

Interrupted Nazarov Cyclizations

Denmark Group Meeting

Jan. 29, 2008

Presenter Matthew T. Burk

Outline

History of the Nazarov cyclization

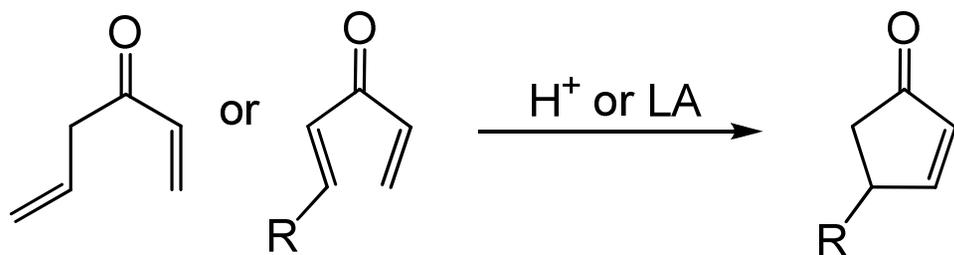
Mechanism

π - Cation Cyclizations

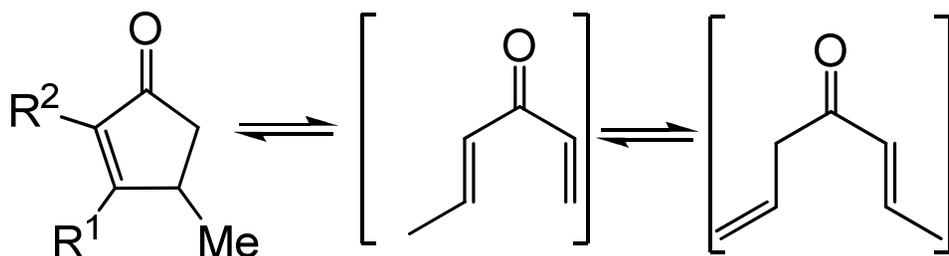
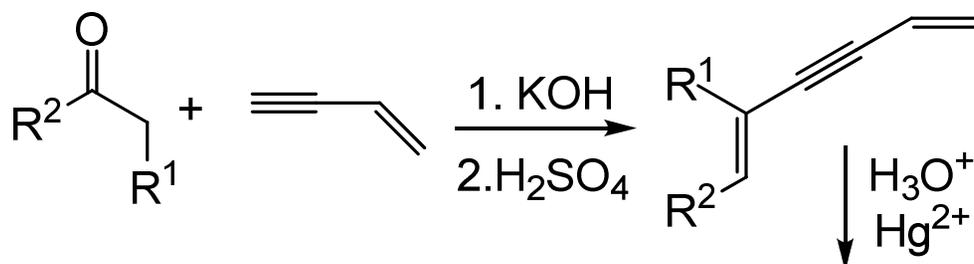
Intermolecular Trapping

Other Precursors

Introduction



Reported in 1941 by I. N. Nazarov

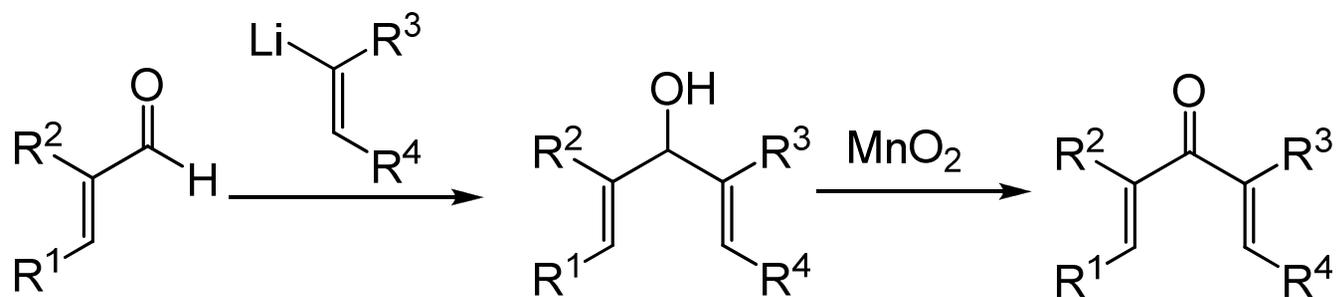


Ivan Nikolaevich Nazarov
(1906-1957)

Russ. Chem. Bull. **1957**, 1177

Nazarov, I. N.; Zaretskaya, I. I. *Izv. Akad. Nauk. SSSR, Ser. Khim.* **1941**, 211
Santelli-Rouvier, C.; Santelli, M. *Synthesis*, **1983**, 429

Typical Preparation of Starting Materials



Simple preparation of simple starting materials.

