

Carbonyl-Olefin Metathesis

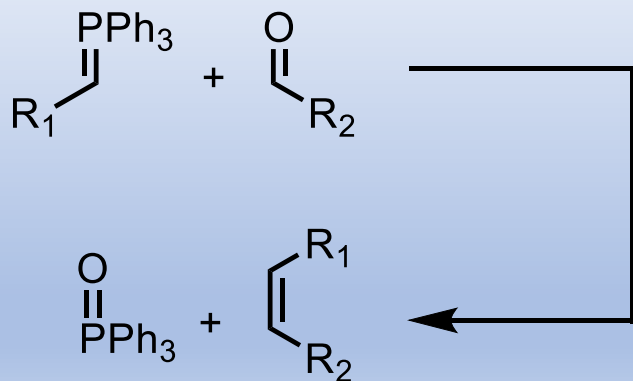
Vincent Kassel

4-20-2021

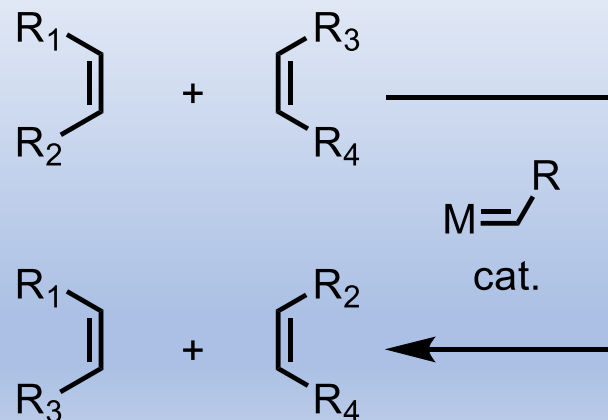


Double Bond Metathesis in Organic Synthesis

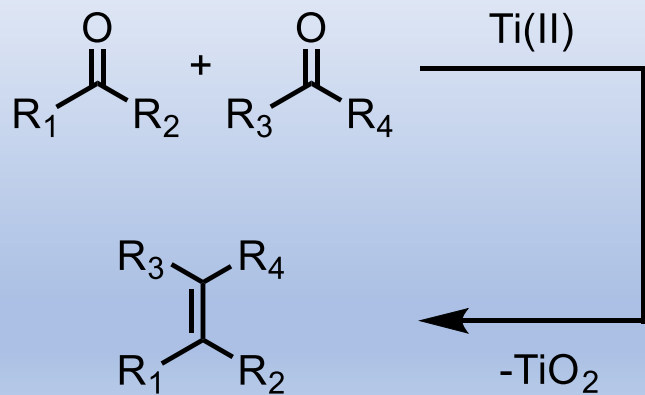
Wittig Olefination:



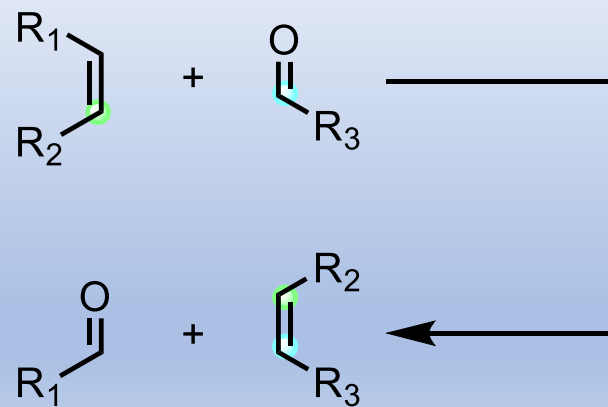
Olefin-Olefin Metathesis:



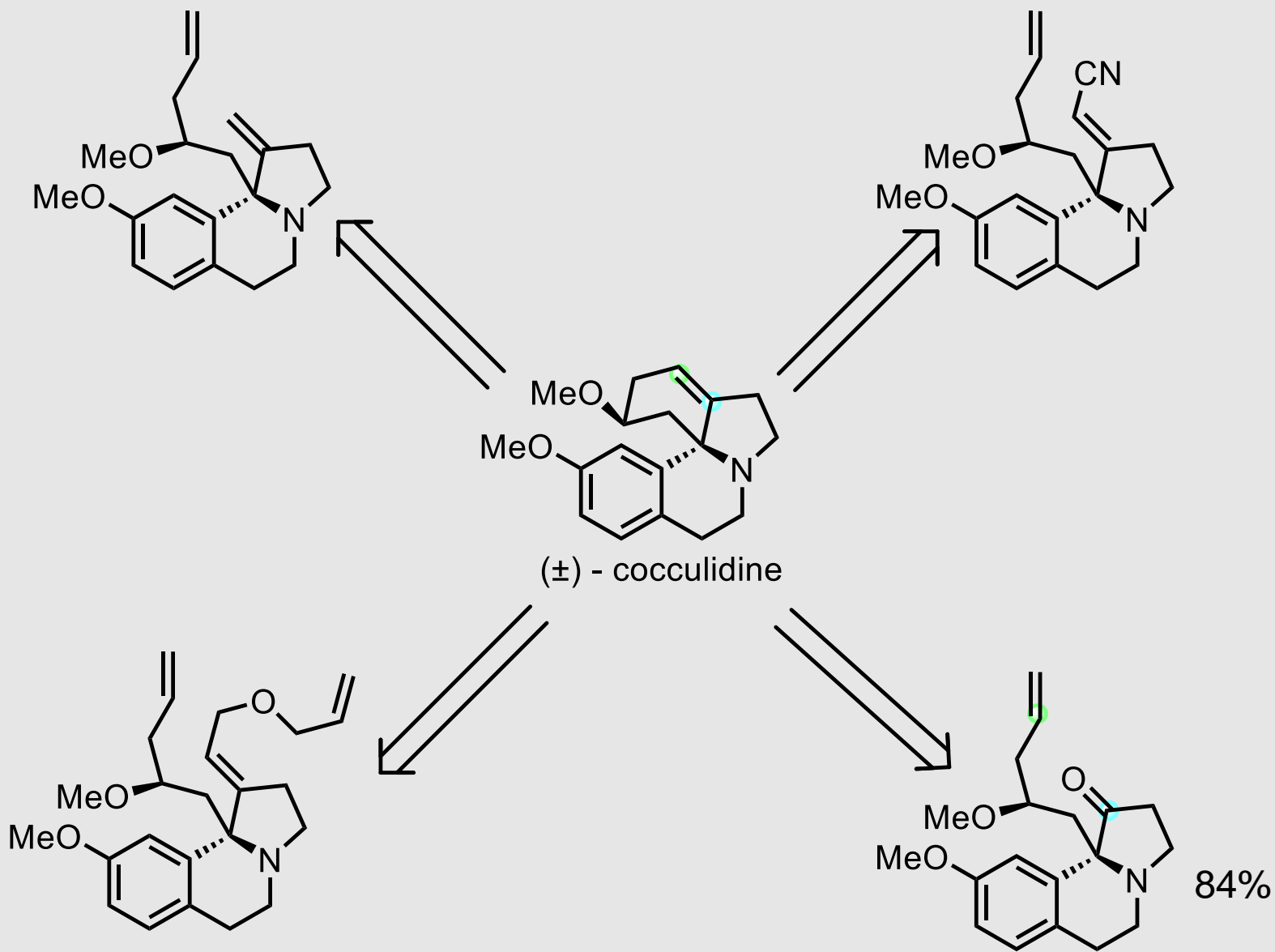
McMurry Coupling: (not formally metathesis)



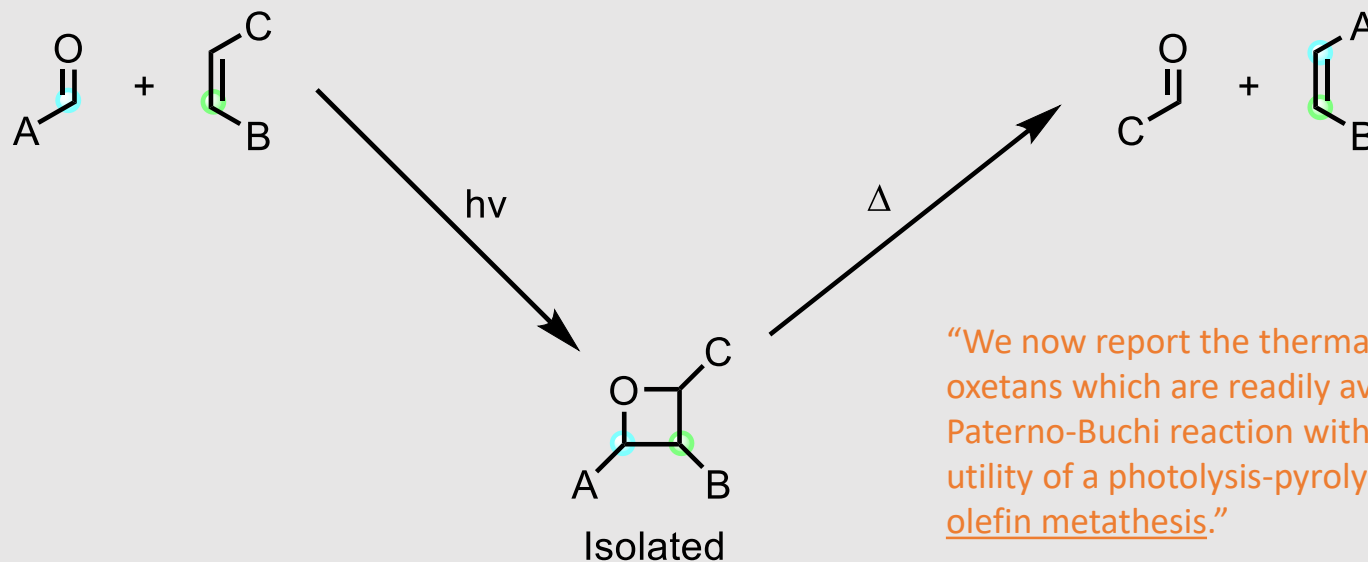
Carbonyl-Olefin Metathesis:



New Disconnections In The Chemist's Toolbox

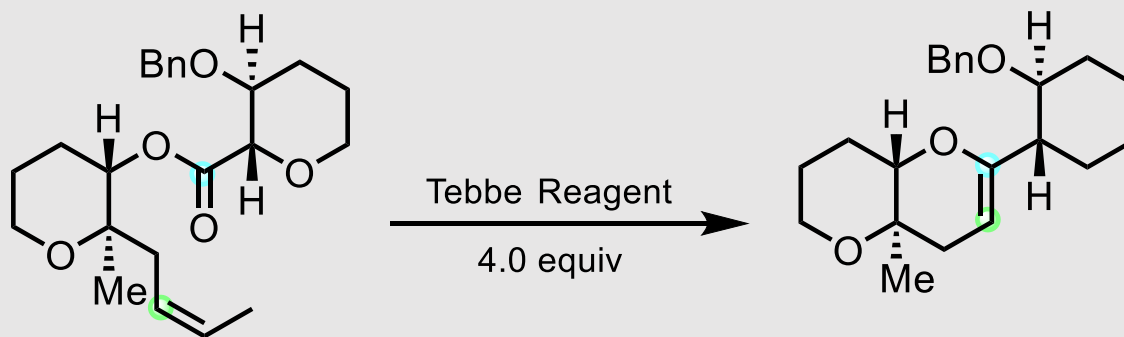


“Formal” vs “Actual” Carbonyl-Olefin Metathesis



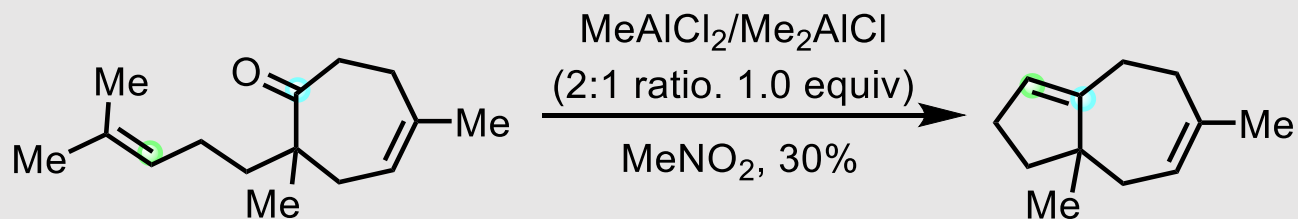
“We now report the thermal transformations of some oxetans which are readily available *via* the photochemical Paterno-Buchi reaction with emphasis on the synthetic utility of a photolysis-pyrolysis sequence involving carbonyl-olefin metathesis.”

Jones II, G. et al. *J. Chem. Soc. Chem. Commun.* **1973**, 374–375.



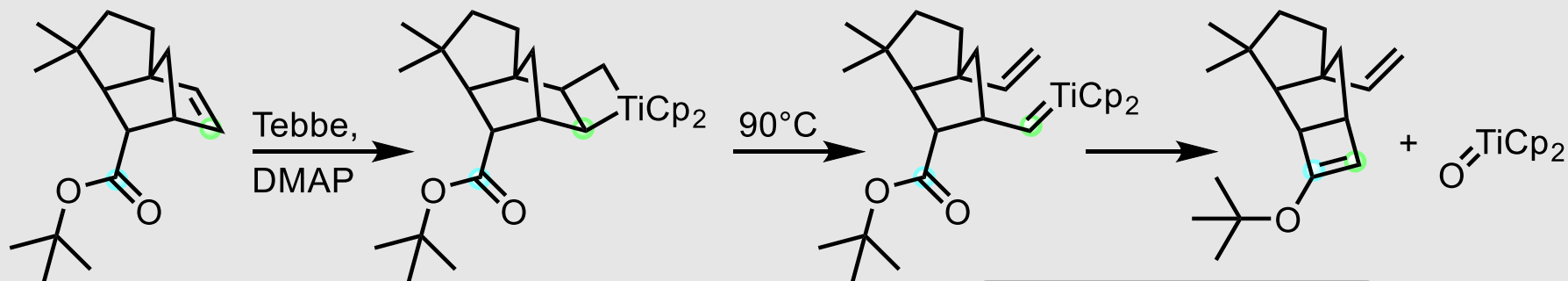
Nicolaou, K.C. et al. *J. Am. Chem. Soc.* **1996**, *118*, 10335-10336.

