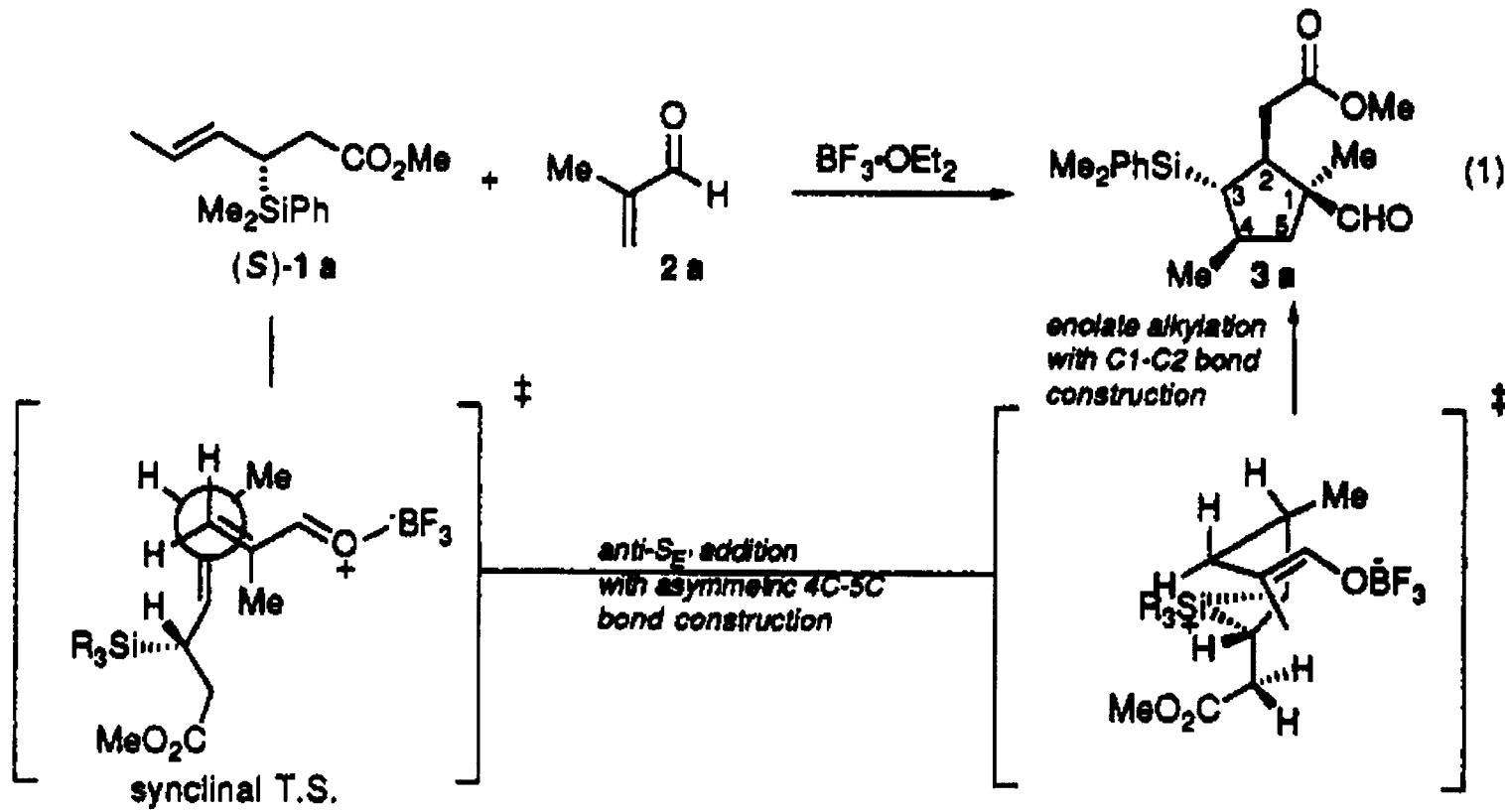


Panek, J.S., Jain, N.F. *JOC*, 1993, 58, 2345

Table I. Asymmetric [3 + 2] Cyclopentane Annulation

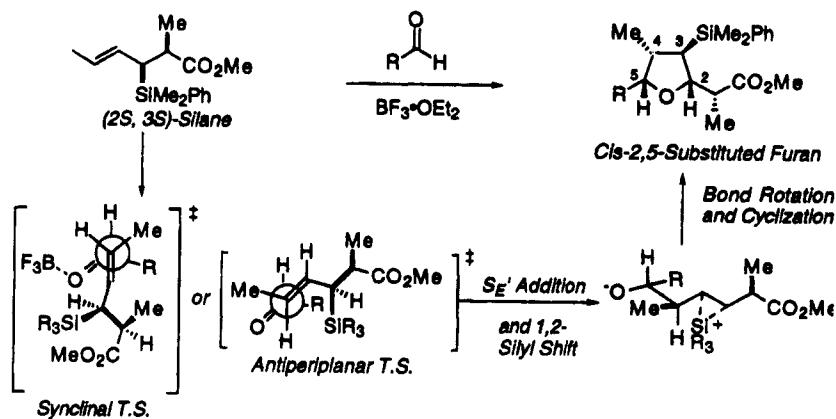
crotyl silyl ether	$\alpha,\beta$ -unsaturated aldehyde/ketone	reaction condns <sup>a</sup> (T (°C)/time (h))	major diast <sup>b</sup>	% yield <sup>c</sup>
		rt/8		93
		rt/12		73
		rt/12		73
		rt/7		65
		rt/5.5		62
		rt/7		51
		rt/7		81
		-10-0/10		56

# Stereochemical Rationale



# Diastereoselective Synthesis of Substituted Furans

entry	aldehydes	(E)-crotylalane	reaction cond. <sup>a</sup>	major diastereomer <sup>b</sup>	yield% <sup>c</sup>	ratio 4,5-syn/anti <sup>d</sup>	ratio %det <sup>e</sup>
1	2a		-78 °C/15h		85	30:1	96
2	2a		-78 °C/15h		85	30:1	96
3	2a		-78 °C/15h		80	30:1	96
4 <sup>f</sup>	2a		-78 °C/15h		68	30:1	96
5	2b		-30 °C/15h		65	30:1	96
6	2b		-30 °C/15h		50	20:1	96



Masse, C.E., Panek, J.S., *CHEMREV*, 1995, 95, 1293  
 Panek, J.S., Yang, M. *JACS*, 1991, 113, 9868

