

Denmark Group Meeting

June 26, 2001

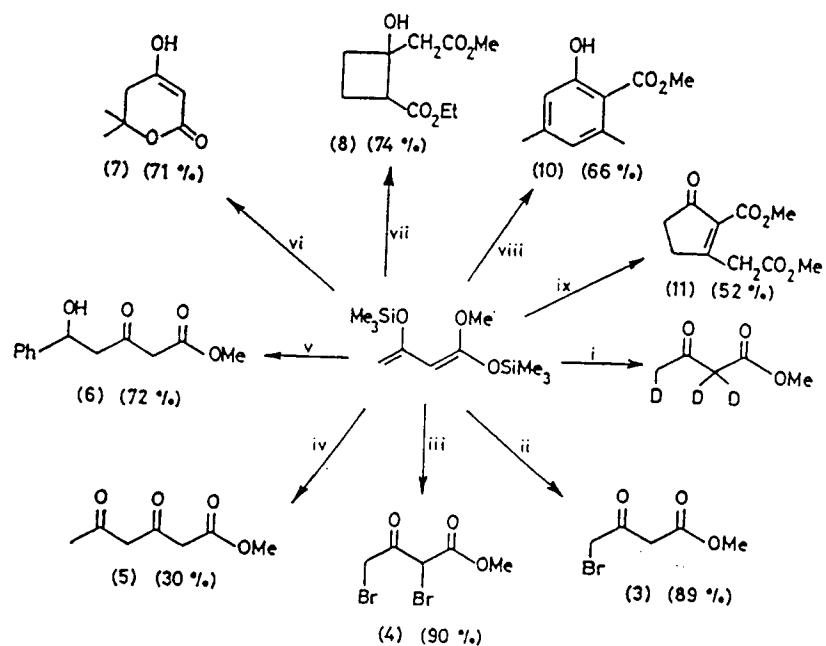
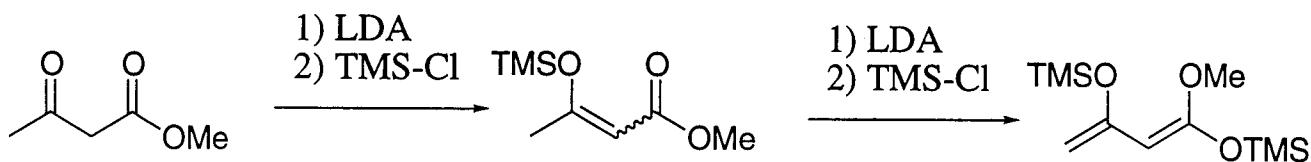
Tyson A. Miller

The Chan Reaction



Chan Diene: Dianion Equivalent of Methyl Acetoacetate

2



Reagents: i, D_2O ; ii, 1 mol Br_2 ; iii, 2 mol Br_2 ; iv, $AcCl$, ether, $-78^\circ C$; v, $PhCHO$, $TiCl_4$; vi, acetone, $TiCl_4$; vii, $CH_2=CHCO_2Et$, $TiCl_4$; viii, $MeC(OSiMe_3)=CHCOMe$, $TiCl_4$; ix, 2 mol $TiCl_4$.

Chan, T.-H., Brownbridge, P. JCS CC (1979) 578.

Keck, G.E., Yu, T. OL (1999) 1 289.
Reports that Chan diene gives
"no synthetically useful results" when
used with $Ti(O-i-Pr)_4$ and (S)-BINOL,
yet . . .

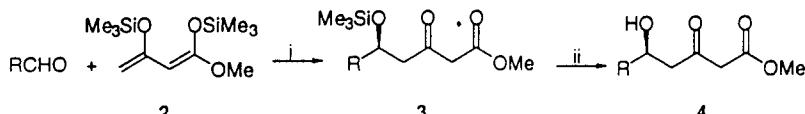


Table 1
Asymmetric aldol reaction promoted by $Ti(IV)/(S)$ -($-$)-BINOL complex

Entry	R	Product	Yield(%) ^a	Ee(%) ^b	$[\alpha]_D$
a	3-Furyl	4a	48	83	-36
b	C_6H_5	4b ^c	19	99	-46
		4b ^d	33	95	-45
c	p -NO ₂ C ₆ H ₄	4c	86	90	-39
d	p -CH ₃ OC ₆ H ₄	4d	54	93	-45
e	n-C ₉ H ₁₉	4e	50	87	-22

a) All the yields refer to isolated chromatographically pure compounds, whose structures were confirmed by ¹H-NMR and ¹³C analysis.¹⁶

b) E.e.s have been determined by HPLC analysis.

c) Yield refers to pure aldol 4b isolated directly after aldol reaction.

d) Yield refers to pure aldol 4b isolated after desilylation procedure.

Soriente, A.; et al. Tet. Asym. (2000) 11 2255.