Early Literature Examples



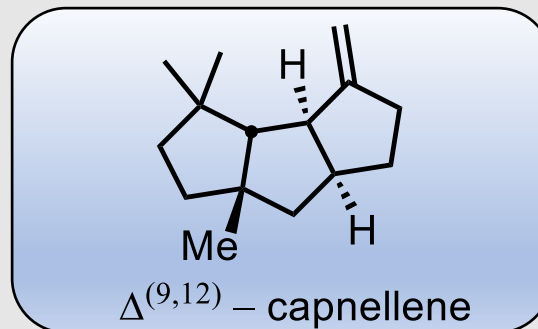
“The formation of the unusual metathesis product presumably results from stepwise cycloaddition to give an oxetane which then opens in the other direction and loses acetone”

Snider, B.B. et al. *J. Org. Chem.* **1984**, *49*, 3988-3994.



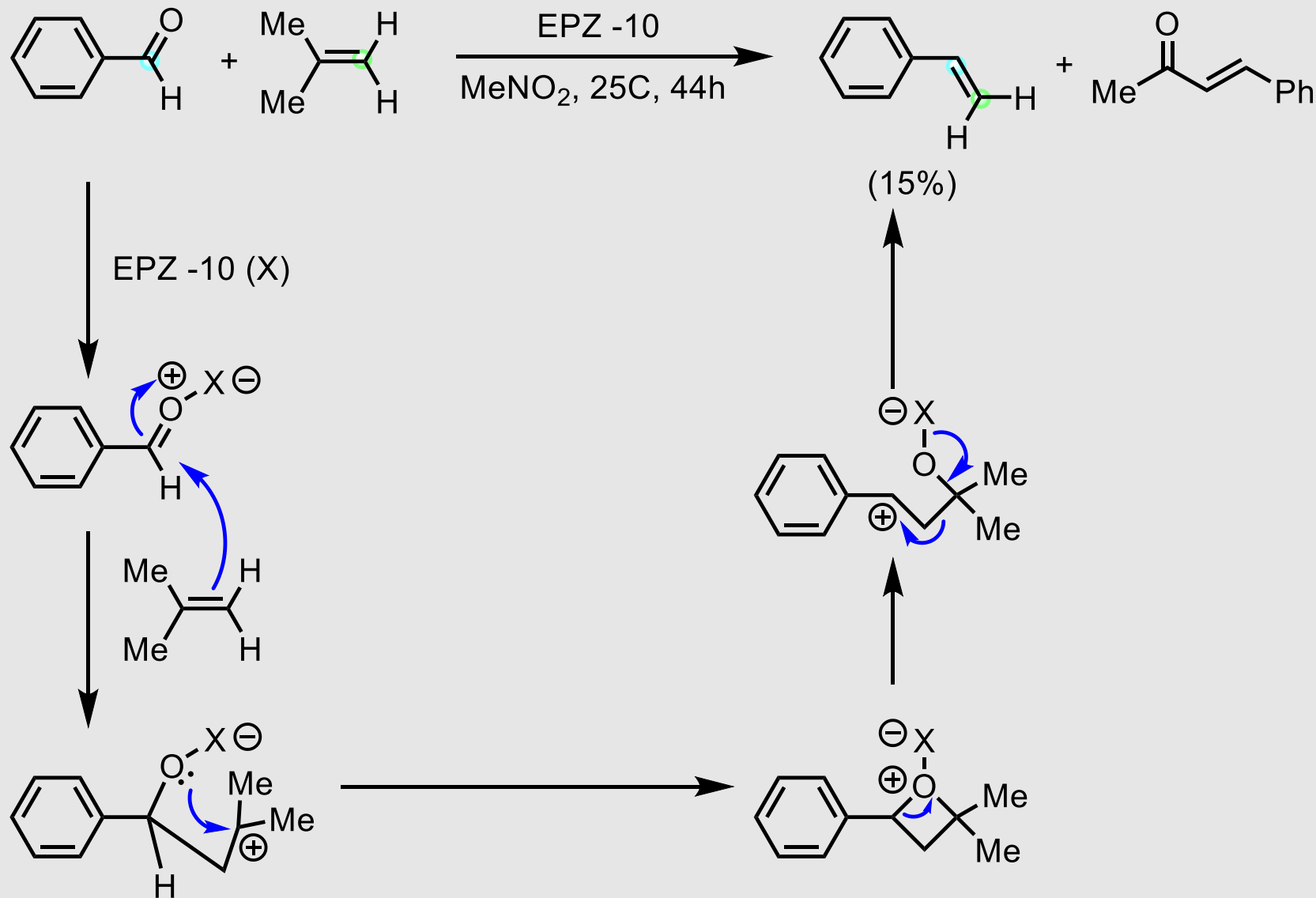
Grubbs R. H. et al. *J. Am. Chem. Soc.* **1986**, *108*, 855-856.

Grubbs R. H. et al. *J. Org. Chem.* **1990**, *55*, 843-862.

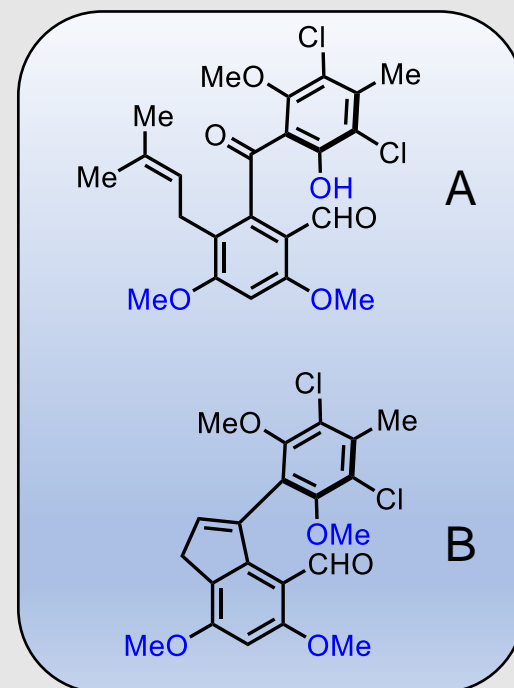
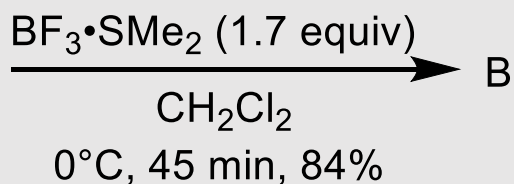
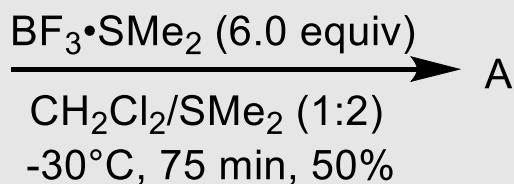
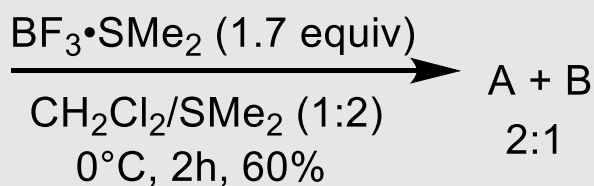
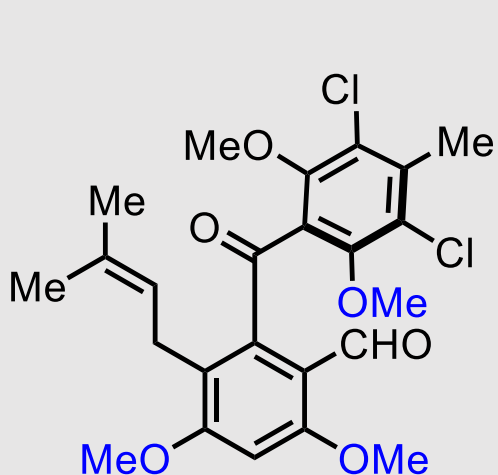
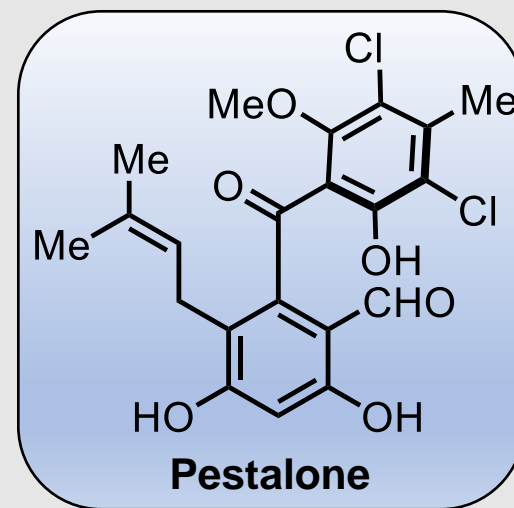
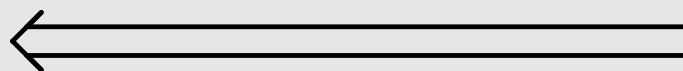
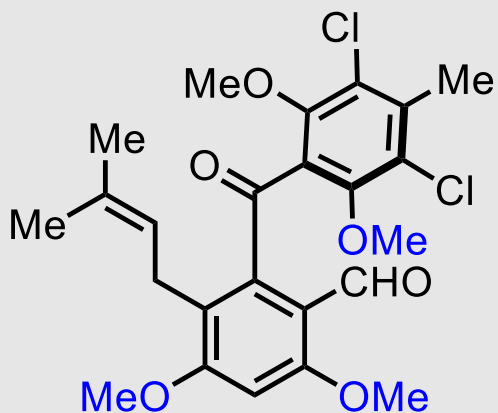


Acid-Catalyzed Olefination of Benzaldehyde

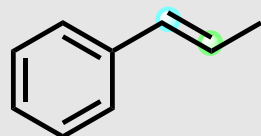
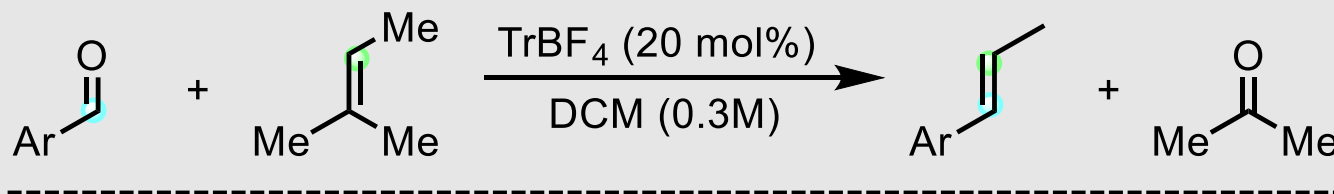
Propose a Mechanism for this Transformation:



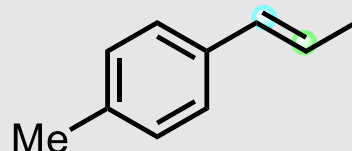
Novel Observation in the Synthesis of Pestalone



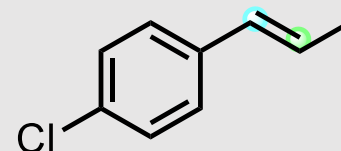
Trans-Selective Carbocation-Catalyzed Olefination of Aldehydes



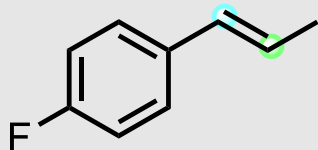
25°C, 30h, 68%



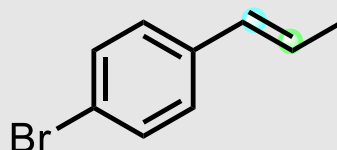
-78°C, 100h, 74%



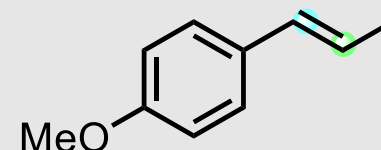
25°C, 30h, 54%



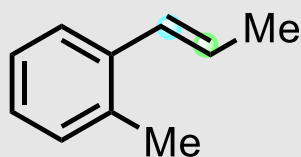
25°C, 30h, 59%



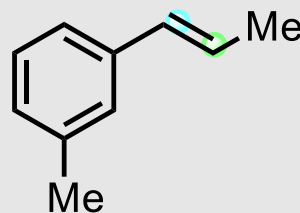
25°C, 30h, 55%



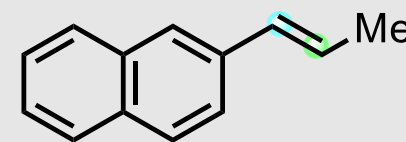
-78°C, 30h, 15%



25°C, 30h, 85%



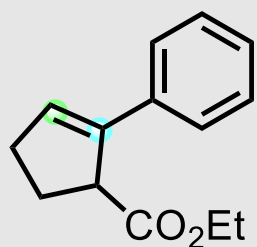
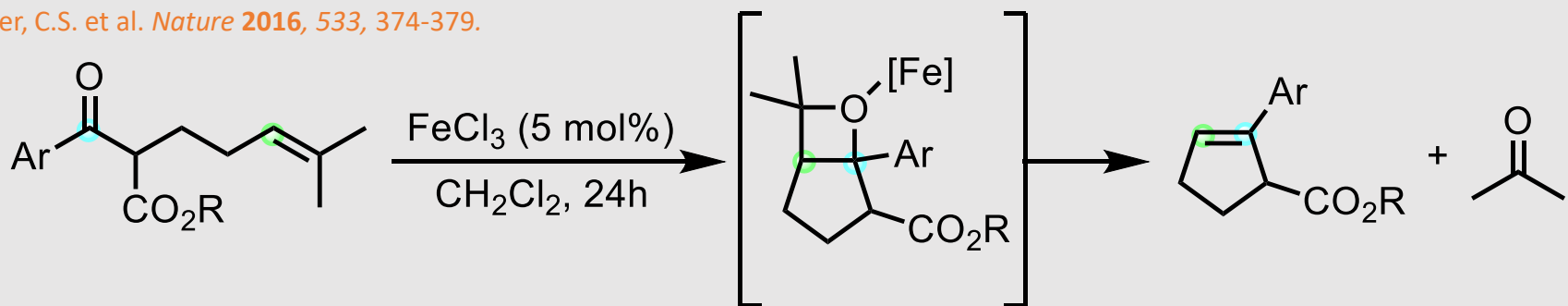
25°C, 30h, 83%



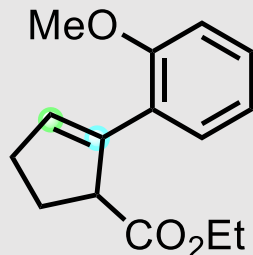
25°C, 30h, 60%

Fe(III)-Mediated Carbonyl-Olefin Metathesis (Schindler)

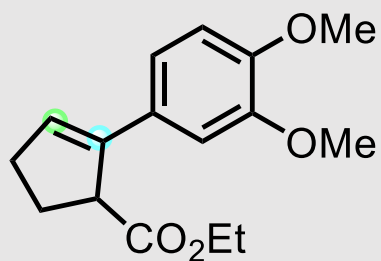
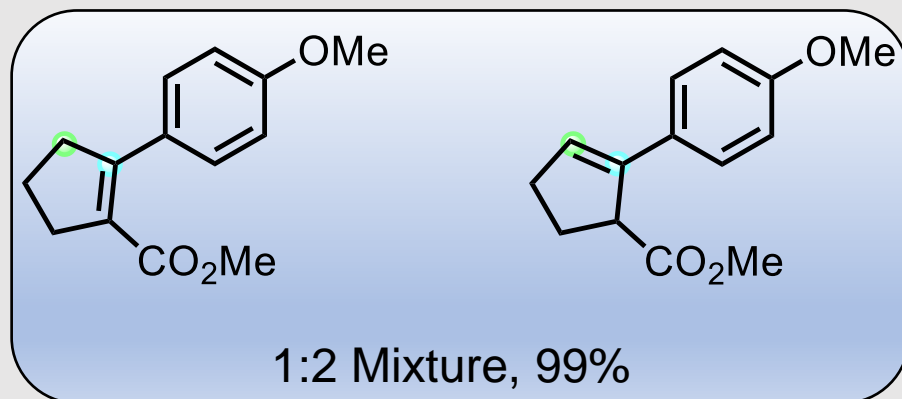
Shindler, C.S. et al. *Nature* 2016, 533, 374-379.