# Lewis Acid Control of Diastereoselectivity

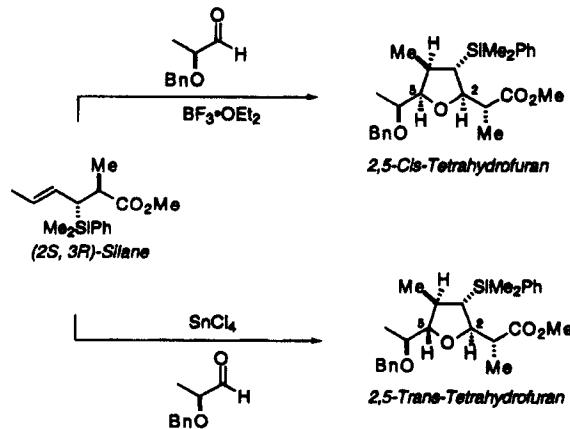
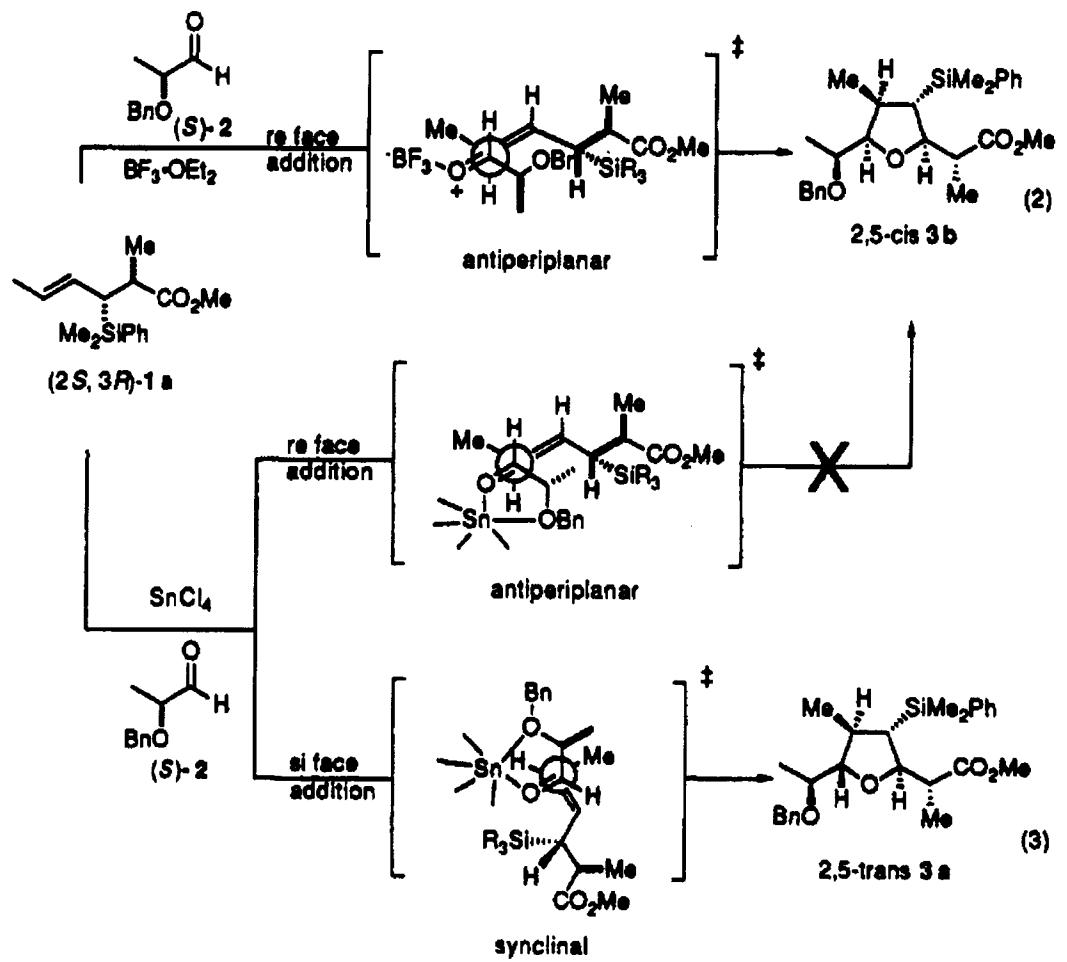


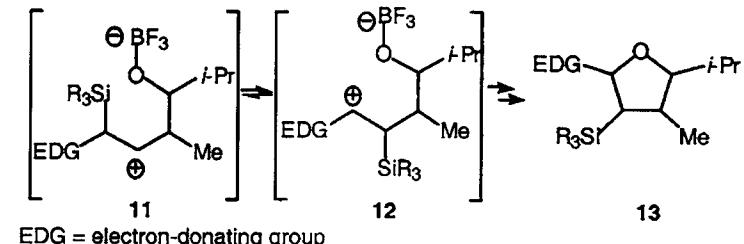
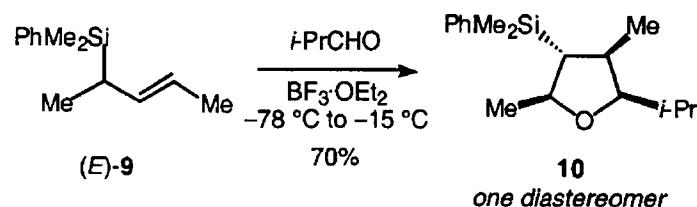
Table I. Double Asymmetric Induction in Additions of Chiral (*E*)-Crotylsilanes to (*S*)-2-(Benzylxy)propanal

entry	aldehyde <sup>a</sup>	(E)-crotylsilane <sup>b</sup>	reaction condn. <sup>c</sup> LA/temp/time	reaction products, <sup>d</sup> (% yield) <sup>e</sup>		ratio anti/syn C4/C5 (furan) <sup>f</sup>
				tetrahydrofuran	homallylic alcohol	
1.	2	(2 <i>S</i> , 3 <i>R</i> )-1a	$\text{SnCl}_4$ -35 °C/12h	3a (75)	4a (<5)	>40:1
2.	2	1a	$\text{BF}_3 \cdot \text{OEt}_2$ -25 °C/12h	3b (50)	4b (<5)	40:1
3.	2	(2 <i>R</i> , 3 <i>S</i> )-1b	$\text{BF}_3 \cdot \text{OEt}_2$ -25 °C/12h	3c (65)	4c (10)	>40:1
4.	2	1b	$\text{SnCl}_4$ -35 °C/12h	3c (36)	4c (38)	>40:1