Mechanism: Cationic Intermediates

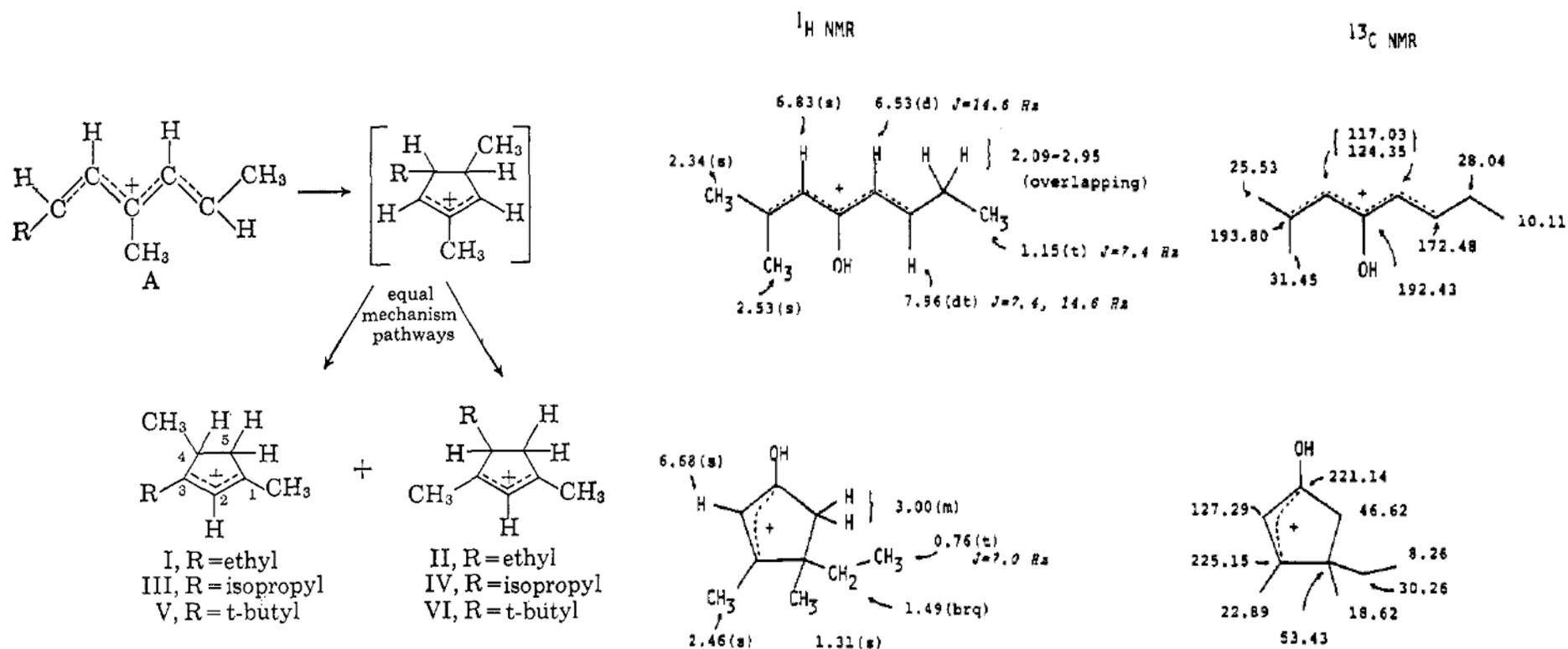
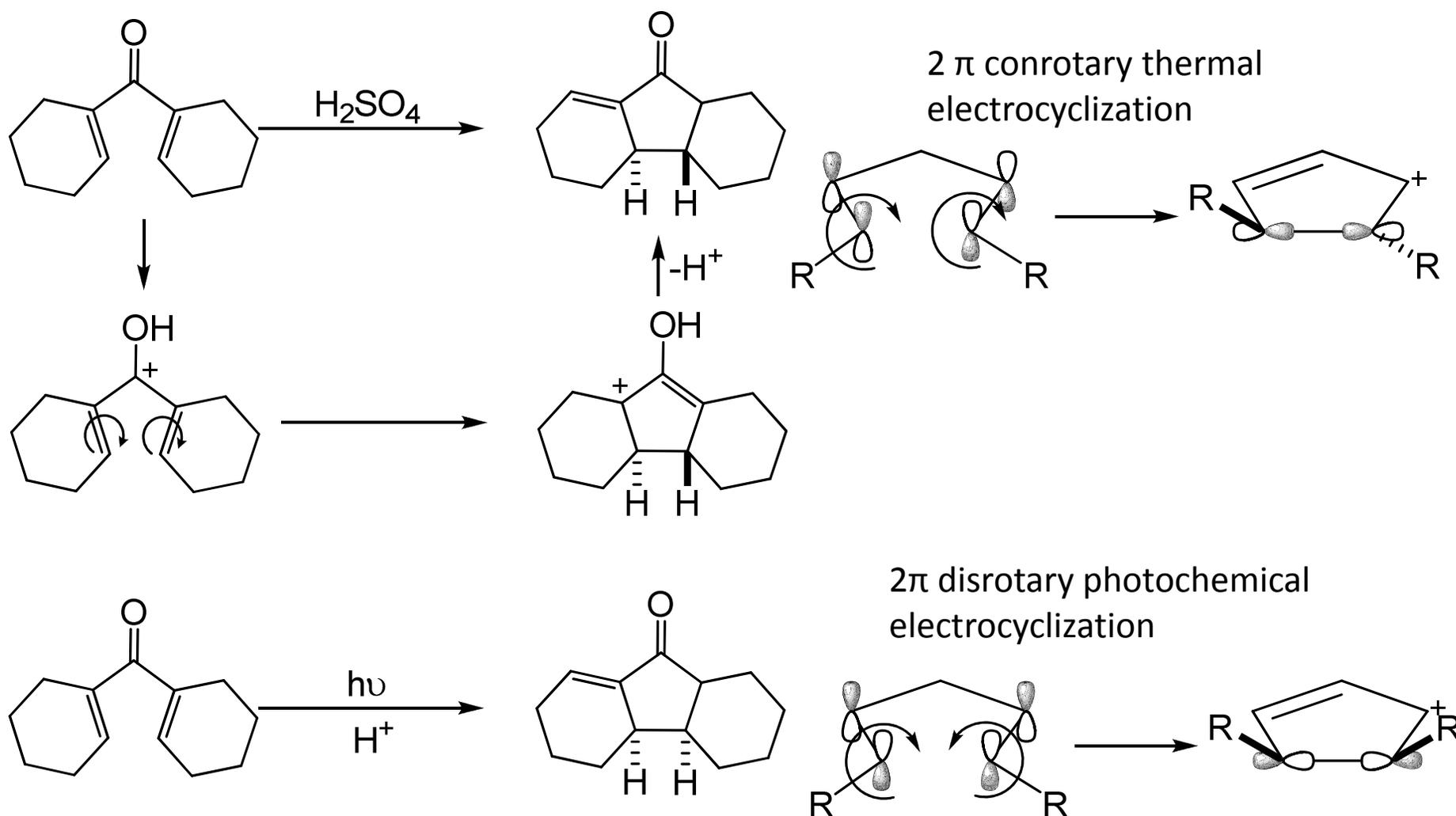
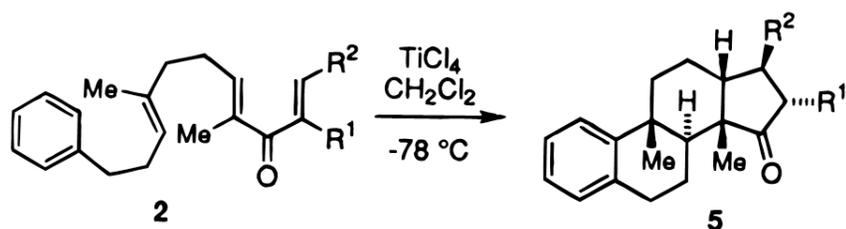