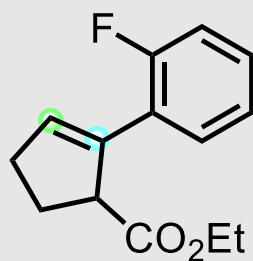
99%



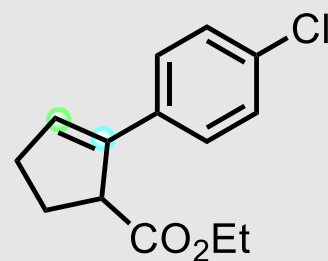
99%



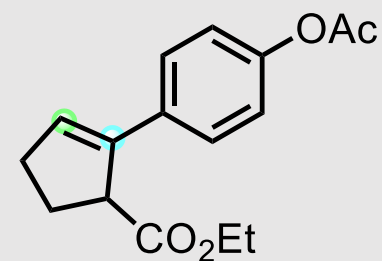
66%



99%



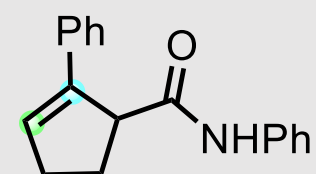
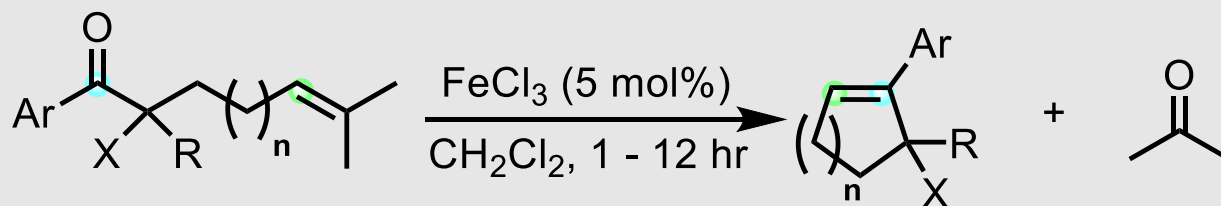
55% r.t.; 99% reflux



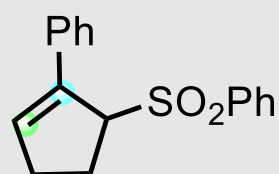
57%

Fe(III)-Mediated Carbonyl-Olefin Metathesis (Schindler)

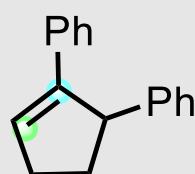
Shindler, C.S. et al. *Nature* 2016, 533, 374-379.



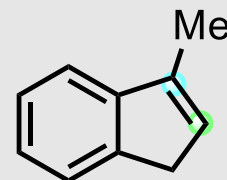
57%
(100% FeCl₃)



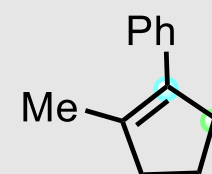
66%



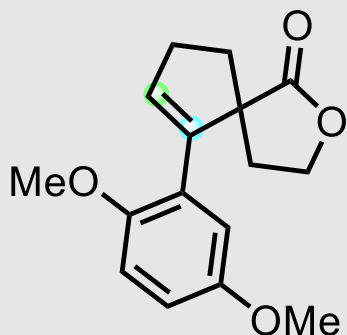
43%



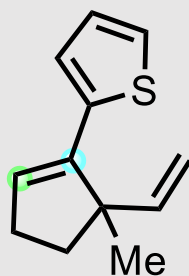
93%



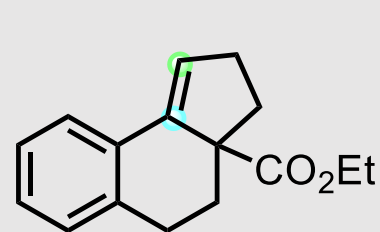
52%



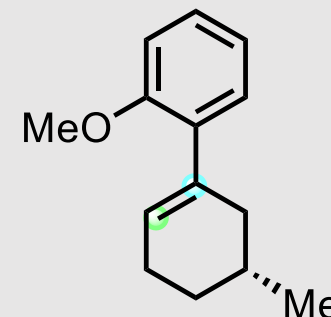
68%



76%



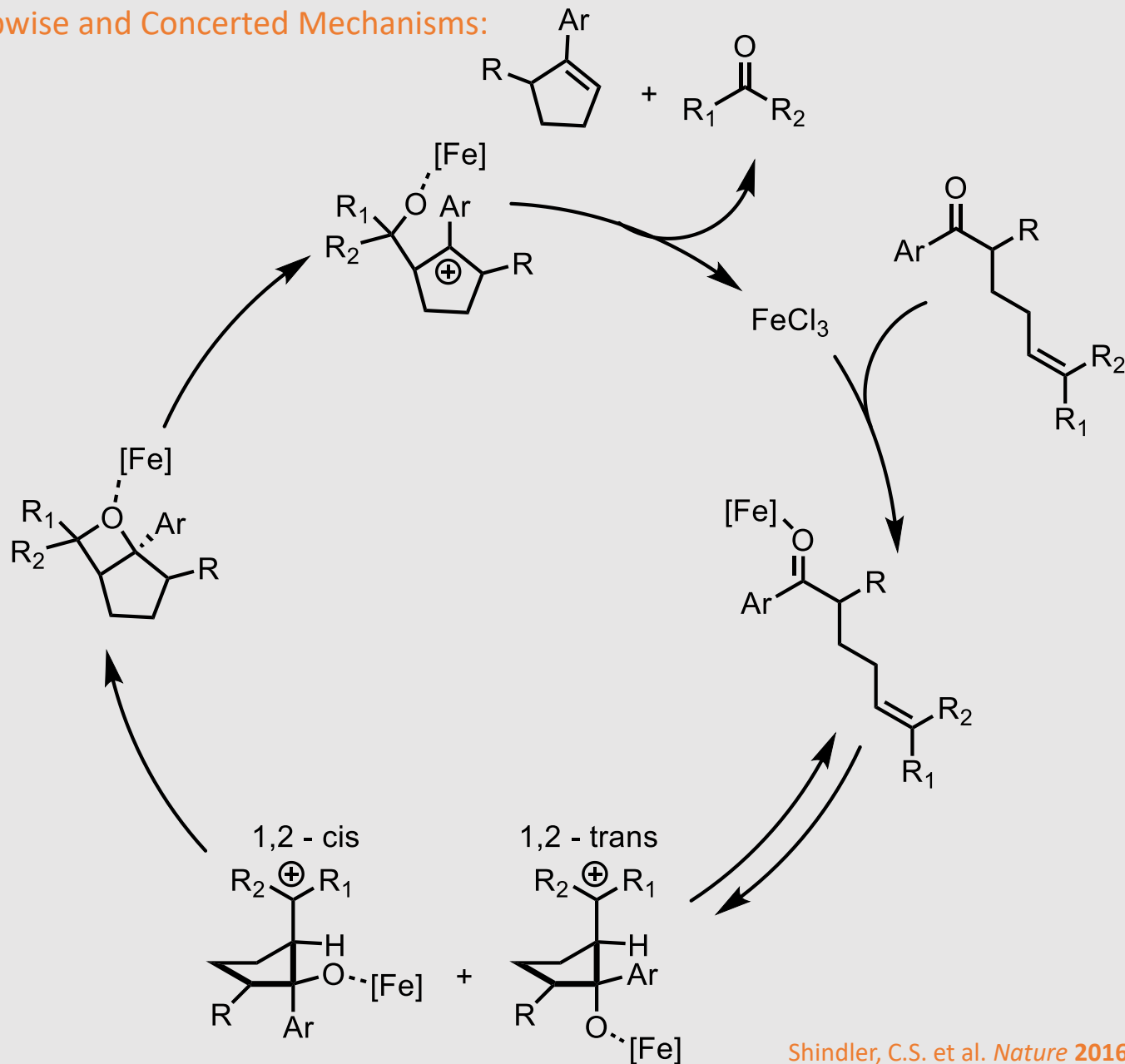
85%



71%

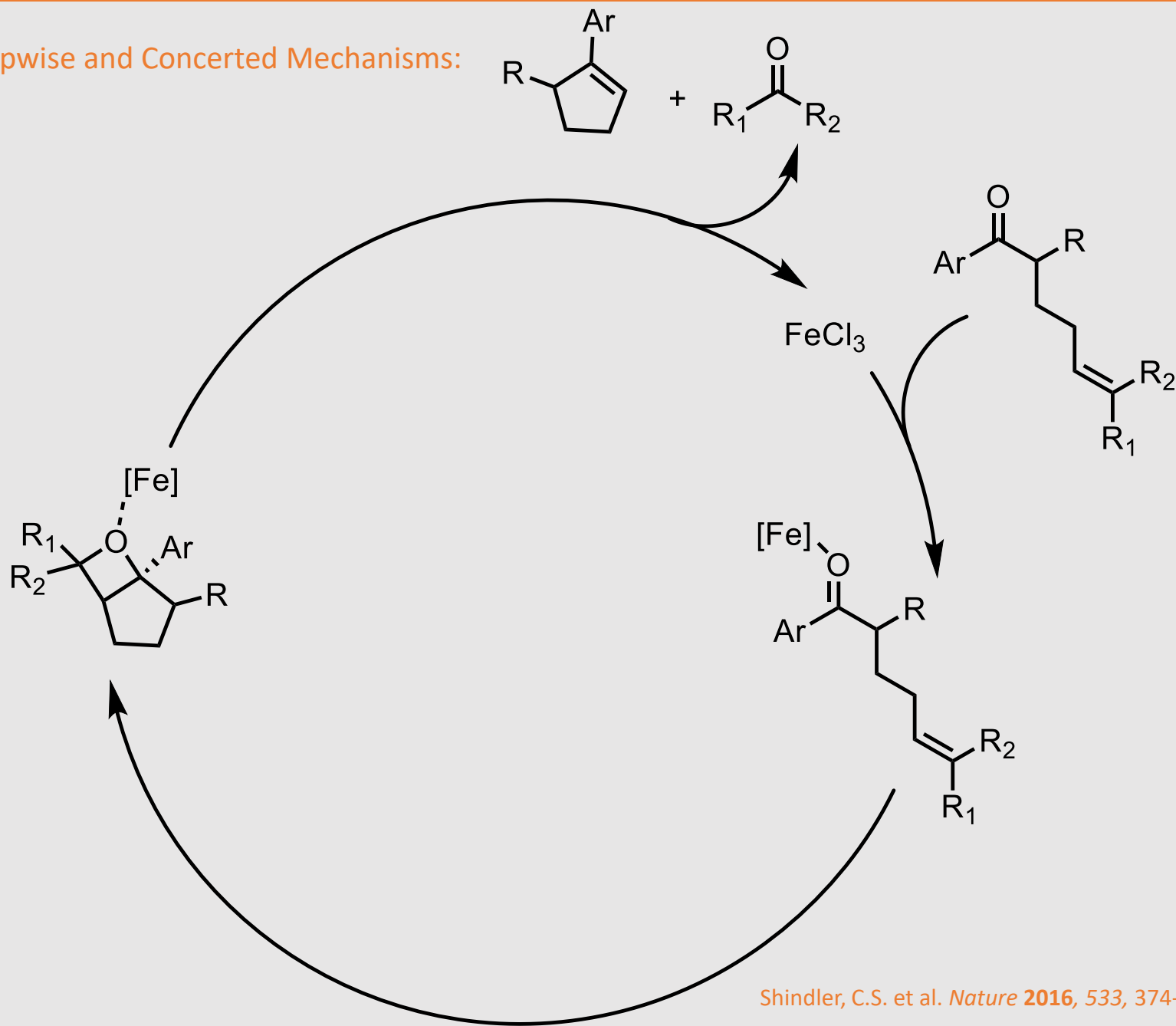
Fe(III)-Mediated Carbonyl-Olefin Metathesis (Schindler)

Proposed Stepwise and Concerted Mechanisms:



Fe(III)-Mediated Carbonyl-Olefin Metathesis (Schindler)

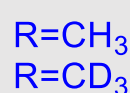
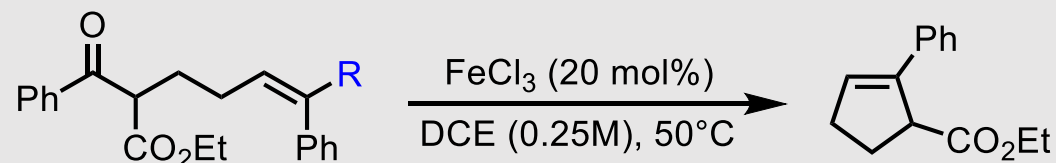
Proposed Stepwise and Concerted Mechanisms:



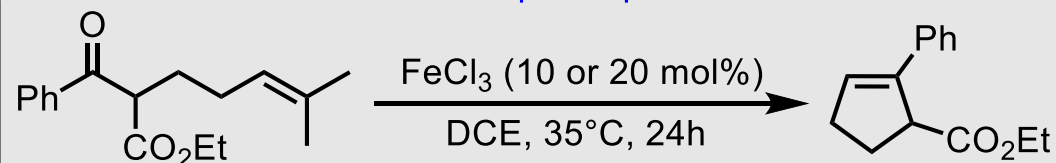
Further Mechanistic Investigations

Takeaway Observations:

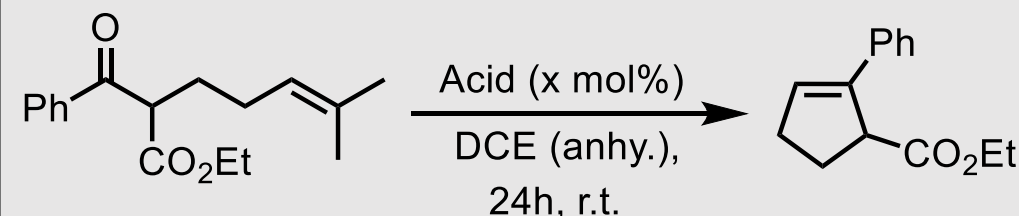
- Success of metathesis is related to Lewis acidity of metal catalyst, Bronsted acids not operative.
- EPR rules out SET events by iron and suggests binding at carbonyl(s).
- Saturation kinetics observed for iron at 10 and 20 mol% FeCl₃ (i.e. turnover-limiting step is impacted by [FeCl₃]).
- The reaction displays an inverse β-SKIE (oxetane formation rate-limiting).
- The π-system of the aryl group facilitates delocalization of the HOMO, conserving orbital symmetry.
- Oxetane fragmentation found to depend on pendant olefin identity.