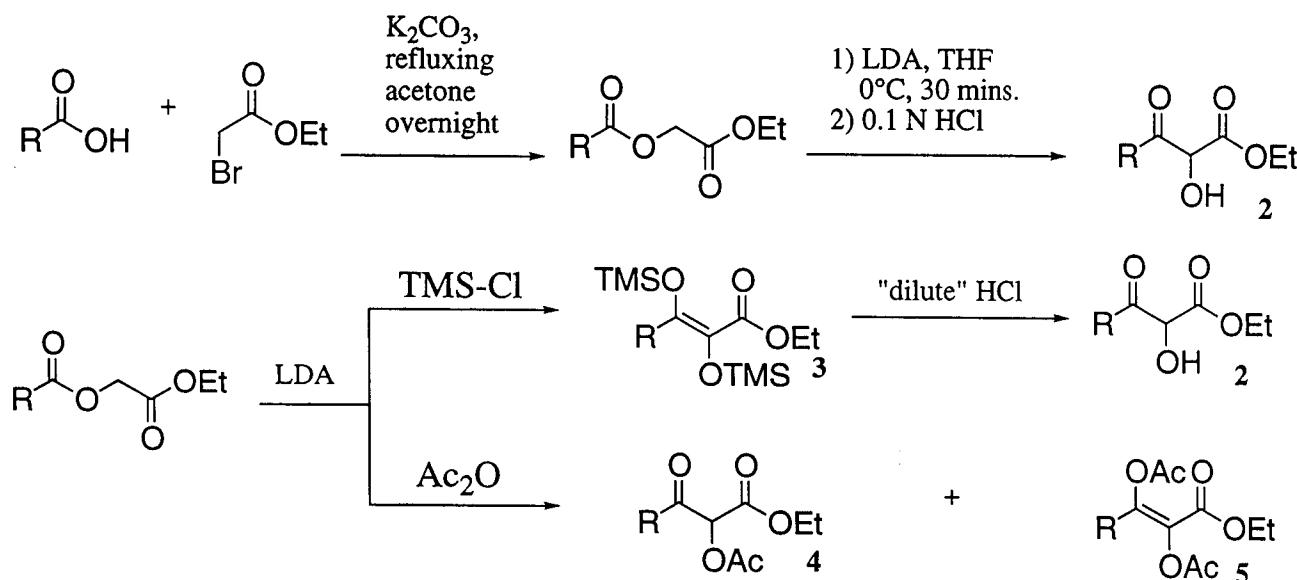
Scheme 1. Synthesis of δ -hydroxy- β -ketoesters 4; reagents:

(i) $Ti(O-i-Pr)_4$, THF, S-($-$)-BINOL, CH_2Cl_2 , $-78^\circ C$, m.s.;

(ii) H^+ /THF, $0^\circ C$

The Chan Reaction

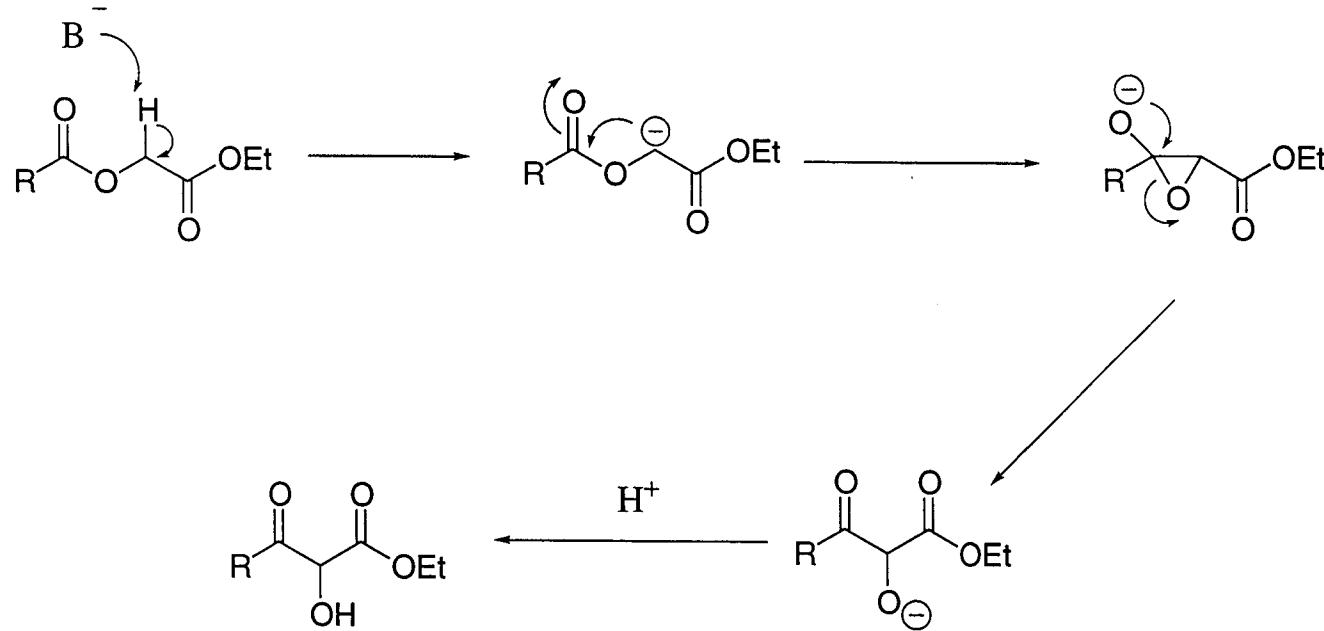
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α -Acyloxyacetate	Rxn Cond.	Product	% Yield Isolated
$R = \text{Ethyl}$	LDA, $0^\circ C$, H $+$	2a	57%
	LDA, $0^\circ C$, TMSCl	3a	87%
	LDA, $0^\circ C$, 1 eq Ac_2O	4a	35%
$R = n\text{-Pentyl}$	LDA, $0^\circ C$, H $+$	2b	52%
	LDA, $0^\circ C$, TMSCl	3b	65%
	LDA, $0^\circ C$, 1 eq Ac_2O	4b	45%
$R = \text{Cyclohexyl}$	LDA, $0^\circ C$, H $+$	2c	65%
	LDA, $0^\circ C$, TMSCl	3c	50%
	LDA, $0^\circ C$, 1 eq Ac_2O	4c	67%
	LDA, $0^\circ C$, 2 eq Ac_2O	5c	59%
$R = \text{sec-Pentyl}$	LDA, $0^\circ C$, H $+$	2d	65%
	LDA, $0^\circ C$, TMSCl	3d	62%
	LDA, $0^\circ C$, 1 eq Ac_2O	4d	65%
	LDA, $0^\circ C$, 2 eq Ac_2O	5d	62%
$R = \text{Phenyl}$	LHMDS, $0^\circ C$, 2 eq Ac_2O	5e	63%