Figure 1. Chemical shifts of carbocation intermediates **2b** \rightarrow **4b** measured in concentrated H_2SO_4 . Me_4Si as an external standard.

Mechanism: Stereochemical Evidence



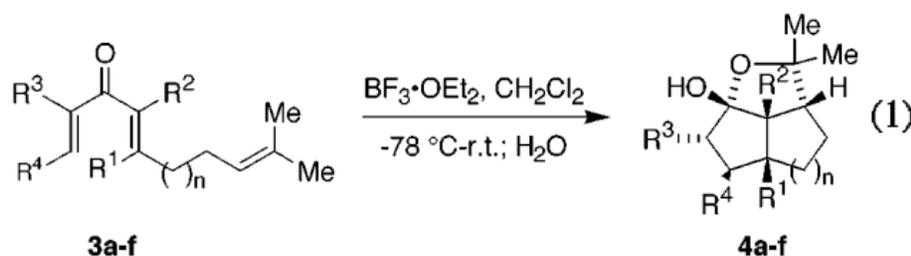
Nazarov – Initiated π -Cation Cyclizations



substrate	R ¹	R ²	product	yield (%) ^b
2a	H	H	5a	— ^c
2b	Me	H	5b	99
2c	Me	Et	5c	73 ^d
2d	—(CH ₂) ₄ —		5d	98
2e	—(CH ₂) ₅ —		5e	74 ^d

10 steps to make **2**

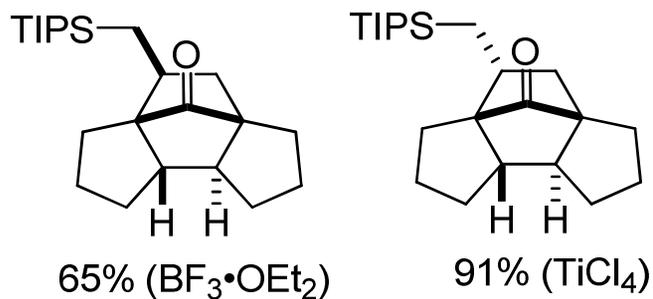
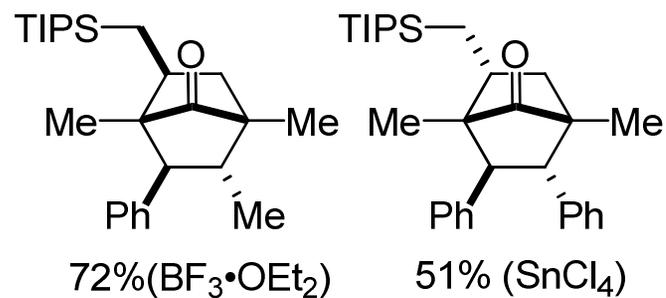
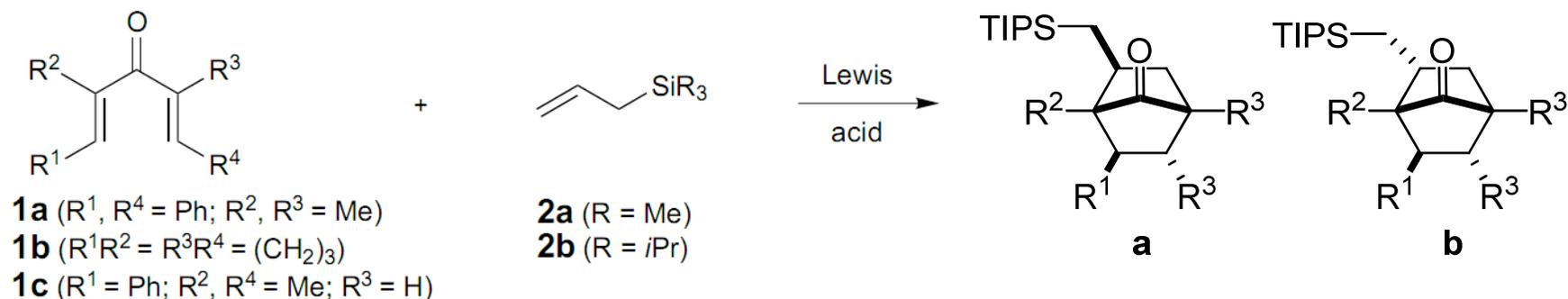
Bender, J. A.; Arif, A. M.; West, F. G. *J. Am. Chem. Soc.* **1999**, *121*, 7443