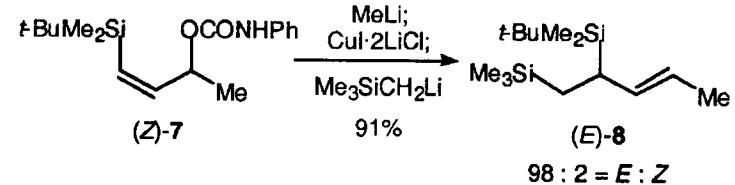
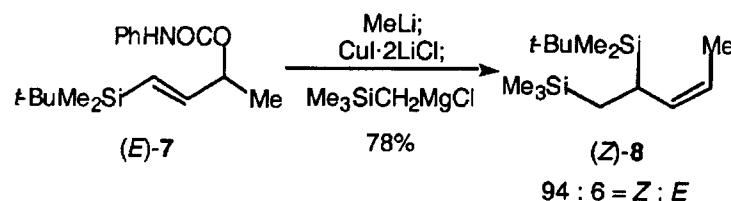
# Rationalization of Diastereoselectivity



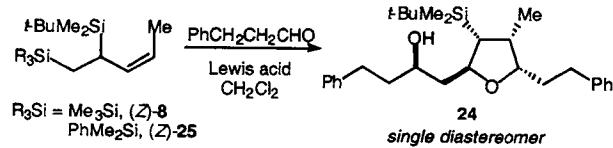
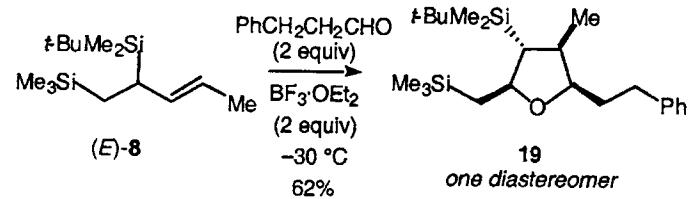
# [3+2] Cyclizations via Silylmethyl Allylic Silanes



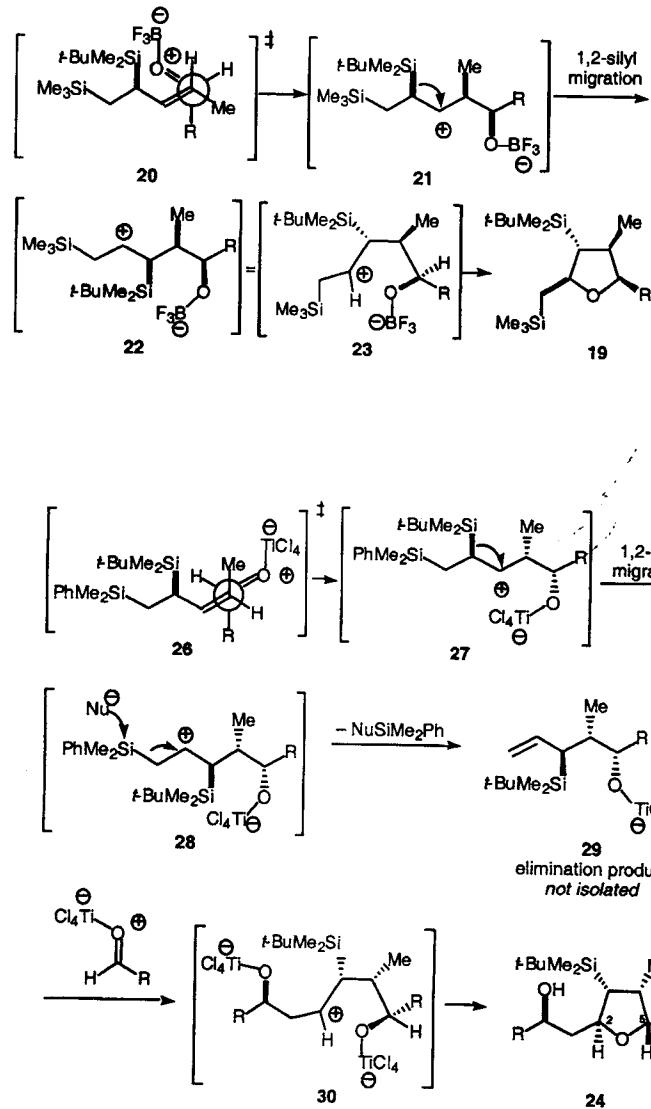
- Silylmethyl allylic silanes have rarely been used in organic synthesis
- Woerpel and coworkers recognized that, upon reaction with aldehydes, as in the case of Panek's work, a 1,2 silylmigration might occur.
- If an electron donating group, such as a methylsilyl group was attached, the cation would be doubly stabilized, pushing the reaction to proceed
- No general preparation for silylmethyl allylic silanes was known