Proposed Mechanism

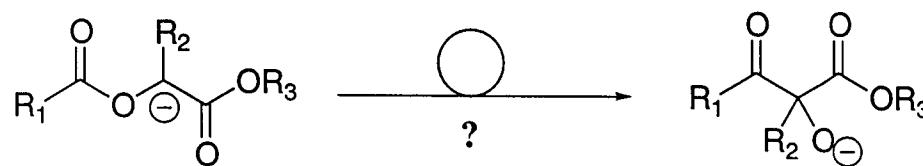
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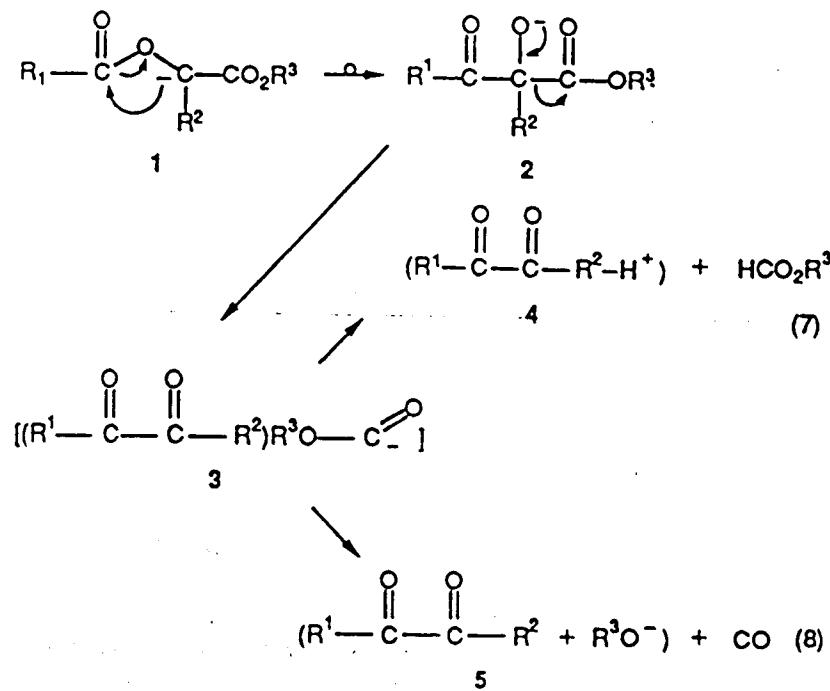
Lee, S.D.; Chan, T.H.; Kwon, K.S. TL (1984) 25 3399.

Gas Phase MS/MS of 1,2-Anionic Rearrangements

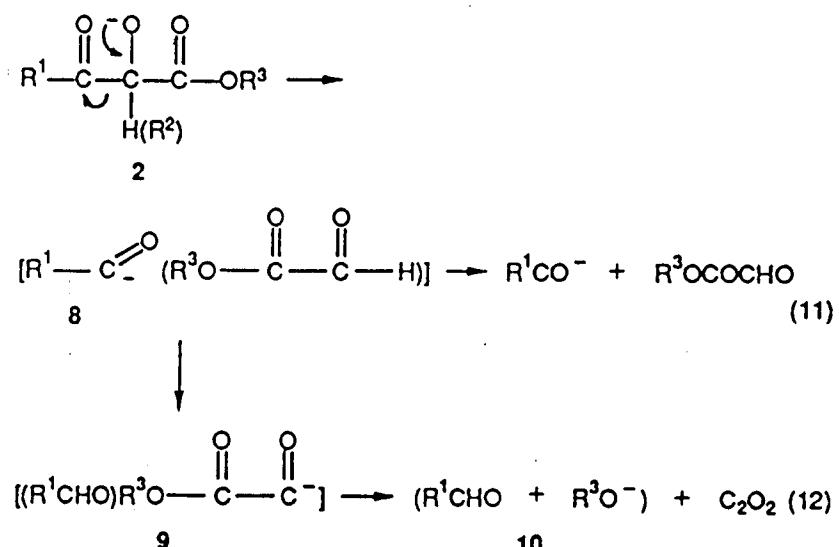
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Scheme II