entry	dienone	R ¹	R ²	R ³	R ⁴	n	yield of 4 ^b (%)
1	3a	H	CH ₃	CH ₃	H	1	75
2	3b	H	CH ₃	C(CH ₃) ₃	H	1	89 ^c
3	3c	H	CH ₃	CH ₂ CH ₃	CH ₃	1	73
4	3d	H	CH ₃	—(CH ₂) ₄ —		1	62
5	3e	H	CH ₃	CH ₃	H	2	42 ^d
6	3f	CH ₃	H	CH ₃	H	1	e

Bender, J. A.; Blize, A. E.; Browder, C. C.; Giese, S.; West, F. G. *J. Org. Chem.* **1998**, *63*, 2430

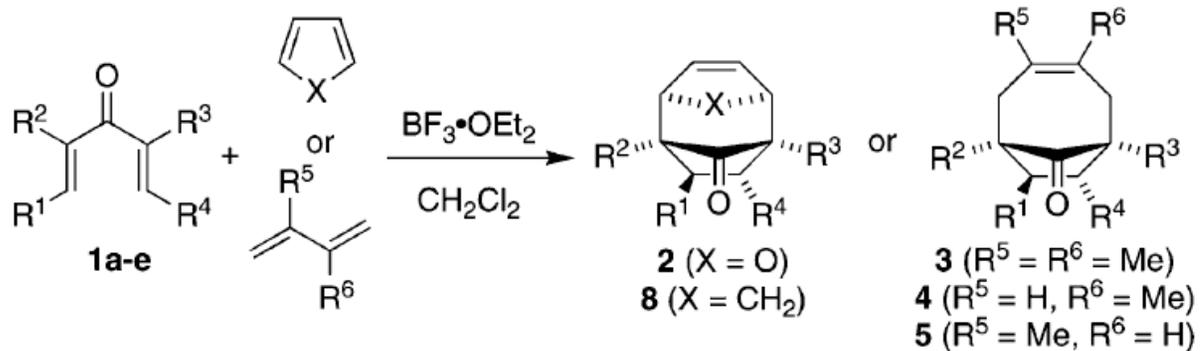
Allyl Silane Nucleophiles



TMS allyl silane gives mixtures of allylation and cycloaddition

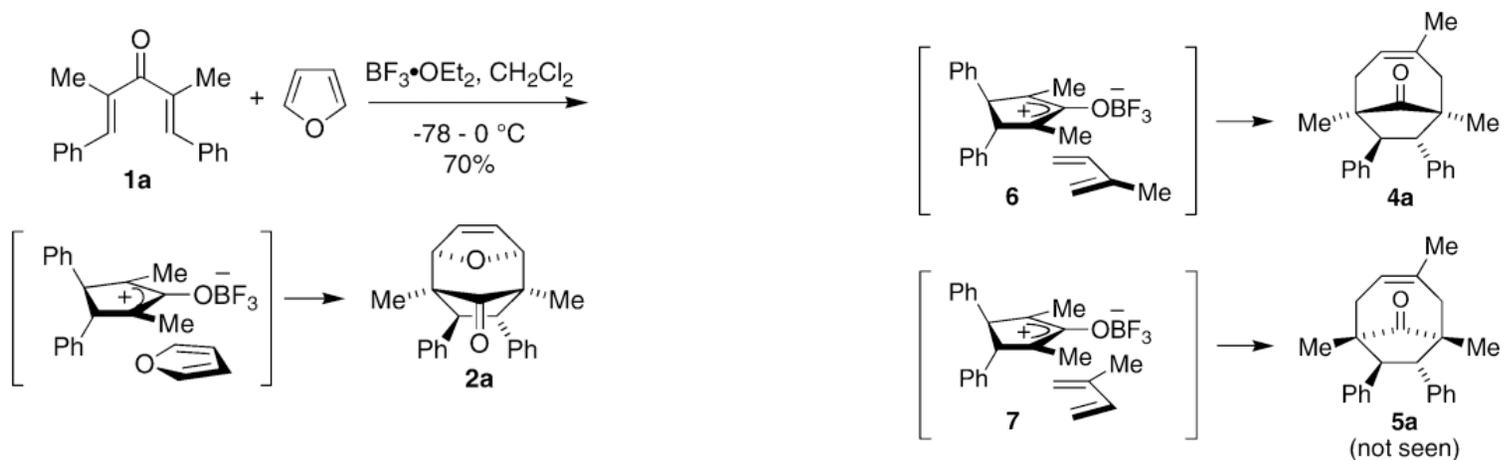
SnCl_4 favors **b**, $\text{BF}_3 \cdot \text{OEt}_2$ favors **a**