$K_H/K_D = 0.65 \pm 0.07$
 $sp^2 \text{ to } sp^3$



Rate order of FeCl₃ = 1.1 ± 0.2



10 mol% FeCl₃ (97%)

99%

10 mol% Anhy. FeCl₃ (98%)

99%

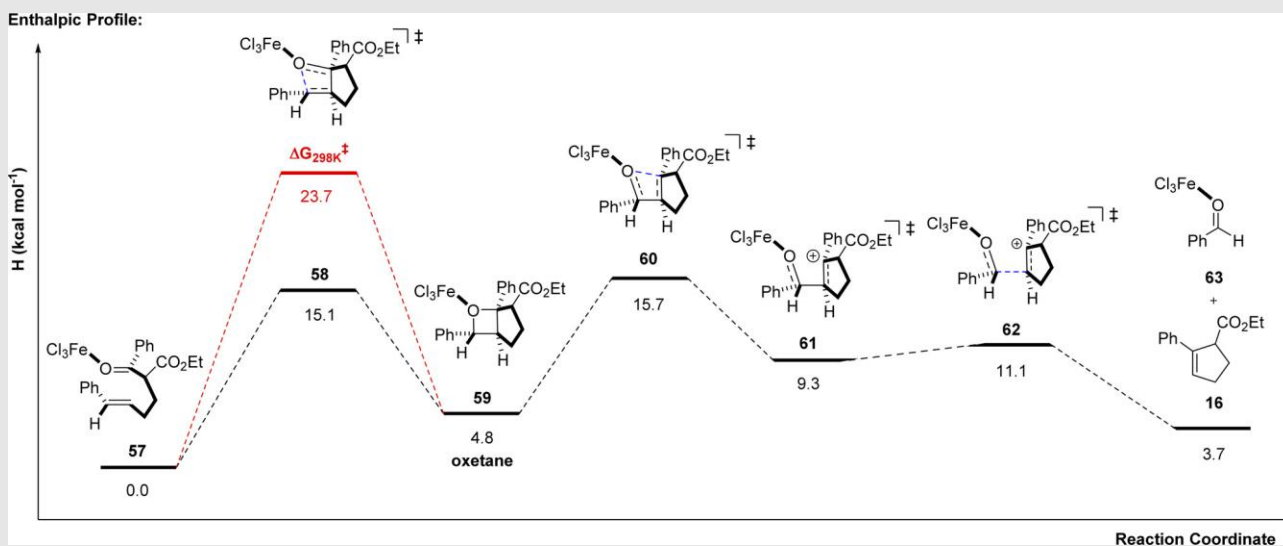
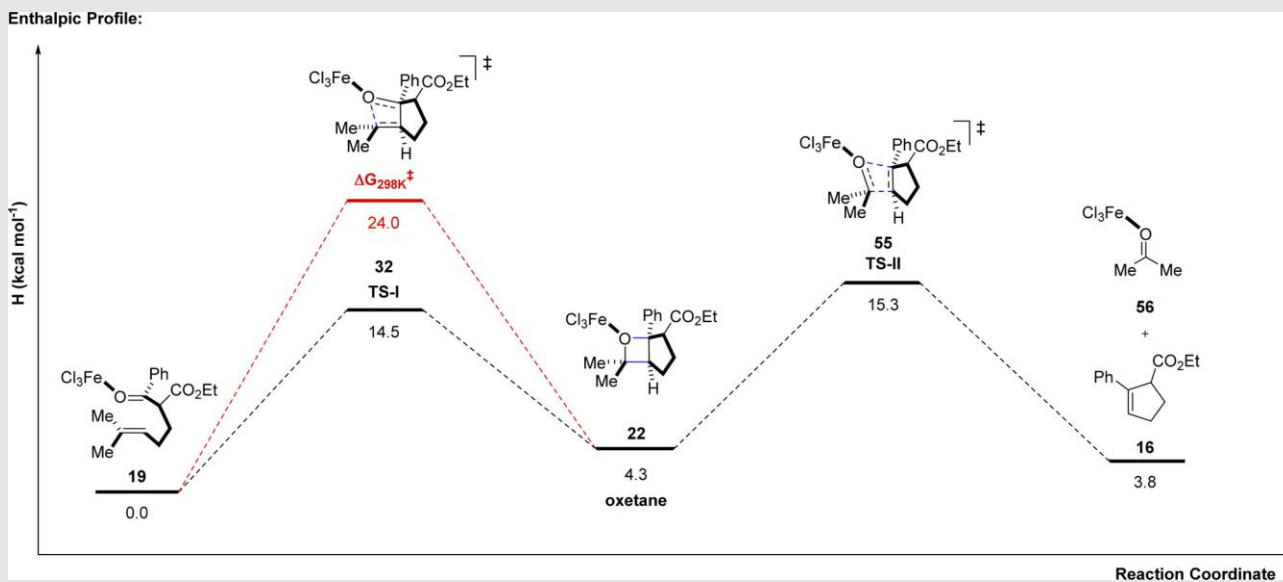
5 mol% HCl in DCE

0%

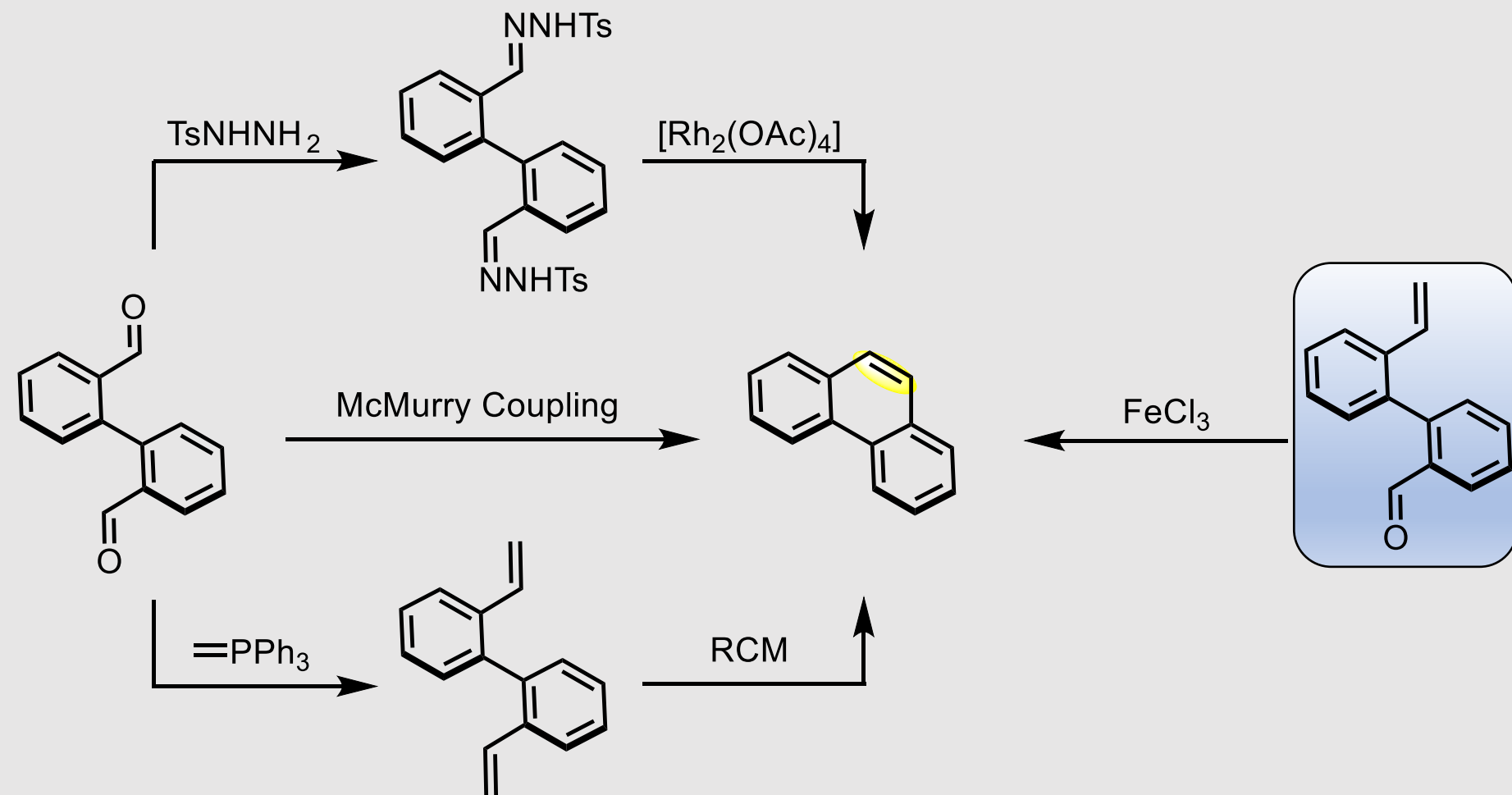
100 mol% HCl in DCE

0%

Fe(III)-Mediated Carbonyl-Olefin Metathesis (Schindler)



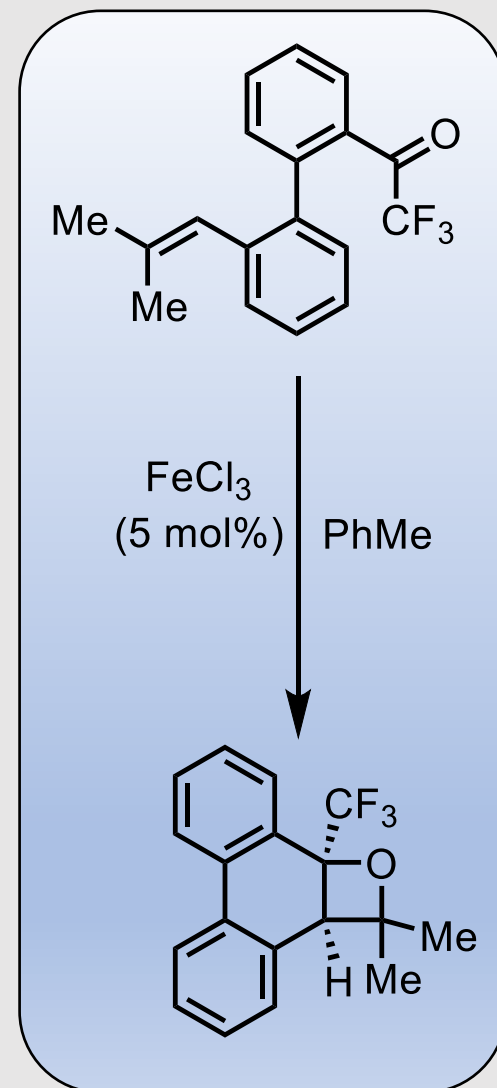
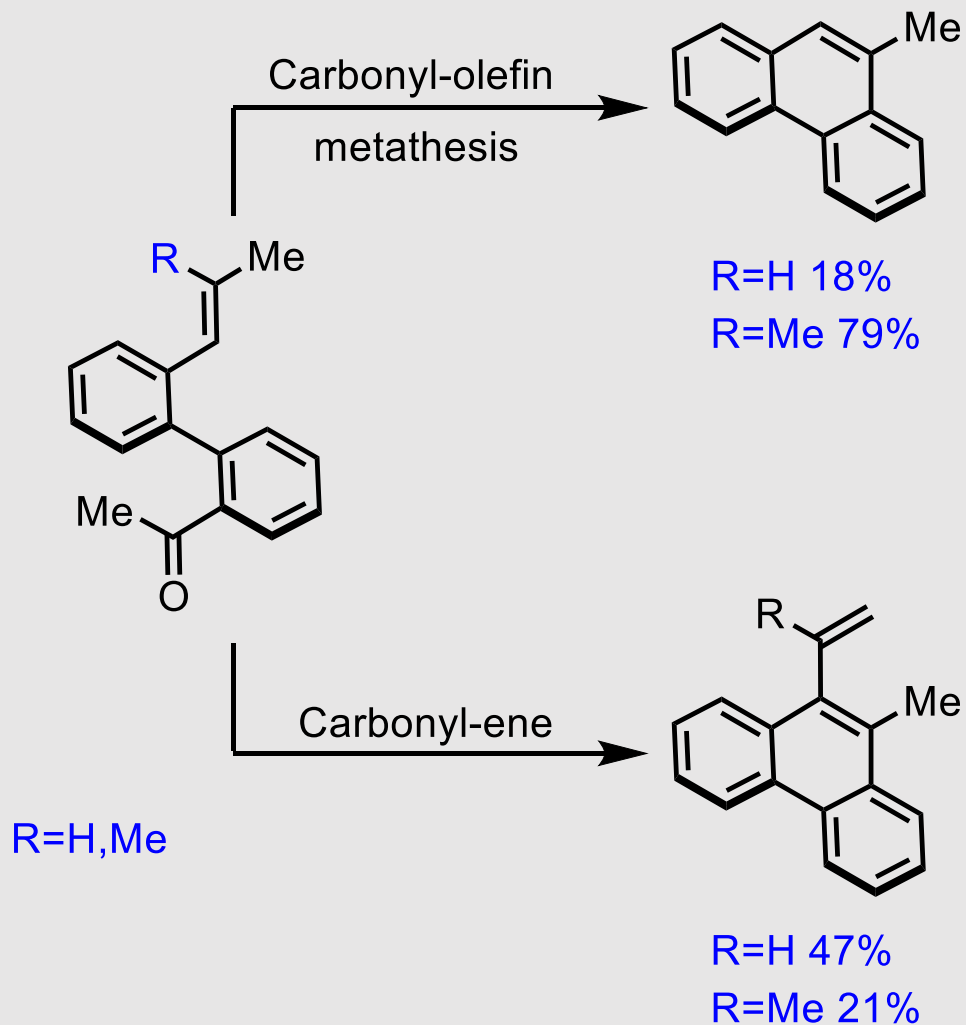
Fe(III)-Mediated Access to Polycyclic Aromatic Compounds