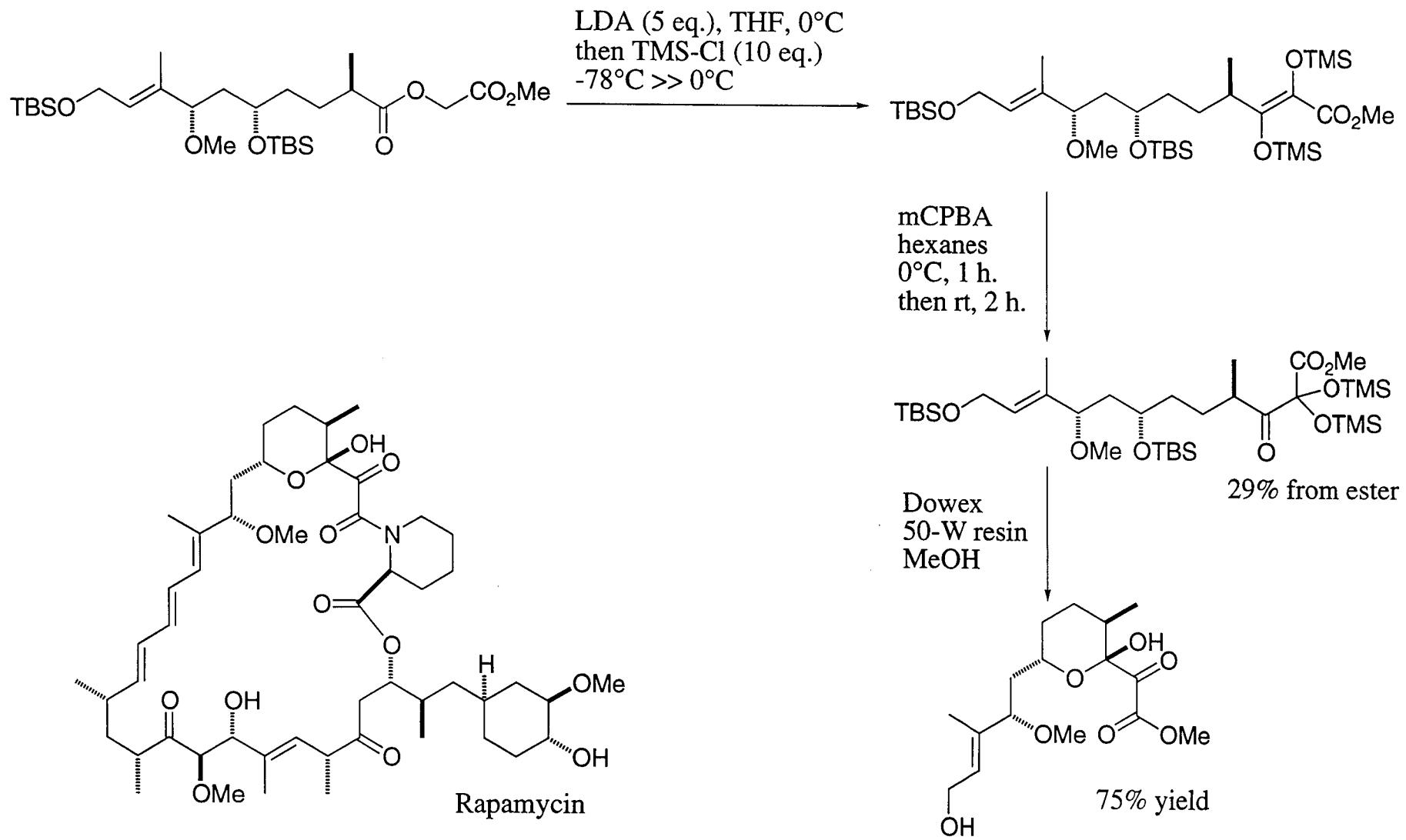
Scheme III



Study supports gas-phase 1,2-anionic rearrangement (also for α -(acyloxy)acetonitriles)

Tricarbonyl Subunit of Rapamycin

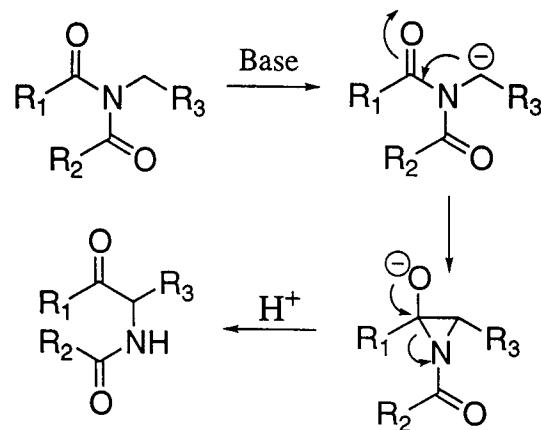
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White, J.D.; Jeffrey, S.C. JOC (1996) 61 2600.