Intermolecular [4+3] Cycloadditions

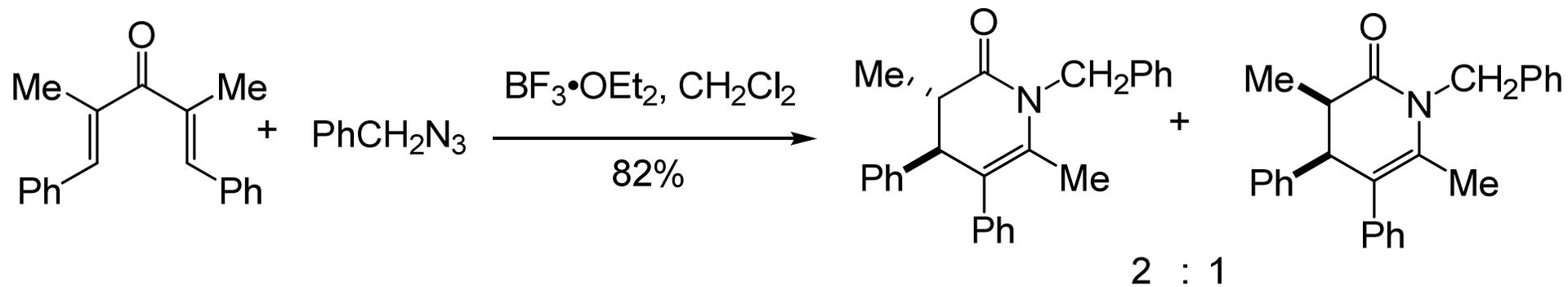


20-93% yield

Unsymmetrical dienes favor **4** mostly >4:1

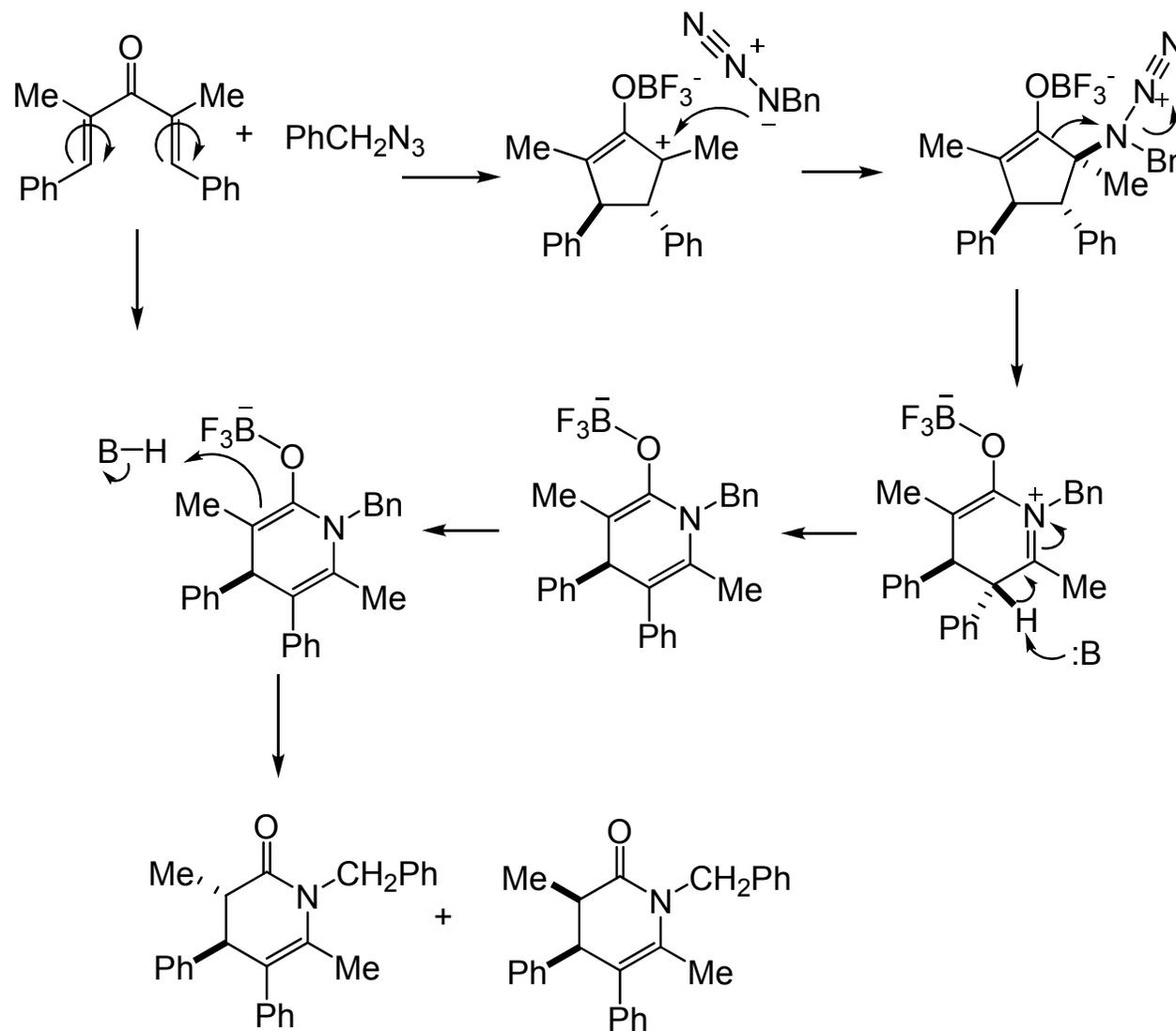


Question



Propose a mechanism for this transformation

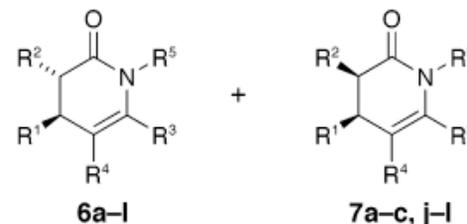