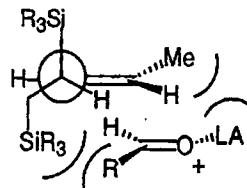
# Synthesis of Substituted Tetrahydrofurans



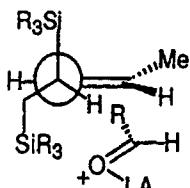
Entry	Substrate	Lewis Acid (Equiv) <sup>a</sup>	Temperature	Time (h)	Yield <sup>b</sup> (%)
1	(Z)-8	$\text{BF}_3\cdot\text{OEt}_2$ (1)	$-78^\circ\text{C} \rightarrow -33^\circ\text{C}$	24	No reaction
2	(Z)-8	$\text{BF}_3\cdot\text{OEt}_2$ (1)	0 °C	20	No reaction
3	(Z)-8	$\text{SnBr}_4$ (1)	$-30^\circ\text{C}$	20	No reaction
4	(Z)-8	$\text{TiCl}_4$ (1)	$-30^\circ\text{C}$	19	Complex mixture
5	(Z)-8	$\text{TiCl}_4$ (1)	$-78^\circ\text{C}$	3	25
6	(Z)-8	$\text{TiCl}_4$ (2)	$-90^\circ\text{C}$	4	41
7	(Z)-25	$\text{TiCl}_4$ (2)	$-90^\circ\text{C}$	4	46



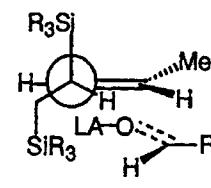
# Difference in Reactivities Between E,Z Olefins