N to C Acyl Migration Reaction of Imides: α -Aminoketones

7



Proceeds with retention of configuration . . .

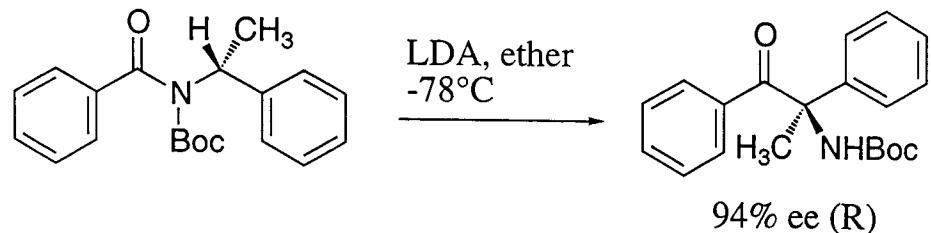
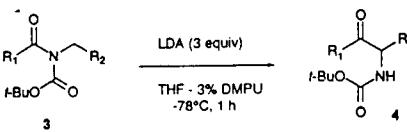


Table I. N-C Acyl Migration Reaction of Imides.

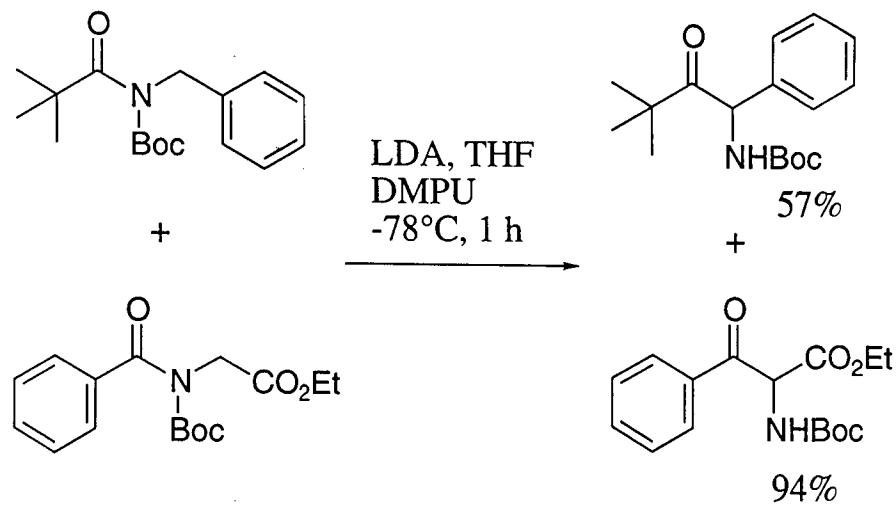


run	R ₁	R ₂	yield (%) ^a
1	Ph-	-Ph	81
2	Ph-	-Ph-OMe	89
3	Ph-	2-furyl	70
4	Ph-	CO ₂ Et	91
5	Ph-	CO ₂ t-Bu	82
6	Ph-	=	81 ^b
7	Ph-	-N ₃	71
8	Ph-	-N ₃	81
9	Ph-	-Ph-O ₂ CH	80
10	MeO-Ph-	-Ph	55
11	t-Bu-	-Ph	98 ^c (58) ^d
12	PhCH ₂ CH ₂ -	CO ₂ Et	53 ^c (57) ^d

a) Isolated yields. b) Yield of the isomerized α,β -unsaturated ketone 5. c) Yields based on the consumed starting materials. d) Yields of the recovered starting material.

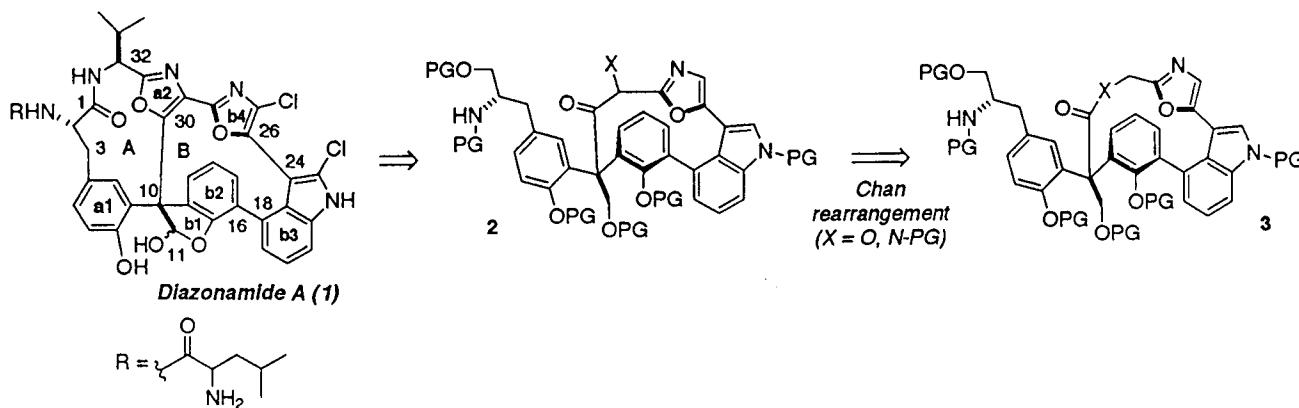
In addition, KH, KHMDS, and s-BuLi with sparteine showed no acyl migrated product.