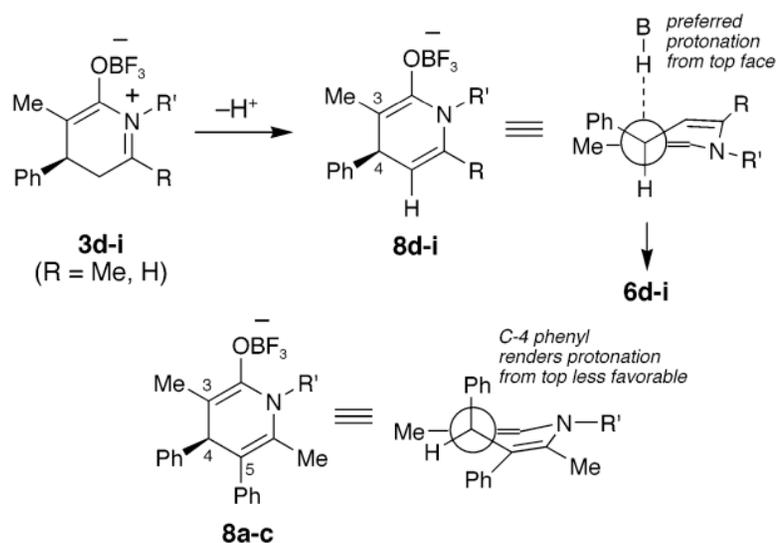
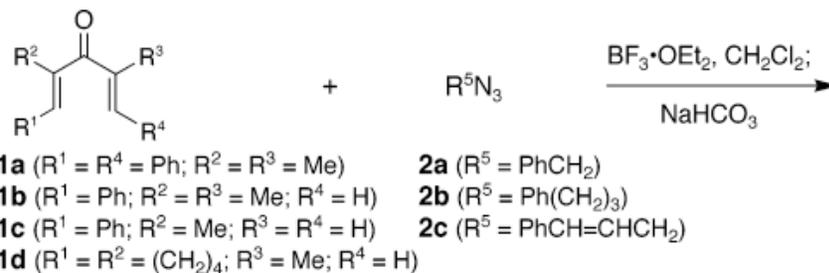
Answer



2 : 1

Song, D.; Rostami, A; West, F. G. *J. Am. Chem. Soc.* **2007**, *129*, 12019

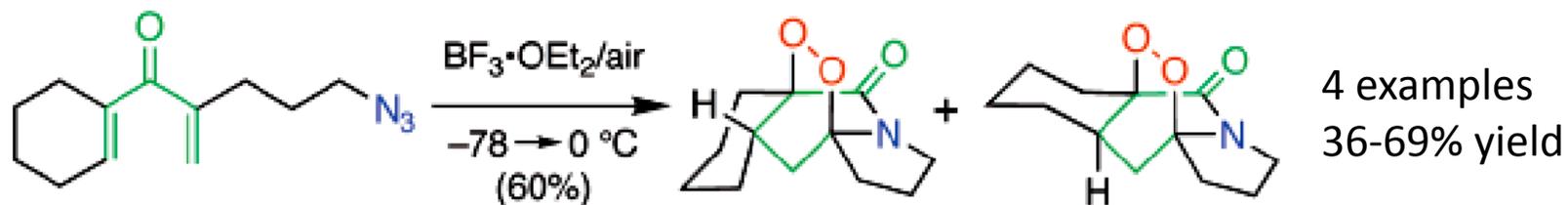
Intermolecular Azide Trapping cont'd.