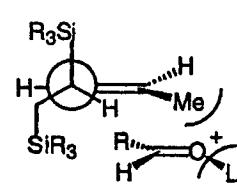
31  
anti-periplanar



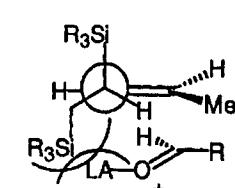
32  
synclinal



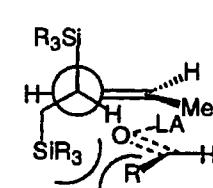
33  
synclinal



34  
anti-periplanar



35  
synclinal



36  
synclinal

## Analysis of E-Olefin TS

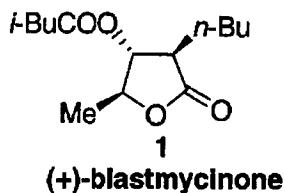
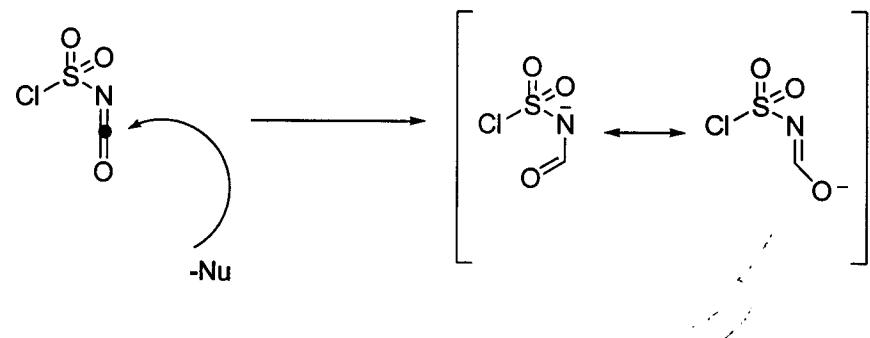
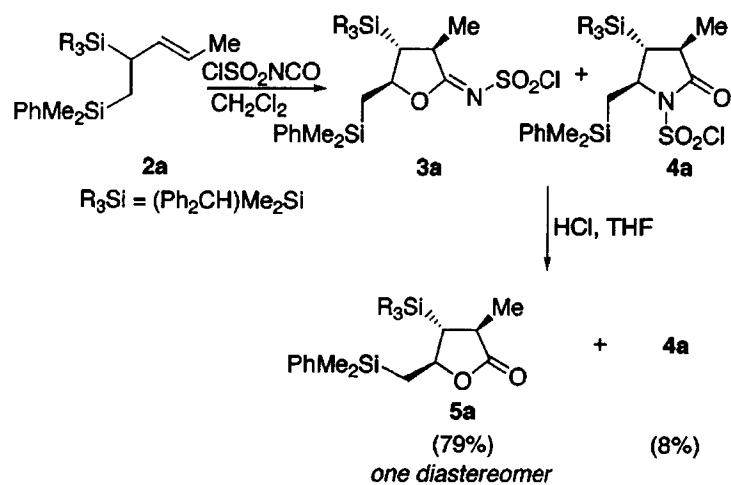
- For E-olefin there are three approaches that lead to product
- Only anti-periplanar is refuted due to sterics

## Analysis of Z-Olefin TS

- In the case of the Z-olefin, all three transition states are disfavored by sterics

# Diastereoselective Synthesis of Lactones

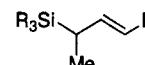
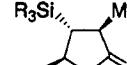
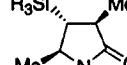
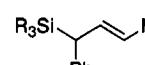
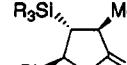
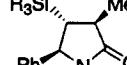
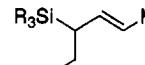
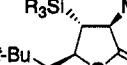
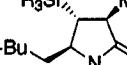
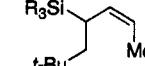
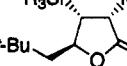
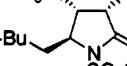
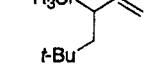
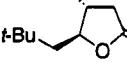
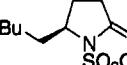
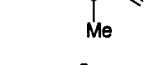
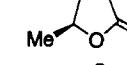
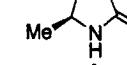
Scheme 1