Crossover experiment shows no intermolecular migration . . .



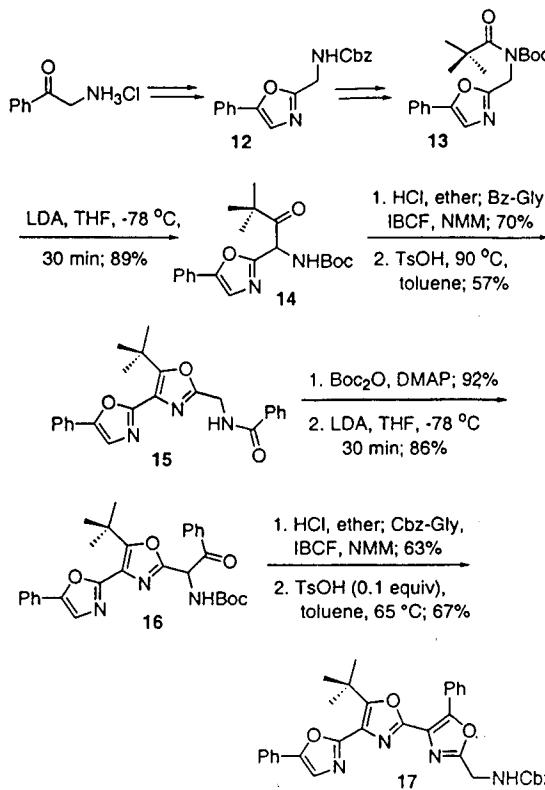
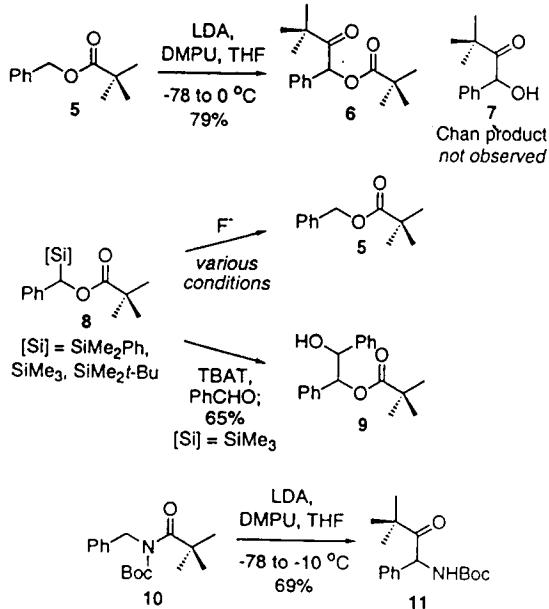
Hara, O.; Ito, M.; Hamada, Y. TL (1998) 39 5537.