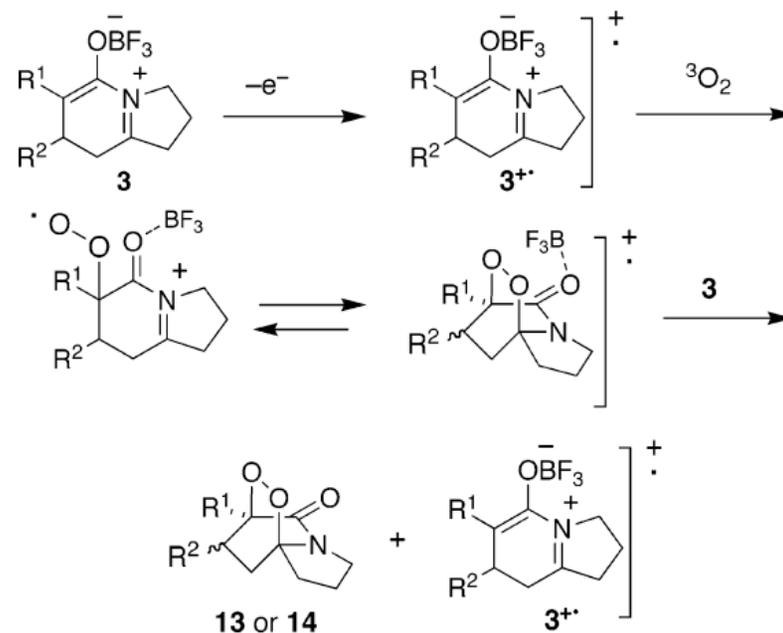
entry	dienone	azide	conditions	products (% yield; ratio) ^b
1	1a	2a	-78 °C/10 min	6a + 7a (82; 2:1)
2	1a	2b	-78 °C/10 min	6b + 7b (78; 2.3:1)
3	1a	2c	-78 °C/10 min	6c + 7c (85; 2:1)
4	1b	2a	-78 °C/0.5 h	6d (75)
5	1b	2b	-78 °C/0.5 h	6e (80)
6	1b	2c	-78 °C/0.5 h	6f (72)
7	1c	2a	0 °C/1 h	6g (62)
8	1c	2b	0 °C/1 h	6h (40)
9	1c	2c	0 °C/1 h	6i (43)
10	1d	2a	0 °C/0.5 h	6j + 7j (80; 2:1)
11	1d	2b	0 °C/0.5 h	6k + 7k (70; 3:1)
12	1d	2c	0 °C/0.5 h	6l + 7l (73; 2.5:1)

^a See Experimental Section for standard procedure. ^b All yields given are based on isolated material after chromatographic purification.

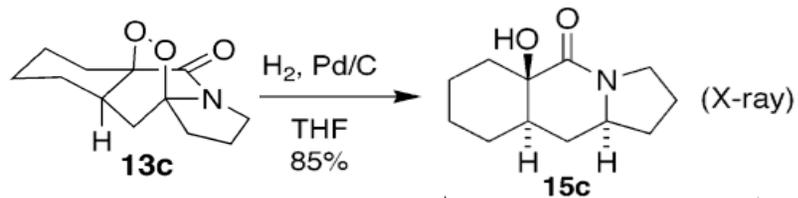
Intramolecular Azide capture



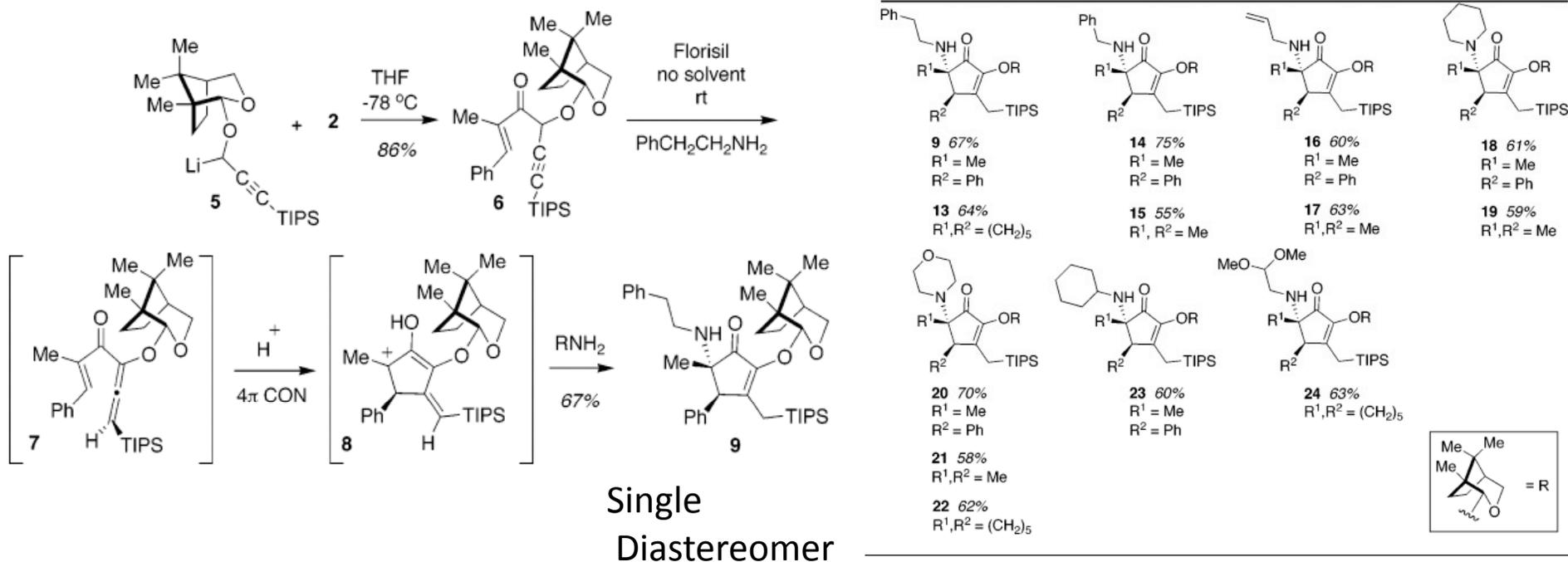
Scheme 4. Cation Radical Chain Mechanism for Peroxide Formation