Salient Observations

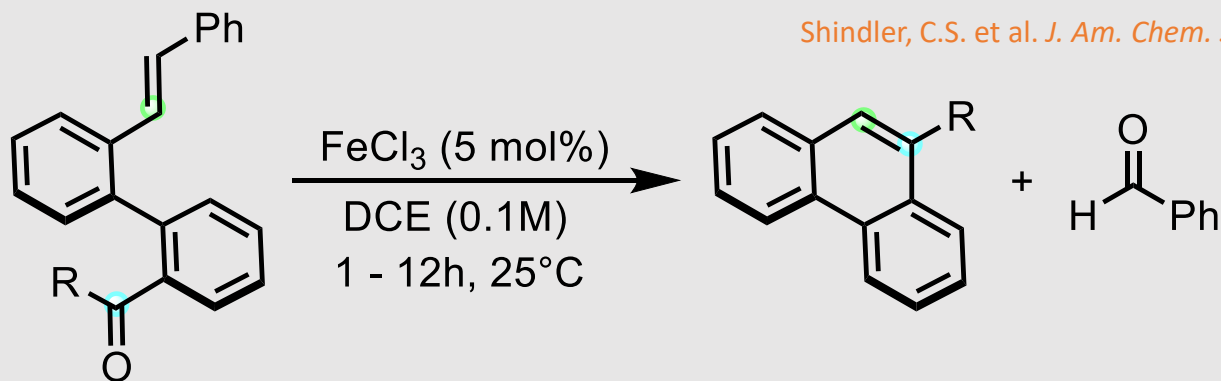
Non-styrenyl olefins compete with carbonyl-ene.

Prenyl previously preferred, aldehydes now work.

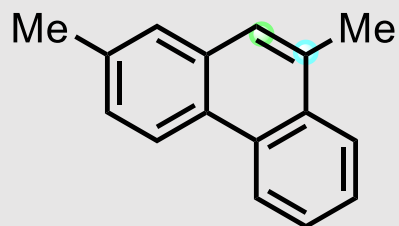


Substrate Examples

Shindler, C.S. et al. *J. Am. Chem. Soc.* **2017**, *139*, 2960-2963.

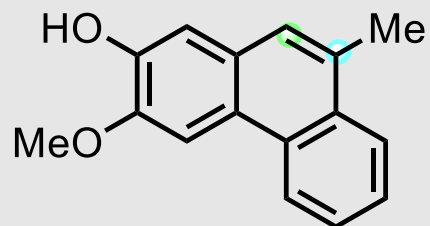


Phenanthrene



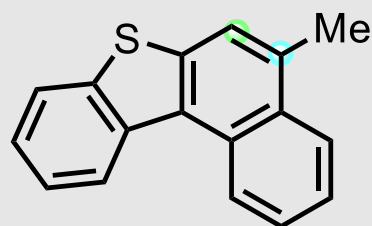
89%

Phenanthrene



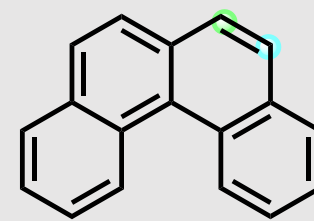
78%

Benzo-thiophene



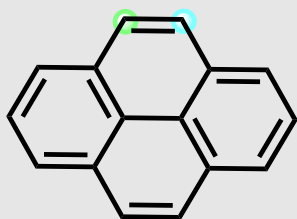
62%*

Benzo-phenanthrene



89%

Pyrene



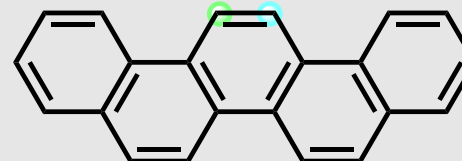
70%

Methyl-chrysene



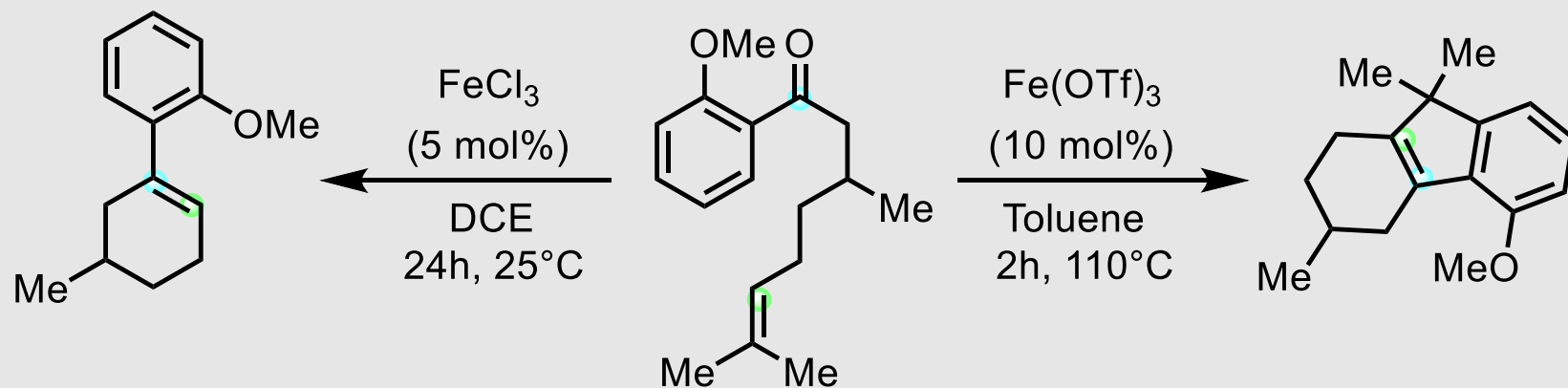
80%

Picene

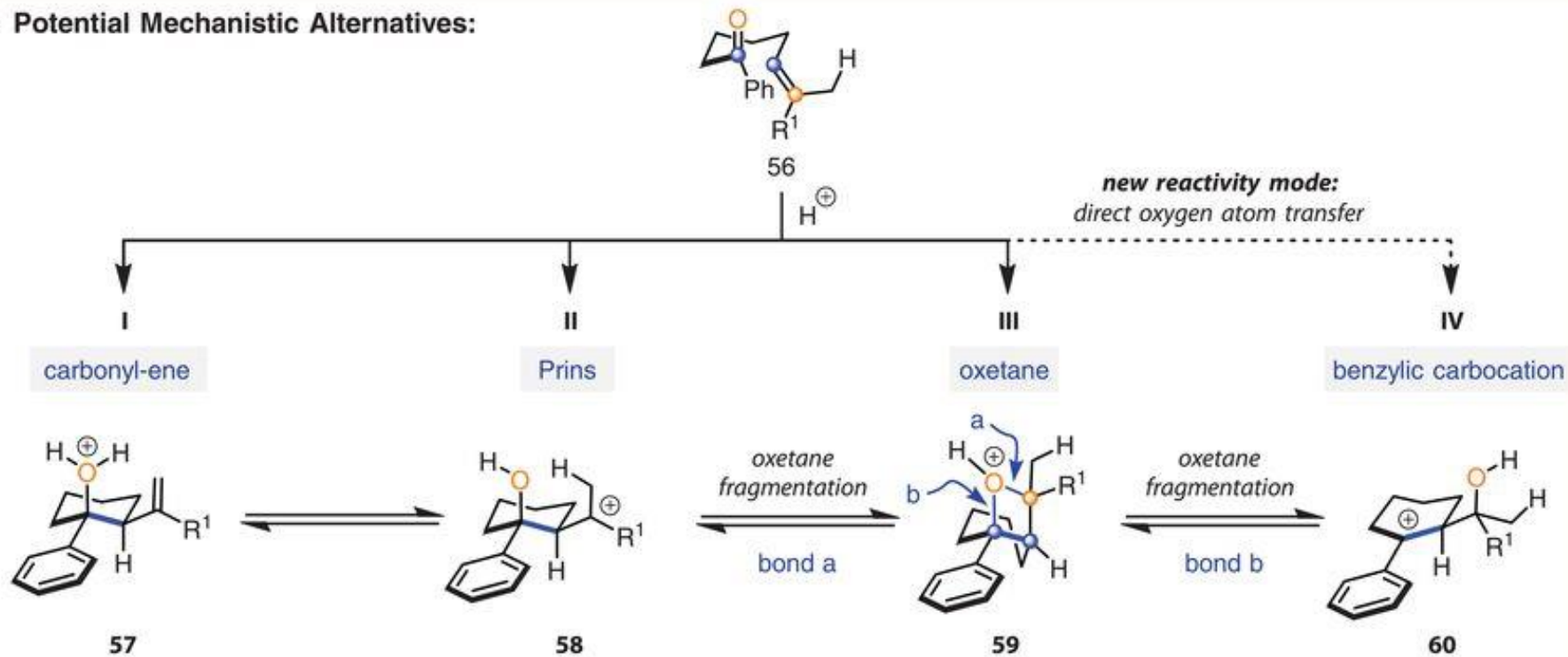


53%

Interrupted Carbonyl-Olefin Metathesis (Tetrahydrofluorenes)

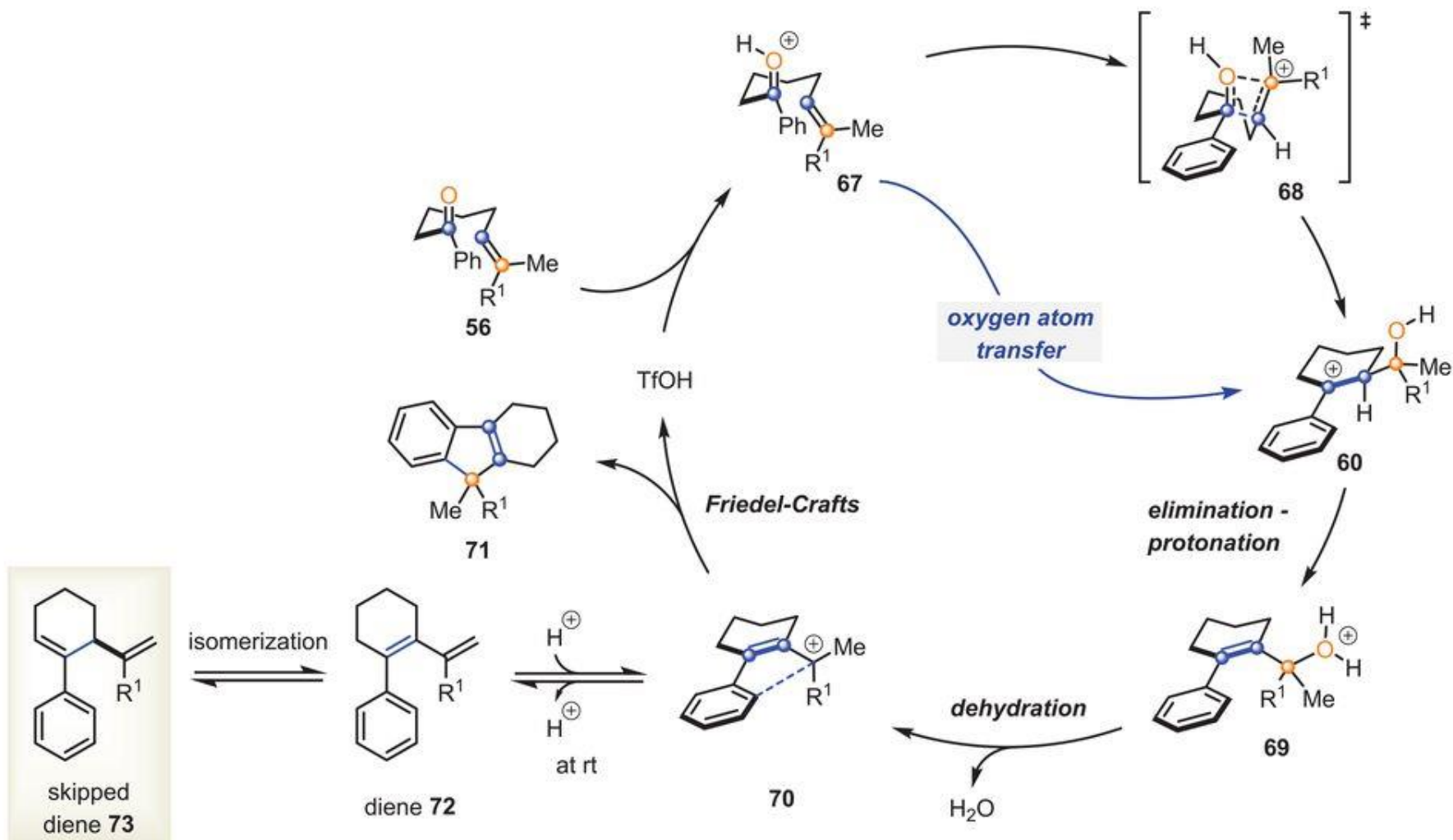


A Potential Mechanistic Alternatives:

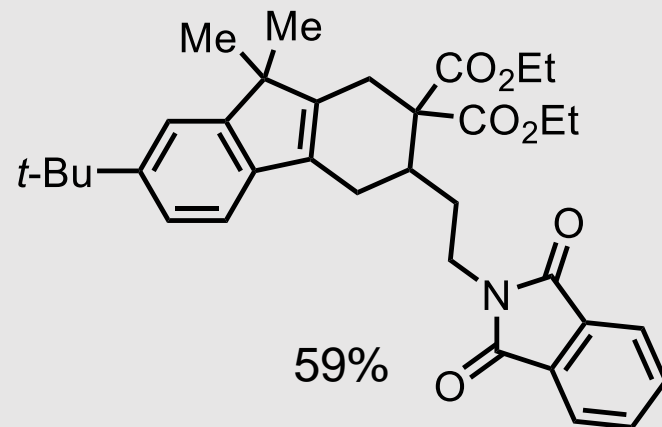
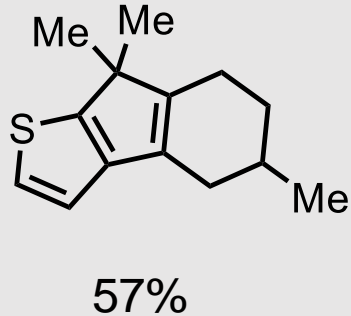
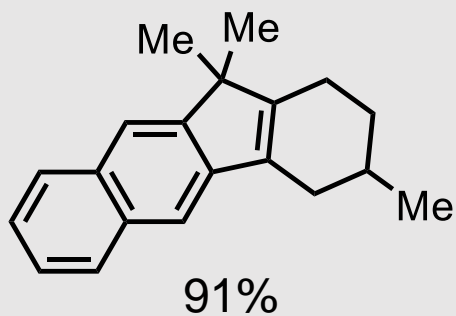
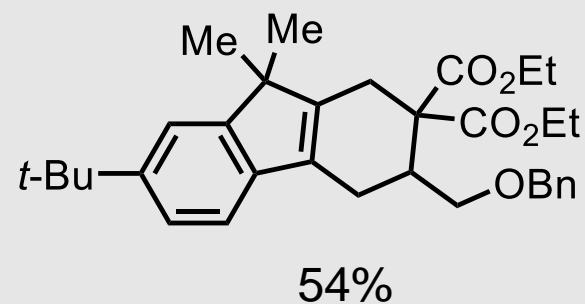
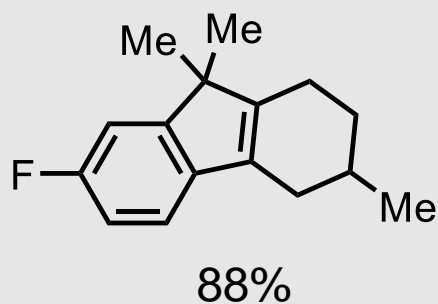
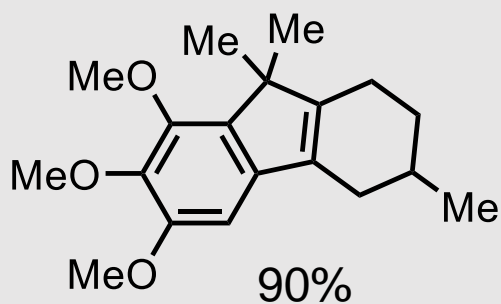
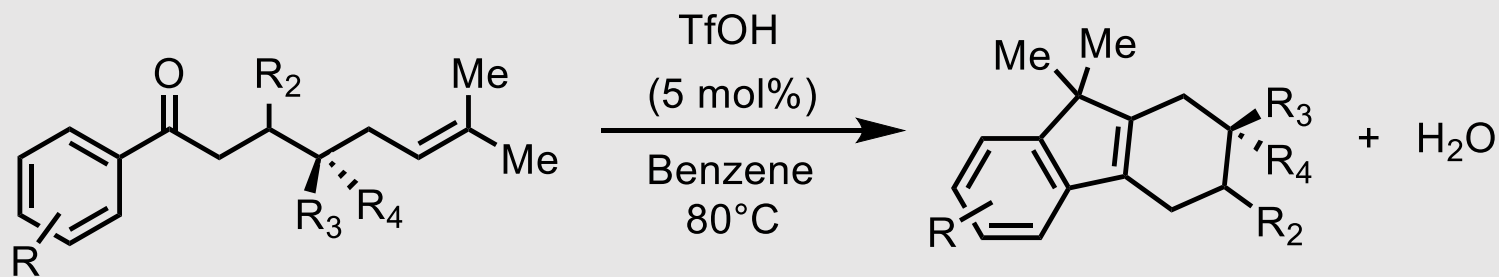


Proposed Mechanistic Route

A Mechanistic Hypothesis for Interrupted Carbonyl-Olefin Metathesis:

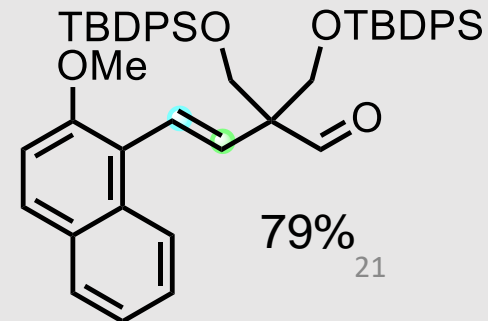
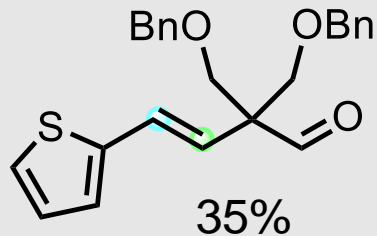
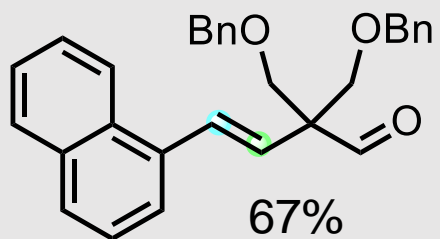
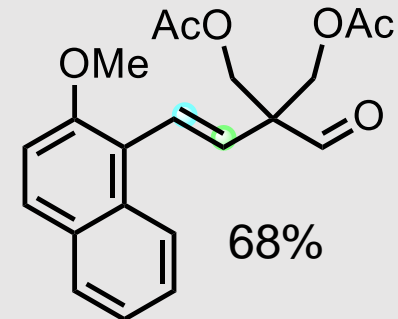
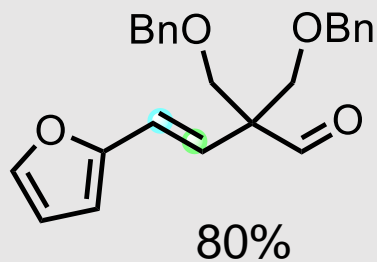
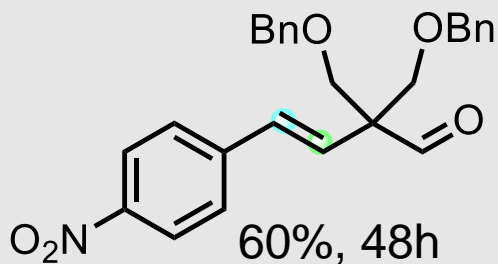
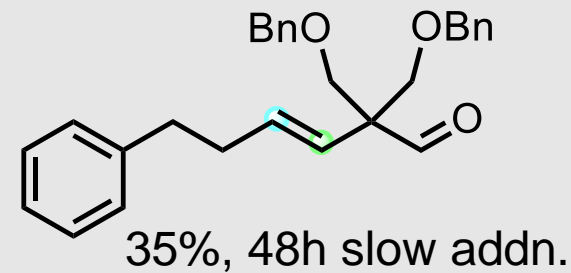
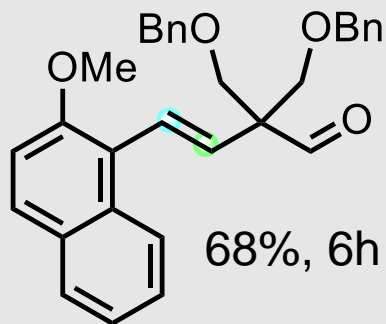
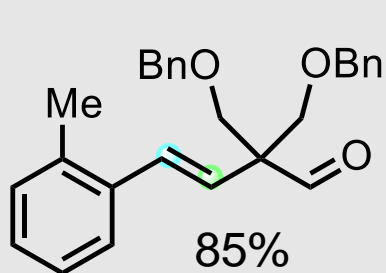
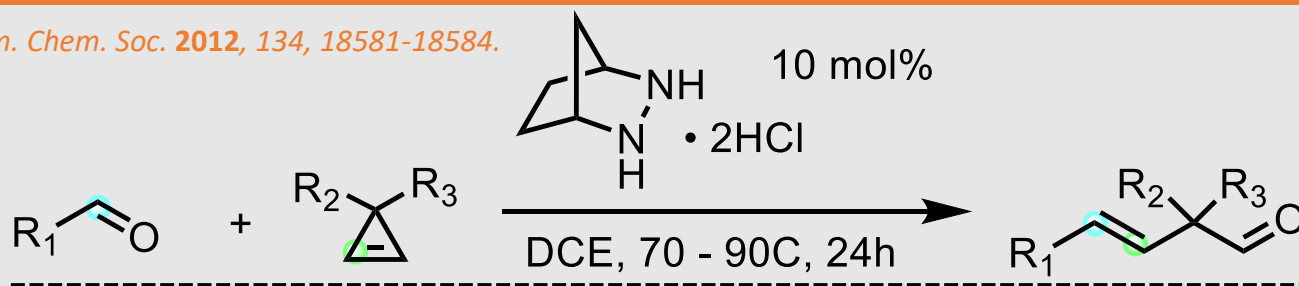


Interrupted Carbonyl-Olefin Metathesis (Tetrahydrofluorenes)



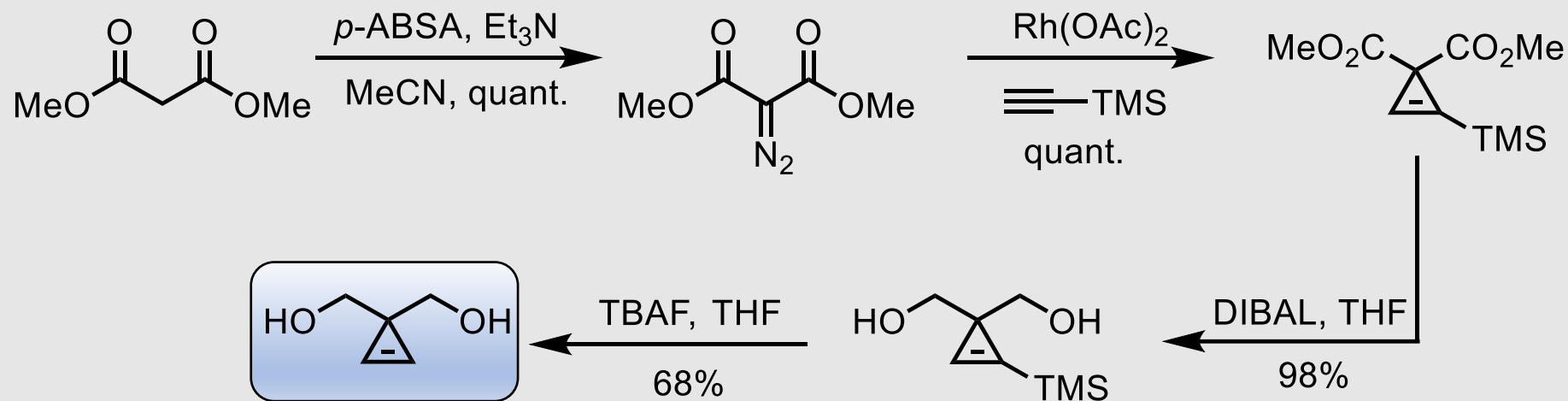
Organocatalytic Carbonyl-Olefin Metathesis

Lambert, T. et al. *J. Am. Chem. Soc.* **2012**, *134*, 18581-18584.

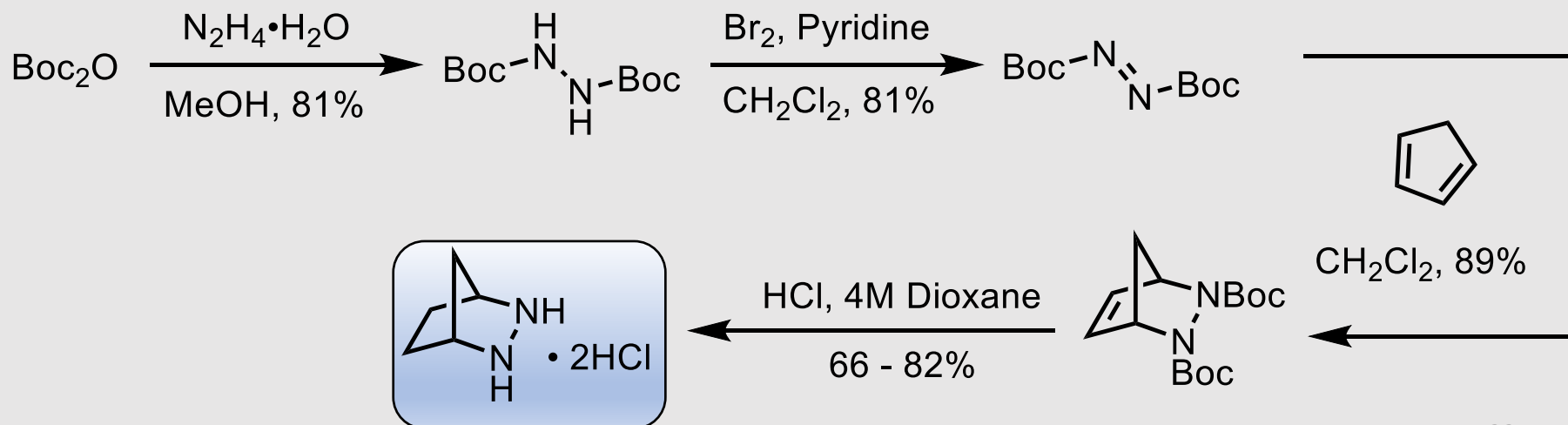


Organocatalytic Carbonyl-Olefin Metathesis

Synthesis of Cyclopropenes (Propose a route):