Polyoxazole Synthesis Towards Diazonamide A

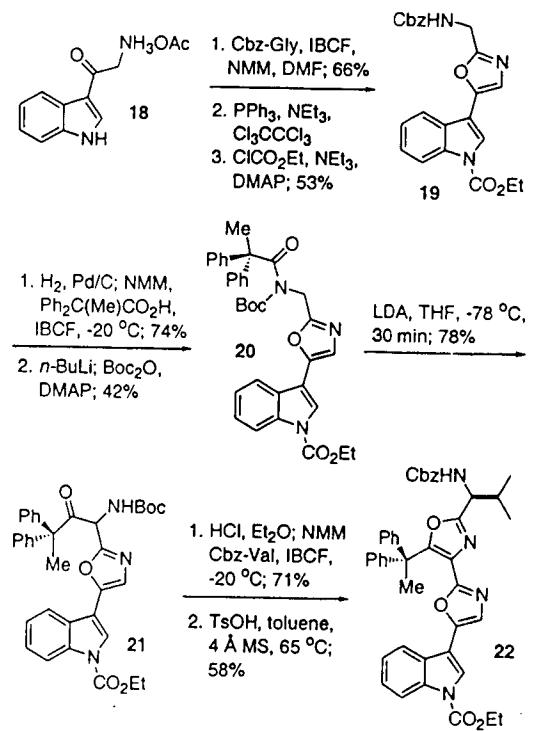


Scheme 2. Polyoxazoles via Sequential Chan Reactions

Scheme 1. Preliminary Model Studies

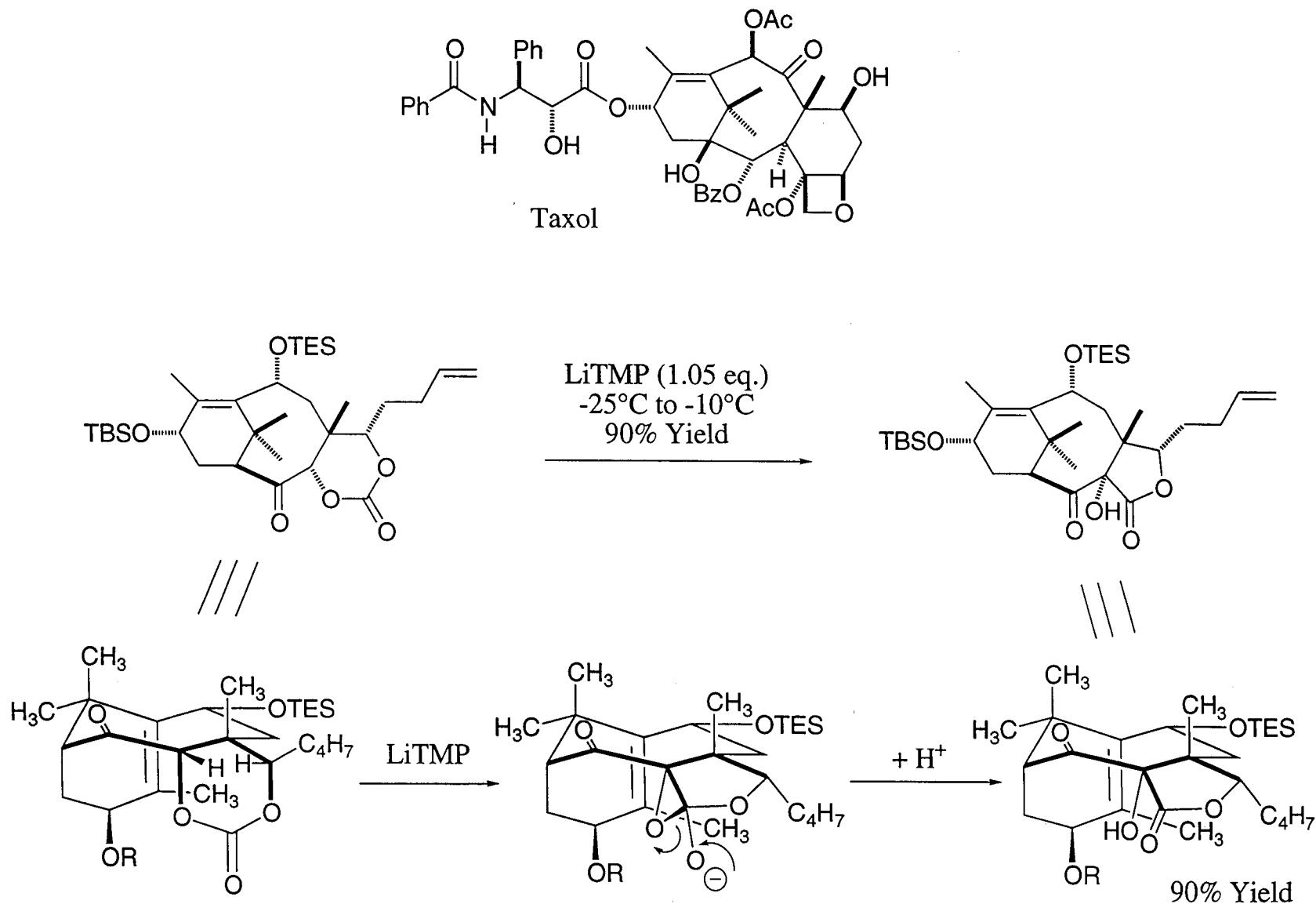


Scheme 3. Synthesis of the Indolyl Bisoxazole Fragment of Diazonamide A



Ring Reduction in Functionalization of Taxol B Ring

9

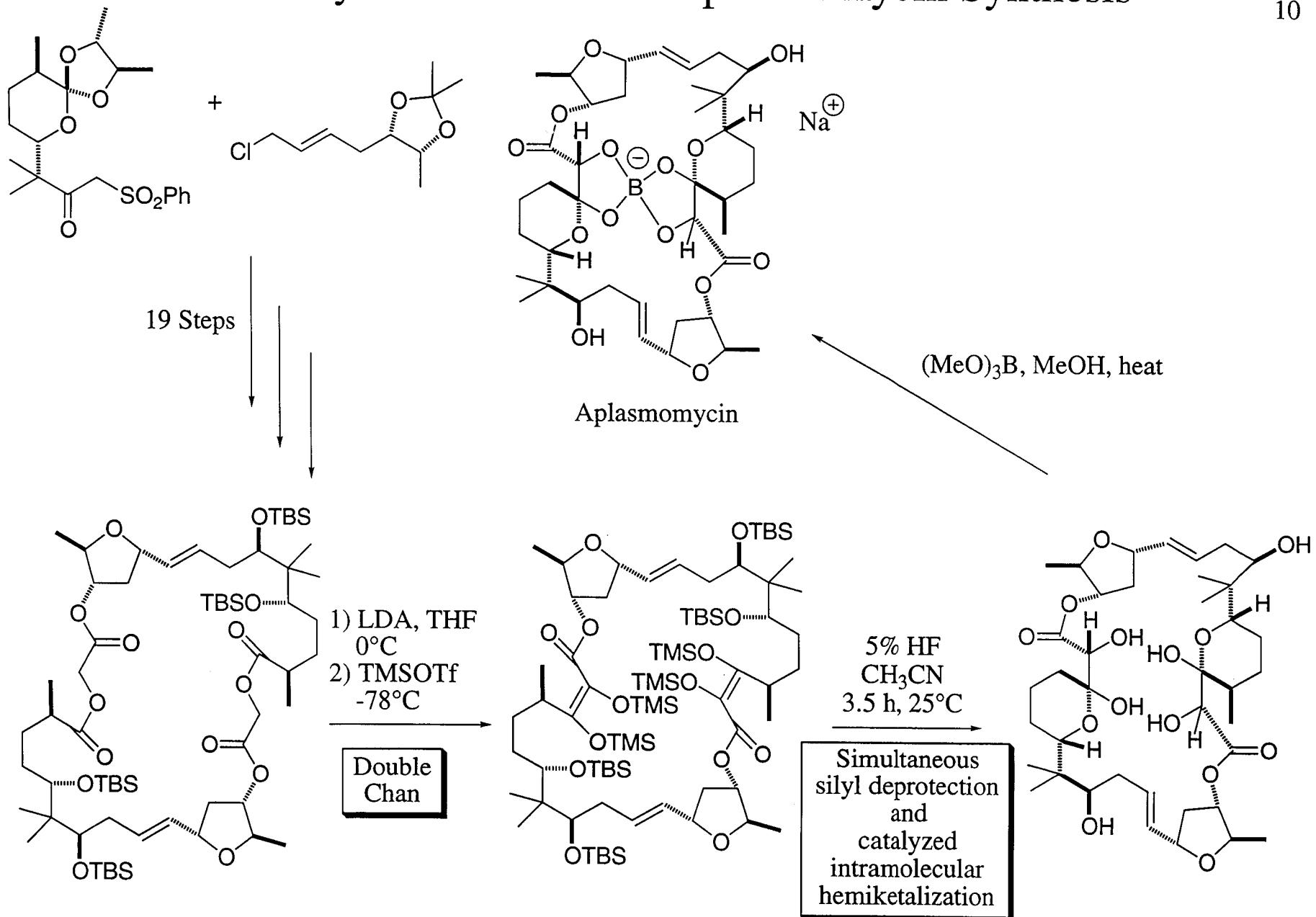


First example of cyclic system and stereoselective!

Holton, R.A.; et al. JACS (1994) 116 1597.

Macrocyclic Reduction in Aplasmomycin Synthesis

10



~~Holton, R.A.; et al. JACS (1994) 116 1597.~~
J.D. White 1794 S4eJ

Conclusion

11

- 1) The Chan reaction was discovered in 1984, but has been used in a very limited number of cases.
- 2) There is evidence for an epoxide formation mechanism, which was substantiated by gas phase studies. In addition, other reaction analogs appear to follow the same mechanistic pathway.
- 3) The reaction has been featured in only a few syntheses (aplasmomycin, boromycin, taxol, rapamycin, diazonamide A).
- 4) The Chan rearrangement has shown to proceed with retention of stereochemistry and with decent enantiomeric excesses, in cases where stereocenters are involved. (taxol subunit, acyclic imides)
- 5) Chan rearrangement products can yield useful synthetic intermediates, including tricarbonyl equivalents, α -hydroxy- β -ketoesters, polyoxazoles, and β -aminoketones.