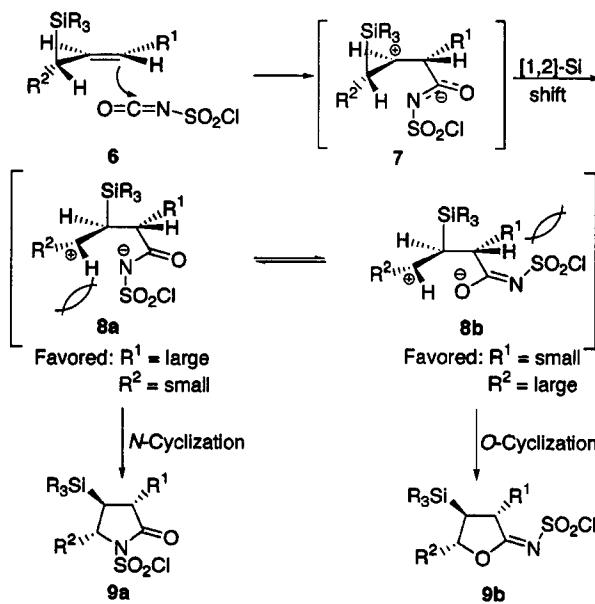


Gamma butyrolactones are found in a variety of natural products

# Studies to Determine O,N selectivity

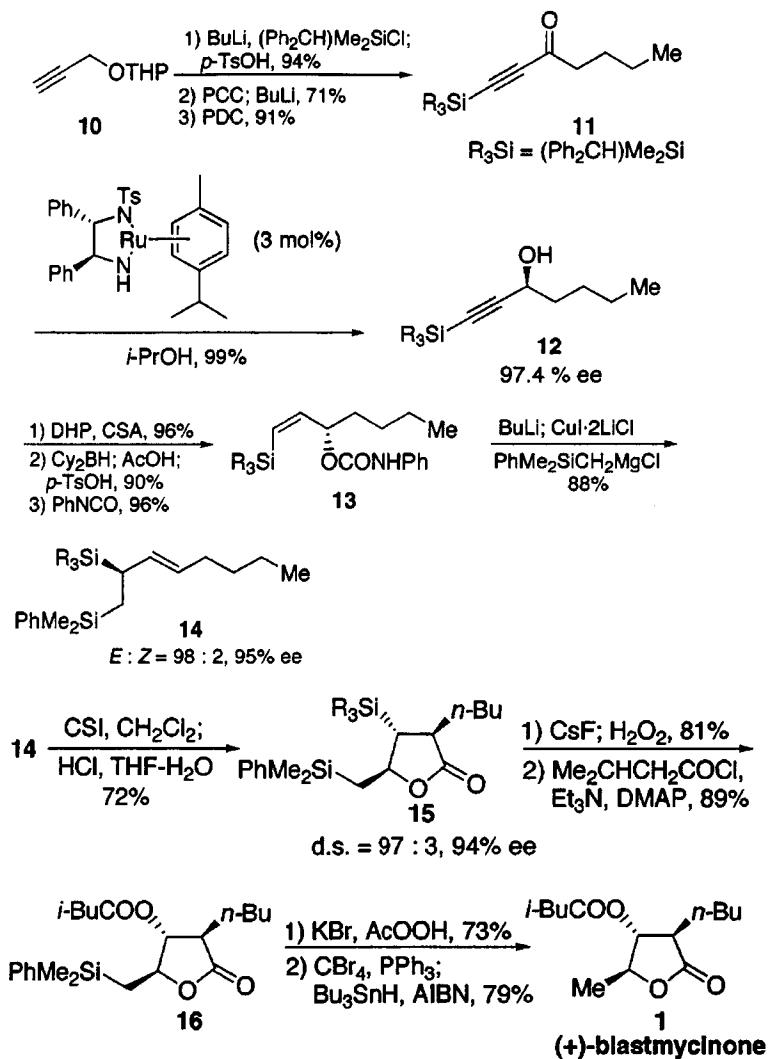
**Table 1.** Annulation Reactions of Allylic Silanes with  $\text{ClSO}_2\text{NCO}^a$

Allylic Silane <b>2<sup>b</sup></b>	[O]/[M] <sup>c</sup>	Products		Yields <sup>d</sup> (Diastereomer ratio) <sup>e</sup>
	1 : 6			16% (95 : 5)
<b>2b</b> (E : Z = 99 : 1) <sup>f</sup>				77% (98 : 2)
	1 : 2			30% (98 : 2)
<b>2c</b> (E : Z > 99 : 1) <sup>f</sup>				57% (99 : 1)
	>20 : 1			61% (> 99 : 1)
<b>2d<sup>g</sup></b> (E : Z = 98 : 2) <sup>f</sup>				1%
	>20 : 1			81% (94 : 6)
<b>2e</b> (E : Z = 4 : 96) <sup>f</sup>				3%
	8 : 1			82% (> 96 : 4)
<b>2f</b>				8%
	1 : 1			37% (91 : 9)
<b>2g</b>				42% (91 : 9)



Woerpel, K.A., Peng, Z. *Org. Lett.* 2001, 3, 675

# Total Synthesis of Blastomycin



- 13% yield over 13 steps
- 3 contiguous stereocenters set selectively in one step
- Methylsilyl allylsilanes are reactive starting material whose chemistry has yet to be fully explored

# Conclusion

- Silicon assisted [3+2] cyclization reactions have been carried out using allenyl, and allyl silanes
- There are predictable, diastereoselective syntheses in the literature for carbocycles as well as heterocycles
- The Danheiser annulation as well as early allyl cyclizations suffer from alkyl silyl groups that are difficult to manipulate
- Later syntheses employ activated silanes that can undergo stereoselective functionalization
- Much work has yet to be accomplished in all examples, primarily in the silyl methyl allylic silanes
- Although well studied, few total syntheses employ this methodology