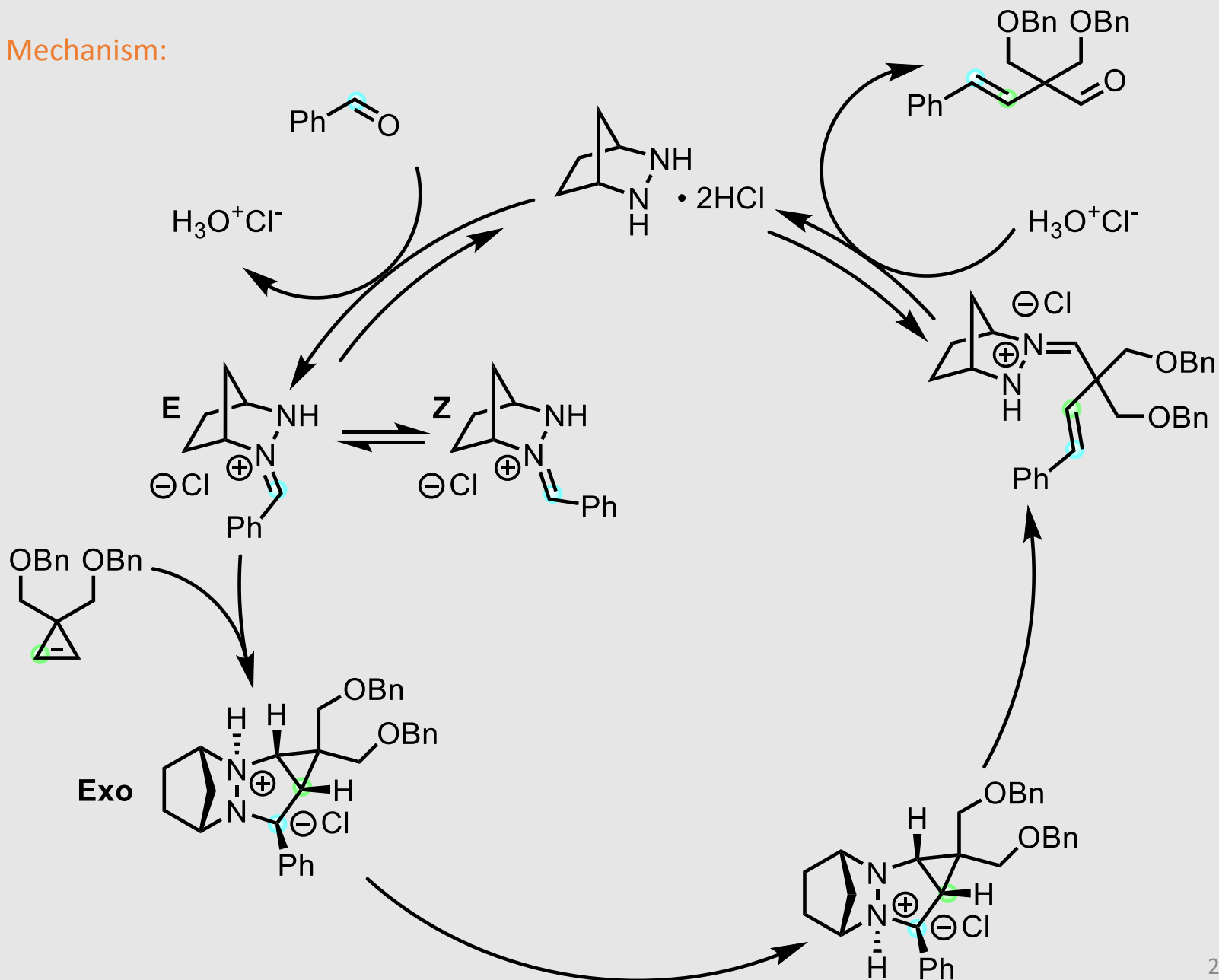


Synthesis of Catalyst (Propose a route):

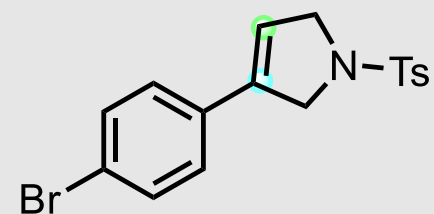
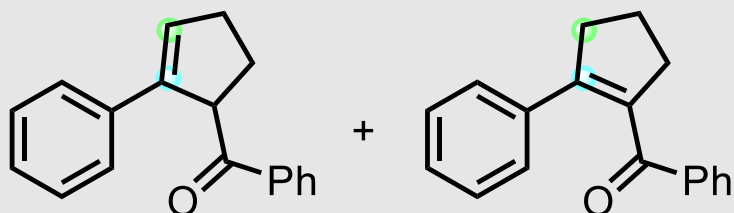
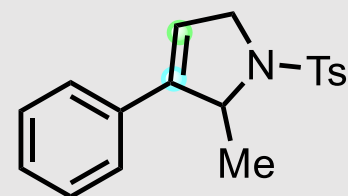
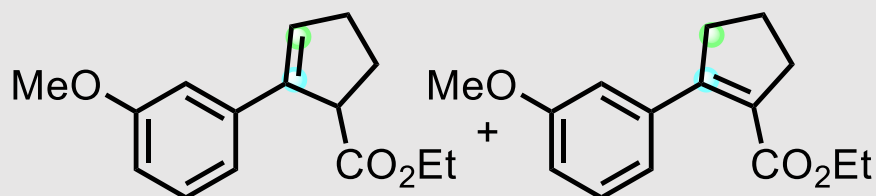
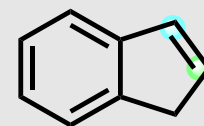
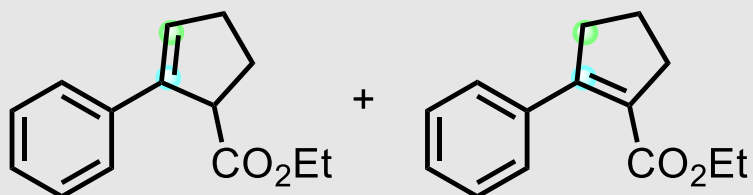
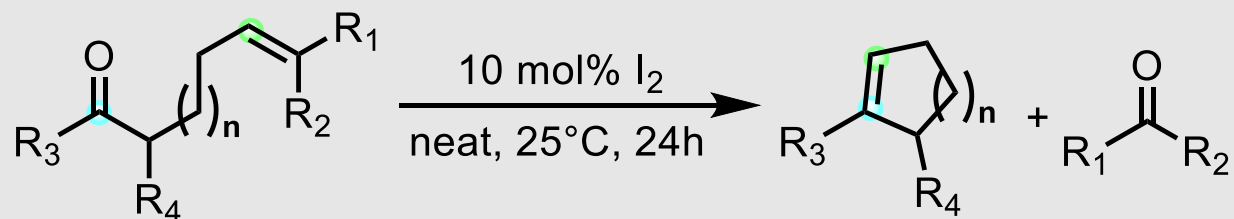


Organocatalytic Carbonyl-Olefin Metathesis

Posited Mechanism:

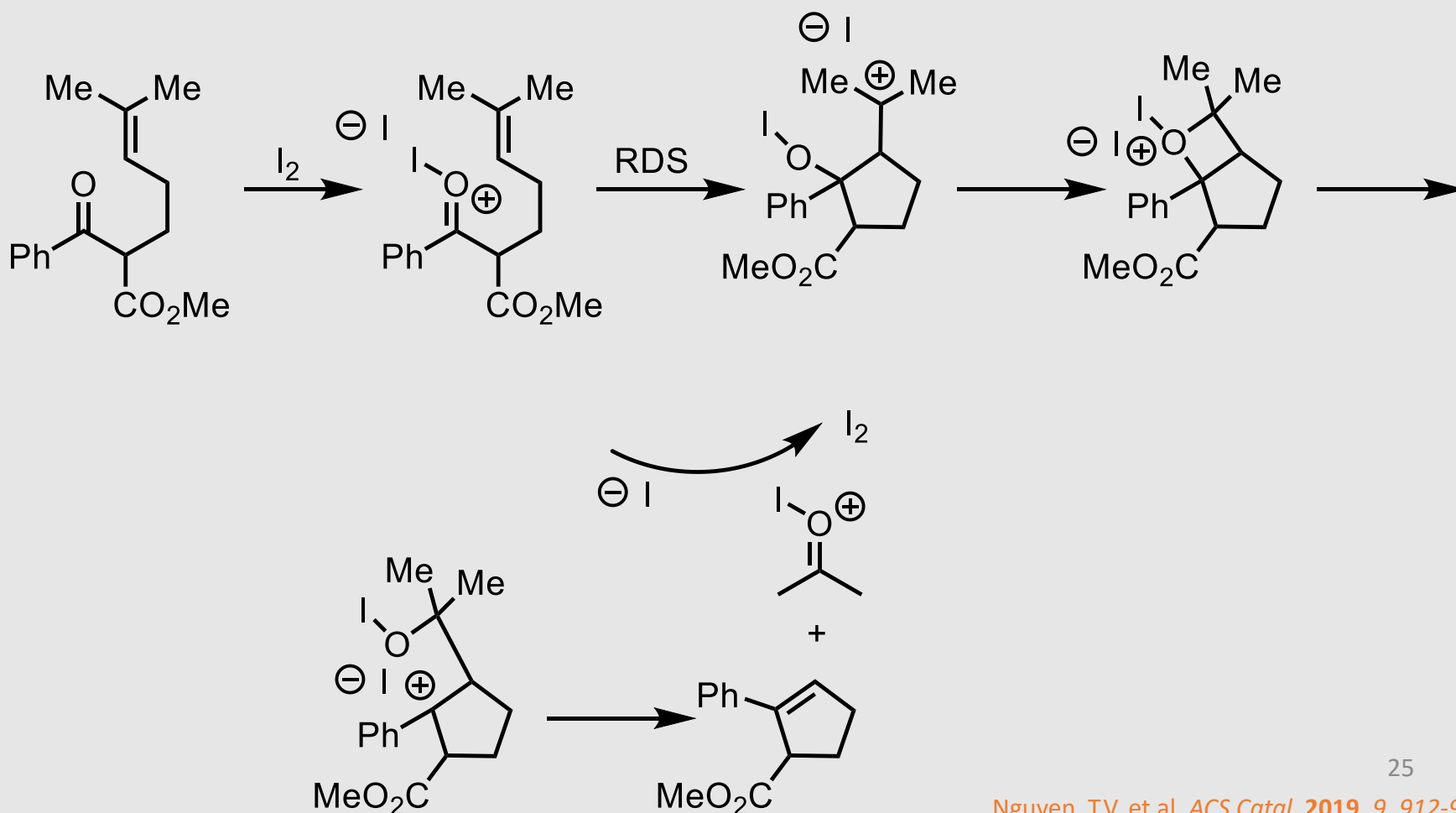
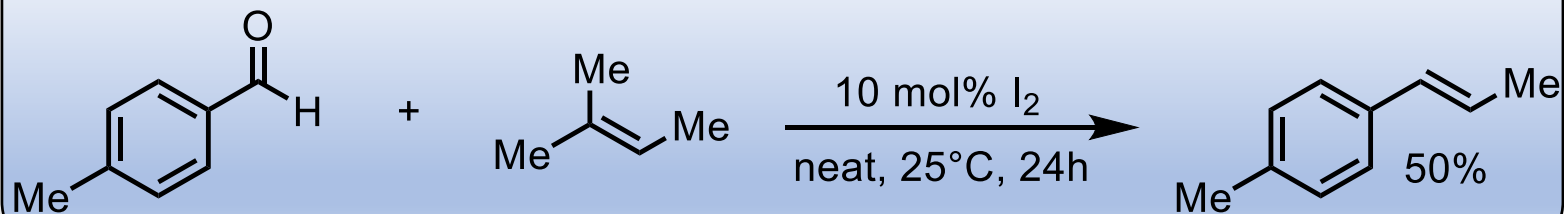


Iodine-Catalyzed Carbonyl-Olefin Metathesis

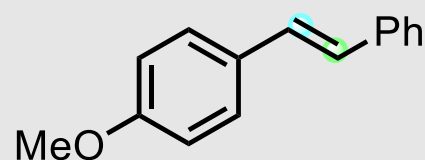
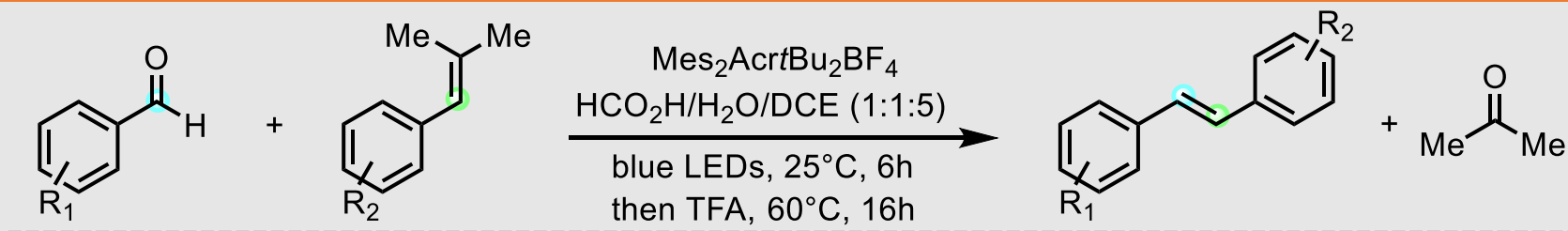


Iodine-Catalyzed Carbonyl-Olefin Metathesis

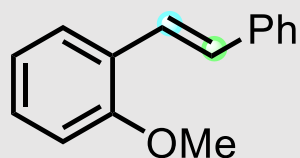
Cross-Metathesis Example:



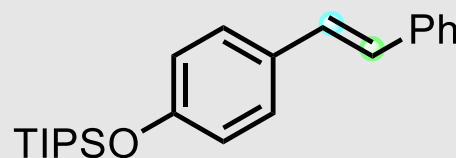
Photoredox Carbonyl-Olefin Metathesis



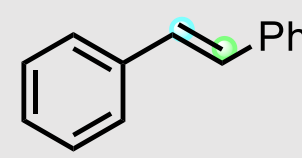
70%; 8:1 E:Z



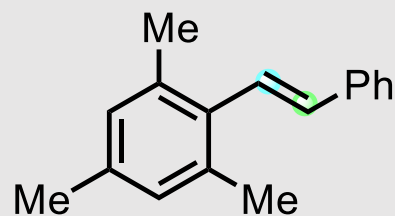
86%; 10:1 E:Z



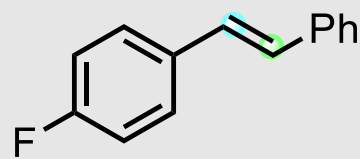
65%; 9:1 E:Z



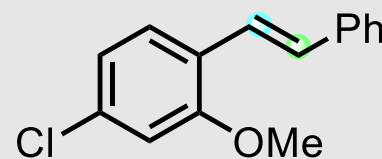
41%; >20:1 E:Z



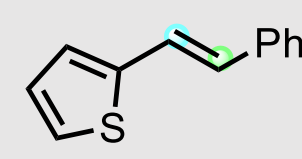
69%; 10:1 E:Z



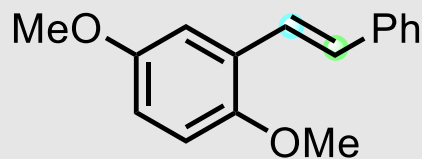
43%; >20:1 E:Z



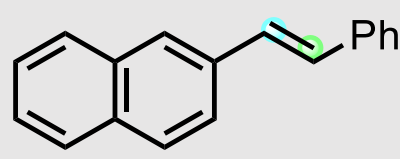
71%; >20:1 E:Z



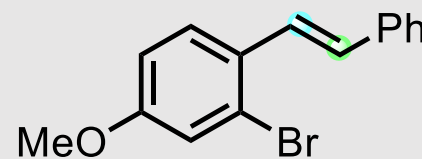
34%; 5:1 E:Z



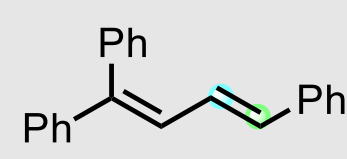
57%; 15:1 E:Z



61%; >20:1 E:Z

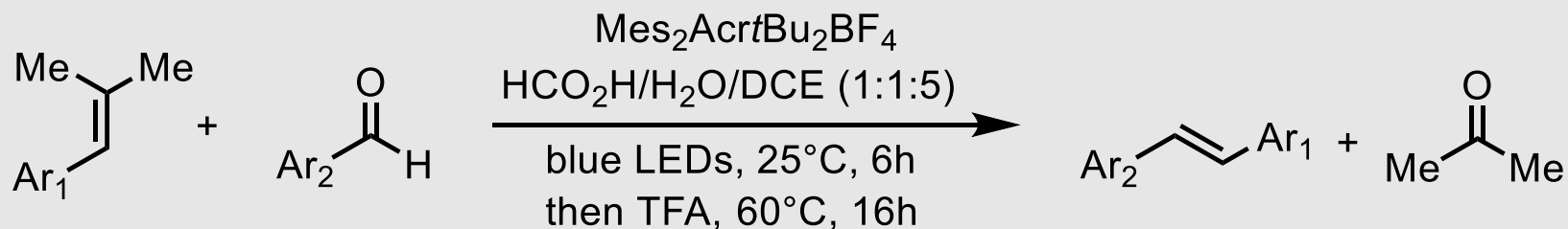


67%; >12:1 E:Z



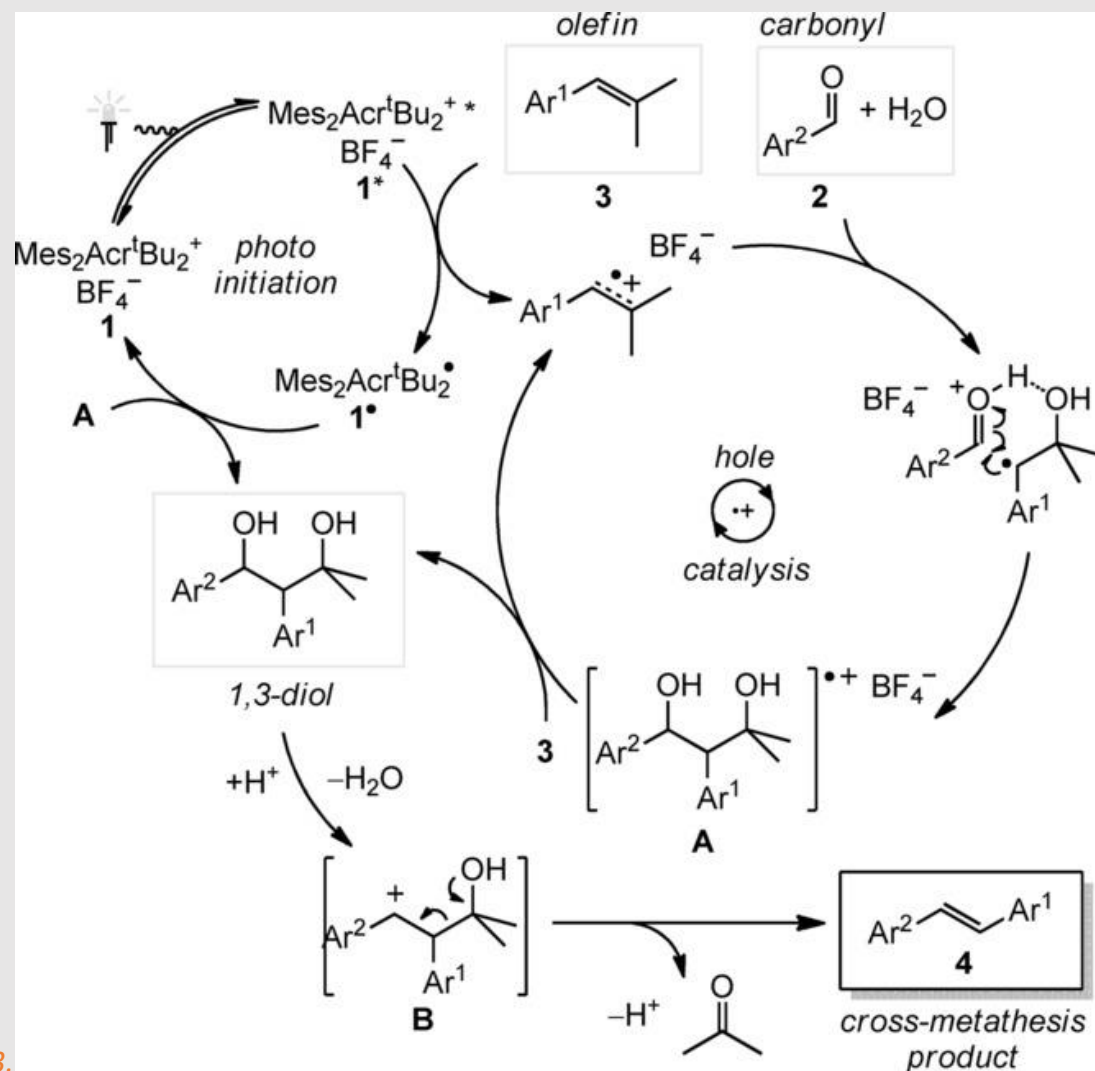
63%; 20:1 E:Z

Photoredox Carbonyl-Olefin Metathesis

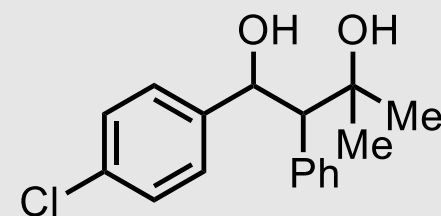
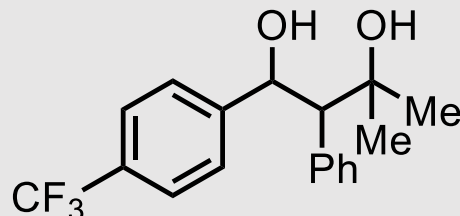
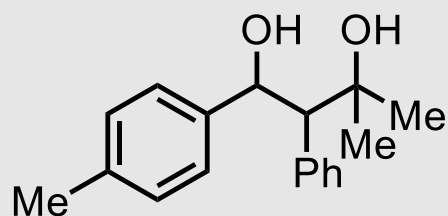
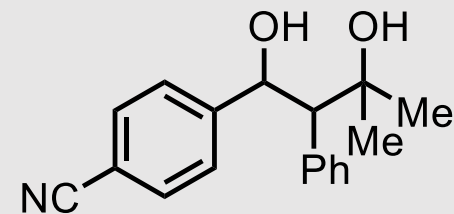
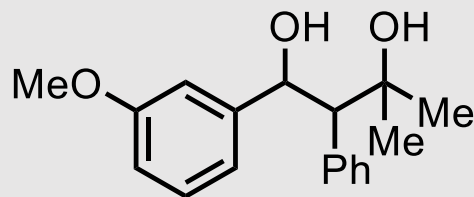
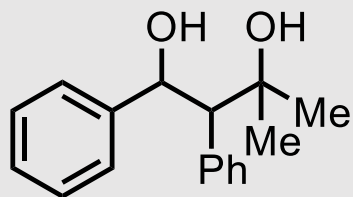
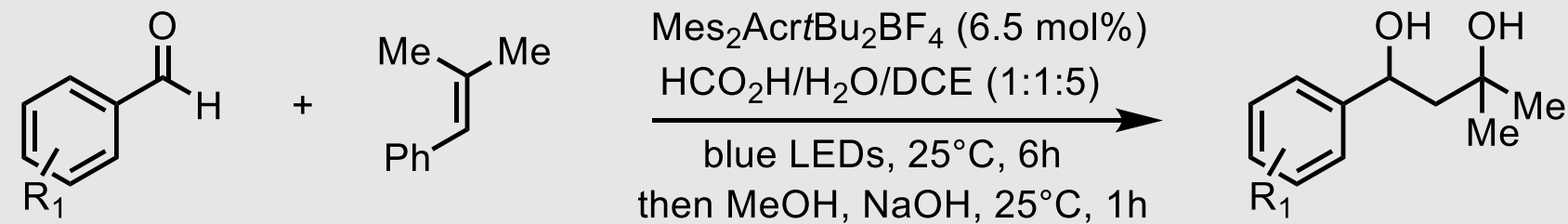


Propose a catalytic cycle.

Hints: This is proposed to operate via hole catalysis. Also, an oxetane is not the putative intermediate.

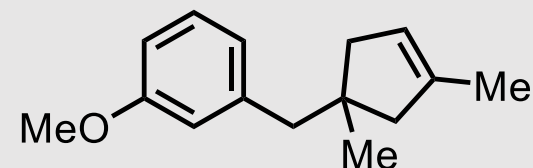
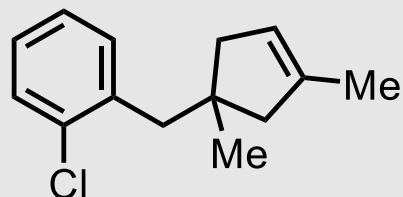
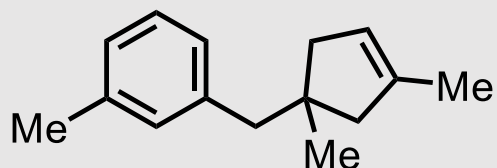
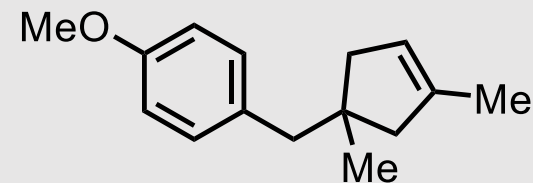
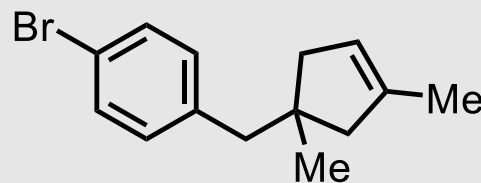
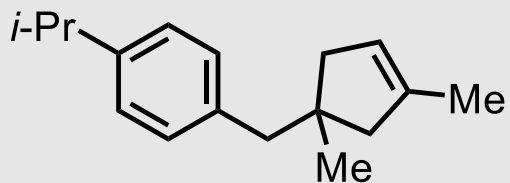
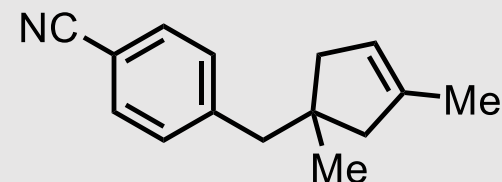
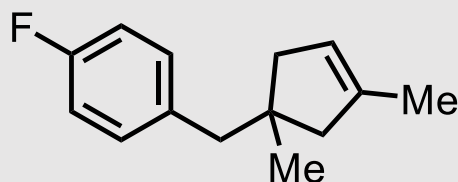
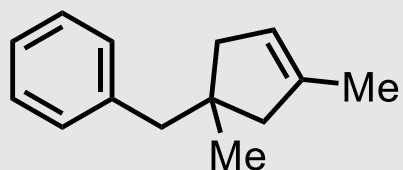
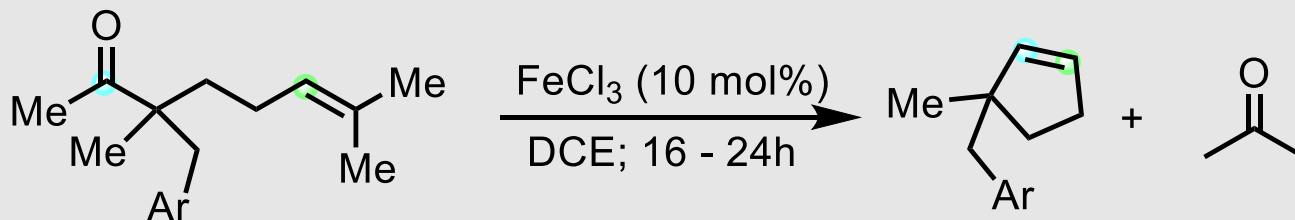


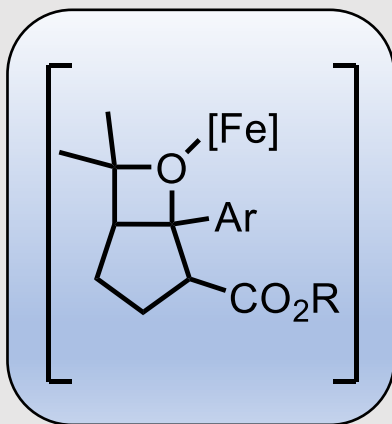
Photoredox Carbonyl-Olefin Metathesis



Fe(III)-Mediated Carbonyl-Olefin Metathesis (Schindler)

Schindler, C.S. Sigman, M.S. et al. *J. Am. Chem. Soc.* **2019**, *141*, 1690-1700.





Future Directions:

- **Imine-carbonyl metathesis via azetidines?**
- **Ways to turnover metal-oxo by-products?**
- **Mechanistic paradigms outside of oxetanes.**
- **More methods involving photoredox processes.**
- **What about McMurry!?**
(Although not actually metathesis)