References

Chan diene work

- Chan, T.H.; Brownridge, P. JCS CC (1979) 578.
Soriente, A.; DeRosa, M.; Villano, R.; Scettri, A. Tet. Asym. (2000) 11 2255.
Keck, G.E.; Yu, T. OL (1999) 1 289.
Brownridge, P.; Chan, T.H.; Brook, M.A.; Kang, G.J. Can. J. Chem. (1983) 61 688.

Original Chan Reaction

- Lee, S.D.; Chan., T.H.; Kwon, K.S. TL (1984) 25 3399.

Gas Phase MS of 1,2-Anionic Rearrangements

- Eichinger, P.C.H.; Hayes, R.N.; Bowie, J.H. JACS (1991) 113 1949.

N to C Acyl Migration with Imides

- Hara, O.; Ito, M.; Hamada, Y. TL (1998) 39 5537.

Chan Reaction used toward synthesis

- Rapamycin: White, J.D.; Jeffrey, S.C. JOC (1996) 61 2600.
Diazonamide A: Wipf, P.; Methot, J.-L. OL (2001) 3 1261.
Taxol: Holton, R.A.; et al (17 others). JACS (1994) 116 1597.
Aplasmomycin: White, J.D.; Vedananda, T.R.; Kang, M.; Choudhry, S.C. JACS (1986) 108 8105.
Boromycin: White, J.D.; et al (7 others). JACS (1989) 111 790.

Chan Reaction used for other reasons

- Fadnavis, N.W.; Vadivel, S.K.; Sharfuddin, M. Tet. Asym. (1999) 10 3675.

Other Ways to Make α -Hydroxy- β -Ketoesters

- Murahashi, S.; et al. (6 others). JOC (1993) 58 2929.
Araneo, S.; Clerici, A.; Porta, O. TL (1994) 35 2213.
Clerici, A.; Clerici, L.; Porta, O. Tetrahedron (1996) 52 11037.

Equilibria of Anions of α -Hydroxy- β -Diketones and α -Ketol Esters

- Rubin, M.B.; Inbar, S. JOC (1988) 53 3355.