Scheme 5. Peroxide Hydrogenolysis

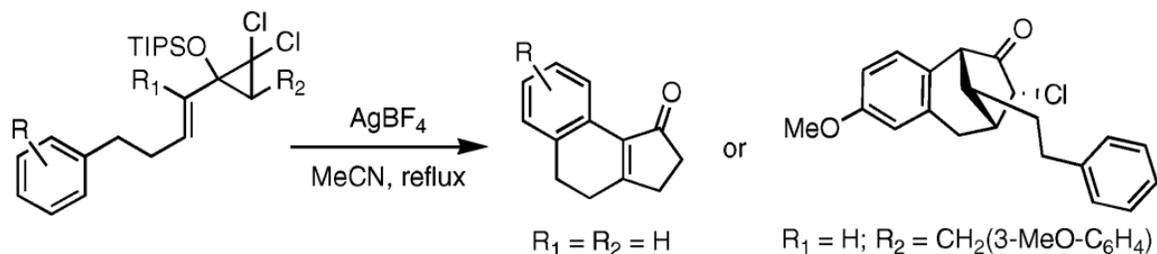


Asymmetric Amine Intercepted Nazarov Cyclization

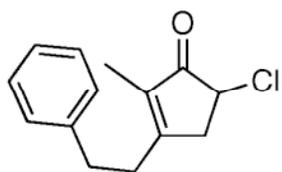


Auxiliary derived from camphor

Electrocyclic Ring Opening/ Interrupted Nazarov

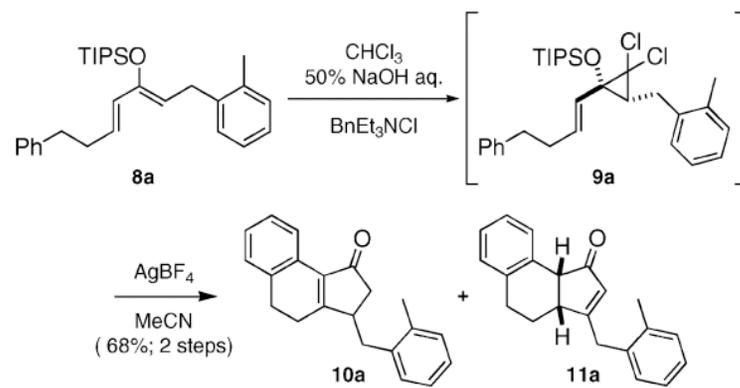
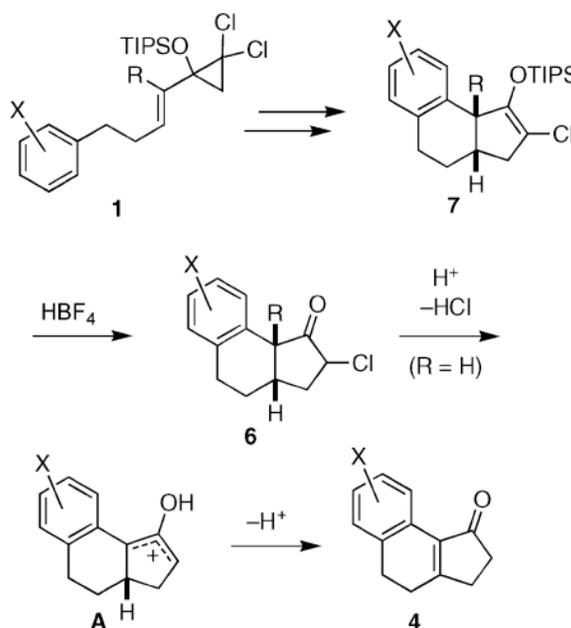


$R^1 = \text{Me}$



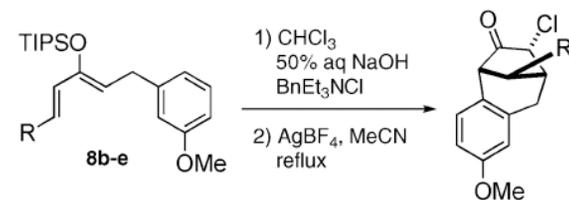
97%

Scheme 2. Proposed Mechanism of Arene Trapping and Elimination

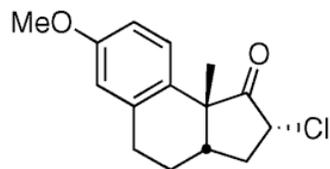


10a

11a



12b: $R = \text{CH}_2\text{CH}_2\text{Ph}$ (57%)
12c: $R = \text{CH}(\text{CH}_3)_2$ (68%)
12d: $R = \text{CH}_2\text{CH}_3$ (81%)
12e: $R = \text{Ph}$ (59%)



89%

Conclusions

Many carbocyclic structures are accessible by interrupted Nazarov cyclization methods

Accessibility of starting materials is an issue for intramolecular cyclization methods.

Reviews:

Santelli-Rouvier, C.; Santelli, M. *Synthesis*, **1983**, 429

Habermas, K. L.; Denmark, S. E.; Jones, T. K. *Org. React.* **1994**, *45*, 1-158

Pellissier, H. *Tetrahedron* **2005**, *61*, 6479

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